

*To my family*

## ACKNOWLEDGEMENTS

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## ABSTRACT

The propellane alkaloids comprise a large class of natural products that possess varying degrees of structural complexity and biological activity. The earliest of these to be isolated was acutumine, a chlorinated alkaloid that has been shown to exhibit selective T-cell cytotoxicity and anti-amnesic properties. Alternatively, the hasubanan family of natural products has garnered considerable attention from the synthetic community in part due to its structural similarities to morphine. While these alkaloids have been the subject of numerous synthetic studies over the last forty years, very few enantioselective total syntheses have been reported to date.

As part of a research program directed towards the synthesis of various alkaloid natural products, we have developed a unified strategy for the preparation of the hasubanan and acutumine alkaloids. Specifically, a highly diastereoselective 1,2-addition of organometallic reagents to benzoquinone-derived *tert*-butanesulfinimines was established, which provides access to enantioenriched 4-aminocyclohexadienone products. This methodology enabled the enantioselective construction of functionalized dihydroindolones, which were found to undergo intramolecular Friedel-Crafts conjugate additions to furnish the propellane cores of several hasubanan alkaloids. As a result of these studies, the first enantioselective total syntheses of 8-demethoxyrunanine and cepharatines A, C, and D were accomplished in 9-11 steps from commercially available starting materials.

More recent efforts have focused on applying the sulfinimine methodology to the synthesis of a more structurally complex propellane alkaloid, acutumine. Extensive studies have determined that a properly functionalized dihydroindolone undergoes a

photochemical [2+2] cycloaddition followed by a lactone fragmentation/Dieckmann cyclization to establish the carbocyclic framework of the natural product. The preparation of more appropriately oxidized propellane intermediates is currently under investigation, and is anticipated to facilitate our synthetic endeavors toward acutumine.

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**LIST OF ABBREVIATIONS**

$[\alpha]_D$	angle of optical rotation of plane-polarized light
Å	angstrom(s)
Ac	acetyl
AIBN	azobisisobutyronitrile
app	apparent
aq	aqueous
Ar	aryl group
atm	atmosphere(s)
Bn	benzyl
Boc	<i>tert</i> -butoxycarbonyl
bp	boiling point
br	broad
Bu	butyl
<i>n</i> -Bu	butyl or <i>norm</i> -butyl
<i>t</i> -Bu	<i>tert</i> -butyl
<i>c</i>	concentration of sample for measurement of optical rotation
<sup>13</sup> C	carbon-13 isotope
<sup>14</sup> C	carbon-14 isotope
/C	supported on activated carbon charcoal
°C	degrees Celcius
calc'd	calculated
CBS	Corey-Bakshi-Shibata

Cbz	benzyloxycarbonyl
cm <sup>-1</sup>	wavenumber(s)
CN	nitrile
conc.	concentrated
Cy	cyclohexyl
d	doublet
<i>d</i>	dextrorotatory
D	deuterium
dba	dibenzylideneacetone
DBU	1,8-diazabicyclo[5.4.0]undec-7-ene
DHP	dihydropyran
DIBAL	diisobutylaluminum hydride
DMAP	4-dimethylaminopyridine
DME	1,2-dimethoxyethane
DMF	<i>N,N</i> -dimethylformamide
DMP	Dess–Martin periodinane
DMSO	dimethylsulfoxide
dppp	1,3-bis(diphenylphosphino)propane
dr	diastereomeric ratio
<i>ee</i>	enantiomeric excess
E <sup>+</sup>	electrophile
<i>E</i>	trans (entgegen) olefin geometry
EDC	<i>N</i> -ethyl- <i>N</i> '-(3-dimethylaminopropyl)carbodiimide

e.g.	for example (Latin: <i>exempli gratia</i> )
eq	equation
ESI	electrospray ionization
Et	ethyl
<i>et al.</i>	and others (Latin: <i>et alii</i> )
FAB	fast atom bombardment
g	gram(s)
h	hour(s)
$^1\text{H}$	proton
$^2\text{H}$	deuterium
$^3\text{H}$	tritium
H-G II	Hoveyda-Grubbs, 2 <sup>nd</sup> generation
HMDS	hexamethyldisilamide or hexamethyldisilazide
HMPA	hexamethylphosphoramide
<i>hν</i>	light
HPLC	high performance liquid chromatography
HRMS	high resolution mass spectrometry
Hz	hertz
i.e.	that is (Latin: <i>id est</i> )
IR	infrared spectroscopy
<i>J</i>	coupling constant
<i>k</i>	rate constant
kcal	kilocalorie(s)

kg	kilogram(s)
L	liter or neutral ligand
<i>l</i>	levorotatory
LA	Lewis acid
LAH	lithium aluminum hydride
LDA	lithium diisopropylamide
m	multiplet or meter(s)
M	molar or molecular ion
<i>m</i>	meta
μ	micro
<i>m</i> -CPBA	<i>meta</i> -chloroperbenzoic acid
Me	methyl
mg	milligram(s)
MHz	megahertz
min	minute(s)
mL	milliliter(s)
mol	mole(s)
MOM	methoxymethyl
mp	melting point
Ms	methanesulfonyl (mesyl)
<i>m/z</i>	mass-to-charge ratio
N	normal or molar
NBS	<i>N</i> -bromosuccinimide

NIS	<i>N</i> -iodosuccinimide
nm	nanometer(s)
NMO	<i>N</i> -methylmorpholine- <i>N</i> -oxide
NMR	nuclear magnetic resonance
NOE	nuclear Overhauser effect
NOESY	nuclear Overhauser enhancement spectroscopy
Nu <sup>-</sup>	nucleophile
<i>o</i>	ortho
<i>p</i>	para
Ph	phenyl
pH	hydrogen ion concentration in aqueous solution
p <i>K</i> <sub>a</sub>	acid dissociation constant
PMB	<i>para</i> -methoxybenzyl
PMHS	polymethylhydrosiloxane
ppm	parts per million
Pr	propyl
<i>i</i> -Pr	isopropyl
<i>n</i> -Pr	propyl or <i>norm</i> -propyl
psi	pounds per square inch
pyr	pyridine
q	quartet
R	alkyl group
<i>R</i>	rectus

ref	reference
$R_f$	retention factor
s	singlet or seconds
<i>S</i>	sinister
t	triplet
TBAF	tetra- <i>n</i> -butylammonium fluoride
TBHP	<i>tert</i> -butyl hydroperoxide
TBS	<i>tert</i> -butyldimethylsilyl
Tf	trifluoromethanesulfonyl
TFA	trifluoroacetic acid
TFAA	trifluoroacetic anhydride
THF	tetrahydrofuran
TLC	thin layer chromatography
TMS	trimethylsilyl
TMSE	2-(trimethylsilyl)ethyl
TOF	time-of-flight
tol	tolyl
Ts	<i>para</i> -toluenesulfonyl (tosyl)
UV	ultraviolet
w/v	weight per volume
v/v	volume per volume
X	anionic ligand or halide
Z	cis (zusammen) olefin geometry