

*Appendix C*

## X-RAY DIFFRACTION REPORTS FROM THE BECKMAN INSTITUTE

X-ray structures for compounds **1-Ir(tma)<sub>2</sub>**, **1b-Ir(tma)<sub>2</sub>**, **1-Ir(py)<sub>2</sub>**, **1-Ir(dmap)<sub>2</sub>**, and **1-Ir(NH<sub>3</sub>)<sub>2</sub>** are listed below, along with full reports and analysis, including experimental setup, selected bond lengths, and other relevant parameters. These reports were produced by Michael W. Day at the Beckman Institute. The reports for **1-Ir(tma)<sub>2</sub>**, **1b-Ir(tma)<sub>2</sub>**, **1-Ir(py)<sub>2</sub>**, which were originally PDF files, have been reformatted for the Microsoft Word program, while the reports for **1-Ir(dmap)<sub>2</sub>**, and **1-Ir(NH<sub>3</sub>)<sub>2</sub>**, which were produced in Word format, have largely just been edited in small ways, and therefore appear somewhat different from the former three reports. The nomenclature in all cases has been altered to match that of the thesis.

CALIFORNIA INSTITUTE OF TECHNOLOGY  
BECKMAN INSTITUTE  
X-RAY CRYSTALLOGRAPHY LABORATORY

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**Crystal Structure Analysis of:**

**1-Ir(tma)<sub>2</sub>**

(shown below)

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Table 1. Crystal data

Figures Minimum overlap

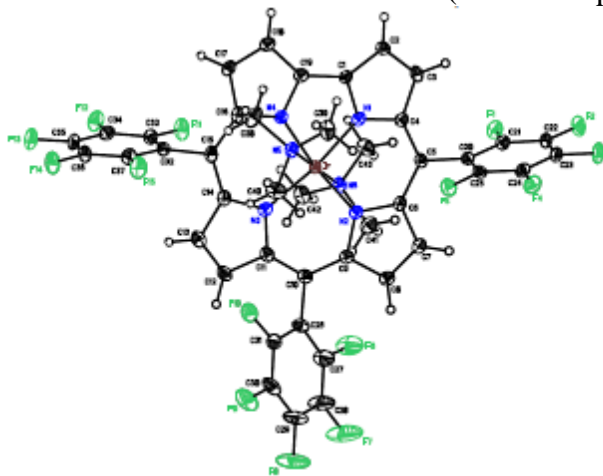
Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters

Table 6. Observed and calculated structure factors (available upon request)



**1-Ir(tma)<sub>2</sub>**

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and have been placed on hold pending further instructions from me. The deposition number is 671270. Ideally the CCDC would like the publication to contain a footnote of the type: “Crystallographic data have been deposited at the CCDC, 12 Union Road,

Cambridge CB2 1EZ, UK, and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 671270”.

Table C-1-1. Crystal data and structure refinement for **1-Ir(tma)<sub>2</sub>** (CCDC 671270)

Empirical formula	C <sub>43</sub> H <sub>26</sub> N <sub>6</sub> F <sub>15</sub> Ir • 0.185(CH <sub>2</sub> Cl <sub>2</sub> )		
Formula weight	1119.61		
Crystallization Solvent	Dichloromethane		
Crystal Habit	Needle		
Crystal size	0.32 x 0.07 x 0.02 mm <sup>3</sup>		
Crystal color	Blue		
<b>Data Collection</b>			
Type of diffractometer	Bruker KAPPA APEX II		
Wavelength	0.71073 Å MoKα		
Data Collection Temperature	100(2) K		
θ range for 9398 reflections used in lattice determination	2.54 to 28.61°		
Unit cell dimensions	a = 8.1456(4) Å	α = 80.628(3)°	
	b = 13.1188(6) Å	β = 80.543(3)°	
	c = 19.8366(8) Å	γ = 89.488(3)°	
Volume	2062.68(16) Å <sup>3</sup>		
Z	2		
Crystal system	Triclinic		
Space group	P-1		
Density (calculated)	1.803 Mg/m <sup>3</sup>		
F(000)	1092		
θ range for data collection	1.57 to 28.65°		
Completeness to θ = 28.65°	99.2 %		
Index ranges	-10 ≤ h ≤ 10, -17 ≤ k ≤ 17, -26 ≤ l ≤ 26		
Data collection scan type	ω scans; 22 settings		
Reflections collected	70274		
Independent reflections	10507 [R <sub>int</sub> = 0.0628]		
Absorption coefficient	3.370 mm <sup>-1</sup>		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7457 and 0.6119		

Table C-1-1 (cont.)

**Structure Solution and Refinement**

Structure solution program	SHELXS-97 (Sheldrick, 1990)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 1997)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	10507 / 0 / 620
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	1.818
Final R indices [ $I > 2\sigma(I)$ , 9104 reflections]	$R1 = 0.0330$ , $wR2 = 0.0548$
R indices (all data)	$R1 = 0.0458$ , $wR2 = 0.0561$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.001
Average shift/error	0.000
Largest diff. peak and hole	1.996 and -1.834 e.Å <sup>-3</sup>

**Special Refinement Details**

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

The crystals contain dichloromethane as a solvent of crystallization. It was refined with partial occupancy in the asymmetric unit with a final refined value of Occ = 0.185.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt), etc., and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles, and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

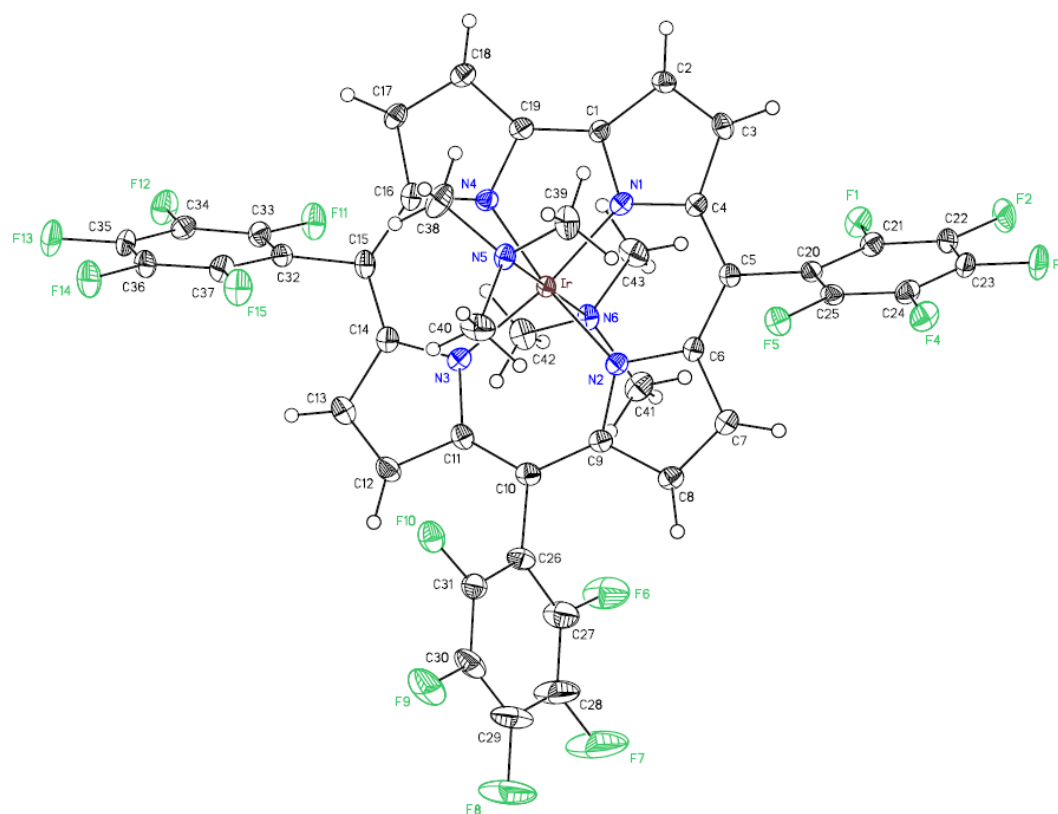


Table C-1-2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1-Ir(tma)<sub>2</sub>** (CCDC 671270).  $U_{eq}$  is defined as the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z	$U_{eq}$	Occ
Ir	1324(1)	2479(1)	7024(1)	12(1)	1
F(1)	2573(2)	6662(2)	6126(1)	22(1)	1
F(2)	1653(3)	8597(2)	5712(1)	29(1)	1
F(3)	-1593(3)	9019(2)	5636(1)	30(1)	1
F(4)	-3922(2)	7467(2)	5984(1)	24(1)	1
F(5)	-3029(2)	5525(2)	6378(1)	20(1)	1
F(6)	-857(3)	3257(2)	9648(1)	47(1)	1
F(7)	-3076(4)	3111(3)	10825(1)	77(1)	1
F(8)	-5838(4)	1883(3)	11010(1)	68(1)	1
F(9)	-6354(3)	799(2)	10003(1)	41(1)	1
F(10)	-4104(2)	887(2)	8849(1)	25(1)	1
F(11)	6068(2)	-569(2)	7429(1)	26(1)	1
F(12)	6967(2)	-2552(2)	7641(1)	27(1)	1
F(13)	4673(3)	-4087(2)	7761(1)	29(1)	1
F(14)	1475(3)	-3621(2)	7630(1)	28(1)	1
F(15)	577(2)	-1642(2)	7363(1)	26(1)	1
N(1)	2027(3)	3495(2)	6203(1)	14(1)	1
N(2)	-342(3)	3405(2)	7438(1)	15(1)	1
N(3)	807(3)	1315(2)	7794(1)	14(1)	1
N(4)	3056(3)	1707(2)	6516(1)	15(1)	1
N(5)	-560(3)	1907(2)	6503(1)	15(1)	1
N(6)	3187(3)	3064(2)	7548(1)	15(1)	1
C(1)	3191(4)	3214(2)	5691(2)	14(1)	1
C(2)	3443(4)	4078(3)	5149(2)	16(1)	1
C(3)	2414(4)	4851(3)	5346(2)	17(1)	1
C(4)	1519(4)	4486(2)	6029(2)	14(1)	1
C(5)	295(4)	4921(2)	6494(2)	14(1)	1
C(6)	-536(4)	4425(3)	7153(2)	14(1)	1
C(7)	-1670(4)	4855(3)	7659(2)	16(1)	1
C(8)	-2129(4)	4108(3)	8221(2)	17(1)	1
C(9)	-1263(4)	3180(3)	8087(2)	14(1)	1
C(10)	-1247(4)	2211(3)	8522(2)	16(1)	1
C(11)	-262(4)	1344(3)	8396(2)	16(1)	1
C(12)	-200(4)	356(3)	8833(2)	20(1)	1
C(13)	890(4)	-230(3)	8481(2)	21(1)	1
C(14)	1562(4)	364(3)	7815(2)	16(1)	1
C(15)	2758(4)	78(3)	7282(2)	17(1)	1
C(16)	3479(4)	701(3)	6654(2)	16(1)	1
C(17)	4585(4)	507(3)	6045(2)	17(1)	1
C(18)	4750(4)	1408(3)	5568(2)	17(1)	1
C(19)	3759(4)	2178(3)	5863(2)	16(1)	1
C(20)	-188(4)	6007(2)	6265(2)	14(1)	1
C(21)	952(4)	6826(3)	6082(2)	17(1)	1
C(22)	504(5)	7828(3)	5870(2)	20(1)	1
C(23)	-1140(5)	8048(3)	5836(2)	20(1)	1
C(24)	-2305(4)	7261(3)	6011(2)	18(1)	1
C(25)	-1832(4)	6264(3)	6217(2)	16(1)	1

C(26)	-2414(4)	2103(3)	9196(2)	20(1)	1
C(27)	-2212(5)	2640(3)	9724(2)	31(1)	1
C(28)	-3327(6)	2570(4)	10326(2)	45(1)	1
C(29)	-4733(6)	1952(4)	10428(2)	41(1)	1
C(30)	-4987(5)	1392(3)	9917(2)	28(1)	1
C(31)	-3829(4)	1461(3)	9321(2)	21(1)	1
C(32)	3288(4)	-1025(3)	7392(2)	15(1)	1
C(33)	4912(4)	-1306(3)	7460(2)	18(1)	1
C(34)	5384(4)	-2315(3)	7577(2)	18(1)	1
C(35)	4227(4)	-3099(3)	7635(2)	19(1)	1
C(36)	2605(4)	-2857(3)	7570(2)	19(1)	1
C(37)	2171(4)	-1838(3)	7440(2)	18(1)	1
C(38)	12(5)	1039(3)	6131(2)	26(1)	1
C(39)	-1127(5)	2734(3)	5990(2)	23(1)	1
C(40)	-2065(4)	1530(3)	7013(2)	26(1)	1
C(41)	2496(4)	3734(3)	8061(2)	22(1)	1
C(42)	4013(4)	2201(3)	7937(2)	21(1)	1
C(43)	4516(4)	3688(3)	7045(2)	23(1)	1
Cl(1)	6878(13)	5635(7)	9770(5)	95(4)	0.185(3)
Cl(2)	9171(10)	7272(6)	9083(4)	71(3)	0.185(3)
C(51)	8800(40)	5970(20)	9329(15)	58(8)	0.185(3)

Table C-1-3. Selected bond lengths [Å] and angles [°] for **1-Ir(tma)<sub>2</sub>** (CCDC 671270)

Ir-N(1)	1.940(3)	N(1)-Ir-N(4)	79.57(11)
Ir-N(4)	1.968(3)	N(1)-Ir-N(3)	172.20(12)
Ir-N(3)	1.970(3)	N(4)-Ir-N(3)	92.69(11)
Ir-N(2)	1.981(3)	N(1)-Ir-N(2)	93.09(11)
Ir-N(6)	2.184(3)	N(4)-Ir-N(2)	172.66(11)
Ir-N(5)	2.186(3)	N(3)-Ir-N(2)	94.64(11)
		N(1)-Ir-N(6)	90.92(11)
		N(4)-Ir-N(6)	89.98(11)
		N(3)-Ir-N(6)	90.11(10)
		N(2)-Ir-N(6)	89.78(11)
		N(1)-Ir-N(5)	89.02(11)
		N(4)-Ir-N(5)	90.67(11)
		N(3)-Ir-N(5)	90.04(11)
		N(2)-Ir-N(5)	89.55(11)
		N(6)-Ir-N(5)	179.32(12)



Table C-1-4. Bond lengths [Å] and angles [°] for **1-Ir(tma)<sub>2</sub>** (CCDC 671270)

Ir-N(1)	1.940(3)	C(14)-C(15)	1.411(5)
Ir-N(4)	1.968(3)	C(15)-C(16)	1.415(5)
Ir-N(3)	1.970(3)	C(15)-C(32)	1.499(5)
Ir-N(2)	1.981(3)	C(16)-C(17)	1.439(5)
Ir-N(6)	2.184(3)	C(17)-C(18)	1.381(5)
Ir-N(5)	2.186(3)	C(18)-C(19)	1.427(5)
F(1)-C(21)	1.351(4)	C(20)-C(21)	1.389(5)
F(2)-C(22)	1.345(4)	C(20)-C(25)	1.392(4)
F(3)-C(23)	1.338(4)	C(21)-C(22)	1.378(5)
F(4)-C(24)	1.350(4)	C(22)-C(23)	1.378(5)
F(5)-C(25)	1.343(4)	C(23)-C(24)	1.371(5)
F(6)-C(27)	1.348(5)	C(24)-C(25)	1.378(5)
F(7)-C(28)	1.349(5)	C(26)-C(27)	1.386(5)
F(8)-C(29)	1.333(5)	C(26)-C(31)	1.398(5)
F(9)-C(30)	1.336(5)	C(27)-C(28)	1.368(6)
F(10)-C(31)	1.340(4)	C(28)-C(29)	1.378(7)
F(11)-C(33)	1.342(4)	C(29)-C(30)	1.387(6)
F(12)-C(34)	1.346(4)	C(30)-C(31)	1.377(5)
F(13)-C(35)	1.338(4)	C(32)-C(37)	1.389(5)
F(14)-C(36)	1.344(4)	C(32)-C(33)	1.392(5)
F(15)-C(37)	1.350(4)	C(33)-C(34)	1.369(5)
N(1)-C(1)	1.366(4)	C(34)-C(35)	1.379(5)
N(1)-C(4)	1.368(4)	C(35)-C(36)	1.378(5)
N(2)-C(9)	1.368(4)	C(36)-C(37)	1.374(5)
N(2)-C(6)	1.383(4)	Cl(1)-C(51)	1.68(3)
N(3)-C(11)	1.363(4)	Cl(2)-C(51)	1.72(3)
N(3)-C(14)	1.383(4)		
N(4)-C(16)	1.357(4)	N(1)-Ir-N(4)	79.57(11)
N(4)-C(19)	1.375(4)	N(1)-Ir-N(3)	172.20(12)
N(5)-C(40)	1.487(4)	N(4)-Ir-N(3)	92.69(11)
N(5)-C(38)	1.487(5)	N(1)-Ir-N(2)	93.09(11)
N(5)-C(39)	1.487(4)	N(4)-Ir-N(2)	172.66(11)
N(6)-C(42)	1.483(4)	N(3)-Ir-N(2)	94.64(11)
N(6)-C(41)	1.489(5)	N(1)-Ir-N(6)	90.92(11)
N(6)-C(43)	1.495(4)	N(4)-Ir-N(6)	89.98(11)
C(1)-C(2)	1.420(4)	N(3)-Ir-N(6)	90.11(10)
C(1)-C(19)	1.437(5)	N(2)-Ir-N(6)	89.78(11)
C(2)-C(3)	1.375(5)	N(1)-Ir-N(5)	89.02(11)
C(3)-C(4)	1.438(5)	N(4)-Ir-N(5)	90.67(11)
C(4)-C(5)	1.424(5)	N(3)-Ir-N(5)	90.04(11)
C(5)-C(6)	1.425(4)	N(2)-Ir-N(5)	89.55(11)
C(5)-C(20)	1.493(4)	N(6)-Ir-N(5)	179.32(12)
C(6)-C(7)	1.431(5)	C(1)-N(1)-C(4)	111.6(3)
C(7)-C(8)	1.364(5)	C(1)-N(1)-Ir	118.0(2)
C(8)-C(9)	1.442(5)	C(4)-N(1)-Ir	130.4(2)
C(9)-C(10)	1.416(4)	C(9)-N(2)-C(6)	110.8(3)
C(10)-C(11)	1.418(5)	C(9)-N(2)-Ir	124.6(2)
C(10)-C(26)	1.494(5)	C(6)-N(2)-Ir	123.9(2)
C(11)-C(12)	1.443(4)	C(11)-N(3)-C(14)	110.2(3)
C(12)-C(13)	1.352(5)	C(11)-N(3)-Ir	125.4(2)
C(13)-C(14)	1.444(5)	C(14)-N(3)-Ir	124.3(2)

C(16)-N(4)-C(19)	112.7(3)	N(4)-C(19)-C(1)	112.8(3)
C(16)-N(4)-Ir	129.7(2)	C(18)-C(19)-C(1)	141.6(3)
C(19)-N(4)-Ir	116.5(2)	C(21)-C(20)-C(25)	115.4(3)
C(40)-N(5)-C(38)	107.0(3)	C(21)-C(20)-C(5)	122.9(3)
C(40)-N(5)-C(39)	106.3(3)	C(25)-C(20)-C(5)	121.7(3)
C(38)-N(5)-C(39)	107.2(3)	F(1)-C(21)-C(22)	117.1(3)
C(40)-N(5)-Ir	110.4(2)	F(1)-C(21)-C(20)	120.0(3)
C(38)-N(5)-Ir	113.6(2)	C(22)-C(21)-C(20)	122.9(3)
C(39)-N(5)-Ir	111.9(2)	F(2)-C(22)-C(23)	119.7(3)
C(42)-N(6)-C(41)	106.2(3)	F(2)-C(22)-C(21)	120.5(3)
C(42)-N(6)-C(43)	107.3(3)	C(23)-C(22)-C(21)	119.8(3)
C(41)-N(6)-C(43)	106.6(3)	F(3)-C(23)-C(24)	120.5(3)
C(42)-N(6)-Ir	110.9(2)	F(3)-C(23)-C(22)	120.3(3)
C(41)-N(6)-Ir	113.9(2)	C(24)-C(23)-C(22)	119.2(3)
C(43)-N(6)-Ir	111.6(2)	F(4)-C(24)-C(23)	119.8(3)
N(1)-C(1)-C(2)	106.5(3)	F(4)-C(24)-C(25)	120.0(3)
N(1)-C(1)-C(19)	112.8(3)	C(23)-C(24)-C(25)	120.2(3)
C(2)-C(1)-C(19)	140.7(3)	F(5)-C(25)-C(24)	117.4(3)
C(3)-C(2)-C(1)	108.2(3)	F(5)-C(25)-C(20)	120.0(3)
C(2)-C(3)-C(4)	108.0(3)	C(24)-C(25)-C(20)	122.6(3)
N(1)-C(4)-C(5)	120.2(3)	C(27)-C(26)-C(31)	115.6(3)
N(1)-C(4)-C(3)	105.7(3)	C(27)-C(26)-C(10)	123.4(3)
C(5)-C(4)-C(3)	134.1(3)	C(31)-C(26)-C(10)	121.0(3)
C(4)-C(5)-C(6)	126.9(3)	F(6)-C(27)-C(28)	117.9(4)
C(4)-C(5)-C(20)	116.8(3)	F(6)-C(27)-C(26)	119.1(3)
C(6)-C(5)-C(20)	116.3(3)	C(28)-C(27)-C(26)	123.0(4)
N(2)-C(6)-C(5)	125.2(3)	F(7)-C(28)-C(27)	120.7(4)
N(2)-C(6)-C(7)	106.0(3)	F(7)-C(28)-C(29)	119.2(4)
C(5)-C(6)-C(7)	128.7(3)	C(27)-C(28)-C(29)	120.1(4)
C(8)-C(7)-C(6)	108.8(3)	F(8)-C(29)-C(28)	120.9(4)
C(7)-C(8)-C(9)	107.7(3)	F(8)-C(29)-C(30)	120.0(4)
N(2)-C(9)-C(10)	123.7(3)	C(28)-C(29)-C(30)	119.2(4)
N(2)-C(9)-C(8)	106.7(3)	F(9)-C(30)-C(31)	120.5(4)
C(10)-C(9)-C(8)	129.6(3)	F(9)-C(30)-C(29)	120.0(4)
C(9)-C(10)-C(11)	128.0(3)	C(31)-C(30)-C(29)	119.5(4)
C(9)-C(10)-C(26)	115.3(3)	F(10)-C(31)-C(30)	117.5(3)
C(11)-C(10)-C(26)	116.7(3)	F(10)-C(31)-C(26)	119.9(3)
N(3)-C(11)-C(10)	123.3(3)	C(30)-C(31)-C(26)	122.6(4)
N(3)-C(11)-C(12)	107.5(3)	C(37)-C(32)-C(33)	115.6(3)
C(10)-C(11)-C(12)	129.2(3)	C(37)-C(32)-C(15)	121.7(3)
C(13)-C(12)-C(11)	107.4(3)	C(33)-C(32)-C(15)	122.7(3)
C(12)-C(13)-C(14)	108.9(3)	F(11)-C(33)-C(34)	117.8(3)
N(3)-C(14)-C(15)	125.0(3)	F(11)-C(33)-C(32)	119.6(3)
N(3)-C(14)-C(13)	106.0(3)	C(34)-C(33)-C(32)	122.6(3)
C(15)-C(14)-C(13)	129.0(3)	F(12)-C(34)-C(33)	120.6(3)
C(14)-C(15)-C(16)	127.8(3)	F(12)-C(34)-C(35)	119.4(3)
C(14)-C(15)-C(32)	115.6(3)	C(33)-C(34)-C(35)	120.0(3)
C(16)-C(15)-C(32)	116.6(3)	F(13)-C(35)-C(36)	120.3(3)
N(4)-C(16)-C(15)	120.0(3)	F(13)-C(35)-C(34)	120.3(3)
N(4)-C(16)-C(17)	105.6(3)	C(36)-C(35)-C(34)	119.5(3)
C(15)-C(16)-C(17)	134.3(3)	F(14)-C(36)-C(37)	121.1(3)
C(18)-C(17)-C(16)	107.9(3)	F(14)-C(36)-C(35)	119.5(3)
C(17)-C(18)-C(19)	108.4(3)	C(37)-C(36)-C(35)	119.4(3)
N(4)-C(19)-C(18)	105.4(3)	F(15)-C(37)-C(36)	117.1(3)
F(15)-C(37)-C(32)	119.9(3)	Cl(1)-C(51)-Cl(2)	115.1(16)
C(36)-C(37)-C(32)	122.9(3)		

Table C-1-5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for **1-Ir(tma)<sub>2</sub>** (CCDC 671270). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$ .

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
Ir	84(1)	112(1)	151(1)	-24(1)	-12(1)	-12(1)
F(1)	125(11)	215(11)	329(12)	-94(9)	-24(9)	-44(8)
F(2)	288(13)	160(11)	402(13)	-42(9)	19(10)	-112(9)
F(3)	410(14)	118(10)	357(13)	-29(9)	-72(11)	63(9)
F(4)	184(11)	255(11)	285(12)	-61(9)	-66(9)	95(9)
F(5)	136(10)	204(11)	250(11)	-13(9)	-45(9)	-31(8)
F(6)	426(16)	668(18)	344(14)	-232(13)	-7(12)	-238(13)
F(7)	790(20)	1220(30)	327(15)	-425(17)	121(15)	-290(20)
F(8)	552(19)	1040(30)	328(15)	-117(16)	267(14)	-123(17)
F(9)	187(12)	430(15)	494(15)	108(12)	95(11)	-67(11)
F(10)	196(11)	218(11)	311(12)	-6(9)	-33(9)	-39(9)
F(11)	145(11)	183(11)	453(14)	-42(10)	-62(10)	-45(9)
F(12)	154(11)	225(11)	423(13)	-77(10)	-45(10)	63(9)
F(13)	290(13)	137(11)	448(14)	-61(10)	-51(10)	37(9)
F(14)	257(12)	173(11)	417(13)	-45(10)	-85(10)	-85(9)
F(15)	154(11)	235(11)	405(13)	-49(10)	-111(10)	-17(9)
N(1)	89(15)	161(15)	172(15)	-51(12)	-29(12)	12(12)
N(2)	128(15)	154(15)	159(15)	-35(12)	-24(12)	10(12)
N(3)	73(14)	172(15)	185(15)	-57(12)	-18(12)	-5(12)
N(4)	122(15)	155(15)	160(15)	-36(12)	0(12)	7(12)
N(5)	119(15)	136(15)	214(16)	-43(12)	-37(13)	-17(12)
N(6)	114(15)	139(15)	198(16)	-15(12)	-38(12)	-12(12)
C(1)	124(18)	136(17)	165(17)	-48(14)	-16(14)	-16(14)
C(2)	134(18)	179(18)	151(17)	-30(14)	-9(14)	-48(14)
C(3)	151(19)	144(17)	214(19)	0(14)	-32(15)	-37(14)
C(4)	105(17)	147(17)	169(17)	-47(14)	-34(14)	-18(13)
C(5)	108(17)	137(16)	171(18)	-22(14)	-31(14)	-18(13)
C(6)	115(17)	135(17)	188(18)	-37(14)	-38(14)	-3(13)
C(7)	119(17)	170(17)	199(18)	-56(14)	-51(14)	9(14)
C(8)	114(18)	202(19)	190(18)	-50(15)	-9(14)	-3(14)
C(9)	96(17)	173(17)	161(17)	-38(14)	-22(14)	-21(14)
C(10)	117(18)	198(18)	174(18)	-33(14)	-10(14)	-38(14)
C(11)	111(17)	166(18)	202(18)	-5(14)	-28(14)	-26(14)
C(12)	171(19)	192(19)	205(19)	20(15)	11(15)	-13(15)
C(13)	190(20)	183(18)	225(19)	13(15)	-42(16)	-21(15)
C(14)	117(18)	151(17)	216(19)	-27(14)	-39(15)	-3(14)
C(15)	132(18)	131(17)	245(19)	-29(15)	-65(15)	-23(14)
C(16)	125(18)	128(17)	237(19)	-37(14)	-64(15)	5(14)
C(17)	132(18)	166(18)	225(19)	-72(15)	-29(15)	23(14)
C(18)	125(18)	199(18)	197(18)	-66(15)	-13(14)	-8(14)
C(19)	103(17)	183(18)	180(18)	-54(14)	-1(14)	-34(14)
C(20)	162(18)	141(17)	129(17)	-26(13)	-23(14)	-9(14)
C(21)	152(18)	172(17)	184(18)	-50(14)	-34(15)	-10(14)
C(22)	260(20)	129(17)	191(18)	-47(14)	-3(16)	-61(15)
C(23)	300(20)	125(17)	177(18)	-31(14)	-35(16)	33(15)
C(24)	166(19)	194(18)	189(18)	-49(15)	-36(15)	58(15)
C(25)	148(18)	170(17)	143(17)	-30(14)	1(14)	-39(14)

C(26)	152(19)	252(19)	163(18)	-15(15)	8(15)	-7(15)
C(27)	290(20)	390(20)	230(20)	-68(18)	0(18)	-77(19)
C(28)	500(30)	670(30)	190(20)	-170(20)	30(20)	-70(30)
C(29)	390(30)	570(30)	200(20)	-20(20)	100(20)	-10(20)
C(30)	190(20)	300(20)	290(20)	80(18)	37(17)	-39(17)
C(31)	210(20)	204(19)	199(19)	-7(15)	-19(16)	54(15)
C(32)	159(18)	140(17)	159(17)	-24(14)	-21(14)	-11(14)
C(33)	166(19)	153(17)	214(19)	-52(14)	-13(15)	-46(14)
C(34)	140(18)	200(18)	200(18)	-48(15)	-10(15)	33(14)
C(35)	220(20)	120(17)	228(19)	-35(14)	-30(16)	38(14)
C(36)	210(20)	146(17)	232(19)	-41(14)	-33(16)	-78(15)
C(37)	130(18)	190(18)	239(19)	-47(15)	-40(15)	12(14)
C(38)	220(20)	250(20)	380(20)	-172(18)	-117(18)	15(16)
C(39)	210(20)	191(19)	320(20)	-32(16)	-135(17)	-18(16)
C(40)	160(20)	380(20)	250(20)	-63(17)	-10(16)	-108(17)
C(41)	190(20)	250(20)	270(20)	-104(16)	-77(16)	-18(16)
C(42)	150(20)	220(20)	280(20)	-6(16)	-93(16)	-12(15)
C(43)	163(19)	270(20)	240(20)	-18(16)	-47(16)	-91(16)
<hr/>						
Cl(1)	1080(80)	550(50)	1040(70)	-190(50)	450(60)	-70(50)
Cl(2)	560(50)	710(60)	840(60)	70(40)	-220(40)	-20(40)
C(51)	600(200)	500(170)	700(200)	-430(150)	-200(160)	20(140)

CALIFORNIA INSTITUTE OF TECHNOLOGY  
BECKMAN INSTITUTE  
X-RAY CRYSTALLOGRAPHY LABORATORY

Date 24 January 2008

**Crystal Structure Analysis of:**

**1b-Ir(tma)<sub>2</sub>**

(shown below)

**For** Investigator: Joshua Palmer ext. 6332

Advisor: H. B. Gray ext. 6500

Account Number: HGB.BP-1-BP.AMOCO

**By** Michael W. Day 116 Beckman ext. 2734

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Contents

Table 1. Crystal data

Figures Minimum overlap

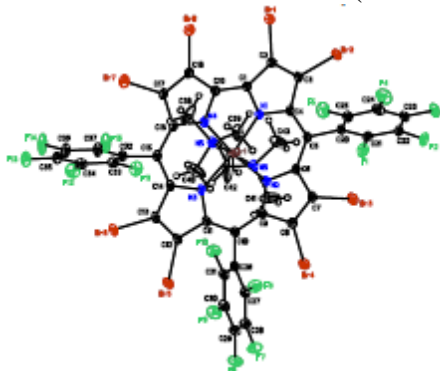
Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters

Table 6. Observed and calculated structure factors (available upon request)



**1b-Ir(tma)<sub>2</sub>**

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and have been placed on hold pending further instructions from me. The deposition number is 675602. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 657602".



Table C-2-1. Crystal data and structure refinement for **1b-Ir(tma)<sub>2</sub>** (CCDC 675602)

Empirical formula	C <sub>43</sub> H <sub>18</sub> N <sub>6</sub> Br <sub>3</sub> F <sub>15</sub> Ir, 3(CH <sub>2</sub> Cl <sub>2</sub> )		
Formula weight	1989.89		
Crystallization Solvent	Dichloromethane		
Crystal Habit	Block		
Crystal size	0.46 x 0.25 x 0.14 mm <sup>3</sup>		
Crystal color	Dark green		
<b>Data Collection</b>			
Type of diffractometer	Bruker KAPPA APEX II		
Wavelength	0.71073 Å MoKα		
Data Collection Temperature	100(2) K		
θ range for 9503 reflections used in lattice determination	2.58 to 43.28°		
Unit cell dimensions	a = 10.5456(5) Å b = 12.9293(6) Å c = 21.2776(9) Å	α = 81.591(3)° β = 76.344(3)° γ = 86.714(3)°	
Volume	2788.0(2) Å <sup>3</sup>		
Z	2		
Crystal system	Triclinic		
Space group	P-1		
Density (calculated)	2.370 Mg/m <sup>3</sup>		
F(000)	1872		
θ range for data collection	1.76 to 43.48°		
Completeness to θ = 43.48°	96.6 %		
Index ranges	-20 ≤ h ≤ 20, -24 ≤ k ≤ 22, -41 ≤ l ≤ 40		
Data collection scan type	ω scans; 24 settings		
Reflections collected	177276		
Independent reflections	40945 [R <sub>int</sub> = 0.0501]		
Absorption coefficient	8.508 mm <sup>-1</sup>		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7485 and 0.4771		

Table C-2-1 (cont.)

### Structure Solution and Refinement

Structure solution program	SHELXS-97 (Sheldrick, 1990)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 1997)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	40945 / 0 / 764
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	1.534
Final R indices [ $I > 2\sigma(I)$ , 28576 reflections]	$R1 = 0.0412$ , $wR2 = 0.0606$
R indices (all data)	$R1 = 0.0793$ , $wR2 = 0.0634$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.004
Average shift/error	0.000
Largest diff. peak and hole	4.350 and -2.801 e.Å <sup>-3</sup>

### Special Refinement Details

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

The crystals contain dichloromethane as a solvent of crystallization—three in the asymmetric unit. One of those three is disordered with the chlorine atoms occupying two orientations while sharing a common carbon atom. The disorder was modeled without restraint, except that the total chlorine population was restrained so that the orientations' occupancy summed to one.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt), etc., and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles, and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

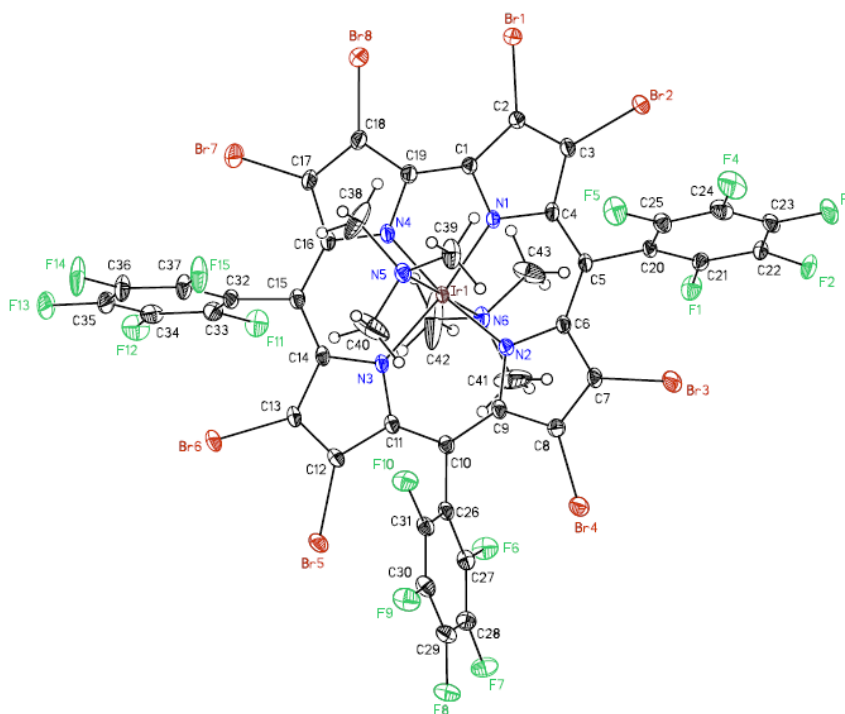




Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1b-Ir(tma)<sub>2</sub>** (CCDC 675602).  $U(\text{eq})$  is defined as the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z	$U_{\text{eq}}$	Occ
Ir(1)	5600(1)	2707(1)	2759(1)	11(1)	1
Br(1)	3909(1)	1493(1)	5558(1)	21(1)	1
Br(2)	2518(1)	-307(1)	4875(1)	21(1)	1
Br(3)	2815(1)	-505(1)	1893(1)	24(1)	1
Br(4)	4474(1)	911(1)	528(1)	20(1)	1
Br(5)	8185(1)	4625(1)	73(1)	19(1)	1
Br(6)	8884(1)	6129(1)	1101(1)	20(1)	1
Br(7)	7691(1)	5625(1)	4165(1)	25(1)	1
Br(8)	5967(1)	3715(1)	5284(1)	21(1)	1
F(1)	4745(1)	-1423(1)	3164(1)	23(1)	1
F(2)	3299(1)	-3156(1)	3442(1)	27(1)	1
F(3)	664(1)	-2985(1)	3816(1)	31(1)	1
F(4)	-531(1)	-1066(1)	3821(1)	33(1)	1
F(5)	909(1)	668(1)	3487(1)	28(1)	1
F(6)	8099(1)	1767(1)	282(1)	23(1)	1
F(7)	8270(1)	1915(1)	-1015(1)	24(1)	1
F(8)	6419(1)	3027(1)	-1548(1)	25(1)	1
F(9)	4470(1)	4040(1)	-792(1)	26(1)	1
F(10)	4310(1)	3925(1)	500(1)	22(1)	1
F(11)	10087(1)	4930(1)	2599(1)	25(1)	1
F(12)	11309(1)	6752(1)	2474(1)	29(1)	1
F(13)	9936(1)	8587(1)	2426(1)	31(1)	1
F(14)	7336(1)	8600(1)	2531(1)	42(1)	1
F(15)	6107(1)	6777(1)	2626(1)	33(1)	1
N(1)	4654(2)	1931(1)	3581(1)	13(1)	1
N(2)	4941(2)	1826(1)	2219(1)	14(1)	1
N(3)	6639(1)	3618(1)	2001(1)	13(1)	1
N(4)	6125(2)	3447(1)	3397(1)	14(1)	1
N(5)	3890(2)	3752(1)	2756(1)	16(1)	1
N(6)	7308(2)	1662(1)	2749(1)	17(1)	1
C(1)	4824(2)	2224(1)	4151(1)	14(1)	1
C(2)	4113(2)	1495(1)	4663(1)	15(1)	1
C(3)	3560(2)	784(1)	4385(1)	15(1)	1
C(4)	3926(2)	1065(1)	3685(1)	13(1)	1
C(5)	3677(2)	624(1)	3152(1)	13(1)	1
C(6)	4142(2)	987(1)	2479(1)	14(1)	1
C(7)	3882(2)	585(1)	1915(1)	16(1)	1
C(8)	4529(2)	1174(1)	1364(1)	15(1)	1
C(9)	5232(2)	1975(1)	1547(1)	14(1)	1
C(10)	6058(2)	2754(1)	1153(1)	14(1)	1
C(11)	6738(2)	3512(1)	1358(1)	13(1)	1
C(12)	7557(2)	4341(1)	979(1)	14(1)	1
C(13)	7880(2)	4931(1)	1397(1)	14(1)	1
C(14)	7305(2)	4479(1)	2058(1)	13(1)	1
C(15)	7380(2)	4814(1)	2654(1)	15(1)	1
C(16)	6835(2)	4323(1)	3292(1)	15(1)	1
C(17)	6849(2)	4550(1)	3935(1)	16(1)	1

C(18)	6142(2)	3799(1)	4387(1)	16(1)	1
C(19)	5653(2)	3094(1)	4043(1)	15(1)	1
C(20)	2869(2)	-329(1)	3331(1)	13(1)	1
C(21)	3447(2)	-1321(1)	3334(1)	15(1)	1
C(22)	2711(2)	-2209(1)	3483(1)	18(1)	1
C(23)	1378(2)	-2127(2)	3662(1)	21(1)	1
C(24)	771(2)	-1146(2)	3665(1)	22(1)	1
C(25)	1523(2)	-275(2)	3496(1)	17(1)	1
C(26)	6188(2)	2829(1)	433(1)	14(1)	1
C(27)	7183(2)	2327(1)	28(1)	16(1)	1
C(28)	7280(2)	2390(2)	-635(1)	17(1)	1
C(29)	6358(2)	2968(2)	-911(1)	18(1)	1
C(30)	5365(2)	3485(2)	-520(1)	18(1)	1
C(31)	5296(2)	3414(1)	136(1)	16(1)	1
C(32)	8063(2)	5808(1)	2612(1)	15(1)	1
C(33)	9384(2)	5832(2)	2583(1)	17(1)	1
C(34)	10026(2)	6748(2)	2517(1)	20(1)	1
C(35)	9323(2)	7686(2)	2494(1)	22(1)	1
C(36)	8009(2)	7688(2)	2544(1)	24(1)	1
C(37)	7387(2)	6761(2)	2597(1)	20(1)	1
C(38)	3731(2)	4500(2)	3240(2)	39(1)	1
C(39)	2666(2)	3168(2)	2920(1)	25(1)	1
C(40)	3956(3)	4381(2)	2113(1)	47(1)	1
C(41)	7656(3)	1137(2)	2154(1)	41(1)	1
C(42)	8466(3)	2229(2)	2770(2)	72(1)	1
C(43)	7110(2)	815(2)	3308(1)	32(1)	1
C(51)	2473(3)	626(2)	9260(1)	36(1)	1
Cl(1)	926(1)	936(1)	9104(1)	40(1)	1
Cl(2)	3616(1)	1580(1)	8896(1)	57(1)	1
C(52)	1202(4)	2774(2)	707(1)	53(1)	1
Cl(3)	1075(1)	3496(1)	1378(1)	50(1)	1
Cl(4)	773(1)	1482(1)	931(1)	58(1)	1
C(53)	10193(4)	6577(2)	5492(2)	53(1)	1
Cl(5A)	9121(2)	5785(1)	5443(1)	54(1)	0.526(2)
Cl(6A)	9884(2)	7892(1)	5425(1)	55(1)	0.526(2)
Cl(5B)	8684(1)	7181(1)	5220(1)	41(1)	0.474(2)
Cl(6B)	10756(2)	7409(1)	5851(1)	54(1)	0.474(2)

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Table C-2-3. Selected bond lengths [Å] and angles [°] for **1b-Ir(tma)<sub>2</sub>** (CCDC 675602)

Ir(1)-N(1)	1.9585(16)	N(1)-Ir(1)-N(4)	79.08(6)
Ir(1)-N(4)	1.9630(14)	N(1)-Ir(1)-N(3)	172.15(6)
Ir(1)-N(3)	1.9870(15)	N(4)-Ir(1)-N(3)	93.08(6)
Ir(1)-N(2)	1.9893(14)	N(1)-Ir(1)-N(2)	93.07(6)
Ir(1)-N(6)	2.1865(15)	N(4)-Ir(1)-N(2)	172.13(6)
Ir(1)-N(5)	2.1918(14)	N(3)-Ir(1)-N(2)	94.77(6)
		N(1)-Ir(1)-N(6)	91.33(6)
		N(4)-Ir(1)-N(6)	89.55(6)
		N(3)-Ir(1)-N(6)	88.99(6)
		N(2)-Ir(1)-N(6)	90.10(6)
		N(1)-Ir(1)-N(5)	89.15(6)
		N(4)-Ir(1)-N(5)	91.07(6)
		N(3)-Ir(1)-N(5)	90.61(6)
		N(2)-Ir(1)-N(5)	89.33(6)
		N(6)-Ir(1)-N(5)	179.27(6)

Table C-2-4. Bond lengths [Å] and angles [°] for **1b-Ir(tma)<sub>2</sub>** (CCDC 675602)

Ir(1)-N(1)	1.9585(16)	C(7)-C(8)	1.358(3)
Ir(1)-N(4)	1.9630(14)	C(8)-C(9)	1.456(2)
Ir(1)-N(3)	1.9870(15)	C(9)-C(10)	1.409(3)
Ir(1)-N(2)	1.9893(14)	C(10)-C(11)	1.421(2)
Ir(1)-N(6)	2.1865(15)	C(10)-C(26)	1.495(2)
Ir(1)-N(5)	2.1918(14)	C(11)-C(12)	1.439(3)
Br(1)-C(2)	1.8634(17)	C(12)-C(13)	1.363(2)
Br(2)-C(3)	1.8684(19)	C(13)-C(14)	1.445(3)
Br(3)-C(7)	1.8656(18)	C(14)-C(15)	1.419(2)
Br(4)-C(8)	1.8733(16)	C(15)-C(16)	1.414(3)
Br(5)-C(12)	1.8721(17)	C(15)-C(32)	1.490(3)
Br(6)-C(13)	1.8726(19)	C(16)-C(17)	1.444(2)
Br(7)-C(17)	1.8717(18)	C(17)-C(18)	1.377(3)
Br(8)-C(18)	1.8603(18)	C(18)-C(19)	1.436(2)
F(1)-C(21)	1.335(2)	C(20)-C(25)	1.379(3)
F(2)-C(22)	1.345(2)	C(20)-C(21)	1.388(2)
F(3)-C(23)	1.333(2)	C(21)-C(22)	1.378(3)
F(4)-C(24)	1.335(2)	C(22)-C(23)	1.369(3)
F(5)-C(25)	1.347(2)	C(23)-C(24)	1.387(3)
F(6)-C(27)	1.343(2)	C(24)-C(25)	1.371(3)
F(7)-C(28)	1.3392(19)	C(26)-C(31)	1.384(2)
F(8)-C(29)	1.332(2)	C(26)-C(27)	1.387(2)
F(9)-C(30)	1.341(2)	C(27)-C(28)	1.380(2)
F(10)-C(31)	1.3449(19)	C(28)-C(29)	1.381(3)
F(11)-C(33)	1.346(2)	C(29)-C(30)	1.381(2)
F(12)-C(34)	1.335(2)	C(30)-C(31)	1.372(2)
F(13)-C(35)	1.337(2)	C(32)-C(33)	1.382(3)
F(14)-C(36)	1.341(2)	C(32)-C(37)	1.389(2)
F(15)-C(37)	1.336(2)	C(33)-C(34)	1.372(3)
N(1)-C(4)	1.357(2)	C(34)-C(35)	1.385(3)
N(1)-C(1)	1.375(2)	C(35)-C(36)	1.364(3)
N(2)-C(9)	1.375(2)	C(36)-C(37)	1.376(3)
N(2)-C(6)	1.378(2)	C(51)-Cl(2)	1.739(3)
N(3)-C(11)	1.373(2)	C(51)-Cl(1)	1.753(2)
N(3)-C(14)	1.384(2)	C(52)-Cl(4)	1.725(3)
N(4)-C(16)	1.356(2)	C(52)-Cl(3)	1.793(3)
N(4)-C(19)	1.367(2)	C(53)-Cl(5A)	1.599(4)
N(5)-C(40)	1.475(3)	C(53)-Cl(6B)	1.618(4)
N(5)-C(39)	1.476(3)	C(53)-Cl(6A)	1.704(3)
N(5)-C(38)	1.488(2)	C(53)-Cl(5B)	1.907(4)
N(6)-C(42)	1.472(3)		
N(6)-C(43)	1.478(3)	N(1)-Ir(1)-N(4)	79.08(6)
N(6)-C(41)	1.484(2)	N(1)-Ir(1)-N(3)	172.15(6)
C(1)-C(19)	1.423(3)	N(4)-Ir(1)-N(3)	93.08(6)
C(1)-C(2)	1.426(3)	N(1)-Ir(1)-N(2)	93.07(6)
C(2)-C(3)	1.386(2)	N(4)-Ir(1)-N(2)	172.13(6)
C(3)-C(4)	1.445(2)	N(3)-Ir(1)-N(2)	94.77(6)
C(4)-C(5)	1.423(2)	N(1)-Ir(1)-N(6)	91.33(6)
C(5)-C(6)	1.419(2)	N(4)-Ir(1)-N(6)	89.55(6)
C(5)-C(20)	1.490(3)	N(3)-Ir(1)-N(6)	88.99(6)
C(6)-C(7)	1.461(2)	N(2)-Ir(1)-N(6)	90.10(6)

N(1)-Ir(1)-N(5)	89.15(6)	N(2)-C(9)-C(8)	105.35(16)
N(4)-Ir(1)-N(5)	91.07(6)	C(10)-C(9)-C(8)	130.06(15)
N(3)-Ir(1)-N(5)	90.61(6)	C(9)-C(10)-C(11)	127.82(15)
N(2)-Ir(1)-N(5)	89.33(6)	C(9)-C(10)-C(26)	116.35(15)
N(6)-Ir(1)-N(5)	179.27(6)	C(11)-C(10)-C(26)	115.76(16)
C(4)-N(1)-C(1)	112.63(15)	N(3)-C(11)-C(10)	123.67(16)
C(4)-N(1)-Ir(1)	129.68(11)	N(3)-C(11)-C(12)	106.17(14)
C(1)-N(1)-Ir(1)	117.38(12)	C(10)-C(11)-C(12)	130.03(15)
C(9)-N(2)-C(6)	112.37(14)	C(13)-C(12)-C(11)	108.37(15)
C(9)-N(2)-Ir(1)	124.19(13)	C(13)-C(12)-Br(5)	122.03(15)
C(6)-N(2)-Ir(1)	123.44(11)	C(11)-C(12)-Br(5)	129.58(13)
C(11)-N(3)-C(14)	111.38(15)	C(12)-C(13)-C(14)	108.59(16)
C(11)-N(3)-Ir(1)	124.90(12)	C(12)-C(13)-Br(6)	122.26(14)
C(14)-N(3)-Ir(1)	123.64(11)	C(14)-C(13)-Br(6)	129.15(13)
C(16)-N(4)-C(19)	113.34(14)	N(3)-C(14)-C(15)	125.53(17)
C(16)-N(4)-Ir(1)	129.08(13)	N(3)-C(14)-C(13)	105.44(14)
C(19)-N(4)-Ir(1)	117.42(12)	C(15)-C(14)-C(13)	129.03(17)
C(40)-N(5)-C(39)	106.81(19)	C(16)-C(15)-C(14)	126.93(17)
C(40)-N(5)-C(38)	106.9(2)	C(16)-C(15)-C(32)	115.67(15)
C(39)-N(5)-C(38)	106.20(17)	C(14)-C(15)-C(32)	117.36(16)
C(40)-N(5)-Ir(1)	112.08(12)	N(4)-C(16)-C(15)	121.45(15)
C(39)-N(5)-Ir(1)	111.87(11)	N(4)-C(16)-C(17)	105.17(16)
C(38)-N(5)-Ir(1)	112.58(12)	C(15)-C(16)-C(17)	133.39(17)
C(42)-N(6)-C(43)	106.8(2)	C(18)-C(17)-C(16)	108.05(15)
C(42)-N(6)-C(41)	107.0(2)	C(18)-C(17)-Br(7)	123.09(13)
C(43)-N(6)-C(41)	106.00(17)	C(16)-C(17)-Br(7)	128.83(15)
C(42)-N(6)-Ir(1)	111.72(12)	C(17)-C(18)-C(19)	108.27(15)
C(43)-N(6)-Ir(1)	112.88(12)	C(17)-C(18)-Br(8)	123.95(13)
C(41)-N(6)-Ir(1)	112.03(12)	C(19)-C(18)-Br(8)	127.70(15)
N(1)-C(1)-C(19)	112.90(16)	N(4)-C(19)-C(1)	113.19(15)
N(1)-C(1)-C(2)	105.70(15)	N(4)-C(19)-C(18)	105.14(16)
C(19)-C(1)-C(2)	141.36(16)	C(1)-C(19)-C(18)	141.56(17)
C(3)-C(2)-C(1)	108.27(15)	C(25)-C(20)-C(21)	116.82(17)
C(3)-C(2)-Br(1)	124.84(15)	C(25)-C(20)-C(5)	122.19(15)
C(1)-C(2)-Br(1)	126.89(13)	C(21)-C(20)-C(5)	120.99(16)
C(2)-C(3)-C(4)	107.76(16)	F(1)-C(21)-C(22)	118.77(16)
C(2)-C(3)-Br(2)	123.21(13)	F(1)-C(21)-C(20)	119.55(17)
C(4)-C(3)-Br(2)	129.02(13)	C(22)-C(21)-C(20)	121.63(17)
N(1)-C(4)-C(5)	120.88(16)	F(2)-C(22)-C(23)	119.87(18)
N(1)-C(4)-C(3)	105.61(14)	F(2)-C(22)-C(21)	120.17(18)
C(5)-C(4)-C(3)	133.51(17)	C(23)-C(22)-C(21)	119.94(17)
C(6)-C(5)-C(4)	126.60(17)	F(3)-C(23)-C(22)	120.14(18)
C(6)-C(5)-C(20)	117.69(15)	F(3)-C(23)-C(24)	120.06(19)
C(4)-C(5)-C(20)	115.70(15)	C(22)-C(23)-C(24)	119.78(18)
N(2)-C(6)-C(5)	126.18(15)	F(4)-C(24)-C(25)	121.18(18)
N(2)-C(6)-C(7)	105.08(15)	F(4)-C(24)-C(23)	119.73(19)
C(5)-C(6)-C(7)	128.74(17)	C(25)-C(24)-C(23)	119.08(18)
C(8)-C(7)-C(6)	108.54(16)	F(5)-C(25)-C(24)	117.90(17)
C(8)-C(7)-Br(3)	122.16(13)	F(5)-C(25)-C(20)	119.40(17)
C(6)-C(7)-Br(3)	129.26(14)	C(24)-C(25)-C(20)	122.69(17)
C(7)-C(8)-C(9)	108.64(15)	C(31)-C(26)-C(27)	116.39(16)
C(7)-C(8)-Br(4)	122.43(14)	C(31)-C(26)-C(10)	120.65(15)
C(9)-C(8)-Br(4)	128.92(14)	C(27)-C(26)-C(10)	122.96(15)
N(2)-C(9)-C(10)	124.59(15)	F(6)-C(27)-C(28)	118.20(15)



F(6)-C(27)-C(26)	119.42(15)	F(12)-C(34)-C(33)	121.32(17)
C(28)-C(27)-C(26)	122.38(16)	F(12)-C(34)-C(35)	119.68(18)
F(7)-C(28)-C(27)	120.86(16)	C(33)-C(34)-C(35)	119.00(18)
F(7)-C(28)-C(29)	119.62(16)	F(13)-C(35)-C(36)	120.29(18)
C(27)-C(28)-C(29)	119.51(16)	F(13)-C(35)-C(34)	119.72(19)
F(8)-C(29)-C(28)	120.62(16)	C(36)-C(35)-C(34)	119.99(19)
F(8)-C(29)-C(30)	120.02(16)	F(14)-C(36)-C(35)	119.39(19)
C(28)-C(29)-C(30)	119.36(16)	F(14)-C(36)-C(37)	120.47(19)
F(9)-C(30)-C(31)	121.11(16)	C(35)-C(36)-C(37)	120.13(18)
F(9)-C(30)-C(29)	118.98(16)	F(15)-C(37)-C(36)	119.48(16)
C(31)-C(30)-C(29)	119.90(16)	F(15)-C(37)-C(32)	119.03(17)
F(10)-C(31)-C(30)	117.99(15)	C(36)-C(37)-C(32)	121.48(18)
F(10)-C(31)-C(26)	119.55(15)	Cl(2)-C(51)-Cl(1)	113.23(14)
C(30)-C(31)-C(26)	122.45(16)	Cl(4)-C(52)-Cl(3)	114.43(17)
C(33)-C(32)-C(37)	116.86(17)	Cl(5A)-C(53)-Cl(6B)	153.0(2)
C(33)-C(32)-C(15)	122.33(16)	Cl(5A)-C(53)-Cl(6A)	121.1(2)
C(37)-C(32)-C(15)	120.81(16)	Cl(6B)-C(53)-Cl(6A)	53.20(12)
F(11)-C(33)-C(34)	118.05(17)	Cl(5A)-C(53)-Cl(5B)	63.30(15)
F(11)-C(33)-C(32)	119.44(17)	Cl(6B)-C(53)-Cl(5B)	108.72(16)
C(34)-C(33)-C(32)	122.48(17)	Cl(6A)-C(53)-Cl(5B)	58.54(12)

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Table C-2-5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for **1b-Ir(tma)<sub>2</sub>** (CCDC 675602). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$ .

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
Ir(1)	107(1)	77(1)	142(1)	-10(1)	-5(1)	-15(1)
Br(1)	276(1)	182(1)	147(1)	-38(1)	18(1)	-67(1)
Br(2)	254(1)	174(1)	170(1)	-11(1)	22(1)	-93(1)
Br(3)	279(1)	253(1)	190(1)	-6(1)	-46(1)	-175(1)
Br(4)	212(1)	223(1)	160(1)	-10(1)	-49(1)	-91(1)
Br(5)	169(1)	202(1)	168(1)	8(1)	4(1)	-76(1)
Br(6)	212(1)	155(1)	208(1)	17(1)	-16(1)	-93(1)
Br(7)	262(1)	216(1)	281(1)	-65(1)	-47(1)	-90(1)
Br(8)	251(1)	195(1)	177(1)	-45(1)	-37(1)	-44(1)
F(1)	154(5)	187(6)	337(6)	-80(5)	-16(5)	28(5)
F(2)	429(8)	105(5)	273(6)	-58(5)	-81(6)	-2(5)
F(3)	395(8)	264(7)	271(6)	11(5)	-39(6)	-252(6)
F(4)	129(6)	456(8)	385(7)	-15(6)	-2(5)	-118(5)
F(5)	172(6)	221(6)	404(7)	-18(5)	-27(5)	56(5)
F(6)	182(6)	279(7)	237(6)	-47(5)	-62(5)	89(5)
F(7)	168(6)	330(7)	220(5)	-92(5)	-4(4)	55(5)
F(8)	256(6)	321(7)	135(5)	4(5)	-19(4)	4(5)
F(9)	243(7)	295(7)	223(6)	25(5)	-87(5)	94(5)
F(10)	166(6)	262(6)	209(5)	-33(5)	-12(4)	89(5)
F(11)	170(6)	234(6)	355(7)	-48(5)	-83(5)	51(5)
F(12)	122(5)	433(8)	319(6)	-62(6)	-43(5)	-108(5)
F(13)	299(7)	249(7)	388(7)	-80(6)	-8(6)	-194(5)
F(14)	266(7)	110(6)	821(11)	-118(7)	11(7)	-20(5)
F(15)	110(5)	137(6)	690(10)	-72(6)	-6(6)	-14(4)
N(1)	141(7)	89(6)	150(6)	-11(5)	-20(5)	-20(5)
N(2)	142(7)	113(7)	143(6)	-15(5)	3(5)	-28(5)
N(3)	107(6)	88(6)	165(6)	-14(5)	0(5)	-23(5)
N(4)	145(7)	103(7)	171(6)	-10(5)	-22(5)	-31(5)
N(5)	138(7)	118(7)	189(7)	-11(5)	-2(5)	7(5)
N(6)	146(7)	112(7)	227(7)	-11(6)	-28(6)	-3(6)
C(1)	150(8)	109(8)	147(7)	-21(6)	-4(6)	-4(6)
C(2)	172(8)	124(8)	148(7)	-19(6)	1(6)	-14(6)
C(3)	140(8)	107(8)	180(8)	-8(6)	6(6)	-16(6)
C(4)	116(7)	86(7)	185(7)	-16(6)	-9(6)	-3(6)
C(5)	110(7)	88(7)	179(7)	-15(6)	12(6)	-6(6)
C(6)	126(7)	107(8)	165(7)	-15(6)	-9(6)	-23(6)
C(7)	142(8)	130(8)	189(8)	-18(6)	-23(6)	-54(6)
C(8)	150(8)	160(9)	154(7)	-27(6)	-29(6)	-25(6)
C(9)	113(7)	112(8)	175(7)	-9(6)	-14(6)	-23(6)
C(10)	114(7)	118(8)	159(7)	-5(6)	-16(6)	-3(6)
C(11)	98(7)	99(8)	174(7)	0(6)	-7(6)	3(6)
C(12)	119(7)	125(8)	159(7)	20(6)	-10(6)	-2(6)
C(13)	105(7)	90(7)	208(8)	6(6)	-21(6)	-9(6)
C(14)	104(7)	85(7)	193(8)	-8(6)	-14(6)	5(6)
C(15)	111(7)	106(8)	203(8)	-22(6)	-5(6)	1(6)
C(16)	123(7)	106(8)	197(8)	-20(6)	-15(6)	-11(6)
C(17)	156(8)	123(8)	213(8)	-56(7)	-20(6)	-26(6)

C(18)	178(8)	122(8)	190(8)	-48(6)	-25(6)	-5(7)
C(19)	155(8)	123(8)	170(7)	-25(6)	-18(6)	-8(6)
C(20)	124(7)	109(8)	151(7)	-18(6)	-2(6)	-37(6)
C(21)	138(8)	140(8)	169(7)	-41(6)	-7(6)	-22(6)
C(22)	274(10)	108(8)	162(8)	-23(6)	-51(7)	-39(7)
C(23)	272(10)	208(10)	144(8)	0(7)	-35(7)	-148(8)
C(24)	148(8)	296(11)	196(8)	-14(8)	-4(7)	-78(7)
C(25)	143(8)	160(9)	193(8)	-9(7)	-12(6)	-13(7)
C(26)	124(7)	124(8)	163(7)	-3(6)	-8(6)	-27(6)
C(27)	123(7)	146(8)	191(8)	-7(6)	-29(6)	2(6)
C(28)	126(8)	191(9)	186(8)	-37(7)	9(6)	-7(7)
C(29)	168(8)	205(9)	147(7)	4(6)	-11(6)	-30(7)
C(30)	146(8)	164(9)	196(8)	25(7)	-40(6)	11(7)
C(31)	127(8)	136(8)	178(7)	5(6)	-2(6)	6(6)
C(32)	120(7)	131(8)	193(8)	-36(6)	-2(6)	-36(6)
C(33)	124(8)	187(9)	189(8)	-39(7)	-7(6)	-14(7)
C(34)	121(8)	295(11)	185(8)	-54(7)	-10(6)	-69(7)
C(35)	216(9)	200(10)	248(9)	-66(8)	8(7)	-125(7)
C(36)	174(9)	132(9)	390(11)	-78(8)	25(8)	-39(7)
C(37)	101(7)	137(9)	342(10)	-61(7)	11(7)	-30(6)
C(38)	255(12)	306(13)	744(19)	-363(13)	-220(13)	142(10)
C(39)	141(9)	173(10)	442(12)	-25(9)	-81(8)	-12(7)
C(40)	314(14)	501(16)	371(13)	260(11)	116(10)	263(13)
C(41)	484(16)	506(16)	183(9)	-82(10)	-62(10)	369(14)
C(42)	148(11)	165(12)	1890(40)	-226(19)	-310(19)	36(9)
C(43)	299(12)	375(13)	201(9)	89(9)	3(8)	183(10)
C(51)	294(13)	344(13)	441(14)	38(11)	-160(11)	-26(10)
Cl(1)	263(3)	385(3)	561(4)	75(3)	-177(3)	-59(2)
Cl(2)	357(4)	439(4)	939(6)	212(4)	-351(4)	-142(3)
C(52)	730(20)	363(16)	400(15)	20(12)	28(14)	-6(15)
Cl(3)	510(4)	625(5)	329(3)	-29(3)	-55(3)	-61(4)
Cl(4)	449(4)	512(5)	669(5)	-5(4)	97(4)	-144(3)
C(53)	680(20)	389(17)	439(16)	-86(13)	63(15)	61(15)
Cl(5A)	495(9)	594(10)	576(9)	-269(7)	-42(7)	-169(7)
Cl(6A)	584(10)	370(8)	548(8)	-87(6)	185(7)	-46(6)
Cl(5B)	313(7)	568(9)	330(6)	-77(6)	-73(5)	78(6)
Cl(6B)	396(8)	577(10)	763(11)	-344(9)	-234(8)	84(7)

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CALIFORNIA INSTITUTE OF TECHNOLOGY  
BECKMAN INSTITUTE  
X-RAY CRYSTALLOGRAPHY LABORATORY

Date 24 January 2008

**Crystal Structure Analysis of:**

**1-Ir(py)<sub>2</sub>**

(shown below)

**For** Investigator: Joshua Palmer ext. 6332

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Contents

Table 1. Crystal data

Figures Minimum overlap

Table 2. Atomic Coordinates

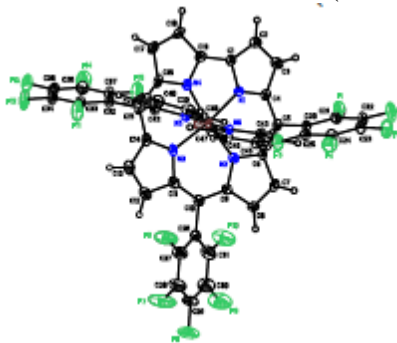
Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters

Table 6. Hydrogen bond distances and angles

Table 7. Observed and calculated structure factors (available upon request)



**1-Ir(py)<sub>2</sub>**

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and have been placed on hold pending further instructions from me. The deposition number is 657603. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 657603".

Table C-3-1. Crystal data and structure refinement for **1-Ir(py)<sub>2</sub>** (CCDC 657603)

Empirical formula	C <sub>47</sub> H <sub>18</sub> N <sub>6</sub> F <sub>15</sub> Ir, 3(CH <sub>4</sub> O)	
Formula weight	1240.00	
Crystallization Solvent	Methanol	
Crystal Habit	Column	
Crystal size	0.31 x 0.12 x 0.07 mm <sup>3</sup>	
Crystal color	Dark red	
<b>Data Collection</b>		
Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoKα	
Data Collection Temperature	100(2) K	
θ range for 9608 reflections used in lattice determination	2.39 to 32.74°	
Unit cell dimensions	a = 48.016(2) Å b = 8.7249(4) Å c = 27.2703(13) Å	β = 124.247(2)°
Volume	9443.7(8) Å <sup>3</sup>	
Z	8	
Crystal system	Monoclinic	
Space group	C2/c	
Density (calculated)	1.744 Mg/m <sup>3</sup>	
F(000)	4864	
θ range for data collection	1.81 to 32.95°	
Completeness to θ = 32.95°	98.5 %	
Index ranges	-73 ≤ h ≤ 72, -11 ≤ k ≤ 13, -40 ≤ l ≤ 41	
Data collection scan type	ω scans; 21 settings	
Reflections collected	112588	
Independent reflections	17484 [R <sub>int</sub> = 0.0721]	
Absorption coefficient	2.938 mm <sup>-1</sup>	
Absorption correction	None	
Max. and min. transmission (calc)	0.8208 and 0.4628	

Table C-3-1 cont.

### Structure Solution and Refinement

Structure solution program	SHELXS-97 (Sheldrick, 1990)
Primary solution method	Patterson method
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 1997)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	17484 / 54 / 695
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	1.520
Final R indices [ $I > 2\sigma(I)$ , 12341 reflections]	$R1 = 0.0441$ , $wR2 = 0.0563$
R indices (all data)	$R1 = 0.0761$ , $wR2 = 0.0580$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.007
Average shift/error	0.000
Largest diff. peak and hole	4.237 and -2.827 e.Å <sup>-3</sup>

### Special Refinement Details

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

The crystals contain methanol as a solvent of crystallization—three in the asymmetric unit. One of those three is disordered between two positions. The disorder was modeled with restraint on the C-O bond distances and on the anisotropic displacement parameters (ADP). In the solvent only C-O distances were restrained to have similar distances, the ADPs were restrained to mimic isotropic behavior, and for the disordered site the population was restrained so the occupancies summed to one.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for

negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt), etc., and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles, and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

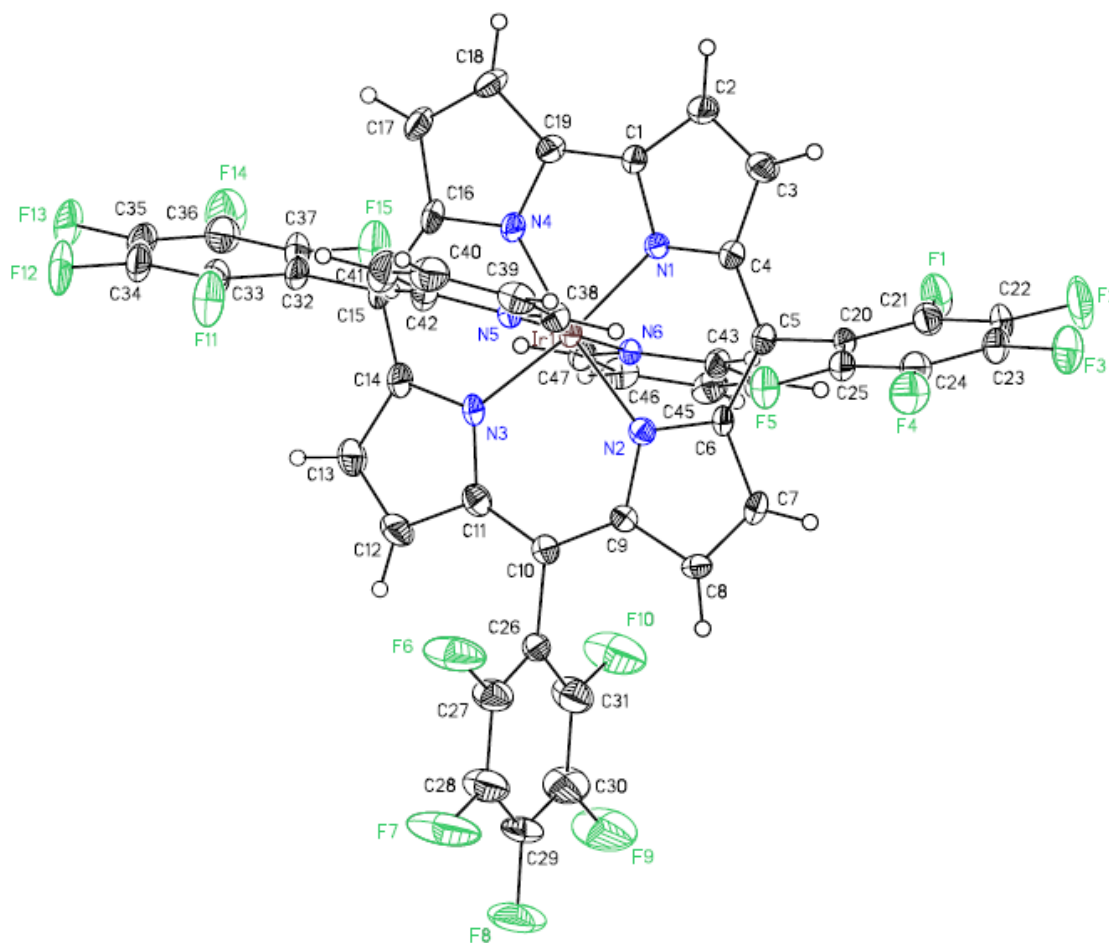


Table C-3-2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1-Ir(py)<sub>2</sub>** (CCDC 657603).  $U_{eq}$  is defined as the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z		$U_{eq}$
Ir(1)	5966(1)	2657(1)	37(1)	17(1)	1
F(1)	5516(1)	3864(2)	1516(1)	36(1)	1
F(2)	5321(1)	2868(2)	2209(1)	50(1)	1
F(3)	5304(1)	-171(2)	2394(1)	45(1)	1
F(4)	5495(1)	-2227(2)	1892(1)	36(1)	1
F(5)	5695(1)	-1256(2)	1201(1)	29(1)	1
F(6)	7064(1)	-1186(2)	1111(1)	60(1)	1
F(7)	7708(1)	-1870(2)	1935(1)	79(1)	1
F(8)	8128(1)	299(2)	2706(1)	67(1)	1
F(9)	7893(1)	3140(2)	2662(1)	79(1)	1
F(10)	7246(1)	3836(2)	1831(1)	66(1)	1
F(11)	6131(1)	1475(2)	-1786(1)	44(1)	1
F(12)	6229(1)	2374(2)	-2618(1)	48(1)	1
F(13)	6261(1)	5398(2)	-2813(1)	44(1)	1
F(14)	6212(1)	7524(2)	-2139(1)	50(1)	1
F(15)	6118(1)	6640(2)	-1299(1)	44(1)	1
N(1)	5548(1)	2875(2)	-7(1)	18(1)	1
N(2)	6212(1)	1834(2)	853(1)	17(1)	1
N(3)	6356(1)	2535(3)	-27(1)	21(1)	1
N(4)	5672(1)	3499(2)	-758(1)	19(1)	1
N(5)	5804(1)	504(2)	-333(1)	19(1)	1
N(6)	6125(1)	4806(2)	395(1)	17(1)	1
C(1)	5268(1)	3338(3)	-542(1)	20(1)	1
C(2)	4995(1)	3231(3)	-476(2)	26(1)	1
C(3)	5117(1)	2679(3)	84(1)	27(1)	1
C(4)	5474(1)	2447(3)	381(1)	19(1)	1
C(5)	5725(1)	1839(3)	948(1)	19(1)	1
C(6)	6070(1)	1600(3)	1170(1)	16(1)	1
C(7)	6342(1)	1145(3)	1757(1)	19(1)	1
C(8)	6631(1)	1139(3)	1778(1)	19(1)	1
C(9)	6552(1)	1570(3)	1205(1)	16(1)	1
C(10)	6765(1)	1702(3)	1000(1)	21(1)	1
C(11)	6678(1)	2154(3)	440(2)	25(1)	1
C(12)	6893(1)	2367(4)	223(2)	38(1)	1
C(13)	6694(1)	2873(3)	-344(2)	35(1)	1
C(14)	6351(1)	2996(3)	-511(2)	26(1)	1
C(15)	6066(1)	3559(3)	-1046(1)	24(1)	1
C(16)	5735(1)	3776(3)	-1179(1)	21(1)	1
C(17)	5416(1)	4175(3)	-1707(1)	25(1)	1
C(18)	5178(1)	4109(3)	-1579(1)	24(1)	1
C(19)	5340(1)	3684(3)	-977(1)	19(1)	1
C(20)	5612(1)	1341(3)	1330(1)	19(1)	1
C(21)	5509(1)	2351(3)	1595(1)	24(1)	1
C(22)	5406(1)	1847(3)	1947(2)	30(1)	1
C(23)	5399(1)	318(4)	2045(2)	30(1)	1
C(24)	5497(1)	-717(3)	1791(2)	25(1)	1
C(25)	5602(1)	-200(3)	1442(1)	23(1)	1

C(26)	7129(1)	1337(3)	1446(1)	21(1)	1
C(27)	7262(1)	-81(3)	1490(2)	33(1)	1
C(28)	7595(1)	-444(4)	1909(2)	42(1)	1
C(29)	7804(1)	637(4)	2292(2)	40(1)	1
C(30)	7686(1)	2067(4)	2269(2)	43(1)	1
C(31)	7352(1)	2408(4)	1847(2)	37(1)	1
C(32)	6117(1)	4024(3)	-1516(2)	24(1)	1
C(33)	6151(1)	2980(3)	-1861(2)	30(1)	1
C(34)	6201(1)	3435(4)	-2291(2)	32(1)	1
C(35)	6219(1)	4956(4)	-2390(2)	32(1)	1
C(36)	6187(1)	6030(4)	-2054(2)	32(1)	1
C(37)	6140(1)	5553(3)	-1626(2)	30(1)	1
C(38)	5692(1)	-525(3)	-113(2)	24(1)	1
C(39)	5561(1)	-1926(3)	-378(2)	29(1)	1
C(40)	5536(1)	-2285(3)	-893(2)	33(1)	1
C(41)	5648(1)	-1245(3)	-1120(2)	35(1)	1
C(42)	5780(1)	122(3)	-832(2)	29(1)	1
C(43)	6110(1)	5276(3)	851(1)	20(1)	1
C(44)	6235(1)	6665(3)	1127(1)	26(1)	1
C(45)	6378(1)	7628(3)	929(1)	27(1)	1
C(46)	6387(1)	7189(3)	451(2)	26(1)	1
C(47)	6258(1)	5777(3)	192(1)	22(1)	1
<hr/>					
C(51)	7183(5)	6190(20)	349(12)	179(11)	0.314(3)
O(51)	7428(5)	5226(18)	363(9)	202(8)	0.314(3)
<hr/>					
C(52)	7911(1)	9259(6)	1017(3)	122(2)	1
O(52)	7660(1)	10415(4)	759(2)	123(2)	1
<hr/>					
C(53)	6840(2)	8859(7)	9592(4)	84(2)	0.686(3)
O(53)	7116(1)	9625(4)	9631(2)	83(2)	0.686(3)
<hr/>					
C(54)	6988(2)	865(6)	8397(3)	131(2)	1
O(54)	7050(1)	1778(4)	8876(2)	133(2)	1

Table C-3-3. Selected bond lengths [Å] and angles [°] for **1-Ir(py)<sub>2</sub>** (CCDC 657603)

Ir(1)-N(4)	1.947(2)	N(4)-Ir(1)-N(1)	79.55(10)
Ir(1)-N(1)	1.953(2)	N(4)-Ir(1)-N(2)	172.23(10)
Ir(1)-N(2)	1.976(2)	N(1)-Ir(1)-N(2)	92.69(10)
Ir(1)-N(3)	1.979(2)	N(4)-Ir(1)-N(3)	92.92(10)
Ir(1)-N(6)	2.052(2)	N(1)-Ir(1)-N(3)	172.47(10)
Ir(1)-N(5)	2.066(2)	N(2)-Ir(1)-N(3)	94.84(10)
		N(4)-Ir(1)-N(6)	91.55(9)
		N(1)-Ir(1)-N(6)	91.87(9)
		N(2)-Ir(1)-N(6)	88.27(9)
		N(3)-Ir(1)-N(6)	88.60(9)
		N(4)-Ir(1)-N(5)	87.80(9)
		N(1)-Ir(1)-N(5)	88.04(9)
		N(2)-Ir(1)-N(5)	92.38(9)
		N(3)-Ir(1)-N(5)	91.41(9)
		N(6)-Ir(1)-N(5)	179.34(12)



Table C-3-4. Bond lengths [Å] and angles [°] for **1-Ir(tma)<sub>2</sub>** (CCDC 671270)

Ir(1)-N(4)	1.947(2)	C(15)-C(32)	1.494(4)
Ir(1)-N(1)	1.953(2)	C(16)-C(17)	1.434(4)
Ir(1)-N(2)	1.976(2)	C(17)-C(18)	1.370(4)
Ir(1)-N(3)	1.979(2)	C(18)-C(19)	1.416(4)
Ir(1)-N(6)	2.052(2)	C(20)-C(25)	1.386(4)
Ir(1)-N(5)	2.066(2)	C(20)-C(21)	1.396(4)
F(1)-C(21)	1.340(3)	C(21)-C(22)	1.376(4)
F(2)-C(22)	1.342(3)	C(22)-C(23)	1.366(4)
F(3)-C(23)	1.335(3)	C(23)-C(24)	1.372(4)
F(4)-C(24)	1.346(3)	C(24)-C(25)	1.382(4)
F(5)-C(25)	1.344(3)	C(26)-C(27)	1.365(4)
F(6)-C(27)	1.340(3)	C(26)-C(31)	1.377(4)
F(7)-C(28)	1.344(3)	C(27)-C(28)	1.381(5)
F(8)-C(29)	1.343(4)	C(28)-C(29)	1.345(5)
F(9)-C(30)	1.348(3)	C(29)-C(30)	1.358(4)
F(10)-C(31)	1.338(3)	C(30)-C(31)	1.381(5)
F(11)-C(33)	1.341(3)	C(32)-C(33)	1.383(4)
F(12)-C(34)	1.344(3)	C(32)-C(37)	1.384(4)
F(13)-C(35)	1.337(4)	C(33)-C(34)	1.383(4)
F(14)-C(36)	1.342(3)	C(34)-C(35)	1.365(4)
F(15)-C(37)	1.347(3)	C(35)-C(36)	1.378(4)
N(1)-C(4)	1.348(3)	C(36)-C(37)	1.369(4)
N(1)-C(1)	1.375(4)	C(38)-C(39)	1.378(4)
N(2)-C(9)	1.368(3)	C(39)-C(40)	1.374(5)
N(2)-C(6)	1.387(4)	C(40)-C(41)	1.368(4)
N(3)-C(14)	1.367(4)	C(41)-C(42)	1.371(4)
N(3)-C(11)	1.378(4)	C(43)-C(44)	1.373(4)
N(4)-C(19)	1.361(3)	C(44)-C(45)	1.373(4)
N(4)-C(16)	1.362(4)	C(45)-C(46)	1.383(4)
N(5)-C(42)	1.343(4)	C(46)-C(47)	1.381(4)
N(5)-C(38)	1.350(4)	C(51)-O(51)	1.425(5)
N(6)-C(43)	1.347(4)	C(52)-O(52)	1.418(4)
N(6)-C(47)	1.353(3)	C(53)-O(53)	1.432(4)
C(1)-C(2)	1.426(4)	C(54)-O(54)	1.410(4)
C(1)-C(19)	1.442(4)		
C(2)-C(3)	1.378(4)	N(4)-Ir(1)-N(1)	79.55(10)
C(3)-C(4)	1.437(4)	N(4)-Ir(1)-N(2)	172.23(10)
C(4)-C(5)	1.424(4)	N(1)-Ir(1)-N(2)	92.69(10)
C(5)-C(6)	1.422(4)	N(4)-Ir(1)-N(3)	92.92(10)
C(5)-C(20)	1.482(4)	N(1)-Ir(1)-N(3)	172.47(10)
C(6)-C(7)	1.440(4)	N(2)-Ir(1)-N(3)	94.84(10)
C(7)-C(8)	1.359(4)	N(4)-Ir(1)-N(6)	91.55(9)
C(8)-C(9)	1.433(4)	N(1)-Ir(1)-N(6)	91.87(9)
C(9)-C(10)	1.421(4)	N(2)-Ir(1)-N(6)	88.27(9)
C(10)-C(11)	1.395(4)	N(3)-Ir(1)-N(6)	88.60(9)
C(10)-C(26)	1.497(4)	N(4)-Ir(1)-N(5)	87.80(9)
C(11)-C(12)	1.462(4)	N(1)-Ir(1)-N(5)	88.04(9)
C(12)-C(13)	1.354(5)	N(2)-Ir(1)-N(5)	92.38(9)
C(13)-C(14)	1.445(4)	N(3)-Ir(1)-N(5)	91.41(9)
C(14)-C(15)	1.413(4)	N(6)-Ir(1)-N(5)	179.34(12)
C(15)-C(16)	1.427(4)	C(4)-N(1)-C(1)	112.4(2)

C(4)-N(1)-Ir(1)	130.07(19)	C(17)-C(18)-C(19)	108.2(3)
C(1)-N(1)-Ir(1)	116.8(2)	N(4)-C(19)-C(18)	106.4(3)
C(9)-N(2)-C(6)	110.5(2)	N(4)-C(19)-C(1)	112.3(3)
C(9)-N(2)-Ir(1)	124.9(2)	C(18)-C(19)-C(1)	141.2(3)
C(6)-N(2)-Ir(1)	124.27(19)	C(25)-C(20)-C(21)	115.5(3)
C(14)-N(3)-C(11)	111.4(2)	C(25)-C(20)-C(5)	120.8(3)
C(14)-N(3)-Ir(1)	124.5(2)	C(21)-C(20)-C(5)	123.7(2)
C(11)-N(3)-Ir(1)	123.62(19)	F(1)-C(21)-C(22)	118.5(3)
C(19)-N(4)-C(16)	111.8(3)	F(1)-C(21)-C(20)	119.4(3)
C(19)-N(4)-Ir(1)	118.0(2)	C(22)-C(21)-C(20)	122.1(3)
C(16)-N(4)-Ir(1)	129.7(2)	F(2)-C(22)-C(23)	119.6(3)
C(42)-N(5)-C(38)	117.3(3)	F(2)-C(22)-C(21)	119.8(3)
C(42)-N(5)-Ir(1)	120.9(2)	C(23)-C(22)-C(21)	120.6(3)
C(38)-N(5)-Ir(1)	121.6(2)	F(3)-C(23)-C(22)	120.6(3)
C(43)-N(6)-C(47)	117.9(2)	F(3)-C(23)-C(24)	120.1(3)
C(43)-N(6)-Ir(1)	121.29(19)	C(22)-C(23)-C(24)	119.3(3)
C(47)-N(6)-Ir(1)	120.8(2)	F(4)-C(24)-C(23)	119.9(3)
N(1)-C(1)-C(2)	105.5(3)	F(4)-C(24)-C(25)	120.4(3)
N(1)-C(1)-C(19)	113.0(3)	C(23)-C(24)-C(25)	119.7(3)
C(2)-C(1)-C(19)	141.4(3)	F(5)-C(25)-C(24)	117.6(2)
C(3)-C(2)-C(1)	108.2(3)	F(5)-C(25)-C(20)	119.7(3)
C(2)-C(3)-C(4)	107.8(3)	C(24)-C(25)-C(20)	122.8(3)
N(1)-C(4)-C(5)	121.2(3)	C(27)-C(26)-C(31)	115.5(3)
N(1)-C(4)-C(3)	106.0(3)	C(27)-C(26)-C(10)	122.6(3)
C(5)-C(4)-C(3)	132.7(3)	C(31)-C(26)-C(10)	122.0(3)
C(6)-C(5)-C(4)	126.1(3)	F(6)-C(27)-C(26)	119.3(3)
C(6)-C(5)-C(20)	117.1(3)	F(6)-C(27)-C(28)	117.6(3)
C(4)-C(5)-C(20)	116.7(3)	C(26)-C(27)-C(28)	123.1(3)
N(2)-C(6)-C(5)	125.4(3)	F(7)-C(28)-C(29)	120.4(3)
N(2)-C(6)-C(7)	106.0(2)	F(7)-C(28)-C(27)	120.1(3)
C(5)-C(6)-C(7)	128.6(3)	C(29)-C(28)-C(27)	119.5(3)
C(8)-C(7)-C(6)	108.4(3)	F(8)-C(29)-C(28)	120.3(3)
C(7)-C(8)-C(9)	108.3(3)	F(8)-C(29)-C(30)	119.8(3)
N(2)-C(9)-C(10)	123.3(3)	C(28)-C(29)-C(30)	119.8(3)
N(2)-C(9)-C(8)	106.8(2)	F(9)-C(30)-C(29)	119.9(3)
C(10)-C(9)-C(8)	129.9(3)	F(9)-C(30)-C(31)	120.3(3)
C(11)-C(10)-C(9)	127.9(3)	C(29)-C(30)-C(31)	119.8(3)
C(11)-C(10)-C(26)	116.4(3)	F(10)-C(31)-C(26)	119.7(3)
C(9)-C(10)-C(26)	115.7(3)	F(10)-C(31)-C(30)	118.0(3)
N(3)-C(11)-C(10)	124.8(3)	C(26)-C(31)-C(30)	122.3(3)
N(3)-C(11)-C(12)	105.9(3)	C(33)-C(32)-C(37)	115.7(3)
C(10)-C(11)-C(12)	129.3(3)	C(33)-C(32)-C(15)	123.0(3)
C(13)-C(12)-C(11)	107.5(3)	C(37)-C(32)-C(15)	121.3(3)
C(12)-C(13)-C(14)	109.1(3)	F(11)-C(33)-C(34)	118.2(3)
N(3)-C(14)-C(15)	125.3(3)	F(11)-C(33)-C(32)	119.7(3)
N(3)-C(14)-C(13)	106.0(3)	C(34)-C(33)-C(32)	122.1(3)
C(15)-C(14)-C(13)	128.6(3)	F(12)-C(34)-C(35)	120.0(3)
C(14)-C(15)-C(16)	127.1(3)	F(12)-C(34)-C(33)	119.7(3)
C(14)-C(15)-C(32)	116.5(3)	C(35)-C(34)-C(33)	120.3(3)
C(16)-C(15)-C(32)	116.4(3)	F(13)-C(35)-C(34)	120.3(3)
N(4)-C(16)-C(15)	120.3(3)	F(13)-C(35)-C(36)	120.4(3)
N(4)-C(16)-C(17)	105.7(3)	C(34)-C(35)-C(36)	119.3(3)
C(15)-C(16)-C(17)	133.9(3)	F(14)-C(36)-C(37)	121.2(3)
C(18)-C(17)-C(16)	107.9(3)	F(14)-C(36)-C(35)	119.4(3)



C(37)-C(36)-C(35)	119.4(3)	C(40)-C(41)-C(42)	119.4(3)
F(15)-C(37)-C(36)	117.5(3)	N(5)-C(42)-C(41)	122.9(3)
F(15)-C(37)-C(32)	119.3(3)	N(6)-C(43)-C(44)	122.8(3)
C(36)-C(37)-C(32)	123.3(3)	C(43)-C(44)-C(45)	119.0(3)
N(5)-C(38)-C(39)	122.2(3)	C(44)-C(45)-C(46)	119.2(3)
C(40)-C(39)-C(38)	119.4(3)	C(47)-C(46)-C(45)	119.2(3)
C(41)-C(40)-C(39)	118.7(3)	N(6)-C(47)-C(46)	121.9(3)

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Table C-3-5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for **1-Ir(py)<sub>2</sub>** (CCDC 657603). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$ .

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
Ir(1)	170(1)	230(1)	136(1)	13(1)	103(1)	8(1)
F(1)	488(14)	323(10)	423(14)	21(8)	345(12)	83(8)
F(2)	695(16)	565(12)	531(15)	27(10)	528(14)	154(10)
F(3)	524(15)	679(13)	383(14)	115(10)	396(14)	18(10)
F(4)	411(12)	362(10)	380(12)	95(9)	266(11)	-37(9)
F(5)	405(12)	266(9)	325(12)	-23(8)	277(11)	-17(8)
F(6)	274(13)	413(12)	699(19)	-202(11)	25(13)	32(9)
F(7)	323(14)	533(13)	950(20)	-92(13)	21(15)	173(10)
F(8)	167(13)	908(16)	600(20)	-107(13)	19(13)	68(11)
F(9)	318(14)	775(15)	830(20)	-448(14)	53(15)	-113(11)
F(10)	357(14)	452(12)	840(20)	-272(12)	138(15)	1(10)
F(11)	701(16)	372(11)	456(15)	-71(9)	462(14)	-80(10)
F(12)	677(15)	607(12)	393(13)	-136(11)	444(13)	-73(11)
F(13)	524(15)	686(13)	313(13)	85(10)	359(13)	-10(10)
F(14)	695(15)	452(11)	523(15)	125(11)	451(14)	32(11)
F(15)	680(16)	386(10)	460(15)	-35(10)	447(14)	-17(10)
N(1)	145(13)	257(13)	150(13)	20(10)	94(12)	1(9)
N(2)	181(14)	153(11)	176(14)	-2(9)	101(12)	7(9)
N(3)	217(13)	287(13)	190(13)	-5(11)	158(12)	7(11)
N(4)	195(14)	243(12)	159(14)	11(10)	122(12)	-11(10)
N(5)	177(14)	258(13)	160(14)	16(10)	106(13)	44(10)
N(6)	155(14)	258(12)	133(13)	37(10)	99(12)	27(10)
C(1)	192(17)	252(15)	160(16)	-3(12)	109(15)	3(12)
C(2)	167(17)	393(17)	202(18)	-19(14)	89(16)	-22(13)
C(3)	190(16)	356(17)	261(17)	34(15)	134(15)	1(14)
C(4)	207(15)	223(14)	163(14)	12(13)	112(13)	12(13)
C(5)	230(17)	200(14)	167(16)	11(11)	136(15)	-12(11)
C(6)	220(17)	155(13)	165(16)	-3(11)	143(14)	1(11)
C(7)	283(19)	176(14)	150(16)	-16(11)	139(16)	-8(12)
C(8)	152(16)	200(14)	165(17)	11(11)	57(14)	17(11)
C(9)	160(16)	145(13)	185(16)	1(11)	101(14)	-12(11)
C(10)	197(17)	264(15)	202(18)	8(12)	125(15)	11(12)
C(11)	210(17)	314(16)	266(18)	-1(13)	166(16)	-16(13)
C(12)	240(18)	640(20)	330(20)	114(19)	209(17)	117(18)
C(13)	320(20)	550(20)	300(20)	86(16)	257(19)	50(16)
C(14)	269(19)	370(18)	225(18)	19(13)	194(17)	-3(13)
C(15)	300(20)	293(16)	179(17)	-13(13)	174(17)	-16(13)
C(16)	275(19)	255(15)	157(17)	-21(12)	157(16)	-46(12)
C(17)	280(20)	327(17)	139(17)	4(13)	115(16)	-29(13)
C(18)	208(18)	318(16)	132(17)	21(12)	54(15)	-35(12)
C(19)	186(17)	208(14)	168(17)	-1(11)	87(15)	-15(11)
C(20)	148(16)	288(16)	151(16)	11(12)	104(14)	11(12)
C(21)	260(17)	262(16)	246(16)	27(14)	164(15)	39(13)
C(22)	300(20)	444(19)	260(20)	-14(14)	216(18)	77(14)
C(23)	290(20)	500(20)	240(20)	76(15)	215(18)	24(15)
C(24)	235(19)	313(17)	250(20)	61(13)	167(17)	-8(13)
C(25)	204(18)	321(17)	191(18)	-29(13)	123(16)	-1(13)

C(26)	162(17)	307(16)	191(18)	12(13)	111(15)	-13(12)
C(27)	210(20)	345(18)	360(20)	-69(15)	106(19)	-32(14)
C(28)	270(20)	390(20)	460(30)	-37(17)	120(20)	73(16)
C(29)	124(19)	600(20)	340(20)	-35(18)	58(19)	34(16)
C(30)	219(19)	480(20)	440(30)	-191(17)	96(19)	-136(16)
C(31)	266(19)	377(19)	440(20)	-70(18)	173(18)	-1(16)
C(32)	236(19)	384(18)	169(18)	6(13)	148(16)	-6(13)
C(33)	330(20)	390(19)	253(19)	-25(14)	209(18)	-59(14)
C(34)	320(20)	490(20)	230(20)	-60(16)	205(18)	-34(15)
C(35)	270(20)	550(20)	200(20)	58(16)	166(18)	-15(16)
C(36)	320(20)	393(19)	300(20)	83(15)	208(19)	12(15)
C(37)	360(20)	373(18)	250(20)	-11(14)	225(19)	37(15)
C(38)	231(19)	258(16)	233(19)	5(13)	140(16)	22(12)
C(39)	227(18)	268(17)	340(20)	1(14)	138(18)	15(13)
C(40)	299(19)	279(17)	320(20)	-65(16)	115(17)	12(15)
C(41)	470(20)	318(18)	280(20)	-36(15)	220(20)	33(15)
C(42)	390(20)	294(17)	240(20)	6(14)	213(19)	51(14)
C(43)	243(18)	232(15)	162(17)	45(12)	129(16)	58(12)
C(44)	330(20)	254(15)	170(17)	-19(13)	131(16)	32(13)
C(45)	298(17)	195(15)	237(17)	-36(14)	103(15)	-8(14)
C(46)	238(17)	230(16)	275(18)	31(13)	121(16)	-21(13)
C(47)	231(18)	266(16)	182(17)	36(12)	129(16)	16(12)
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C(51)	1790(120)	1790(120)	1800(120)	-110(40)	1020(70)	30(40)
O(51)	2030(90)	1980(90)	2020(90)	0(40)	1130(60)	-20(40)
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C(52)	990(30)	1390(30)	1380(40)	-20(30)	730(30)	260(30)
O(52)	1170(30)	1270(20)	1180(30)	50(20)	630(20)	240(20)
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C(53)	990(40)	840(30)	820(40)	-120(30)	580(30)	-190(30)
O(53)	950(30)	880(30)	720(30)	-40(20)	500(30)	130(20)
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C(54)	1430(40)	1350(40)	1220(40)	-90(30)	800(30)	-10(30)
O(54)	1610(30)	1240(20)	1190(30)	-120(20)	810(30)	180(20)

Table C-3-6. Hydrogen bonds for **1-Ir(py)<sub>2</sub>** (CCDC 657603) [Å and °]

D-H...A	d(D-H)	d(H...A)	d(D...A)	∠(DHA)
O(51)-H(51)...O(54)#1	0.84	2.25	2.791(19)	122.4
O(52)-H(52)...O(54)#2	0.84	1.89	2.714(6)	168.1
O(53)-H(53)...O(52)#3	0.84	2.04	2.777(7)	146.7
O(54)-H(54)...O(51)#1	0.84	2.25	2.791(19)	122.7

Symmetry transformations used to generate equivalent atoms:

#1 -x+3/2, -y+1/2, -z+1

#2 -x+3/2, -y+3/2, -z+1

#3 x, y, z+1