Chapter 7

Panel of Cytochrome P450 BM3 Variants to Produce Drug Metabolites and Diversify Lead Compounds

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A. Abstract

Herein we demonstrate that a small panel of variants derived from cytochrome P450 BM3 from *Bacillus megaterium* covers the breadth of reactivity of human P450s by producing 12 of 13 mammalian metabolites for two marketed drugs, verapamil and astemizole, and one research compound. The most active enzymes support preparation of individual metabolites for preclinical bioactivity and toxicology evaluations. Underscoring their potential utility in drug lead diversification, engineered P450 BM3 variants also produce novel metabolites by catalyzing reactions at carbon centers beyond those targeted by animal and human P450s. Production of a specific metabolite can be improved by directed evolution of the enzyme catalyst. Some variants are more active on the more hydrophobic parent drug than on its metabolites, which limits production of multiple-hydroxylated species, a preference that appears to depend on the evolutionary history of the P450 variant.

B. Introduction

Selective C-H oxidation represents one of the great challenges for which synthetic chemists find only substrate-specific solutions (1 - 4). Breakthroughs in selective C-H oxidation methodology could benefit drug discovery, among other fields, by enabling rapid and parallel analog construction for a specific molecular scaffold. Such improvements in efficiency would greatly increase the number and variety of compounds that could be produced and raise the likelihood of identifying effective therapeutic agents.

Many synthetic strategies for C-H oxidation rely on a reactive intermediate that plays upon subtle differences in C-H bond strength (1–5 kcal mol⁻¹) to achieve regioselectivity (5). Owing to the large number of C-H bonds in most bioactive chemicals, identifying a reagent that can react at one C-H bond in preference to all others can be difficult, or even impossible. Nature solves the selectivity problem by incorporating discrete molecular recognition elements into enzyme catalysts so that they can use specific enzyme–substrate interactions to impart reactivity to a specific C-H bond.

Cytochromes P450 (CYPs) are a large superfamily of heme-containing C-H oxidation enzymes. In humans, CYPs play key roles in drug metabolism and clearance. An ability to predict how potential pharmaceuticals will be metabolized will better equip us to identify derivatives with improved biological activity, solubility, toxicity, stability, or bioavailability (*6*). In fact, FDA guidelines indicate that uniquely human metabolites and metabolites present at disproportionately higher levels in humans as compared to the animal species used during standard toxicology testing may require safety assessment before beginning large-scale clinical trials (7). Preparation of these metabolites at sufficient scale for evaluation is not trivial and often requires a *de novo* synthesis for each metabolite. Biosynthetic methods that employ purified human CYPs or crude liver microsomes are not much better, because human CYPs are poorly stable, membrane-bound, multi-component protein systems that generally exhibit low reaction rates. As an alternative to using human CYPs as biocatalysts for metabolite production, we and others have focused on soluble, bacterial P450 BM3 (also known as CYP102A1) (8) as a C-H oxidation platform (9 – 11). Derived from *Bacillus megaterium*, P450 BM3 has properties that greatly facilitate its engineering and use in synthesis: it can be expressed at high levels in *E. coli* (~ 12% dry cell mass), and, unlike nearly all other CYPs, its hydroxylase, reductase, and electron-transfer domains are all in one contiguous polypeptide chain. This last feature might contribute to its relatively high activity (> 1,000 turnovers per min) on its preferred fatty acid substrates (8). Like most CYPs implicated in anabolic pathways, P450 BM3 is substrate specific, and hydroxylates a C₂₀ fatty acid over a C₁₂ fatty acid with more than 200-fold higher efficiency (*12*).

We chose three structurally diverse drug compounds with known patterns of mammalian CYP-dependent clearance to evaluate whether P450 BM3 variants can catalyze similar C-H oxidations. Verapamil is a calcium channel blocker used in the treatment of hypertension and arrhythmia (13). Astemizole is a potent H₁-histamine receptor antagonist used for treatment of common sinus allergy symptoms (14). The third compound, LY294002, is an antiproliferative agent that inhibits phosphatidylinositol 3-kinase, a target with potential for treatment of malignancies (15). We report here that a small collection of P450 BM3 variants can produce nearly all the known human (or rat) metabolites for each of the three drugs. Within each set of active enzyme variants, we identified several that produce selected metabolites in yields and

activities suitable for preparative scale synthesis. We also identified variants that generate metabolites not produced by rat liver microsome controls or known to be human metabolites; the results demonstrate the ability of this C-H oxidation platform to target a range of carbon centers.

C. Results

C.1. Selection of P450 variants

From our extensive collections we selected 120 P450 BM3 variants that had previously demonstrated activity on substrates not in the wild-type enzyme's repertoire. These variants were constructed by using a variety of commonly implemented genetic diversification techniques, including error-prone PCR and targeted mutagenesis of active site residues. We also included variants derived from structure-guided recombination of BM3 and its homologs, CYP102A2 and CYP102A3 (*16*). These enzymes had been selected based on their activities toward a variety of substrates, including straight-chain alkanes, cycloalkanes, alkyl ethers (17 - 19), aromatic compounds (20), derivatives of fatty acids (21), and drug compounds (10, 22).

We also distinguished between enzyme variants that function as monooxygenases and use NADPH to reduce the catalytic Fe^{III} center and peroxygenases that contain only the heme domain and use H_2O_2 in place of O_2 and NADPH. Each has its own catalytic and operational advantages and disadvantages that have been described elsewhere (21). Although we had evidence that the presence of a reductase domain impacts activity favorably (20), we were unsure what effect, if any, this domain would have on regioselectivity. The variants, their sequences relative to wild type, and the criteria for their prior selection from mutant libraries are described in Appendix E.

C.2. Reactions with human and bacterial CYPs

Verapamil and astemizole undergo extensive biotransformation by the major human CYPs to produce demethylated and dealkylated products (23) by hydroxylation of activated carbon centers adjacent to heteroatoms (summarized in Tables 7.1 (a), 7.2 (a)). Experiments

conducted with rat liver microsomes *in vitro* indicate that LY294002 metabolism is confined to single- and double- alkyl hydroxylations of the morpholine ring (Table 7.3), a pattern also observed in other drugs possessing this moiety. Reactions of the 120 P450 BM3 variants with each of the three drugs were assessed by HPLC. Metabolite products were evident in the HPLC traces for a significant number of variants (43 exhibited some activity on verapamil, 42 on astemizole, and 18 on LY294002); the metabolites and their distributions were subsequently characterized by using LCMS and MS/MS. Metabolite structure assignments were aided by comparison of MS and MS/MS spectra with spectra of known metabolites obtained in our laboratory using rat microsomal systems.

C.3. Verapamil

Table 7.1 summarizes the performance of selected enzymes on verapamil (all 43 active enzymes and their product distributions are listed in Appendix E). For example, entry 4 displays the activity of chimera 22313333 on verapamil. It converts 34% of the starting material into an assortment of products, the distribution of which is 41% **2**, 15% **5**, 20% **6**, 9% **7**, and 15% **10**. Variants 2C11 and 9C1 furnished the widest array of products, with six and eight individual metabolites, respectively. In particular, 9C1 produced several metabolites that had undergone two hydroxylation events (**1**, **4**, and **8**). Variant 22313333 (entry 4) was the only enzyme capable of producing double hydroxylation product **5**, while 32313233 (entry 6) best produced norverapamil (3). Interestingly, the addition of a reductase as a fusion to the 32313233 heme domain (entry 7) rendered the enzyme unable to produce new metabolite **7**. Variant 7-11D (entry 13) differs by no more than two mutations from the variants in each of entries 9–12, but its product distribution more closely parallels chimera 21312332. The 9-10A family of enzymes

was the most active, with conversions exceeding 30% (entries 9–12), and produced metabolite **7** with excellent selectivity.

Table 7.1: Conversion of verapamil to: (a) its most abundant^[a] human metabolites, and (b) new metabolites by different P450 BM3 variants

(a)	OMe				OMe			OM	e	NHR ²	(b))		OMe					ОMе	
R ¹ 0	\searrow	\sim	^ r	N~~		OMe	R ¹ O	\bigcirc	\checkmark			F	¹ 0		\sim	~	N			OR⁴
	NĆ	\succ	F	₹ ²				Ν	ć					NĆ	\succ		R^2	R^3		
		M	w	R¹	R ²	_		M _w	R^1	R ²			M _w	R^1		R ²	R	3	R⁴	other
Vera	pamil	45	54	Ме	Me		4	276	н	Me		7	440	Me		Me	Н	I	н	
	1	42	26	н	н		5	276	Me	н	5	8	456	Me		Me	O	н	н	
	2	44	10	н	Me		6	290	Me	Me		9	470	Me		Me	O	н	Me	
	3	44	10	Me	н						1	0	470	Me		Me	н		Me	ОН
	•		Va	riant ^[b]		% (Conv	versio	n 1	2	3	4	5	6	7	8	9	10	-	
		1		2C1.	1		2:	5		8	28			20	8		8	28		
		2		9 <i>C1</i> [[]	c]		3	1	3		29	3		6	13	3		12		
		3		D6H1	10		73	8			26	8		31				24		
		4	2	223133	333		34	4		41			15	20	9			15		
		5		223132	231		4	3		46				33	5			16		
		6		323132	233		24	4			50			21	13			17		
		7	32	31323	3-R1		2:	5			40			44				16		
		8	2	213123	332		6				17			33	17			33		
		9	9-	-10A A	78F		30	0		10					83					
		10	9-	-10A A	82L		5	1						2	94					
		11	9-	-10A F	787L		4	9						6	94					
		12		12-10	С		34	4			6			6	79		3			
	_	13		7-111	D		2	1			28			29	14			29	_	

^[a] Defined as > 1% abundance following oral ¹⁴C verapamil administration in humans (23). ^[b] Variants shown in italics were selected for activity on propranolol (22), variants in bold are chimeras (16), variants in normal type were selected for activity one alkanes (18 - 19) chimeras were written according to fragment composition: 32313233-R1, for example, represents a protein that inherits the first fragment from parent CYP102A3, the second from CYP102A2, the third from CYP102A4 and so on. R1 connotes a fusion to the reductase domain from parent A1. Chimera fusions were used as monooxygenases; chimera heme domains were used as peroxygenases.

^[c] Because not all P450 BM3 oxidation products could be identified, product distribution totals can be less than 100%.

The enzymes that best mimicked human CYP reactivity on verapamil were derived from

two variant families. Three had been isolated by directed evolution for activity on propranolol

(22), while the remaining four were chimeras (16, 24). The BM3 variants that best produced new

metabolites were derived from the alkane-hydroxylating 9-10A variant (18 - 19). These enzymes catalyzed a new regioselective demethylation reaction at position \mathbb{R}^4 and a new benzylic oxidation at position \mathbb{R}^3 . Even without optimization of the reaction conditions, some enzymes were highly active on verapamil; D6H10 (entry 3), for example, transformed verapamil into metabolites at 78% conversion and a total turnover number (TON) greater than 1500.

The enzymes possess contrasting degrees of regioselectivity. Some were unselective and produced a spectrum of metabolites (e.g., propranolol-evolved enzymes 2C11, 9C1, D6H10). By contrast, chimera 22313231 produced metabolite **2**, chimera 32313233-R1 made dealkylated compound **6** (entries 5 and 7), and many variants produced new metabolites **7** and **10** (entries 8–13) with sufficient selectivity (\geq 30%) for larger scale production without further optimization. To demonstrate that useful quantities of metabolites can be produced, we used purified 9-10A F87L to produce metabolite **7** (9.4 mg) from verapamil (25 mg) in 39% yield (1560 TON, 0.025 mol% catalyst).

C.4. Astemizole

Activity of the enzyme panel on astemizole resulted in the seven metabolites described in Table 7.2 (all 42 active enzymes and their product distributions are listed in Appendix E). Variants DE10, 21313311, 22313333, and 32313233 (entries 1–3 and 9) produced dealkylated metabolite **14** in preference to other compounds. The identity of the residue at position 78 had a strong impact on the product distribution within the 9-10A backbone; mutations A78F, A78T, and A78S (entries 4, 5, and 12) produced metabolites **11**, **13**, and **15** as the most abundant respective products. Furthermore, 41-5B (entry 8), a 9-10A family member that contains the A78F mutation (in addition to A82G and A328V) also favors demethylated **11**. Variants 9-10A

A82S, and 9-10A F87L both produced metabolite **13** in good yield and selectivity (entries 6 and 7). Two chimeric monooxygenases, 32313233-R1 and 32312333-R1 (entries 10 and 11), were effective in generating new aromatic hydroxylation product **15**.

Table 7.2: Conversion of astemizole to: (a) its most abundant^[a] human metabolites, and (b) new metabolites by different P450 BM3 variants

(a)	•	(=	_			(b)			OF =	H				
) N H	N N) NH	N N F	OMe	Ŷ		N)—н ()	(F	OMe	N	
	M _w	R	¹ R ²		M _w			M _w		other				M _w
Astemizole	458	М	e H	14	324	15		474					17	350
11	444	H	н			16		490		ОН				
12	460	H	H OH											
13	474	М	e OH											
	_		Variant ^[b]		% Conversion	11	12	13	14	15	16	17	-	
		1	DE10		9				56				_	
		2	21313311		10	20			40			40		
		3	22313333		9	22			56			22		
		4	9-10A A78	F	21	48		38			14			
		5	9-10A A78	Т	27	37	7	44			4			
		6	9-10A A82	S	30	20	7	70			3			
		7	9-10A F87	L	36		3	88			6			
		8	41-5B		5	100								
		9	32313233		11				45	27		27		
		10	32313233-F	R1	16				13	69				
		11	32312333-F	R1	78				4	67		9		
		12	9-10A A78	S	15	13	7			80				
		13	9-10A A82	L	49	24				61	4			
		14	9-10A A82	Ι	32	31	6	63						
		15	9-104 F87	Δ	31			23	16	15				

The 9-10A-derived monooxygenases were the most adept at producing human metabolites of astemizole (entries 4–8). However, these enzymes were unable to dealkylate astemizole and produce metabolite **14**. Oxidation of this more hindered C-H bond was best accomplished by the propranolol active variants and chimeric peroxygenases (entries 1, 2, and 3, respectively). A new benzylic site not hydroxylated by human CYPs was targeted by six

chimeric BM3 enzymes to produce **17**. A second benzimidazole site was hydroxylated (as demonstrated by metabolites **15** and **16**), which is different from that observed in human metabolites **12** and **13**. Discrimination of these sterically and electronically identical sp² C-H bonds is virtually impossible by using traditional transition metal catalysts and emphasizes the power of molecular recognition as a regiocontrolling element.

Small changes in the substrate channel can affect the regioselectivity of aromatic C-H oxidation. For instance, the presence and position of a single methyl group in Thr78 vs. Ser78 (entry 5 vs. 12) and Ile82 vs. Leu82 (entry 14 vs. 13) were sufficient to bias oxidation at C6 instead of at another position on the benzimidazole ring. Product **16** has undergone two oxidations. Although this metabolite was observed in reactions with ten variants, it was produced in greater than 10% abundance by only one variant (entry 4). New metabolite **15** is a candidate for scale-up and could be produced at high conversion (> 25% selectivity) by 15 variants of different lineages (Table E.4 in the Appendix E). Even without optimization of reaction conditions, several enzymes showed good activity towards astemizole. Variant 32312333- R1, for example, converted 78% astemizole into metabolites (70 TON).

C.5. LY294002

The most abundant metabolites of LY294002 produced by rat liver microsomes are detailed in Table 7.3. Both single hydroxylation products **19** and **20** were identified in BM3-variant reactions with very good regioselectivity. Aminoalcohol **18** is the sole metabolite of all three drugs not observed in the reactions. Derivative **18** requires two oxidations and might not appear simply due to the low conversion in these 96-well plate reactions.

Table 7.3: Conversion of LY294002 by different P450 BM3 variants into the most abundant^[a] metabolites produced by rat liver microsomes



^[a] Define as > 1% abundance following in vitro reaction of LY294002 with rat liver microsomes. ^[b] Variants in normal type were selected for activity on alkanes (18 - 19).

The P450s catalyzing the single hydroxylations were monooxygenases derived from 9-10A (Table 7.3). Although several variants could produce metabolite **19**, only 9-10A F87V oxidized the other position of the morpholine ring and made derivative **20** (entry 3). The remaining twelve metabolite-producing variants were also monooxygenases of both 9-10A and chimeric origin, and produced a metabolite with Mw=238; this indicates morpholine loss and addition of water to the bis-aryl backbone. In all cases, LY294002 conversion was very low (< 15%) and correlated with the low rat liver microsome conversion for LY294002 (40%) relative to microsome activity on verapamil (85%) and astemizole (85%). LY294002 better evades both mammalian and bacterial P450-catalyzed C-H oxidation than verapamil and astemizole under the same conditions.

C.6. Activities on singly-hydroxylated metabolites

Of the twelve mammalian metabolites produced by the P450 BM3 variants, eight were made with sufficiently high selectivity to enable preparative scale production without further optimization (2, 3, 6, 11, 13, 14, 19, and 20). Of the seven new metabolites to which structures could be assigned, four were made with sufficiently high regioselectivity to enable preparative scale production (7, 10, 15, and 17). All eleven of the highly produced metabolites arose from single oxidations. Seven of the eight remaining metabolites are the products of two or more oxidations and are present as minor components in mixtures with the singly-hydroxylated products. Of the 69 individual reactions that underwent appreciable conversion (\geq 10% drug consumed) only five generated products of multiple C-H oxidations in \geq 10% abundance.

The remainder contained single-oxidation products, including mixtures of singlyoxidized species. The lack of bishydroxylated products could reflect a regiochemical effect in which the second C-H site is less reactive. In this case, one of the two possible singlehydroxylation products would be favored. Sixteen variants produced only one singlehydroxylation product; reaction mixtures from ten additional variants contained two (metabolites were present in 1:1 to 3:1 ratios in 6/11 reactions). It is also possible that the relative lack of bishydroxylation products reflects discrimination on the part of the enzymes in which the BM3 binds and hydroxylates the parent drug in preference to its metabolites. Because BM3-catalyzed C-H oxidation increases polarity by either unmasking heteroatoms or formally substituting hydrogen with a hydroxyl group, production of polyhydroxylated metabolites could be disfavored, for example, if the enzyme prefers a more hydrophobic substrate.

To assess enzyme activity on the metabolites produced by a single oxidation, all 120 enzymes were incubated with purified demethylated metabolites norverapamil (3) and desmethylastemizole (26) (11). Product distributions and extents of reaction were determined by HPLC. In every case in which an enzyme was active on both parent drug and metabolite,

regioselectivities were unchanged; this indicates that the metabolite and parent drug bind in similar orientations. Comparison of each enzyme's activity on the parent drug and the associated metabolite showed that enzymes from the alkane-evolved 9-10A lineage tended to be more active on astemizole and verapamil than on their metabolites (true of 49 out of 55 active enzymes; Table 7.4).

Table 7.4: Preference for more hydrophobic parent drug over its metabolite depends on evolutionary history of the variant.

D450 Variant family	Number of P450s more active	Number of P450s more active				
P430 Variant failing	on parent drug	on metabolite				
Alkane-selected lineage	48/55	6/55				
Propranolol-selected and chimera lineages	18/30	12/30				

Because neither the N-methyl group of verapamil nor the O-methyl group of astemizole is the preferred site for C-H oxidation by the alkane-evolved variants, this bias could reflect a preference for the more hydrophobic substrate. In contrast, when this pairwise reaction matrix is analyzed across the propranolol-evolved and chimeric P450 BM3 lineages, there is no statistical difference from random at the 95% confidence interval ($0.43 \le p = 0.60 \le 0.77$).

C.7. Directed evolution can improve metabolite production

None of the 120 members of this catalyst panel had been selected for activity on any of the three drugs. Thus, any initial activity represents a promiscuous activity; such side activities are often easy to improve by directed evolution (27 - 28). Of the 103 enzymes that reacted with these substrates, 9-10A F87L possessed the best combination of activity and regioselectivity (Table 7.2, entry 7); its reaction with astemizole, which produces metabolite **13** with 88% selectivity. We screened 2,000 variants made by error-prone PCR of the 9-10A F87L gene using a colorimetric screen for products of aromatic hydroxylation (29) and identified three new

sequences with improved metabolite production (Table 7.5). The new variants improved the conversion to 51-52% while preserving high regioselectivity (~ 80% for metabolite **13**).

1.5.	110	Judetion of user	m2010 metabon	1000 J J 1011 O D M
		Variant	% Conversion	% Selectivity 13
	1	9-10A F87L	34	88
	2	E4D D68G	51	80
	3	C205R D338G	51	80
	4	E4D H92Q	52	75

Table 7.5: Production of astemizole metabolites by 9-10A F87L variants

D. Discussion

A small, 120-member panel of P450 BM3 variants captured nearly all of the mammalian P450 scope of reactivity by producing 12 of 13 known metabolites. In their ability to mimic human CYPs, the P450 BM3 variants demonstrated considerable versatility, activating C-H bonds of varying strength (90–105 kcal mol⁻¹) and steric encumbrance (sp² vs. sp³ and 1° vs. 2° carbon centers). We were able to assign structures to seven new metabolites (Table 7.1 (b), 7.2 (b)), all of which have undergone oxidation at new carbon centers.

Cytochrome P450 enzymes are versatile catalysts, the biological activities of which are encoded in a wide range of primary sequences. The 26 CYPs, for which the crystal structures have been solved, possess as little as 15% sequence identity (*30*), but share a highly conserved fold. Emphasizing the versatility of the P450 fold, P450cam and P450cin catalyze the oxidation of isosteres camphor and cineole. Although the substrates are nearly identical, the two enzymes differ not only in sequence (27% identity), but also in the structure of their active sites, most notably a complete lack of the B' helix in P450cin (*31 – 32*). Whereas large changes in sequence are possible, only small perturbations are necessary to produce significant changes in function.

For example, CYP2A4 and CYP2A5 differ by only eleven amino acids, yet catalyze hydroxylations on structurally dissimilar coumarin and testosterone substrates (*33*).

Only 57 human enzymes (34) are responsible for known CYP-dependent drug metabolism, and a single enzyme, CYP3A4, accounts for > 50% of the burden for xenobiotic CYP-mediated clearance (35). Because CYPs can be broadly or narrowly specific, we and others have speculated that it should be possible to take advantage of the high native activity of P450 BM3 and use mutation to either relax or shift its substrate specificity in order to generate useful C-H oxidation catalysts (11, 36 - 39). For example, BM3 variant 9-10A, which is 13 mutations away from the wild type, exhibits broad activity across short- and medium-chain alkanesactivity that is low or completely absent in its wild-type parent (18 - 19). Furthermore, 9-10A could be respecialized to hydroxylate propane, preferring it over alkanes that differ by a single methylene group (40). We also showed previously that a variant of 9-10A was able to hydroxylate drug-like compounds efficiently and selectively (10). Here, we wanted to determine whether variants of 9-10A and other BM3-derived enzymes could cover or even exceed the broad substrate range of mammalian CYPs. Within any catalyst panel, both extremes of regioselectivity can be useful: enzymes that already possess the desired selectivity can be used to produce individual metabolites, whereas less selective enzymes can be used to survey metabolite possibilities. Both can serve as starting sequences for directed evolution to enhance activity or tune selectivity. That a systematic and general evolutionary algorithm can be used for catalyst improvement is a particularly appealing aspect of DNA-encoded reagents. Complementary optimization studies used in traditional synthesis methods usually rely on chemical intuition to improve a catalyst and require a good understanding of the catalytic mechanism.

When we examined conversion of the parent drug versus its demethylated metabolite, we noted that one enzyme family consistently converted more of the parent drug than the demethylated and more polar metabolite. These enzymes tend to catalyze single hydroxylations, while enzymes evolved for activity on propranolol and the chimeric variants often catalyze bis-hydroxylations. Structure–activity relationships of this type should help in the future to select enzymes that are most likely to react with as-of-yet-untested substrates and could also help predict product profiles.

E. Conclusion

This panel should enable rapid identification and production of relevant quantities of the human metabolites of drug candidates for pharmacological and toxicological evaluations in preclinical species (41 - 42). Although we have highlighted the potential of these enzymes to accelerate preparation of metabolites for pharmacological and toxicological testing, this enzyme panel is likely also to be useful further upstream in the drug-development process as general reagents for lead diversification. Reagents that rely on molecular recognition will always be restricted in their scope of use. However, because the functionality of small molecules is not evenly distributed across all possible molecular architectures (43), it should be worthwhile to engineer P450-derived reagents that are active on privileged scaffolds that reside in these densely functional regions of structure space. The plurality of C-H sites targeted by this small P450 BM3 variant set—including and extending human P450 metabolism—augurs well for the development of a truly general panel of C-H oxidation catalysts.

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