Analyses of atmospheric CHF₂Cl, heavy ozone, HDO and CH₃D from ATMOS spectra

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The most important fact about Spaceship Earth: an instruction book didn't come with it. -Buckminster Fuller

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Abstract

Mixing ratio measurements of tropospheric CHF₂Cl and stratospheric heavy ozone (16O18O16O and 16O16O18O), HDO and CH₃D were derived from Atmospheric Trace Molecule Spectroscopy Experiment (ATMOS) spectra. The ATMOS instrument is a 0.01 cm⁻¹ resolution, solar-absorption Fourier-transform spectrometer with a frequency response of 600-4800 cm⁻¹ in the mid-infrared. It has recorded ground-based spectra over Table Mountain Facility (TMF), Wrightwood CA (34.4°N, 117.7°W, 2.2 km altitude) between 1985 to 1990, and space-based measurements during the four shuttle missions Spacelab 3 (April/May 1985), ATLAS-1 (March, 1992), ATLAS-2 (April, 1993) and ATLAS-3 (November, 1994).

The measured column burden of CHF₂Cl over TMF showed an exponential increase rate of $(6.7\pm0.5)\%$ yr⁻¹ from October, 1985 to July, 1990. The current uncertainty in historical CHF₂Cl emissions was found to be too large for CHF₂Cl measurements to be used to infer adequately either its lifetime or the OH field.

Enrichments of heavy ozone (with respect to $^{48}O_3$) were measured from both space and ground-based observations. Average enrichments between 26 to 2.6 mb inclusive (\approx 25 to 41 km) were (15 ± 6)% for $^{16}O^{18}O_3$, (10 ± 7)% for $^{16}O^{18}O_3$, and (13 ± 5)% for $^{50}O_3$ (1σ standard deviation). Enrichments increased slightly with altitude, but with no discernable latitudinal variability. From TMF, an average total column $^{16}O^{16}O^{18}O_3$ enrichment of (17 ± 4)% (1σ standard deviation) was determined, with no discernable seasonal variation. Possible biases in the spectral intensities that affect the determination of absolute enrichments are discussed, but any corrections to these intensities would not affect the observed lack of latitudinal and seasonal variability.

From space-based measurements, stratospheric mixing ratios of CH_3D from 100 mb to 17mb (\approx 15 to 28 km) and HDO from 100 mb to 10 mb (\approx 15 to 32 km) were inferred. The average lifetime of CH_3D was found to be (1.19 \pm 0.02) times that of CH_4 , with an average of (1.0 \pm 0.1) molecules of stratospheric HDO produced for each CH_3D destroyed (1 σ combined precision and systematic error), indicating that the rate of stratospheric HDO production is approximately equal to the rate of CH_3D destruction.

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Chapter I

Introduction

This work is concerned with three studies of atmospheric composition using the Atmospheric Trace Molecule Absorption Experiment (ATMOS) spectrometer, and the bulk of this thesis is three independent papers that have been either accepted or submitted to Geophysical Research Letters. The first paper, in chapter II, deals with measurement of the secular increase of CHF₂Cl (HCFC-22). Chapter III investigates the stratospheric abundance of the heavy ozone isotopomers ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O, and their anomalous enrichment (that is, their abundance above that expected statistically from standard isotopic ratios of ¹⁸O to ¹⁶O). Chapter IV is concerned with stratospheric measurements of the water isotopomer HDO and the methane isotopomer CH₃D.

All three papers have a bearing on the prediction of changes in the terrestrial atmosphere. CHF₂Cl is a so-called replacement gas for the chlorofluorocarbons CF₂Cl₂ (CFC-12) and CFCl₃ (CFC-11). The manufacture of CHF₂Cl is controlled by the Montreal Protocol and subsequent amendments, which restricts production of anthropogenic gases whose photolytic products destroy ozone (see, for example, Ko et al. [1994]). Heavy ozone has shown anomalously large enrichments in the stratosphere using several observation techniques. Enrichments have also been reproduced in the laboratory, but to date this has had no satisfactory scientific explanation. This problem has been somewhat more than an academic curiosity because its existence suggests regions of ozone chemistry that are not fully understood. Understanding heavy ozone enrichment may help in better understanding ozone chemistry in general. The third paper on HDO and CH₃D relates to chemistry, but as discussed by Moyer et al. [1995] and summarized later in this chapter, measurements of the stratospheric HDO/H₂O ratio may help elucidate mechanisms of tropospheric-stratospheric transport of water. As discussed in WMO [1995] and Wennberg et al. [1995], ozone destruction in the lower stratosphere is caused to a large extent by catalytic cycles involving HO_x (=OH +HO₂), which in turn has its origin in the reaction of $O(^{1}D)$ with $H_{2}O$ and CH_{4} . Changes in the amount of stratospheric water injected by, say, changes in tropospheric humidity due to global warming could affect the concentration of HO_x in the lower stratosphere. Better understanding of the mechanisms of cross-tropopause transport of water would help in model predictions of global change.

As submissions to Geophysical Research Letters are brief by necessity, this introductory chapter attempts to give more detail on the ATMOS spectrometer and data reduction methods used to analyze its output, and then to place the three aforementioned papers in a broader scientific context.

1.1 The ATMOS experiment

1.1.1 Rationale

Concern over possible changes to the earth's stratosphere in the late 1970's, particularly the destruction of ozone by chlorofluorocarbons and nitrogen oxides from emissions of proposed supersonic aircraft, prompted the development of research programs to better understand the roles of chemistry, radiation and dynamics in earth's stratosphere. Distributions of the so-called source, sink, reservoir and active species involved in ozone destruction were not well known at the time. Experimental validation of the stratospheric inventory of gases is required as both input to and tests for models, not only to calculate the current state of the atmosphere, but also to gain confidence in their predictive capabilities using plausible anthropogenic or natural forcings. As a contribution towards this end, the Atmospheric Trace Molecule Spectroscopy Experiment (ATMOS) program was developed, and as stated by *Farmer et al.* [1987], "the primary objective of the ATMOS experiment is to measure simultaneously as many of the minor and trace species as possible and to determine their spatial distribution."

1.1.2 Spectrometer design and operation

The ATMOS instrument (Figure 1.1.1) is a modified Fourier-transform infrared (FTIR) interferomer. It has a spectral response of 600 - 4800 cm⁻¹ (2.2 - 16μm), from which spectral intervals are selected using optical bandpass filters (Table 1.1). Most of the minor species in the stratosphere (e.g., those in the part-per-million range) and many of the trace species (e.g., those in the part-per-billion range and less) have observable rotational-vibrational spectral transitions in the mid-infrared. A broad band spectrometer such as the ATMOS instrument is particularly useful as the spectra it obtains provide measurements on many gases of interest. Designed for use in the payload bay of the Space Shuttle, the ATMOS instrument makes measurements in the solar occultation mode. Using the sun as a light source, absorption spectra through the Earth's limb at different tangent altitudes are

obtained during an orbital sunrise or sunset. It has a scan time of 2.2 seconds, which for a 4 minute occultation event, provides about 100 measurements ranging in altitude from over 200 km to cloud top levels. The vertical spacing of these observations correspond to 4 km at the tangent altitude, decreasing to 2 km near 20 km due to refraction effects. With a shuttle distance of about 2000 km to the tangent point, selectable instrument field-of-views, ranging from 1 to 2.8 mrad, limit the vertical resolution from about 2 to 5 km. A simplified illustration of ATMOS measurement geometry is given in Figure 1.1.2.

A full description of the ATMOS instrument is given by Farmer et al. [1987]. A block diagram of ATMOS instrument components is given in Figure 1.1.3 and a schematic illustration of the optical components is given in Figure 1.1.4. The suntracker uses visible wavelength photodiode sensors on its outside face such that the sensor's field-of-view is centered on the solar disk. Solar radiation enters the foreoptics to a mirrored field stop with selectable arpertures corresponding to instrument field-of-views of 1, 1.4, 2.0, and 2.8 mrad. Light reflected by the field stop is passed to an on-board camera which records the positioning of the instrument on the sun. Light passing through the field stop is transferred to the interferometer's KBr beamsplitter at an incidence angle of 45°. The transmitted and reflected light from the beamsplitter are passed to one of two moving cat'seye retroreflectors, each consisting of a paraboloid primary mirror and a slightly convex secondary mirror. In order to minimize the distance travelled by the mirrors, the beams are double-passed down each arm of the interferometer using a retroreflector. An optical path difference of ±50 cm for the two light beams is achieved by continuously moving the retroreflectors in opposite directions at 50 cm s⁻¹. The light rays returning from the retroreflectors are recombined in the beamsplitter and the resulting intensity-modulated radiation is transferred to the exit optics, where the light passes through a selectable spectral filter (see Table 1.1) and is focused on a HgCdTe detector cooled to 77K.

In addition to the solar radiation, a 0.633 µm beam from a HeNe laser is passed parallel to the solar radiation in the interferometer and is independently sampled. The laser is used for two reasons. First, sampling of the laser fringes allows precise velocity control for the retroreflectors using electronic feedback loops. Second, sampling of the analog output by the HgCdTe detector is selectably triggered every second or third laser fringe. The output from the detector is passed to a low gain channel to record the central fringe, and a high gain channel to record the interferogram. These signals are digitized, combined with instrument housekeeping data, and are stored on board the shuttle in a flight recorder or telemetried for recording on earth.

Processing of the interferograms is described by *Norton and Rinsland* [1991]. The interferograms are then corrected for phase and detector non-linearity errors, and fast Fourier transformed to power spectra. For space-based measurements, these power spectra were ratioed against exo-atmospheric solar spectra (obtained with each occultation) to remove solar spectral features and absorptions of residual H₂O and CO₂ within the instrument, producing pure atmospheric transmission spectra. Retrieved spectra from several tangent altitudes from bandpass filter 1 is illustrated in Figure 1.1.5.

1.1.3 ATMOS profile measurements from the space shuttle

The ATMOS instrument first flew aboard the space shuttle as part of the scientific payload of the Spacelab 3 mission of April 29 - May 7, 1985. Nineteen atmospheric observations were completed until operations were terminated on the third day of the mission due to a pressure leak in the reference HeNe laser housing. Despite the early cessation of observations, data were obtained over the ≈32°N and ≈44°S latitude bands. Figure 1.1.6 illustrates the latitude and longitude of the 30 km tangent point locations of the Spacelab 3 and subsequent flights. A full discussion of ATMOS on Spacelab 3 can be found in *Farmer et al.* [1987].

The next flights of the ATMOS instrument were on the Atmospheric Laboratory for Applications and Science (ATLAS) series beginning with ATLAS-1, March 24 - April 3, 1992. During the ATLAS-1 mission, approximately 41 sunset occultations were made between 55°S and 24°S, and 52 sunrise occultations between 31°S and 29°N. The timing of ATLAS-1 is important as it occurred about 9 months after the eruption of Mount Pinatubo, which injected large quantities of SO₂ into the lower stratosphere which subsequently formed a sulfuric acid aerosol layer (e.g., *Winkler and Osborn* [1992]). While the presence of the layer allowed analysis of the mid-infrared extinction by aerosol below about 30 km [*Rinsland et al.*, 1994], operationally it had the deleterious effect of causing the ATMOS suntracker, which uses visible wavelength photodiode sensors, to lose lock on the sun at tangent altitudes below about 25 km in the tropics; the aerosol was too opaque in the visible wavelengths for the limb viewing geometry. Consequently, many sunrise observations from the southern subtropics to northern subtropics at tangent altitudes below about 25 km could not be made, although southern mid-latitude measurements were largely unaffected because of a lower aerosol loading.

The third flight of ATMOS was on the ATLAS-2 mission, April 8 - 16, 1993. The mission timing and shuttle orbit were arranged to allow sunrise observations to be made in high northern latitudes inside and outside the Arctic polar vortex during the boreal spring. About 65 sunrise occultations were obtained around 65°N and about 39 sunset occultations were obtained between 50°S and 9°S. The vortex measurements proved exceptionally useful for measurement of CH₃D because, as discussed in Chapter IV, the descent (and compression) of high altitude air to lower altitudes within the vortex allowed measurement of air of low CH₃D mixing ratio (≤0.6 x 10-9) that could not be measured directly at midlatitudes. Much as had happened during ATLAS-1, many observations in the southern mid-latitudes below about 25 km were lost due to increased sulfuric acid aerosol opacity from the Mt. Pinatubo eruption.

The fourth shuttle flight of ATMOS was on the ATLAS-3 mission, November 3-14, 1994. Here, the timing and orbit of the mission was arranged for measurements inside and outside the Antarctic polar vortex in the late austral spring, and from the northern subtropics to northern mid-latitudes. Approximately 83 sunrise occultations were obtained at about 68°N and 96 sunset occultations were obtained from 5°N-48°N. As was the case with ATLAS-2, descent inside the vortex allowed measurement of air with low CH₃D mixing ratios. By time of ATLAS-3, the opacity of the Pinatubo aerosol had sufficiently decayed so as not to affect instrument suntracker performance, and sunset measurements in the northern sub-tropics and mid-latitudes were achieved at altitudes down to 10 km and below.

1.1.4 ATMOS observations at Table Mountain

After the Spacelab 3 mission of April/May 1985, the ATMOS instrument was briefly at Table Mountain Facility (TMF; 34.4°N, 117.7°W, 2.23 km altitude) in October, 1985 to test it for a planned re-flight. The instrument was returned to TMF in December, 1986 because of the interruption of shuttle flights after the Challenger disaster. Data were taken when possible until July, 1990 when the instrument was prepared for the ATLAS-1 flight. The site at TMF has a clear view of the eastern horizon, and its altitude of 2.23 km eliminates much of the interference from tropospheric H₂O, CH₄ and CO₂ infrared absorptions. Unfortunately, the site can be quite humid at times, and its proximity to Los Angeles makes it susceptible to smog spilling over the intervening San Gabriel mountains in the afternoon. Data were gathered at TMF on some 51 days between 1985 and 1990, and have been used in measurement of several stratospheric trace gases [Gunson and Irion, 1991], and CHF₂Cl and heavy ozone as described in this thesis.

1.1.5 ATMOS data analysis methods

The ATMOS project data analysis software has been described by *Norton and Rinsland* [1991]. Atmospheric temperature, pressure and molecular density profiles are

derived for the spectra within each occultation. The procedure for this is briefly outlined in Appendix A. These temperature-pressure profiles were mapped onto a 150 layer model atmosphere for each occultation, each layer 1 km thick and homogeneous, along with assumed mixing ratio profiles for some 50 gas species which contribute to infrared absorption in the stratosphere.

To determine the mixing ratio of a target gas at a particular tangent pressure (or altitude), synthetic spectra were calculated for the ray path from the sun to the instrument and fitted to the observed spectra over selected spectral intervals (e.g., those in the lower panels of Figure 1.1.5). The synthetic spectra were calculated line-by-line using the ATMOS spectral linelist for both target and background gases for each model layer, except that for certain gases (e.g., N₂O₅, CFCl₃, CF₂Cl₂, CHF₂Cl), laboratory measured cross-sections were used directly [*Brown et al.*, 1995]. (Note that at the time the studies described in Chapter II were made, empirical spectral line parameters were being used for CHF₂Cl, but these had been verified against laboratory cross-sections of *Varanasi* [1992] and *McDaniel et al.* [1991].) Following the description of *Abrams et al.* [1995], the spectrum within a particular layer can be modelled by:

$$T_s(v_i,z) = \exp[-\Sigma_j \sum_k \kappa_{ijk} n_k \vartheta_{jk} \chi g_k]$$
 (E1.1)

where: T_s = synthetic (i.e., modelled) transmission

 v_i = frequency at spectral grid point i (cm⁻¹)

z = altitude (km)

 κ_{iik} = absorption of the jth gas at the kth layer at frequency v_i

 n_k = total molecular density in the kth layer (molecules cm⁻³)

 ϑ_{ik} = assumed mixing ratio of the jth gas in the kth layer

 χ is a scale factor for ϑ_{jk} and equals unity for a non-target gas $g_k = \text{slant path through the layer (cm)}.$

Least-squares techniques were used to adjust the mixing ratio scale factor, χ , to achieve:

$$\Sigma_{i} \frac{\partial T_{s}(v_{i},z)}{\partial \chi} R_{i} = 0$$
 (E1.2)

where R_i is the residual, equal to the observed minus the synthetic transmission:

$$R_{i} = \left(T_{o}(v_{i},z) - T_{s}(v_{i},z)\right)$$
 (E1.3)

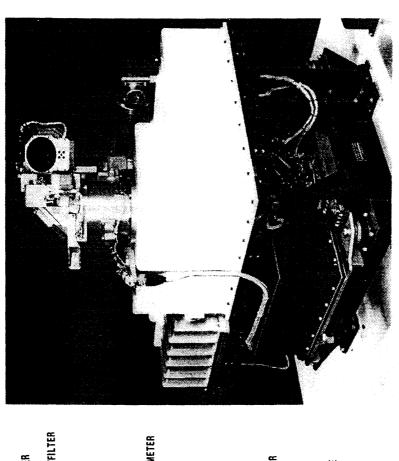
where T_0 is the observed transmission. Note that the partial derivative in (E1.2) acts as a weighting function. The retrieval of a vertical mixing ratio profile was begun by fitting a spectral microwindow at the highest altitude for which there is discernable absorption by the target gas. In an onion-peeling technique, successively lower spectra were fitted in a similar fashion (while keeping mixing ratios above at the just determined values). This was repeated until a pre-determined lower altitude cutoff is reached, usually determined on the basis of the line opacity. Where several microwindows were evaluated for a target gas, a weighted average was made of the resultant individual profiles:

$$\chi_{mn} = \frac{\sum_{n} \left(\chi_{n} / \varepsilon_{n}^{2} \right)}{\sum_{n} \left(1 / \varepsilon_{n}^{2} \right)}$$
 (E1.4)

where ε_n is the estimated precision error of the retrieved mixing ratio (discussed further in Appendix A). To avoid "staircases" from the atmospheric model's vertical resolution being smaller than the tangent height spacing of the measurements, resultant profiles were then smoothed by a five-point polynomial (corresponding to about 3 km in the vertical), while preserving the total column amount of the gas above the lowest retrieval height. This

smoothed profile was used as the assumed profile for a second retrieval. The entire process was repeated twice more, with the final profile produced on the third iteration interpolated on an atmospheric pressure grid of 12 levels per decade in mb.

For ground-based column spectra, the procedures were somewhat different. As the ATMOS instrument has a very fast scan time (2.2 s), several measurements were taken over a small range of solar zenith angles, and resulting spectra were averaged to increase signal-to-noise ratios. An appropriate atmospheric model was selected (e.g., the *U. S. Standard Atmosphere* [1976]), and the column absorption calculated using the 150 km layer model. Column densities were derived by scaling assumed trace gas mixing ratio profiles to best fit selected spectral microwindows. Error sources and treatment are discussed briefly in Appendix A.



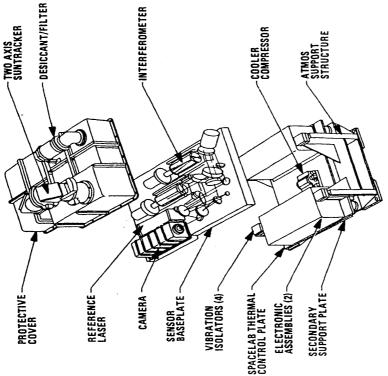


Figure 1.1.1: The ATMOS Fourier Transform Interferometer. The schematic on the left illustrates the optical and electronic subsystems, while a

photograph of the assembled instrument is on the right.

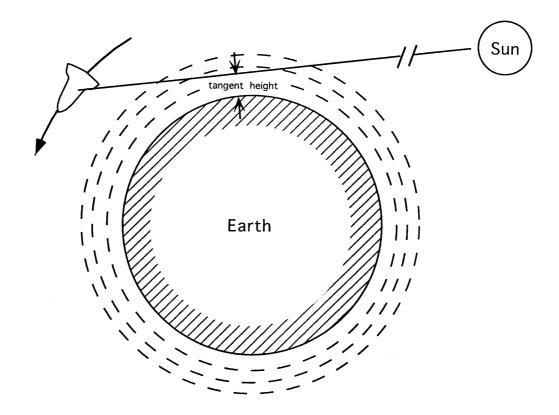


Figure 1.1.2: Simplified viewing geometry by ATMOS during a shuttle occultation.

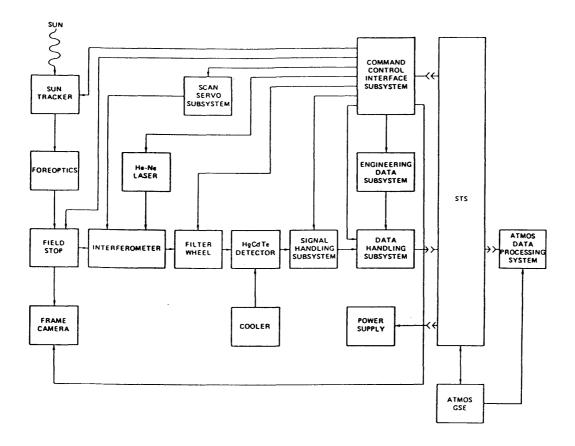


Figure 1.1.3: Block diagram of ATMOS instrument components (from Farmer et al. [1987])

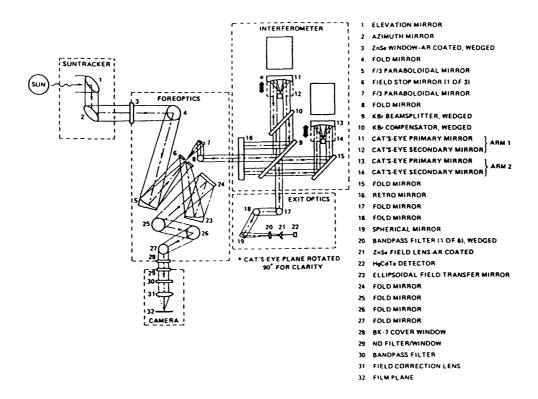
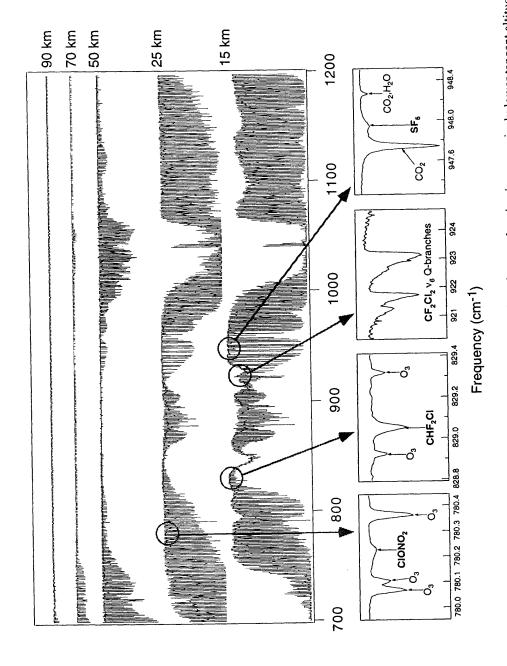


Figure 1.1.4: Schematic diagram of ATMOS optical components (from Farmer et al. [1987]).



observations while the lower panels show small spectral intervals ("microwindows") from which gas mixing ratios are determined. For clarity, vertical Figure 1.1.5: Typical ATMOS filter 1 spectra. The upper panel shows the same broad spectral region in successively lower tangent altitude

scales differ among displayed spectra.

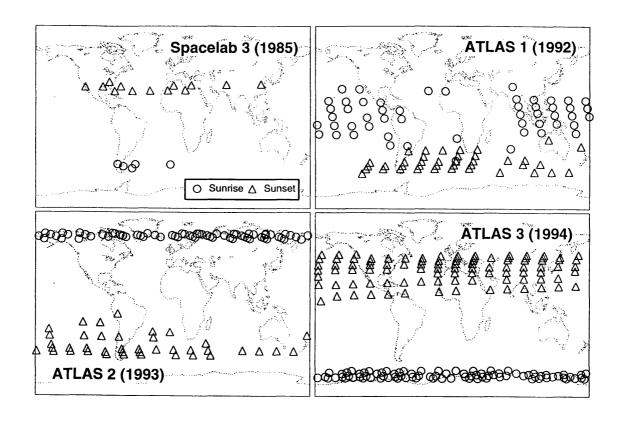


Figure 1.1.6: Distribution of ATMOS observations during each of four Space Shuttle missions. Spacelab 3: April 29-May 7, 1985; ATLAS-1: March 24-April 3, 1992; ATLAS-2: April 8-16, 1993; ATLAS-3: November 3-14, 1994.

1.2 CHF₂Cl and other replacement gases

As replacements for the fully halogenated chlorofluorocarbons (CFCs; e.g., CF₂Cl₂ and CCl₃F) and halons (e.g., CF₃Br and CF₂ClBr), hydrochlorofluorocarbons (HCFCs) such as CHF₂Cl (HCFC-22) and CH₃CClF₂ (HCFC-142b) have been increasingly emitted into the atmosphere. As the CFCs and halons are almost insoluble and chemically inert in the troposphere, their lifetimes are long compared to their tropospheric residence time. Their only effective sink is destruction by ultraviolet radiation in the stratosphere, where their chlorinated and brominated photolysis products can destroy ozone. On the other hand, HCFCs are hydrogenated, and can be destroyed by tropospheric OH. Their tropospheric lifetime is therefore much shorter, and their ability to reach the stratosphere is greatly reduced. (See, for example, *Kaye et al.* [1994].)

CHF₂Cl is the most widely used of the HCFCs finding its major application as a refrigerant [*Midgley and Fisher*, 1993]. It is expected to have only about 5% the ozone destruction capability of CFCl₃ when compared on the basis of unit mass emitted in the troposphere [*WMO*, 1995]. However, that HCFC emissions like those of CHF₂Cl are less harmful to stratospheric ozone does not make them benign, and the necessity to not just slow the growth of atmospheric chlorine loading but actually reduce it led to restrictions on the production of HCFCs, including that of CHF₂Cl, being added to the Montreal Protocol [*Rowlands*, 1993]. It therefore is important to monitor the rate of growth of CHF₂Cl to characterize and help predict its contribution to atmospheric chlorine loading.

A second reason to evaluate the trend in CHF₂Cl is that CFCs and HCFCs can have significant greenhouse absorptions. On a molecule-per-molecule basis, these gases can have a greater impact on atmospheric radiative forcing than those of CO_2 or CH_4 as many of their absorption bands (including some of CHF_2Cl) occur in the relatively transparent "atmospheric window" between 8 and 13 μ m. The long-term amount of forcing by a gas depends not just on its total spectral absorption in the infrared, but also its

lifetime in the troposphere. For example, in a simplified scenario assuming current CO₂ levels and ignoring feedbacks, the radiative forcing of 1kg of CHF₂Cl released into the atmosphere is about 4300 times that of 1kg of CO₂ integrated over 20 years [WMO, 1995]. While their tropospheric concentrations may be much less than CO₂, radiative forcing by cumulative CFC and HCFC absorptions can be significant compared to the extra forcing caused by increasing levels of CO₂ [Ramanathan, 1988], and therefore justifies the monitoring of the growth of CHF₂Cl and other halocarbons.

Another reason to study CHF₂Cl growth in particular is to evaluate its utility in determining the global OH field. Historically, emission estimates and field measurements of methyl chloroform (CH₃CCl₃) have been used in models to simultaneously evaluate CH₃CCl₃ lifetime and the OH field; CH₃CCl₃ is a completely anthropogenic gas whose major sink is reaction with OH. (See *Prinn et al.* [1992] and references therein). It was suggested that gases with accurate emission histories and well calibrated measurements other than CH₃CCl₃ could be used for calculation of the OH field [*Prather and Spivakovsky*, 1990]. CHF₂Cl could be evaluated for such a purpose as it is also of completely anthropogenic origin, and its primary sink is reaction with OH.

Prior to 1993, calibration problems between spectroscopic methods and *in situ* techniques had prevented agreement on the atmospheric concentration of CHF₂Cl [*Kaye et al.*, 1994]. Therefore, updated spectroscopic cross-sections for CHF₂Cl in the infrared [*McDaniel et al.*, 1991; *Varanasi*, 1992] and new calibration scales for gas chromatography-mass spectrometry techniques [*Montzka et al.*, 1993] prompted reanalysis of the global increase of CHF₂Cl, and comparison between *in situ* and spectroscopic results. From ATMOS results at Table Mountain (described in Chapter II), the column CHF₂Cl amounts had increased by an exponential growth rate of $(6.7\pm0.5)\%$ from October, 1995 to July, 1990. These results were found to be slightly lower than, but within experimental error of the *in situ* results of *Montzka et al.* [1993] and infrared

column measurements at Kitt Peak, Arizona and the Jungfraujoch, Switzerland [Zander et al., 1994]. However, modelling studies by Margaret Brown, described in Chapter II, indicated that the current uncertainty in CHF₂Cl emission estimates prevented adequate constraint of either the CHF₂Cl lifetime or the OH field, and so the use of CHF₂Cl measurement to infer OH levels offered no advantages over using CH₃CCl₃.

Continued tropospheric and stratospheric measurement of CHF₂Cl is warranted to evaluate any changes in its growth rate from changes in its emission levels or even changes in global OH field. As mentioned previously, many CFCs and HCFCs have significant absorptions in the infrared, and ATMOS spectra have been used to infer stratospheric increases in CF₂Cl₂ and CF₃Cl [Zander et al., 1995] and CHF₂Cl [Rinsland et al., 1995]. It is expected that as the concentration of other replacement HCFCs increase, their infrared absorptions will become significant (and measurable). Figure 1.2.1 illustrates an ATMOS/ATLAS-3 space-based spectral microwindow at a tangent altitude of 6.4 km, superimposed on which are modelled absorptions of CH₃CF₂Cl (HCFC-142b), the tropospheric concentration of which is currently increasing at 30% yr-1 [WMO, 1995]. Synthetic spectra assume current, and 10 and 30 times current mixing ratios. The increasing absorption expected of CH₃CF₂Cl is clearly apparent, and the illustration shows one of the many roles that infrared spectroscopy can play in monitoring secular changes in the atmosphere.

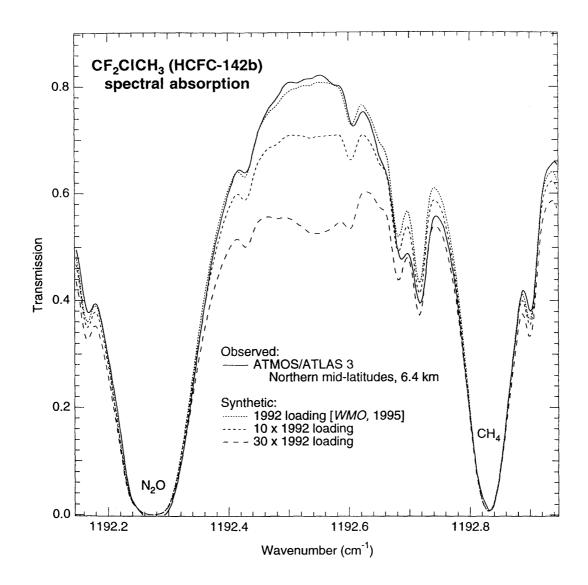


Figure 1.2.1: Modelled absorptions of CF₂ClCH₃ at about 1192 cm⁻¹. Computed absorptions of CF₂ClCH₃ are shown at a 1992 tropospheric mixing ratio of 3.5 pptv [WMO, 1995], 35 pptv and 105 pptv. For comparison, a November, 1994 spectrum from ATMOS/ATLAS-3 at tangent altitude of 6.4 km is displayed.

1.3 A note on the measurement of atmospheric isotopomers

It is often the case that the abundance of a trace isotopomer (an isotopically substituted species such as ¹⁶O¹⁶O¹⁸O or ¹⁶O¹⁸O¹⁶O), in and of itself, is not of interest. Rather it is the ratio of such an abundance to that of the "regular" isotopomer, e.g. ^{[16}O¹⁶O¹⁸O]/^{[48}O₃], and how and where that ratio varies from a standard. (The standard used for the D/H and ¹⁸O/¹⁶O ratios is usually that of Standard Mean Ocean Water, SMOW, which reports the standard D/H ratio as 155.76x10-6 and the ¹⁸O/¹⁶O ratio as 2.005x10-³ [*IUPAC*, 1983].) Investigations of why such ratios vary can reveal subtle processes in geophysics and geochemistry. For example, isotopically labeled water has long been used in better understanding mechanisms of precipitation and its global patterns (e.g., *Dansgaard* [1964]), and validation of global circulation models [*Jouzel et al.*, 1991]. Data of excess carbon-14 (from nuclear weapon tests) have been used in evaluating advection and diffusion coefficients in the stratospheric modelling [*Shia et al.*, 1989]. A review of the mechanisms of isotopic fractionation and their effects on different planetary atmospheres can be found in *Kaye* [1987].

Enrichments (or depletions) of isotopomers from standard ratios are often expressed in percentages:

Enrichment (%) =
$$\frac{R_{\text{sample}} - R_{\text{standard}}}{R_{\text{standard}}} \times 100$$
 (E1.5)

where R_{sample} is the isotopic abundance ratio in a sample (e.g. D/H in water), and $R_{standard}$ is the isotopic abundance calculated from a standard ratio. A slightly different notation widely used (particularly for water isotopes and used in Chapter IV of this work) is the so-

called delta notation, which expresses isotopic deviations on a per mil basis rather than a percentage:

$$\delta (\%c) = \frac{R_{\text{sample}} - R_{\text{standard}}}{R_{\text{standard}}} \times 1000$$
 (E1.6)

Chapter III and Section 1.4 of this introduction involve studies of stratospheric heavy ozone, while Chapter IV and Section 1.5 deal with isotopomers of water and methane. These two issues pose different scientific problems. In the case of heavy ozone, there have been numerous atmospheric, laboratory and theoretical studies done, but there is little apparent agreement on its stratospheric fractionation pattern or the mechanisms by which ozone is isotopically fractionated at all. ATMOS measurements provide the most extensive set of stratospheric heavy ozone observations, and may help to characterize any vertical, latitudinal and seasonal variation of the enrichments. For deuterated water and methane, the fractionation microphysics are thought to be well understood, but stratospheric measurement has been lacking, particularly for CH₃D. The study described in Chapter IV represents the most extensive set of stratospheric observations for these species, which may provide a powerful constraint for model calculations of water transport from the troposphere to the stratosphere.

1.4 Heavy ozone

This section reviews previous experimental and theoretical work done on the phenomenon of heavy ozone enrichment, expanding on the brief introduction on the subject given in Chapter III. Analysis results of ATMOS spectra are discussed.

1.4.1 Previous atmospheric measurements of heavy ozone

Stratospheric heavy ozone enrichment (which, for the purposes of this section, refer to those of ¹⁶O¹⁸O₁, ¹⁶O¹⁸O¹⁶O and their sum in the form of ⁵⁰O₃, except where otherwise noted) was first reported by *Mauersberger* [1981] using a balloon-borne mass spectrometer. (Note that in determining concentrations of heavy ozone, mass spectrometry does not distinguish between symmetric and asymmetric isotopomers.) *Mauersberger* [1981] found stratospheric enrichments of 0-40% in ⁵⁰O₃, with a braod maximum between 28 and 38 km and a minimum at 20 km. The error was 15% at 30km, increasing above and below. (Stratospheric measurements are graphically summarized in Figure 3.1 on page 88.) Using mass-spectrometry, significant enrichments were again reported from two later balloon flights by *Mauersberger* [1987].

Using far-infrared thermal infrared emission measurements from a balloon-borne platform, *Abbas et al.* [1987a] were able to measure the individual absorptions of the $^{16}O^{18}O^{16}O$ and $^{16}O^{16}O^{18}O$ isotopomers. The $^{16}O^{16}O^{18}O$ enrichment had a maximum of $(40 \pm 14)\%$ at 37km, decreasing to $(-2 \pm 12)\%$ at 29 km, and increasing to $(20 \pm 15)\%$ at 25km. The $^{16}O^{18}O^{16}O$ enrichment was about $(18 \pm 15)\%$ at 25 km increasing to $(60 \pm 11)\%$ at 33 km. The finding of seemingly higher enrichments for $^{16}O^{18}O^{16}O$ than $^{16}O^{16}O^{18}O$ appears contradictory to laboratory production experiments of heavy ozone (discussed below) that tend to show symmetric isotopomers less enriched than asymmetric isotopomers [*Morton et al.*, 1989, 1990; *Anderson et al.* 1989; *Mauersberger et al.*, 1993].

Carli and Park [1988], also using thermal emission measurements, reported a ⁵⁰O₃ enrichment of (10±10)% between 27 and 45km, which did not significantly vary with altitude. While they could measure individual ¹⁶O¹⁸O¹⁶O and ¹⁶O¹⁶O¹⁸O lines, they combined the measurements and did not report the individual enrichments. From balloon flights, *Schueler et al.* [1990] collected samples cryogenically, which were later analyzed by mass spectrometry.for both ⁵⁰O₃ and ⁴⁹O₃. In the first flight, ⁵⁰O₃ was enriched 12–14%, and ⁴⁹O₃ was enriched by 9–11%. In the remaining two flights, the enrichment was nearly mass-independent at 8–9%. The errors ranged from 0.5–1.5% for ⁵⁰O₃.

Spectroscopic methods in the mid-infrared have been used to measure the abundance of the two $^{50}\text{O}_3$ isotopomers. *Rinsland et al.* [1985], using a ground-based Fourier-transform infrared (FTIR) spectrometer, found the column enrichments of $^{16}\text{O}^{18}\text{O}$ and $^{16}\text{O}^{18}\text{O}^{16}\text{O}$ over Kitt Peak, Arizona to be $(11 \pm 11)\%$ and $(5 \pm 7)\%$ respectively, noting that no column enrichment could be shown within the range of the error bars. *Goldman et al.* [1989], from two balloon-based measurements using a Fourier transform infrared spectrometer, reported column averaged enrichments above 37 km of $(20 \pm 14)\%$ and $(16 \pm 8)\%$ for $^{16}\text{O}^{18}\text{O}^{16}\text{O}$, and $(40 \pm 18)\%$ and $(25 \pm 12)\%$ for $^{16}\text{O}^{18}\text{O}^{16}\text{O}$.

Krankowsky et al. [1995] reported the first seasonal study of heavy ozone enrichment and the first reported study of its behavior in the troposphere. They analyzed urban air samples using a mass spectrometer over a period of 12 months. They found that the ⁵⁰O₃ enrichment was about 8 to 10% and did not vary with O₃ concentration. They could not find any seasonal variation or significant correlation with temperature. These results were consistent with laboratory measurements at similar temperatures and pressures.

To summarize, it seems to be well established that stratospheric enrichments of heavy ozone exist. However, the magnitude and vertical structure of this enrichment are a

source of confusion. An obvious question remains: Do the dissimilar obtained by the different techniques, as illustrated in Figure 3.1, reveal a highly variable structure as suggested by *Mauersberger et al.* [1993], or are the profiles dissimilar due to instrumental error and/or calibration bias?

1.4.2 Laboratory studies on heavy ozone

To our knowledge, the first laboratory study of heavy ozone enrichment was reported by *Sander et al.* [1977], using ultraviolet photolysis of oxygen. Results were variable; in one experimental run, they reported an exceedingly high enrichment of 95% for $^{50}\text{O}_3$, but all other runs were below 10%. Since that study, no other laboratory enrichments higher than about 20% have been reported.

Heidenreich and Thiemens [1983] synthesized heavy ozone with an electric discharge in a liquid N_2 cooled vessel. They found that both $^{50}O_3$ and $^{49}O_3$ were enriched almost equally (1-4%), while from mass-dependent fractionation chemistry, one would expect $^{49}O_3$ to have only about half the enrichment of $^{50}O_3$. To reduce the effect of isotopic exchange between O and O_2 , Heidenreich and Thiemens [1986] essentially repeated their previous experiment, but at a temperature of 77K. Again, nearly equal enrichments for $^{49}O_3$ and $^{50}O_3$ were found. They proposed that three-body ozone formation was responsible for the observed enhancements:

$$O + O_2 + M \rightarrow O_3 + M \tag{R1.1}$$

and suggested a symmetry effect, with the C_s (i.e., asymmetric) reaction intermediates $^{49}O_3^*$ and $^{50}O_3^*$ having longer lifetimes than the C_{2v} (i.e., symmetric) $^{48}O_3^*$. Since the C_s isotopomer has twice the number of rotational states as the C_{2v} isotopomer, more energy levels are made available for those of the collision partners in an inverse predissociative state, and the collisions may become "stickier." (See *Hertzberg* [1966], pg. 455.) *Yang and Epstein* [1987ab] also suggested a symmetry effect in the predissociation of "vibrationally hot" ozone.

Anderson et al. [1989] investigated symmetry effects by using a tunable diode laser and a mass spectrometer, and separately measured the enrichments of symmetric and asymmetric $^{50}O_3$ produced by electric discharge. It was found that the $^{16}O^{16}O^{18}O$ isotopomer accounted for 4/5ths of the enrichment, not 2/3rds as one would expect on purely statistical grounds. The symmetries of the ozone products were further investigated by *Morton et al.* [1989], who synthesized ozone with an electric discharge, and measured the resultant products from atomic masses 48 (i.e., $^{16}O^{16}O^{16}O$) to 54 (i.e., $^{18}O^{18}O^{18}O$). Results suggested the "more asymmetric" an ozone isotopomer (e.g. $^{16}O^{16}O^{18}O$ more asymmetric than $^{16}O^{18}O^{16}O$ or $^{16}O^{16}O^{17}O$), the greater the enrichment. For example, a maximum enrichment of (20.3 ± 0.2) % was found for $^{51}O_3$ (which would be primarily formed from one each of the ^{16}O , ^{17}O , and ^{18}O isotopes with very little $^{17}O^{17}O^{17}O$), while a slight depletion (0.9 ± 0.2) % was found for the symmetrical $^{18}O^{18}O^{18}O$.

In an attempt to identify the reactions controlling enrichment, *Morton et al.* [1990] dissociated ozone in the Chappuis band (500-700 nm), so that resultant oxygen and oxygen radicals were in their ground electronic states. A small amount of ozone was photodissociated within a bath of O₂ to see if enrichment was produced by the reaction

$$O(^{3}P) + O_{2}(^{3}\Sigma_{g}) + M \rightarrow O_{3} + M$$
 (R1.2)

which is dominant at low O_3 and high O_2 concentrations. Pure ozone was photodissociated to measure isotope effects in the reaction:

$$O + O_3 \rightarrow 2O_2 \tag{R1.3}$$

which dominates at low pressures and high O_3 concentrations. Enrichments were observed when O_2 was present in significant quantities, but no enrichments were seen when the pure ozone was dissociated. Thus, it was concluded that (R1.2) was a reaction producing enrichment. Morton et al. also investigated the temperature dependence of the $^{50}O_3$ enrichments via (R1.2), and found them linearly increasing from 7.5% at 100K to 17.5% at 400K. They also observed a decreasing enrichment with pressures above about

0.5 atm at a temperature of 321K (behavior later confirmed by *Thiemens and Jackson* [1990]). As discussed by *Morton et al.* [1990] and *Thiemens* [1992], if a longer-lived activated complex for the heavy isotopomers is used to explain the enrichment, that is, the reaction proceeds by the mechanism,

$$O + O_2 \leftrightarrow O_3^* \tag{R1.4}$$

$$O_3^* + M \rightarrow O_3 + M \tag{R1.5}$$

with heavier O_3^* longer lived than unsubstituted O_3^* , then enrichment should not be affected by pressure until deviation from third-order to second-order reaction kinetics begins to occur. (This change from third to second-order kinetics at high pressures had been shown for regular ozone formation, and was suggested to occur by the complexing of O or O_2 with M (e.g., N_2):

$$O + M \leftrightarrow OM$$
 (R1.6)

$$OM + O_2 \rightarrow O_3 + M \tag{R1.7}$$

or:

$$O_2 + M \leftrightarrow O_2 M$$
 (R1.8)

$$O_2M + O \rightarrow O_3 + M \tag{R1.9}$$

with reactions (R1.7) or (R1.9) rate-limiting [Croce de Cobos and Troe, 1984]). Such a change in the reaction kinetics affecting the enrichments would be expected near about 8 atm at the temperature used in the experiment of Morton et al. [1990], much higher than where decreasing enrichment was observed to begin. From this, Thiemens [1992] concluded that theories explaining enrichments from differential lifetimes of the metastable states were therefore unlikely to be correct. (Although we note that new results reported by Anderson and Mauersberger [1995], discussed below, may lead to new insights into such activated complexes, and different interpretations of the results of Morton et al. [1990].)

Wen and Thiemens [1991] found a non-mass-dependent depletion effect in the thermal decomposition of heavy ozone. At a high temperature (110°C), thermally

decomposing ozone by the reaction

$$O_3 + M \rightarrow O + O_2 + M \tag{R1.10}$$

produces isotopically heavy product O_2 , enriched in ¹⁸O and ¹⁷O to a maximum of about 2%. The reason for this behavior was not understood.

Mauersberger et al. [1993] reported results from an experiment that was similar to two experiments by Morton et al. [1989, 1990] in that it examined enrichments of all possible molecular weights of ozone, and molecular oxygen and oxygen radicals were kept in their electronic ground state. Results indicated that heavy ozone enrichment occurred only in the asymmetric isotopomers, with asymmetric ⁵¹O₃ enriched by 18%, other isotopomers enriched between about 8% and 14%, and symmetric ¹⁸O¹8O¹8O and ¹⁷O¹7O¹7O depleted. Comparison of these results with those of Morton et al. [1989] showed that the enrichments produced by Mauersberger et al. [1993] using the visible light source were slightly less (≈1-4%) that which would be obtained using an electric discharge. This suggested the ground-state three-body reaction (R1.2):

$$O(3P) + O_2(3\sum_g) + M \rightarrow O_3 + M$$
 (R1.2)

was the predominant mechanism producing enrichment, but additional processes involving higher energy states could be involved in the enrichment of asymmetric heavy ozone.

It thus appears that the asymmetric isotopomers are enriched through the ground-state three-body ozone formation reaction (R1.2), but what other electronic states participate in the enrichment of symmetric and asymmetric isotopomers remains unclear. As will be seen below, the laboratory measurements have nevertheless played a strong role in disproving, or at least discounting, many of the attempts to explain theoretically the enrichments. Still, with the exception of the single run of *Sander et al.* [1977] previously noted, and which has not since been repeated, none of the laboratory measurements

reproduce the high enrichments seen in the stratosphere (e.g., the ⁵⁰O₃ enrichment of 40% seen by *Mauersberger* [1981]).

1.4.3 Theoretical studies on heavy ozone enrichments

In an attempt to explain enrichments produced in an electric discharge, *Thiemens and Heidenreich* [1983] proposed that the cause of the enrichment was due to optical shielding by ¹⁶O¹⁶O. The selective removal of lines from the Schumann-Runge bands resulted in preferential dissociation of ¹⁸O¹⁶O and ¹⁷O¹⁶O. However, this was discounted on theoretical grounds by *Kaye and Strobel* [1983] and experimentally by *Yang and Epstein* [1987a]. Isotopic effects of O₂ dissociation should be obscured by isotopic exchange between atomic O and molecular O₂. In detailed theoretical calculations, *Kaye and Strobel* [1983] argued that the rapidity of the exchange reaction

$$^{18}O + ^{32}O_2 \leftrightarrow ^{16}O + ^{34}O_2$$
 (R1.11)

would prevent a heavy ozone enrichment, and predicted a slight depletion of heavy ozone at stratospheric conditions (≈ -3.2% at 241.5K and 35 km altitude). *Kaye* [1986] examined theoretically the effect of isotopic substitutions on the ozone formation reaction:

$$O + O_2 + M \rightarrow O_3 + M \tag{R1.1}$$

and the oxygen exchange reaction

$$O + O_2 \rightarrow O_2 + O \tag{R1.12}$$

and found the reaction rates affected only slightly. Calculations predicted a small depletion of $^{50}\text{O}_3$ (\approx -4.3% at 300K), although model uncertainties and approximations added considerable uncertainty. Kaye [1986] suggested that the effect of such model uncertainties and approximations may be important in understanding (then current) laboratory results.

Bates [1986] proposed that by taking account of the symmetry numbers (as they relate to the partition functions) in the three-body formation of ozone (R1.1), the formation reaction producing asymmetric heavy ozone would have twice the rate of regular ozone.

For symmetric heavy ozone, the rate would be one half times that of regular ozone. The net result was that heavy ozone was naturally enriched by 2/3. *Anderson and Kaye* [1987] pointed out that Bates had neglected that in an activated complex of an ozone isotopomer, the complex can decay into molecular and atomic oxygen isotopically distinct from the reactants, e.g.

$$^{16}O + ^{16}O^{18}O \rightarrow ^{16}O^{16}O^{18}O^*$$
 (R1.13)

$$^{16}O^{16}O^{18}O^* + M \rightarrow ^{18}O + ^{16}O^{16}O + M$$
 (R1.14)

Any enrichment of heavy ozone solely by symmetry effects would thus be eliminated. Bates [1987] accepted their conclusion.

A novel mechanism for mass-independent isotopic fractionation of ozone was proposed by Valentini [1987], which involved curve-crossing between photo-fragmentary $O_2(^1\Delta_g) + O(^1D)$ and $O_2(^3\Sigma_g) + O(^3P)$. Now, homonuclear $^{16}O^{16}O(^3\Sigma_g)$ can only have odd-J rotational states, but $^{18}O^{16}O(^3\Sigma_g)$ can have odd and even-J rotational states. So only $^{16}O^{16}O(^1\Delta_g)$ in odd-J rotational states can cross to $^{16}O^{16}O(^3\Sigma_g)$, while heteronuclear $^{17}O^{16}O$ and $^{18}O^{16}O$ in both odd and even-J rotational states can cross from the $^{1}\Delta_g$ to the $^{3}\Sigma_g$ state. Thus, via the curve crossing channel, the yield of heteronuclear $^{17}O^{16}O(^3\Sigma_g)$ and $^{18}O^{16}O(^3\Sigma_g)$ were both twice that of homonuclear $^{16}O^{16}O(^3\Sigma_g)$. The controlling effect was symmetry, not isotopic mass. The $O_2(^3\Sigma_g)$ reservoir would be enriched in heavy isotopes, the $O_2(^1\Delta_g)$ reservoir depleted, and since there are no bound O_3 states that correlate with $O_2(^1\Delta_g) + O(^1D)$ or $O_2(^1\Delta_g) + O(^3P)$, the product ozone would be enriched in heavy isotopes.

For explaining stratospheric heavy ozone enrichment, the mechanism of *Valentini* [1987] was deficient in some respects. First, as pointed out by Valentini the stratospheric $O(^3P) + O_2(^3\Sigma_g)$ recombination rate, and the production rate of $O_2(^1\Delta_g)$ via O_3 dissociation, are both very slow compared to $O_2(^1\Delta_g)$ quenching. Isotopic fractionation among O and O_2 reactants dissipates faster than a heavy ozone enrichment can be

produced. Valentini [1987] estimated that stratospheric enrichment of O_3 and $O_2(^3\Sigma_g)$ would only be about 10^{-5} with this scheme. Second, the curve-crossing mechanism of Valentini [1987] would predict that resulting heavy ozone isotopomers would be partitioned two-thirds in $^{16}O^{16}O^{18}O$ and one-third in $^{16}O^{18}O^{16}O$, as would be statistically expected. This turned out to be contrary to the laboratory results of Anderson et al. [1989], Morton et al. [1990], and Mauersberger et al. [1993], who found the asymmetric isotopomers preferentially enriched. Third, Morton et al. [1990], had enriched heavy ozone with molecular oxygen in the ground $^3\Sigma_g$ state, not the excited $^1\Delta_g$ state, and no curve-crossing mechanism was required.

Symmetry again played a role in papers by *Bates* [1988, 1990] in which it was suggested that at steady-state conditions, metastable states of asymmetric heavy ozone, $^{16}O^{18}O^*$ are preferentially formed. Distinguishing between the states $^{16}O^{\sim 16}O^{18}O^*$ and $^{18}O^{\sim 16}O^{16}O^*$, where \sim signifies a newly formed bond and the energy is non-randomized, he argued that the rates of formation of these two isotopomers are additive as long as they can be distinguished from one another. He suggested that the rate of formation of $^{16}O^{16}O^{18}O$ is $(1+\delta)$ times that of regular ozone, where δ is the "rate additivity", given by the formula:

$$\delta = \frac{1}{\tau_{D}} \int_{0}^{\tau_{D}} \exp(-t/\tau_{R}) dt$$

$$= \left\{ \frac{\tau_{R}}{\tau_{D}} \right\} \left\{ 1 - \exp\left(-\frac{\tau_{D}}{\tau_{R}}\right) \right\}$$
(E1.3)

where τ_D and τ_R are the dissociation and energy randomization time constants respectively. Bates [1990] presented calculations estimating that δ =0.74. This scheme would lead to enrichments in asymmetric heavy ozone only. The predicted enrichments were consistent with some of the higher enrichments seen in the stratosphere, but could not account for the lower enrichments seen in the laboratory. *Bates* [1990] suggested that the lower enrichments in the laboratory were due to the reaction:

$$^{50}O_3 + ^{32}O_2 \rightarrow ^{48}O_3 + ^{34}O_2$$
 (R1.15)

either in the gas phase or on the walls of the surface vessel. This was disputed by *Thiemens* [1992], because (R1.13) would be a mass-dependent reaction while in the laboratory, low (<5%) mass-independent enrichments of $^{50}O_3$ and $^{49}O_3$ have been produced. To account for enrichments in asymmetric heavy ozone, *Bates* [1990] proposed "flips" of a terminal oxygen atom on an activated ozone complex, e.g.

$$^{16}\text{O}^{\sim 16}\text{O}^{18}\text{O}^* \to ^{16}\text{O}^{\sim 18}\text{O}^{16}\text{O}^*$$
 (R1.16)

$$^{16}\text{O} \sim ^{18}\text{O}^{16}\text{O}^* \rightarrow ^{18}\text{O} \sim ^{16}\text{O}^{16}\text{O}^*$$
 (R1.17)

$$^{18}\text{O}^{\sim 16}\text{O}^{16}\text{O}^* \to ^{16}\text{O}^{\sim 16}\text{O}^{18}\text{O}^*$$
 (R1.18)

In later Monte Carlo simulations, Morgan and Bates [1992], presented calculations indicating that because of such flips, the ratio of enrichment of the symmetric isotopomer to that of the asymmetric isotopomer should be about 0.38, which was in good agreement with the result of Anderson et al. [1989] of 0.40 ± 0.17 . Experimentally, however, Larsen et al. [1992] presented evidence that ozone formation is primarily "end-on," although measurement error could allow for a 15% cyclic intermediate channel.

An explanation for some of the enrichment of heavy ozone may have been provided by *Miller et al.* [1994]. In a series of experiments, Miller et al. found that in the photodissociation of ozone at 226 nm:

$$O_3 + hv \rightarrow O_2(^{1}\Delta_g) + O(^{1}D)$$
 (R1.19)

$$\rightarrow O_2(^3\Sigma_g) + O(^3P) \tag{R1.20}$$

the distribution of vibrational states of $\mathrm{O}_2(^3\Sigma_g)$ was remarkably bi-modal, with peaks at

 ν =14 and ν =27. They theorized that such "vibrationally hot" $O_2(^3\Sigma_g)$ can quickly react further with oxygen in a net production sequence for ozone:

$$O_3 + hv \rightarrow O_2(^3\Sigma_g, \nu > 26) + O$$
 (R1.21)

$$O_2(^3\Sigma_g, \nu > 26) + O_2 \rightarrow O_3 + O$$
 (R1.22)

$$2(O_2 + O) \rightarrow 2O_3$$
 (R1.23)

Net:
$$3 O_2 \rightarrow 2 O_3$$

As discussed by Miller et al., the new mechanism can have implications for heavy ozone enrichment. Recalling the results of Valentini [1987], $^{16}O^{18}O(^{1}\Delta_{g}) + O(^{1}D)$ is twice as likely to "curve-cross" to $^{16}O^{18}O(^{3}\Sigma_{g}) + O(^{3}P)$ than $^{16}O^{16}O(^{1}\Delta_{g}) + O(^{3}P)$ will cross to $^{16}O^{16}O(^{3}\Sigma_{g}) + O(^{3}P)$. Thus the channel (R1.20) is more favorable for a photodissociating heavy ozone molecule than for a regular ozone molecule. Note also that at least one of the oxygen atoms in the photodissociating O_{3} on the left hand side of (R1.21) is incorporated into the ozone on the right-hand side of (R1.22). The combined effect gives a heavy O atom greater probability to stay within the pool of O_{3} and O_{2} and not be "scrambled out" by the O-O₂ exchange reaction (R1.11). Reactions (R1.21) and (R1.22) can thus cycle to enrich heavy ozone until a termination reaction, e.g., $O + O_{3} \rightarrow 2O_{2}$, is reached. Miller et al. presented calculations that accounted for a 5% enrichment at 40 km decreasing to 0.6% at 55 km. It is again noted that the reaction scheme of (R1.21) through (R1.23) has not been verified experimentally.

Theoretical studies have, to date, been unable to fully explain the laboratory or atmospheric studies, although some research of late may prove fruitful. Recently, Anderson and Mauersberger [1995] reported evidence for three electronically excited states of ozone at an energies slightly above the dissociation energies of the ground state. They discussed the possibility that these low-lying states may be involved in the three-body

ozone formation reaction, and speculated that these excited states may have a role in heavy ozone enrichment. The impact of this new research into ozone chemistry, and heavy ozone enrichment, remains to be determined, however.

1.4.4 Results of ATMOS analyses

It would appear that little scientifically would be gained by analysis of a single ATMOS occultation, other than to add to the set of disparate measurements already made. However, the sheer number of available occultations from the space-based ATMOS data set, their latitudinal range, and the number of measurement days of the ATMOS instrument from Table Mountain allow two outstanding questions to be answered. First, how variable are the vertical profiles of heavy ozone enrichment? Second, does this enrichment vary by latitude or season?

Recalling the laboratory work of *Morton et al.* [1990], who measured relatively little variation within the range of temperatures and pressures typical of the stratosphere, the finding of a low variability by ATMOS might be evidence for the three-body ozone formation reaction (R1.2) primarily operating to enrich stratospheric heavy ozone. It is also useful to characterize any variation of heavy ozone as the photolysis products of enriched heavy ozone can be used in other atmospheric studies. *Yung et al.* [1991] proposed that the observed $\approx 0.5\%$ ¹⁸O enrichment in CO₂ [*Gamo et al.*, 1989; *Thiemens et al.*, 1991] could be used as an indirect measure of O(¹D) in the stratosphere considering the reaction scheme:

$$^{18}O(^{1}D)$$
 (from $^{16}O^{16}O^{18}O$ photolysis) + $CO_2 \rightarrow CO_3^*$ (R1.22)

$$CO_3^* \to C^{18}O^{16}O + {}^{16}O$$
 (R1.23)

The "heaviness" of the $^{50}\text{O}_3$ photolysis products can also be used as a tracer for other stratospheric reactions. *Yung et al.* [1995] lists reactions possibly producing HNO₃ and H₂SO₄ (via photolysis of COS) enriched in heavy oxygen because their production

pathway involves reaction with O_3 . Analyses of such reaction schemes will be greatly aided if the vertical structure of the heavy ozone enrichment is well defined.

Results of analyses of ATMOS space-based spectra, described in Chapter III, show an average (15±6)% enrichment for ¹⁶O¹⁶O¹⁸O, (10±7)% for ¹⁶O¹⁸O¹⁶O, and (13±5)% for ⁵⁰O₃ (1σ standard deviation). Results also indicate that the vertical and latitudinal variability of the ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O enrichment is relatively small. This suggests a relatively small sensitivity of the enrichments to temperature and pressure representative of the stratosphere, in qualitative agreement with the laboratory results of *Morton et al.* [1990]. Column measurements of the ¹⁶O¹⁶O¹⁸O enrichment over TMF, with an average of (17±4)%, show little seasonal variability within the uncertainty, in qualitative agreement with the tropospheric measurements of *Krankowsky et al.* [1995]. On an absolute basis, however, both the ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O enrichments found in the ATMOS study are, on average, higher than what is expected if stratospheric heavy ozone enrichment were completely (or even primarily) caused by the three-body ozone formation reaction (R1.2), and no other formation mechanism has been shown to create ozone in appreciable amounts in the 25-40 km altitude range of our measurements.

It is therefore difficult to quantitatively reconcile the results described here with those measured previously in the laboratory. This suggests the possibility that the line intensities used for the spectroscopic analyses of ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O may be biased too low, resulting in heavy ozone retrievals biased too high. A laboratory double-check of these line intensities is warranted to determine if a scaling of the results given in Chapter III is justified. If the current intensities are found to be reasonably accurate, then the ATMOS results indicate that there are currently unknown processes involved in the creation (or destruction) of stratospheric ozone. Other than this, it appears that research should be directed towards better understanding the unknown, but perhaps fundamental, details in the three-body formation of ozone. If laboratory and stratospheric measurements

have not completely elucidated the reasons for the enrichment, they have at least narrowed the range of the problem. Nevertheless, the anomaly of heavy ozone enrichment remains today as much a challenge to explain as it has been for the past 15 years.

1.5 HDO, CH₃D and implications for tropospheric-stratospheric exchange

The study of atmospheric HDO and CH₃D cannot be separated from that of H₂O and CH₄. Some characteristics of H₂O, CH₄, HDO and CH₃D in the stratosphere are briefly reviewed. Competing theories of troposphere-stratosphere exchange and the implications of the measurement of the D/H ratio in water on the validity of these theories are outlined. Finally, ATMOS results of deuterated water and methane and their implications are presented and further areas for study suggested.

1.5.1 Stratospheric water and methane

The hydrogen loading of the stratosphere is effectively 2(H₂O+2CH₄+H₂), as the abundance of other hydrogen-containing gases is very small compared to this sum. In an indirect way, these gases play a role in regulating stratospheric ozone, as reaction with O(¹D) is an important source of the OH radical which attacks ozone (e.g., *Brasseur and Solomon* [1986]). Stratospheric water plays another important indirect role in ozone depletion. It is incorporated into sulfuric acid aerosol, leading to heterogeneous surface reactions producing HOCl and Cl₂ from reaction of ClONO₂ with H₂O and HCl, respectively. The diurnally varying odd nitrogen species N₂O₅ is converted to HNO₃ by reaction with H₂O on such aerosols, and this indirectly affects ozone destruction because NO₂ (from N₂O₅ photolysis) sequesters ClO into ClONO₂. Many reactions sequestering N₂O₅, and converting chlorine reservoir and sink species to active species, occur readily on frozen ice, HNO₃ hydrates and supercooled and frozen sulfate aerosols in the cold polar night, leading to the Antarctic ozone hole in the austral spring [*WMO*, 1995].

Being long-lived in the mid-to-lower stratosphere, water is controlled by a combination of transport and methane oxidation (e.g., *Brasseur and Solomon* [1986]). However, the stratosphere is exceedingly dry, with H₂O mixing ratios of only a few partsper-million in the lower stratosphere. Excluding extremely cold and dehydrated regions within the Arctic and Antarctic vortices, the mixing ratio of stratospheric H₂O tends to be a

minimum at the tropical tropopause or in a region up to 2 km above. This water vapor minimum, the hygropause, is believed to be a consequence of a seasonal variation in the mixing ratio of water entering the stratosphere. The static stability of the lower stratosphere coupled with the slow rate of ascent allows this "seasonal signature" minimum to be transported upwards with time, and discernable for as long as a year [Mote et al., 1995; Boering et al., 1995; Abbas et al., 1995a; Holton et al., 1995].

Above the tropical hygropause, and throughout the rest of the stratosphere (excluding the vortices), the H_2O mixing ratio increases with altitude as the air becomes older, and CH_4 and H_2 are oxidized by OH, $O(^1D)$ and, for CH_4 , Cl. However, in the mid to lower stratosphere at all latitudes except in the dehydrated regions of the polar vortices, the sum $2CH_4+H_2O$ has been shown on average to be about constant. The increase in H_2O is about equal to twice the decrease in CH_4 , with the H_2 mixing ratio (≈ 0.5 ppmv) varying little. This is not to say that H_2 is unreactive. Rather, its production from CH_4 (via a short-lived CH_2O intermediate) and destruction by oxidation roughly balance each other [Dessler et al., 1994; Abbas et al., 1995b]. In order to use ATMOS stratospheric measurements to estimate the average mixing ratio of H_2O as air enters the stratosphere, the *in situ* contribution of CH_4 oxidation to the water mixing ratio must be subtracted (as had been done by Abbas et al., [1995a] with ATMOS/ATLAS-3 data).

Analogously to CH₄ and H₂O, stratospheric HDO is controlled by transport and oxidation of CH₃D. Likewise, the contribution of CH₃D loss must be subtracted to estimate the entering HDO mixing ratio (and the D/H ratio in water), as previous measurements of the deuterium content of stratospheric water have shown that the D/H ratio in water increases with altitude [Pollock et al., 1980; Rinsland et al., 1984, 1991; Abbas et al., 1987b; Carli and Park, 1988; Dinelli et al., 1991]. It has not been established, however, that the sum CH₃D+HDO is about constant as is the case for 2CH₄+H₂O; the destruction of CH₃D may not be one-to-one with the production of HDO.

The reaction pathway from CH₃D to HDO and HD is that the D/H branching may not follow simple statistics and HD production may not be in balance with its oxidation. Indeed, in the only previous report of both stratospheric CH₃D and HDO measurements (ATMOS Spacelab 3 measurements by *Rinsland et al.* [1991]), data were too sparse to quantitatively relate the destruction of CH₃D to the creation of HDO.

1.5.2 Mechanisms of exchange across the tropical tropopause

Theories explaining tropospheric-stratospheric transport must quantitatively account for the dryness of the lower stratosphere, but exactly how air is dehydrated as a parcel rises through the troposphere and crosses the tropopause is a matter of debate. As discussed in Holton et al. [1995] and Moyer et al. [1995], widespread gradual uplift of air in the tropics across the tropopause cannot be the only method of water transport into the stratosphere; a large resultant cirrus cloud cover is not seen, and the average tropical tropopause temperatures are too warm for the low stratospheric mixing ratios observed. Other theories have suggested gradual uplift in more isolated (and colder) regions [Newell and Gould-Stewart, 1981] or rapid convection by cumulonimbus towers penetrating the tropopause [Danielsen, 1982]. The former scenario represents a slow, quasi-equilibrium process. The latter can be a non-equilibrium process as water can be advected rapidly up from the boundary layer and deposited as ice in the stratosphere (although under certain conditions, stratospheric dehydration can occur if stratospheric water is entrained into ice crystals large enough to fall out). These two scenarios can have very different implications for the deuterium content of stratospheric water.

1.5.3 The use of HDO in evaluating tropical cross-tropopause transport

The substitution of a deuterium atom for a hydrogen atom in water significantly decreases its vapor pressure in a manner that is a strong function of temperature, particularly at temperatures representative of the mid-to-upper troposphere (e.g., *Kaye* [1987]; *Dansguaard* [1964]). In the troposphere, the lower vapor pressure of HDO

compared to H₂O isotopically fractionates water vapor as it ascends to regions of lower temperatures, i.e., the water becomes "lighter," and the HDO/H₂O ratio decreases as HDO is preferentially condensed. As discussed by *Moyer et al.* [1995], the slow uplift of air through the tropopause will tend to strongly deplete water in deuterium. On the other hand, the deuterium content of the water may be higher in rapid convective updrafts because it can be carried in the form of lofted ice particles, or because of kinetic fractionation effects under highly supersaturated conditions prevent equilibrium isotopic partitioning. Measurements of the HDO/H₂O ratio in stratospheric water can help discriminate between competing theories of water transport from the troposphere to the stratosphere; any theory that successfully explains the amount of entering water must simultaneously explain the amount of HDO.

1.5.4 ATMOS HDO and CH₃D results

As discussed in Chapter IV, ATMOS measurements from Spacelab 3, and ATLAS-1, -2, and -3 were used first to characterize the relationship of CH₃D to CH₄, and showed that stratospheric methane becomes progressively enriched in deuterium as the CH₄ mixing ratio decreases. This was expected since the reaction rate constants of CH₃D with OH and Cl are less than those for CH₄. From ATMOS HDO measurements and CH₃D mixing ratios inferred from co-located CH₄, it was found that the stratospheric production of HDO is roughly equal to the destruction of CH₃D; a slope of –(1.0±0.1) was found for a plot of HDO vs CH₃D mixing ratio (1σ combined systematic and random error). Stratospheric changes in the HD mixing ratio were therefore constrained to be ±0.1 molecules created per molecule of CH₃D destroyed, assuming negligible amounts of total D in other molecules.

The HDO and inferred CH₃D mixing ratios were combined with ATMOS H₂O and CH₄ mixing ratios [*Abbas et al.*, 1995a] to derive stratospheric D/H ratios in water corrected for the hydrogen and deuterium contributions of methane oxidation [*Moyer et al.*, 1995]. Results showed that water entering the stratosphere was depleted in deuterium

(using the SMOW standard) by an average of $68\pm8\%$ (1σ combined systematic and random error), and showed no latitudinal or seasonal variation. Modelling results by *Moyer et al.* [1995] indicated that the gradual uplift of water in the troposphere and through the tropopause should result in an 80-90% depletion of deuterium, but ATMOS results were consistent with water vapor experiencing rapid convection to at least upper tropospheric levels.

The isotopic signal in water was shown to provide a powerful constraint on theories of water transport, and ATMOS data can still be further analyzed for this purpose. Due to the limited precision of the measurements and few number of observations made directly in the tropics from ATLAS-3 (and none available near the tropical tropopause from ATLAS-1 due to Pinatubo aerosol), ATMOS results (from filters 2 and 9) were not able to be used to determine a seasonal signal in HDO akin to the behavior of the hygropause, which would have helped in determining the exact mechanism of water transport across the tropopause. With improved spectral line parameters for weak HDO lines in the v_2 band (in filter 3), it may be possible to combine and average results from filters 2, 3 and 9 within narrow latitude bands and determine the seasonal variation. The use of these weak lines would extend the range of HDO measurement below about 100 mb down to the 300 mb level, near the level of the tropopause in the mid-latitudes. Examination of the D/H ratio in water at this level may give information on the extent and effect of tropospheric incursions of water into the lower mid-latitude stratosphere by Rossby waves (see *Holton et al.* [1995]).

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Table 1.1: ATMOS data summary

Filter	Number of Occultations				Spectral	Molecules
	SL3	AT1	AT2	AT3	Bandpass (cm-1)	
1	1 SR 4 SS	6 SR 7 SS	11 SR 6 SS		600-1180	48O ₃ , 16O16O18O, 16O18O16O, N ₂ O, HNO ₃ , CCl ₃ F, CCl ₂ F2, CHF ₂ Cl, ClONO ₂ ,SF ₆ , HNO ₄ , CCl ₄ , C ₂ H ₂
2	1SR 3SS	1 SR 1 SS	9 SR 7SS		1100-2000	⁴⁸ O ₃ , N ₂ O, CH ₄ , HNO ₃ , H ₂ O, NO, NO ₂ , N ₂ O ₅ , HDO, H ₂ 1 ⁷ O, H ₂ 1 ⁸ O, CF ₄
3	1 SR 4SS	15 SR 14 SS	22 SR 9 SS	29 SR 34 SS	1580-3340	⁴⁸ O ₃ , N ₂ O, CH ₄ , HNO ₃ , H ₂ O, HCl, NO, NO ₂ , OCS, HCN, H ₂ 1 ⁷ O, H ₂ 1 ⁸ O, CH ₃ D, CO, C ₂ H ₆ , CH ₃ Cl
4	1 SR 2 SS	8 SR 5 SS	10 SR 7 SS	13 SR 15 SS	3150-4800	N_2O , CH_4 , H_2O , HF
9		17 SR 14 SS	1 SS	14 SR 17SS	600-2450	⁴⁸ O ₃ , ¹⁶ O ¹⁶ O ¹⁸ O, ¹⁶ O ¹⁸ O ¹⁶ O, N ₂ O, CH ₄ , HNO ₃ ,H ₂ O, NO, NO ₂ , CCl ₃ F,CCl ₂ F ₂ , CHClF ₂ , ClONO ₂ , N ₂ O ₅ , SF ₆ , OCS, HDO, H ₂ ¹⁷ O, H ₂ ¹⁸ O, CO, HNO ₄ , CCl ₄ , C ₂ H ₂ , CF ₄
12			11 SR 9 SS	27 SR 30 SS	600-1400	⁴⁸ O ₃ , ¹⁶ O ¹⁶ O ¹⁸ O, ¹⁶ O ¹⁸ O ¹⁶ O, N ₂ O, HNO ₃ , CCl ₃ F, CCl ₂ F ₂ , CHF ₂ Cl, ClONO ₂ ,SF ₆ , HNO ₄ , CCl ₄ , C ₂ H ₂ , CF ₄

Note: SR=sunrise, SS=sunset, SL3=Spacelab 3, AT1=ATLAS-1, AT2=ATLAS-2, AT2=ATLAS-2

Chapter II

Increase in atmospheric CHF₂Cl (HCFC-22) over southern California from 1985 to 1990

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Abstract.

Column densities of CHF₂Cl (HCFC-22) have been measured over Table Mountain Facility (TMF), Wrightwood, California (34.4°N) using the Atmospheric Trace Molecule Spectroscopy (ATMOS) Fourier-transform infrared (FTIR) spectrometer. Between October 1985 and July 1990, the exponential column increase rate was (6.7±0.5)% yr⁻¹. Additionally, column measurements of CHF₂Cl over McMurdo Sound, Antarctica (78°S) in September and October 1986 by the MarkIV FTIR spectrometer were used to derive a south-north interhemispheric ratio of (0.86±0.08). Model calculations investigated the feasibility of using CHF₂Cl column measurements with a predicted global OH field to determine a globally averaged chemical lifetime for CHF₂Cl, or equivalently, an estimate of the OH field using a predicted lifetime. The current uncertainty in historical CHF₂Cl emissions is too large for CHF₂Cl to be used to infer adequately either the lifetime or the OH field.

2.1 Introduction

Concern over the ozone layer has required the increasing use of hydrochlorofluro-carbons (HCFCs) as an alternative to chloroflurocarbons (CFCs). Unlike CFCs, HCFCs react with OH radicals, shortening their tropospheric lifetimes (e.g., 15.8 years calculated for CHF₂Cl vs. 55 years for CFCl₃ and 116 years for CF₂Cl₂ [WMO, 1991, pg. 6.7]). With its shorter tropospheric lifetime, the contribution of CHF₂Cl to inorganic chlorine in the stratosphere has been estimated to be quite small (1%) [Weisenstein et al., 1992]. Nevertheless, with the increasing use of CHF₂Cl, accurate prediction of changes in stratospheric ozone and global climate warrants its continued monitoring.

Among recent reports, *Montzka et al.* [1993], using in-situ measurements, determined a compounded rate of increase of (7.3±0.3)% yr-1 in the global mean ground-level concentration of CHF₂Cl between mid-1987 through 1992. Total column measurements reported by *Zander et al.* [1994] showed an exponential increase of (7.0±0.35)% yr-1 over the Jungfraujoch, Switzerland (46.5°N) from 1986 to 1992, and (7.0±0.23) % yr-1 over Kitt Peak, Arizona (31.9°N) from 1980 to 1992. In this study we report measurements taken by the ATMOS spectrometer over Table Mountain and calculate the column accumulation rate between October, 1985 and July, 1990.

Column measurements by the MkIV FTIR spectrometer over McMurdo Sound in 1986 were used to evaluate the interhemispheric ratio of CHF₂Cl, but these columns were not directly comparable to those in northern mid-latitudes because of the lower Antarctic tropopause. Instead, a simple and self-contained comparison was made by ratioing the CHF₂Cl column to that of N₂O, a long-lived tropospheric source gas whose vertical mixing ratio profile is similar in shape to that of CHF₂Cl. As some 90% of the atmospheric N₂O resides in the troposphere, the N₂O column can act as a surrogate for the tropospheric air

burden, and we use such CHF₂Cl/N₂O ratios to estimate the CHF₂Cl interhemispheric ratio.

We discuss the use of CHF₂Cl measurement for inference of the global OH field. Current estimates of OH are based on scaling the OH field produced by chemical transport models to reproduce observed surface concentrations of methyl chloroform, CH₃CCl₃ (e.g. *Spivakovsky et al.* [1990]; *Prinn et al.* [1992]). Differences in the inferred OH fields are discussed in terms of the error in historical emissions of CHF₂Cl.

2.2 Measurements and data analysis

The ATMOS instrument was designed for stratospheric observations from the Space Shuttle [see *Farmer et al.*, 1987], but is also suitable for ground-based column density measurements of various gases [see *Gunson and Irion*, 1991]. It has an unapodized resolution is 0.01 cm⁻¹ and its fast response time allows several interferograms to be taken within a narrow range of solar zenith angles. For this study, 8 to 41 successive, double-sided interferograms were transformed with the resulting spectra averaged, giving a signal-to-noise ratio in the resulting spectrum of at least 280. For analyses described here, the Mark IV FTIR spectrometer returned data with an unapodized spectral resolution of 0.008 cm⁻¹ and signal-to-noise ratios varying between 100 and 1200. Further details of this instrument, as well as experimental conditions at McMurdo Sound, may be found in *Toon et al.* [1989] and *Toon* [1991].

Analysis software used for column retrievals from both instruments was described by *Norton and Rinsland* [1991]. Constituent and physical parameters were mapped into a 150 layer model atmosphere, each layer homogeneous and 1 km thick. For the ATMOS/-TMF retrievals, the *U. S. Standard Atmosphere* [1976] was used for the temperature and pressure profile while a combination of daily radiosonde and satellite data were used for McMurdo Sound retrievals [see *Toon et al.*, 1989]. Synthetic spectra were calculated for

small frequency intervals containing the target gas, and an assumed mixing ratio profile was scaled until the square of the residual (observed - calculated absorptions) was minimized. Spectral parameters were from the ATMOS main and supplemental linelists, described by Brown et al. [1987]. The 2v₆ Q-branch at 829.05 cm⁻¹ was used for CHF₂Cl analyses (Figure 2.1), while N₂O lines were the same as in *Toon et al.* [1989]. Initial mixing ratio profiles are shown in Figure 2.2. The TMF CHF₂Cl profile was based on ATMOS Spacelab3 northern hemisphere retrievals in the stratosphere [Zander et al., 1987], with an assumed gradient in the troposphere. The tropospheric gradient used for the TMF profile resulted in better spectral fits to the data compared to a constant mixing ratio profile, however, no statistically significant bias in the derived columns was detected. We found that using a constant CHF₂Cl tropospheric mixing ratio profile for McMurdo Sound, and a constant tropospheric N₂O profile for both TMF and McMurdo, produced satisfactory fits for their respective spectra. Column N₂O for McMurdo Sound is from Toon et al. [1989]. Where two or more averaged spectra were available for a particular day, each spectrum was analyzed individually, and the resulting columns averaged. Estimated errors are described in Table 2.1.

2.3 Instrumental results and discussion

CHF₂Cl column results for Table Mountain, McMurdo Sound and, for comparison, Kitt Peak results (*Zander et al.* [1994]) are shown in Figure 2.3. To facilitate comparison of Kitt Peak and TMF columns, retrieved values were divided by the U.S. Standard Atmosphere [1976] pressure (in atmospheres) appropriate to the observation altitude. (Same day measurements of N₂O were not available from Kitt Peak.) An exponential fit to the TMF column data indicates an increase of (6.7±0.5)% yr-1, slightly lower than the figure of (7.0±0.23)% yr-1 for Kitt Peak. No significant systematic difference can

be seen between TMF and Kitt Peak results during the time period in which they overlap. Surface mixing ratios were determined directly from the scaled a priori profiles. Calculations for Table Mountain show an increase in the ground-level mixing ratio from (70 ± 6) pptv in October, 1985 to (99 ± 8) pptv in August, 1990, the linear increase being (5.9 ± 0.4) pptv yr⁻¹ (1 σ error). This agrees within error with the figure of (6.3 ± 0.3) pptv yr⁻¹ reported by *Montzka et al.* [1993] for the period between 1987 and December, 1992.

Before using the TMF data to infer the interhemispheric ratio of CHF₂Cl, we first evaluated its suitability for determining an average northern hemispheric CHF₂Cl column by comparison to latitudinally weighted in situ data previously published by Montzka et al. [1993]. Rather than using ground-level concentrations estimated from the columns, a more robust comparison may be with ratios of CHF₂Cl to N₂O. Figure 2.4 shows the CHF₂Cl/N₂O column ratios from this work. Also shown are the in situ measurements of CHF₂Cl, reported by Montzka et al., ratioed to estimated N₂O concentrations, obtained either from monthly averaged measurements of N₂O (prior to July, 1988) or extrapolations of equations fitted to those measurements (on or after July, 1988; see *Prinn et al.*, [1990]). These N₂O measurements were from Cape Meares, Oregon (45°N) in the northern hemisphere and Cape Grim, Tasmania (41°S) in the south. Good consistency is seen among latitudinally averaged northern hemispheric in situ measurements, in situ measurements from Niwot Ridge (40°N), and the ATMOS/TMF results. Using the average of the McMurdo Sound observations and the fitted line for the ATMOS data, we calculate a southnorth hemispheric ratio of (0.86±0.08). This is in good agreement with the figure of (0.88±0.04) derived from Montzka et al. [1993], who reported an interhemispheric CHF₂Cl difference of (13±1) pptv and a 1992 mean southern hemisphere mixing ratio of (95.2 ± 2) pptv.

2.4 Model methodology and results

Model results were obtained using the 2-D chemical transport model developed by Tung [1982; 1986], Yang et al. [1991] and Olaguer et al. [1991]. The model domain extends from 90°S to 90°N with eighteen latitudinal bands and from 0 to 56 km with 24 vertical levels. The model has been validated for tracers which are sensitive to stratospheric and tropospheric chemical and transport parameters [Yang et al., 1991; Olaguer et al., 1992; Brown, 1993]. Destruction of CHF₂Cl by photodissociation and by chemical reaction with OH and O(1D) were modeled. Concentrations of CHF₂Cl were set to zero at the beginning of 1949, when the atmosphere is expected to have none. The model was run forward to the end of 1992 using emission estimates of Jesson [1980] prior to 1970, and Midgley and Fisher [1993] from 1970 to 1991. For 1992, an emission of 213 x 106 kg is used (obtained by linear extrapolation of prior emission estimates). The uncertainty in emissions subsequent to 1970 were estimated to be +12.5% and -8.5% [Don Fisher, private communication, 1993]. The model has been shown to produce modeled atmospheric concentrations of methyl chloroform consistent with observations [Brown, 1993]. The globally averaged lifetime of methyl chloroform was calculated to be 7.3 years, longer than previous estimates of 5.7 years [Prinn et al., 1992] and 6.2 years [Spivakovsky et al., 1990]. The model calculated CHF₂Cl lifetime here, 19 years, is expected to be longer than that derived using the OH fields of Prinn et al. and Spivakovsky et al. Here, for CHF₂Cl, we demonstrate the relationship between the lifetime and the OH field as well as the sensitivity of the column density to the emissions.

Six experiments (E1-E6) were performed and modeled column densities of CHF₂Cl were compared with observed column densities. The results are summarized in Figure 2.3. Experiment E1, using emission estimates described above, overestimates column densities at both observation sites. These results reflect uncertainty in emission estimates and the

model OH field. Experiments E2-E5 test the sensitivity of the model results to these uncertainties. Modifications to the OH field and the CHF₂Cl emission, as well as average concentrations for all model experiments are summarized in Table 2.2. The model calculated globally averaged CHF₂Cl lifetime in experiments E1 and E2 is 19 years. The scaling of the OH field in experiments E3 and E4 yields a model calculated globally averaged CHF₂Cl lifetime of 15.5 years, consistent with the estimate in *WMO* [1991], while that for E5 yields a CHF₂Cl lifetime of 13.5 years, consistent with the estimate of *Montzka et al.* [1993]. Experiments E2, E4 and E5 all yield results in agreement with observed column densities in the northern hemisphere.

2.5 Conclusions

CHF₂Cl column retrievals from TMF indicate an exponential increase rate of (6.7±0.5)% yr-1 from October, 1985 until July, 1990. Using McMurdo Sound measurements, we calculate a south-north interhemispheric ratio of (0.86±0.08). The method of ratioing the CHF₂Cl and N₂O columns has been shown to be useful for comparing columns of tropospheric gases at different locations. Model results demonstrate that the difference between the modeled and observed CHF₂Cl column densities can be attributed to uncertainty either in the emissions or in the OH concentration. Reduction of the uncertainty in the CHF₂Cl emissions may allow the use of CHF₂Cl observations in adequately constraining the OH field.

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Table 2.1: Error sources for a single spectrum

	CH	N ₂ O	
	ATMOS Table Mtn	MarkIV McMurdo	ATMOS Table Mtn
Random Errors (%)			
Temperature/Pressure	3	3	2
Signal-to-noise	1	1	1
Assumed VMR profile	4	5	2
Zero transmission offset	1	1	1
Interfering absorptions	6	2	2
Total Random Error	8	6	4
Systematic Errors (%)			
Line strengths	10	10	5
Spectral fitting	3.5	3.5	3.5
Total Systematic Error	13.5	13.5	8.5

Table 2.2: Model calculated CHF₂Cl average concentrations at the end of 1992

Experiment	E1	E2	E3	E4	E5
Change in Global OH			+20%	+20%	+40%
Change in CHF ₂ Cl		-10%		-5%	
Global (pptv)	115	104	107	102	101
S. hemisphere (pptv)	105	98	100	95	94
N. hemisphere (pptv)	123	111	115	109	108
Global troposphere	120	108	111	106	105
Global stratosphere	98	88	90	86	85
Rate of increase (b)	6.6	5.9	6.0	5.7	5.6
CHF ₂ Cl lifetime	19	19	15.5	15.5	13.5

⁽a) As compared to experiment E1. Reduction in CHF₂Cl emissions is after 1970.

⁽b) Average global rate of increase is calculated from 1982 to 1992.

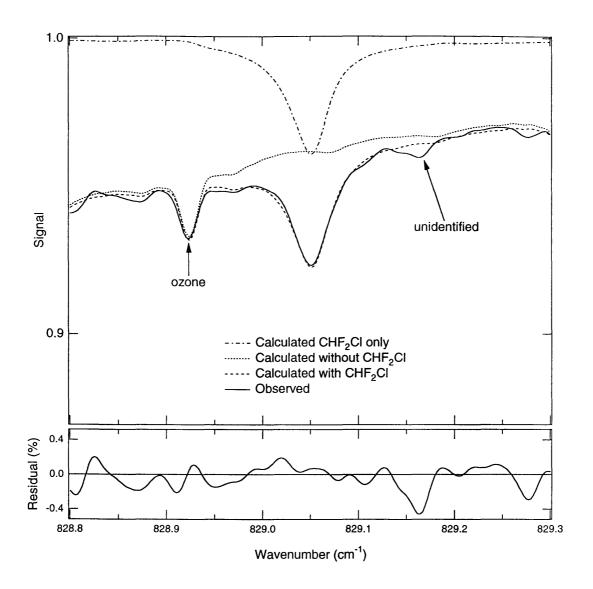


Figure 2.1: Typical ATMOS observed and calculated CHF₂Cl spectra.

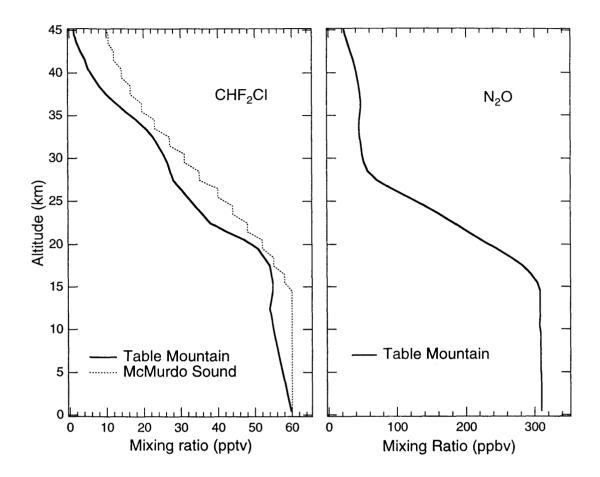


Figure 2.2: Assumed CHF₂Cl and N₂O mixing ratio profiles.

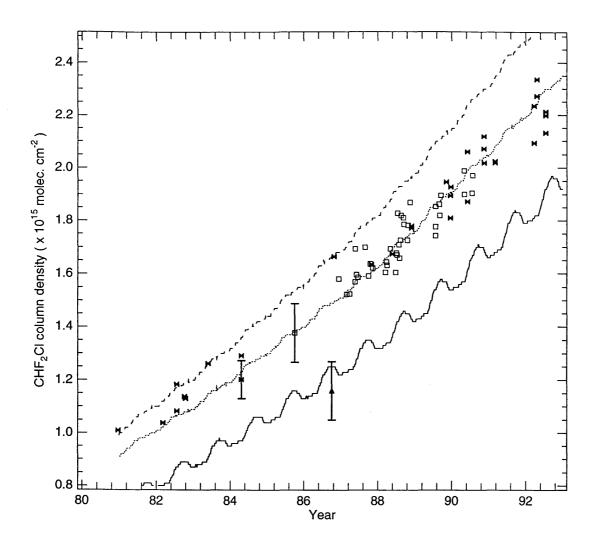


Figure 2.3: CHF₂Cl column retrievals (scaled to sea-level) and results of model simulations.

Observations: □ Table Mountain (this work), ▲ McMurdo Sound (this work), ▶ Kitt Peak

(Zander et al., 1993). Modelling experiments: - - -E1 North, E5 North, E5 South. See

Table 1 for random and systematic errors for Table Mountain and McMurdo measurements. The datum for MarkIV-McMurdo Sound is the average of 42 measurements taken in September and October, 1986; the error bar represents their 9% standard deviation. Error bars for other data sets indicate the average fractional random error. For clarity, only modelling experiments E1 (north) and E5 (north) are superimposed on the northern hemisphere results. These two experiments represent the range of results, however, experiments E2 and E4 gave results similar to E5. For the southern hemisphere, only experiment E5 is shown, although E2 and E4 gave similar results.

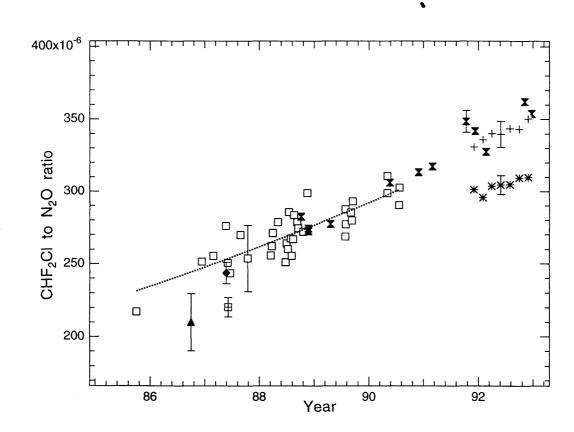


Figure 2.4: CHF₂CI/N₂O column and in-situ ratios. Column measurements (this work): ☐ ATMOS, Table Mountain, CA, ▲ Mark IV, McMurdo Sound, Antarctica. Ground-level in-situ measurements (Montzka et al.[1993]): ▼ Niwot Ridge, CO. SAGA cruises: ◆ N. hemisphere, ⊞ S. hemisphere. Weighted averages: + N. hemisphere, ℋ S. hemisphere. Error bars indicate the average fractional random error for their respective data sets. The error for the in situ data of Montzka et al. includes an assumed 2% random error for the N₂O concentration. Line fitted for ATMOS/TMF measurements only.

Chapter III

Heavy ozone enrichments from ATMOS infrared solar spectra

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Abstract

Vertical enrichment profiles of stratospheric $^{16}O^{16}O^{18}O$ and $^{16}O^{18}O^{16}O$ have been derived from space-based solar occultation spectra recorded at 0.01 cm $^{-1}$ resolution with the ATMOS (Atmospheric Trace MOlecule Spectroscopy) Fourier-transform infrared (FTIR) spectrometer. The observations, made during the Spacelab 3 and ATLAS-1, -2, and -3 shuttle missions, cover polar, mid-latitude and tropical regions. Between 26 to 2.6 mb inclusive (\approx 25 to 41 km), average enrichments, weighted by molecular $^{48}O_3$ density, of (15±6)% were found for $^{16}O^{18}O$, (10±7)% for $^{16}O^{18}O^{16}O$, and (13±5)% for $^{50}O_3$ (1 σ standard deviation). Enrichments increased slightly with altitude, however no latitudinal variability was apparent. From a series of ground-based measurements by the ATMOS instrument at Table Mountain, California (34.4°N), an average total column $^{16}O^{16}O^{18}O$ enrichment of (17±4)% (1 σ standard deviation) was determined, with no significant seasonal variation discerned. Possible biases in the spectral intensities that affect the determination of absolute enrichments are discussed. However, any corrections to the these intensities would probably involve a scaling of the enrichments, and would not affect the observed lack of latitudinal and seasonal variability.

3.1 Introduction

Stratospheric enrichment of ⁵⁰O₃ was first reported by *Mauersberger* [1981], who found enrichments ranging from 0% to 40% using a balloon-borne mass spectrometer. (For the purposes of this paper, % enrichment = $[R_{\rm obs}/R_{\rm std}$ -1] x 100, where $R_{\rm obs}$ is the observed abundance ratio of the heavy isotopomer to the regular isotopomer, and $R_{\rm std}$ is the standard ratio. For $^{50}\text{O}_3$ enrichment, $R_{\text{std}} = 5.97 \times 10^{-3}$, or three times the natural abundance ratio of ¹⁸O to ¹⁶O, ignoring a very small abundance of ¹⁶O¹⁷O¹⁷O. See IUPAC, [1983]). Further stratospheric enrichment of ⁵⁰O₃ have been reported based on mass spectrometry [Mauersberger, 1987], far-infrared emission spectroscopy [Abbas et al., 1987; Carli and Park, 1988], and cryogenic grab-sampling followed by massspectrometry [Schueler et al., 1990]. These measurements have sometimes shown little consistency with each other. Figure 3.1 summarizes previous measurements of ${}^{50}\mathrm{O}_3$ enrichment profiles as well as averaged results of the analyses presented here. Column enrichments have also been determined using mid-infrared solar-absorption Fouriertransform spectrometry. Rinsland et al. [1985] determined a column enrichment of $(11\pm11)\%$ and $(5\pm7)\%$ for $^{16}O^{16}O^{18}O$ and $^{16}O^{18}O^{16}O$, respectively. Goldman et al. [1989], from two balloon-based observations, found column enrichments above 37 km of $(20\pm14)\%$ and $(40\pm18)\%$ for $^{16}O^{16}O^{18}O$, and $(16\pm8)\%$ and $(25\pm12)\%$ for $^{16}O^{18}O^{16}O$. Enrichment of ⁵⁰O₃ from about 8 to 10% in tropospheric urban air was reported by Krankowsky et al. [1995] who found no apparent variation in the enrichment with tropospheric O₂ mixing ratios

Heavy ozone enrichment has also been observed in laboratory measurements. Anderson et al. [1989] found ozone created using an electric discharge (which can produce excited states of O and O_2) enriched primarily in the asymmetric isotopomers, such as $^{16}\mathrm{O}^{18}\mathrm{O}$. Morton et al. [1990] found that enrichment can occur via the reaction of ground-state atomic and molecular oxygen, $\mathrm{O}(^3\mathrm{P}) + \mathrm{O}_2(^3\Sigma_\mathrm{g}) + \mathrm{M} \to \mathrm{O}_3 + \mathrm{M}$, the so-called Chapman formation reaction, and under these conditions, Mauersberger et al. [1993] found enrichment totally in the asymmetric isotopomers. Miller et al. [1994] proposed that the reaction of vibrationally hot O_2 ($\mathrm{v} \geq 26$) with another O_2 molecule can lead to heavy ozone enrichment via preferential potential energy curve-crossing of heteronuclear O_2 from the $\mathrm{O}_2(^1\Delta_\mathrm{g})$ to the $\mathrm{O}_2(^3\Sigma_\mathrm{g})$ state (see Valentini [1987]). However, appreciable production of such vibrationally hot O_2 was predicted to occur only above about 35 km, and the enrichment produced about 5% at 40 km.

While Miller et al. may have provided an explanation for some of the enrichment in the mid-stratosphere, the bulk of the enrichment lacks a generally accepted theoretical explanation. Indeed, detailed statistical mechanical analyses by *Kaye and Strobel* [1983] and *Kaye* [1986] predicted a slight *depletion* of ⁵⁰O₃ under stratospheric conditions. This discrepancy between observation and theory indicates that current understanding of ozone formation is incomplete (see, for example, *Anderson et al.* [1992]). However, a promising advance in finding an enrichment mechanism is the recent discovery of several electronically excited states of ozone near the dissociation threshold [*Anderson and Mauersberger*, 1995]. It may be possible that the formation of ozone via the Chapman mechanism goes through one or more of these states whose quantum properties (such as lifetime) are dependent on the isotopic composition of the reactants.

From Figure 3.1, it is difficult to discern any consistent vertical gradients in previously published measurements, and data are lacking on either seasonal or latitudinal effects in the stratosphere. In this study, we take advantage of the high vertical and latitudinal range of the ATMOS data set from the Spacelab3 and ATLAS series missions to

determine if there are substantial vertical or latitudinal gradients. We also evaluate the ¹⁶O¹⁶O¹⁸O column enrichments from ground-based spectra from Table Mountain Facility (TMF), Wrightwood, CA, (34.4°N, 117.7°W, 2.2 km altitude) to determine if there is a seasonal variation.

3.2 Data acquisition

Information about the ATMOS instrument and its use on the shuttle can be found in *Gunson et al.* [1995]. The spectral filters used for analyses described here, filters 1, 9 and 12, had ranges of 650-1100 cm⁻¹, 650-2450 cm⁻¹, and 625-1400 cm⁻¹ respectively. The number of vertical profiles used for this study were 4 from Spacelab 3, 39 from ATLAS-1, 29 from ATLAS-2 and 87 from ATLAS-3. Using the ATMOS instrument from Table Mountain Facility, ground-based total column measurements were made on 48 separate days from October, 1985 through July, 1990, usually at solar zenith angles corresponding to about 1, 5 and 10 airmasses. As the ATMOS instrument has a very fast response time (2.2 seconds per scan), between 8 and 41 successive, double-sided interferograms were able to be taken within a narrow range of solar zenith angles, which, after transformation and averaging, produced spectra with a signal-to-noise ratio of at least 280 to 1. Only data taken in the morning from TMF were analyzed, as tropospheric smog from nearby Los Angeles could be advected over the intervening mountains and over the site in the afternoon [see *McDermid and Walsh*, 1991].

3.3 Data analysis

ATMOS analysis techniques for space-based observations have been described by Norton and Rinsland [1991]. (See also Gunson et al. [1995] on the analyses of $^{48}O_3$ from ATMOS space-based observations.) The narrow, unresolved v_1 band Q-branch centered at 1090.35 cm⁻¹ was analyzed for $^{16}O^{16}O^{18}O$, as were several lines of the v_3 band P-branch for $^{16}O^{18}O^{16}O$ (see Figure 2.2 and Table 2.1). Interfering $^{48}O_3$ lines were fitted

before final fitting of ¹⁶O¹⁸O¹⁶O or ¹⁶O¹⁶O¹⁸O was attempted. It was not possible to use all of the ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O features used by *Rinsland et al.* [1985] and *Goldman et al.* [1989] because, at the spectral resolution of the ATMOS instrument, there was too much interference by other molecules to successfully fit some of the heavy ozone lines. Analyses of enrichments were made in between tangent pressures of 2.6 and 26 mb because only in this region could suitable lines be found that simultaneously had reasonable signal above the noise, were unsaturated and reasonably free from interference by other molecules. Spectral parameters for all lines were from the ATMOS linelist [*Brown et al.*, 1995], which currently incorporates the ozone parameters given by *Flaud et al.* [1990]. Error sources for stratospheric measurements are discussed by *Gunson et al.* [1995], although we note here that the systematic error caused by uncertainties in the line intensities are 4% for ⁴⁸O₃, 20% for ¹⁶O¹⁶O¹⁸O and 10% for ¹⁶O¹⁸O¹⁶O.

The analysis procedure for the TMF ground-based retrievals differed in that an assumed vertical ozone mixing ratio profile was scaled by a single multiplicative factor until a best fit was obtained between observed and calculated spectra. Table 3.2 describes the lines used for TMF retrievals. Assumed vertical ozone profiles were created using monthly-averaged profiles from the JPL Lidar on TMF [McDermid, 1993]. Vertical temperature and pressure profiles were adapted from daily National Meteorological Center data and merged with the U.S. Standard Atmosphere [1976] for higher altitudes. These profiles were then adjusted so that the atmosphere was in hydrostatic equilibrium. To achieve better consistency in the retrievals among the different airmasses sampled within a day, the height registration of the assumed ozone profiles was shifted up or down 0-3 km. In a similar fashion to the spaced-based measurements, neighboring lines of $^{48}O_3$, CO_2 and H_2O were fitted before final fitting of the $^{16}O^{16}O^{18}O$ absorption was attempted. Measurement of column $^{48}O_3$ by the ATMOS instrument at TMF was previously reported

by Gunson and Irion [1991], however, as some of the ${}^{48}O_3$ lines used in that study may have been saturated at high airmasses, the ${}^{48}O_3$ columns have been re-analyzed for this study. The ${}^{16}O^{18}O^{16}O$ column from TMF was not retrieved because interference by neighboring ${}^{48}O_3$, H_2O and CO_2 lines was too large to achieve acceptable fits. Error sources for the TMF retrievals are listed in Table 3.3.

It should be noted that in the determination of the ¹⁶O¹⁶O¹⁸O line intensities by Camy-Peyret et al. [1986], and incorporated into the ATMOS spectral linelist [Brown et al., 1995], the concentration of ¹⁶O¹⁶O¹⁸O in the sample cell was not directly measured. Rather, the line intensities of ¹⁶O¹⁸O¹⁶O were derived from theory, and the ¹⁶O¹⁸O¹⁶O concentration in the sample cell calculated from measurement of the spectral lines. The ¹⁶O¹⁶O¹⁸O concentration was then assumed to be twice that of ¹⁶O¹⁸O¹⁶O (see Flaud et al., [1986]). This makes an implicit assumption that any enrichment of ¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O would be equal. Camy-Peyret et al. produced their ozone sample using an electric discharge in a liquid-N₂ cooled vessel. However, in subsequent production of heavy ozone by Anderson et al. [1989], also using an electric discharge, but at room temperature, the ¹⁶O¹⁶O¹⁸O produced carried about twice the enrichment of ¹⁶O¹⁸O¹⁶O. It is thus possible that the heavy ozone sample used by Camy-Peyret et al. had a ¹⁶O¹⁸O to ¹⁶O¹⁸O¹⁶O abundance ratio greater than 2:1, so the line intensities derived for ¹⁶O¹⁶O¹⁸O may have been systematically over-estimated, resulting in an under-estimated ¹⁶O¹⁶O¹⁸O abundance when these parameters are used for stratospheric spectra. (We note, however, that the effect of the different reaction temperatures of the Anderson et al. and Camy-Peyret et al. experiments on the enrichment partitioning between ¹⁶O¹⁸O and $^{16}\mathrm{O}^{18}\mathrm{O}^{16}\mathrm{O}$ is not known.) If we assume that the $^{50}\mathrm{O}_3$ enrichment in the Camy-Peyret et al. experiment was 4% (in the upper range of the enrichments found by Heidenreich and Thiemens [1983] under similar conditions) and the enriched ¹⁶O¹⁶O¹⁸O⁻¹⁶O¹⁸O¹⁶O

partitioning of the was 4:1 (as in *Anderson et al.* [1989]), then the error in the ¹⁶O¹⁶O¹⁸O line intensities caused by assuming a 2:1 partitioning is a relatively small 2%.

3.4 Results

Figure 3.3 illustrates vertical enrichment profiles for ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O, averaged within latitude bands covered by ATMOS over four shuttle missions. At least three measurements were used to calculate each datum point. The enrichment profiles tend to increase slightly with altitude, and while the standard deviations can be quite high, there does not appear to be any systematic latitudinal variations. Figure 3.1 illustrates previously published ⁵⁰O₃ enrichment measurements and the average ATMOS ⁵⁰O₃ enrichment profile (assumed the sum of one-third the ¹⁶O¹⁸O¹⁶O enrichment and two-thirds the ¹⁶O¹⁶O¹⁸O enrichment). The ATMOS average profile was determined by averaging results across constant pressure surfaces and assigning an approximate altitude. Weighted by the ⁴⁸O₃ density, the globally averaged ⁵⁰O₃ between 2.6 and 26 mb inclusive is (13±5) %, while that for ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁶O are (15±6)% and (10±7)%, respectively (1σ standard deviation).

Figure 3.4 shows the $^{16}O^{16}O^{16}O$ columns measured above TMF The random error for the $^{16}O^{16}O^{18}O$ enrichment data points (lower panel) is, on average, 9%. While the ozone column varies substantially, no seasonal variation in the enrichment can be discerned. This is in qualitative agreement with a lack of variability in tropospheric $^{50}O_3$ enrichments found by *Krankowsky et al.* [1995]. Ignoring systematic error, the average column enrichment is $(17\pm4)\%$ (1 σ standard deviation), in good agreement with the $^{16}O^{16}O^{18}O$ enrichment derived from the ATMOS stratospheric profiles.

3.5 Conclusions

We have analyzed ATMOS stratospheric spectra for the enrichments of ¹⁶O¹⁸O and ¹⁶O¹⁸O, and ground-based spectra for the column enrichment of ¹⁶O¹⁶O¹⁸O. We

can find no discernable seasonal variability in the ¹⁶O¹⁶O¹⁸O column enrichment, nor can any latitudinal variation be detected in the vertical enrichment profiles. There appears to be a slight increase with altitude, possibly an effect of increasing temperatures and/or decreasing pressure. These observations suggest the enrichment of heavy ozone in the stratosphere is relatively constant, and only weakly regulated by temperature and pressure.

Although previous studies have not ruled out other enrichment processes, laboratory measurements have shown the most significant enrichment in the Chapman reaction, with O and O2 in the electronic ground state. Noting that the Chapman mechanism has been shown to produce no enrichment for the ¹⁶O¹⁸O¹⁶O isotopomer, our determination of a ¹⁶O¹⁸O¹⁶O enrichment of (10±7)% (1σ standard deviation) suggests that the ¹⁶O¹⁸O¹⁶O line intensities may be biased too low by 10% with respect to the line intensities of ⁴⁸O₃. This is within the expected systematic error, although we note it may be possible that other unknown processes are operating to enrich stratospheric ¹⁶O¹⁸O¹⁶O, much as ozone produced in an electric discharge becomes enriched in ¹⁶O¹⁸O¹⁶O. Furthermore, since the ¹⁶O¹⁶O¹⁸O line intensities derived by *Camy-Peyret et al.* [1986] were determined making use of the ¹⁶O¹⁸O¹⁶O line intensities, the ¹⁶O¹⁶O¹⁸O line intensities may also be too low. Laboratory measurement of the ¹⁶O¹⁶O¹⁸O and ¹⁶O¹⁸O¹⁸O spectral line intensities are necessary for verification of the absolute stratospheric enrichments described here. Since Flaud et al. [1986] and Camy-Peyret et al. [1986] reported good model fittings to their laboratory spectra, such laboratory measurements would probably produce a constant, corrective scaling of the line intensities (and thus the enrichments), but would not affect the precision of the results described here.

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Table 3.1: Spectral intervals used for heavy ozone profile analyses from space

Molecule	Window center frequency (cm ⁻¹)	Window width (cm ⁻¹)	Line center(s) (cm ⁻¹)	Line intensity (cm molecule ⁻¹) at 296 K	Ground state energy (cm ⁻¹)	Temperature sensitivity of intensity (%/K at 220K)
¹⁶ O ¹⁶ O ¹⁸ O	1090.445	0.55	unresolved	$2.14x10^{-23}$ (a)	203 (b)	-0.1
	975.27	0.12	975.2503 975.2838	1.69x10 ⁻²⁴ 2.30x10 ⁻²⁴	546 495	1.0 0.7
	981.715	0.14	981.6756 981.7107	3.54x10 ⁻²⁴ 4.46x10 ⁻²⁴	363 328	0.4 0.3
¹⁶ O ₁₈ O ₁₆ O	985.09	0.44	984.9062 984.9790 985.0321 985.1031 985.1578 985.2171	4.32x10 ⁻²³ 5.48x10 ⁻²³ 2.20x10 ⁻²³ 4.78x10 ⁻²³ 3.32x10 ⁻²³ 5.10x10 ⁻²³	298 264 395 279 334 268	0.2 0.1 0.5 0.1 0.3 0.1
	990.422	0.14	990.3889 990.3895 990.3918	$1.55 \times 10^{-23} \\ 1.74 \times 10^{-23} \\ 1.74 \times 10^{-23}$	593 567 567	1.3 1.3 1.1

Notes:

⁽a) Sum of intensities between 1090.1 and 1090.6 cm⁻¹.

⁽b) Average weighted by the intensity of the individual lines.

Spectral intervals used for column density analyses from Table **Table 3.2:** Mountain

Molecule	Line Center (a) (cm-1)	Line Intensity (x 10 ²² cm molec. ⁻¹)	Ground State Energy (cm ⁻¹)	Temperature Sensitivity of Intensity %/K at 220K	Airmass Range
	1095.1008	5.19	310.3	0.2	< 1.5
	1114.8233	1.03	77.1	-0.5	3-10
1123.4234	6.85	120.3	-0.4	<3	
¹⁶ O ¹⁶ O ¹⁶ O	1126.2511	2.49	42.9	-0.5	<5
1163.42	1140.9448	1.02	190.2	-0.1	3-10
	1163.4222	2.57	253.9	0.04	<5
	1176.1047	1.41	353.3	0.3	3-10
¹⁶ O ¹⁶ O ¹⁸ O	(b)	2.14(c)	203 (d)	-0.1(d)	all

Notes:

Notes:
(a) Interval width for all ¹⁶O¹⁶O¹⁶O windows was 0.16 cm⁻¹. The window for ¹⁶O¹⁶O¹⁸O was 1090.35 cm⁻¹ with a width of 0.5 cm⁻¹.
(b) Several unresolved lines between 1090.1 and 1090.6 cm⁻¹
(c) Sum of intensities between 1090.1 and 1090.6 cm⁻¹
(d) Average weighted by the intensities of the individual lines.

Table 3.3: Error sources and resulting percentage uncertainties in retrieved column abundances from Table Mountain for a single spectrum

Random Error Source	¹⁶ O ₃	¹⁶ O ¹⁶ O ¹⁸ O
Finite signal-to-noise Error in 100% transmission level Error in pressure-temperature profile Error in fitting interfering lines Error in assumed O ₃ vertical distribution	0.1 -1 2 3 1 5	0.1 - 2.6 2 5 3 5
Random error root sum of squares	6.2 - 6.3	7.9 - 8.4
Systematic Error Source		
Uncertainties in line intensity parameters Retrieval algorithm ¹	4 <5	20 <5
Total systematic error	<7	<21

^{1.} This error tends to cancel out when retrieved mixing ratios are used to determine the enrichment.

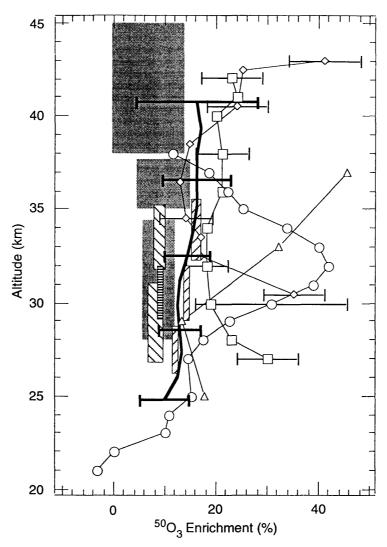


Figure 3.1: Previously reported and globally averaged ATMOS space-borne measurements of ⁵⁰O₃ enrichment profiles. The error bars on the ATMOS measurements are the 1σ standard deviations, and do not include systematic error. ATMOS altitudes are approximate and ATMOS ⁵⁰O₃ enrichments are assumed to be the sum of two-thirds the ¹⁶O¹⁶O¹⁸O enrichment and one-third the ¹⁶O¹⁸O¹⁶O enrichment. For clarity, error bars for many of the data points have been omitted, and altitudes for *Mauersberger* [1987], flight a, have been shifted upwards by 0.5 km. Data from *Carli and Park* [1988] are the range of their measurements. *Mauersberger* [1981]: ○; *Mauersberger* [1987], flight a: ◇ , flight b: □; *Abbas et al.* [1987]: △; *Carli and Park* [1988]: ■ ; *Scheueler et al.* [1990] flight I: □ , flight II: □ , flight III: □ , fligh

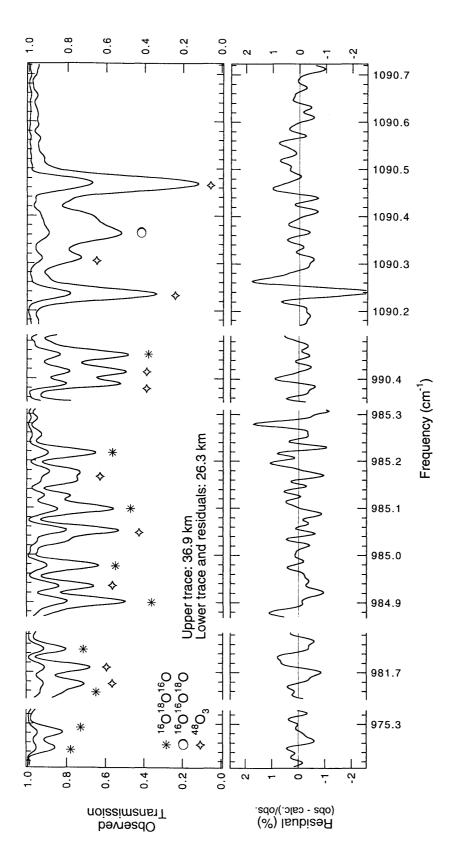
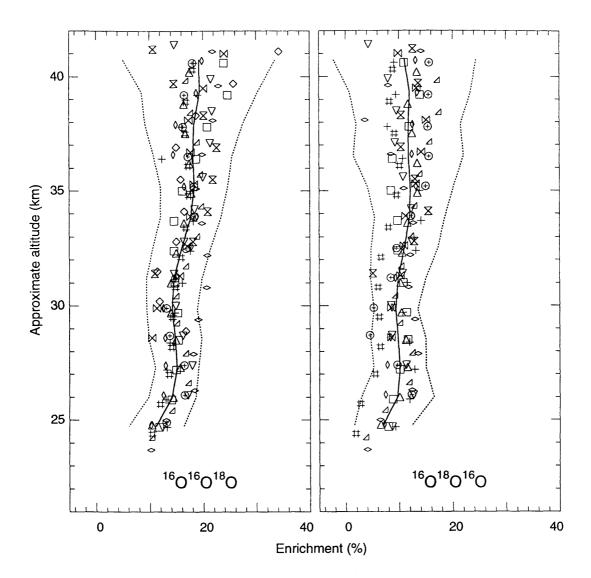


Figure 3.2: Spectral intervals used for space-based profile retrievals of ¹⁶O ¹⁸O and ¹⁶O ¹⁸O on The residual in the bottom panel was calculated as the fractional difference between the observed and the (not shown) calculated spectrum at 26.3 km.



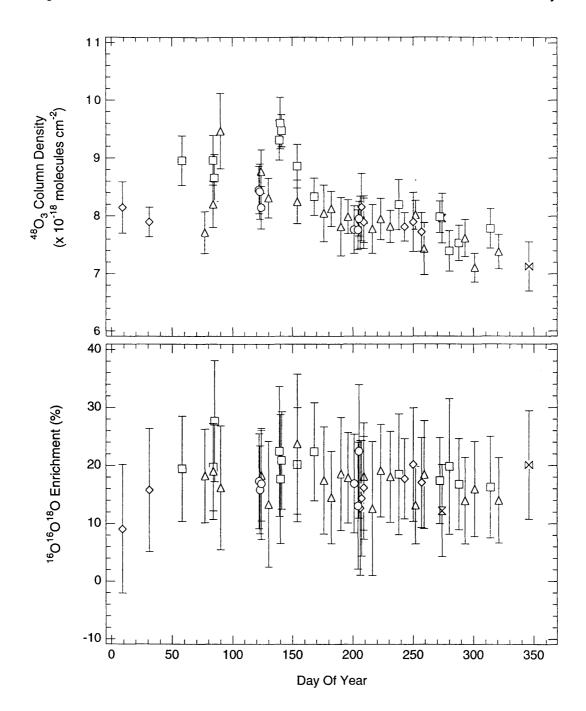


Figure 3.4: $^{48}O_3$ columns and column $^{16}O^{16}O^{18}O$ enrichments from Table Mountain. The error bars do not include systematic error (see text and Table 3.3). \times 1985; \bowtie 1986; \square 1987; \triangle 1988; \diamondsuit 1989; \bigcirc 1990.

Chapter IV

Stratospheric observations of CH₃D and HDO from ATMOS infrared solar spectra: Enrichments of deuterium in methane and implications for HD

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Abstract

Stratospheric mixing ratios of CH₃D from 100 mb to 17mb (≈15 to 28 km) and HDO from 100 mb to 10 mb (≈15 to 32 km) have been inferred from high resolution solar occultation infrared spectra from the Atmospheric Trace MOlecule Spectroscopy (ATMOS) Fouriertransform interferometer. The spectra, taken on board the Space Shuttle during the Spacelab 3 and ATLAS-1, -2, and -3 missions, extend in latitude from 70°S to 65°N. We find CH₃D entering the stratosphere at an average mixing ratio of (9.9±0.8)x10⁻¹⁰ with a D/H ratio in methane $(7.1\pm7.4)\%$ less than that in Standard Mean Ocean Water (SMOW) $(1\sigma \text{ combined precision and systematic error})$. In the mid to lower stratosphere, the average lifetime of CH₃D is found to be (1.19±0.02) times that of CH₄, resulting in an increasing D/H ratio in methane as air "ages" and the methane mixing ratio decreases. We find an average of (1.0±0.1) molecules of stratospheric HDO are produced for each CH₃D destroyed (1 σ combined precision and systematic error), indicating that the rate of HDO production is approximately equal to the rate of CH₃D destruction. Assuming negligible amounts of deuterium in species other than HDO, CH₃D and HD, this limits the possible change in the stratospheric HD mixing ratio below about 10mb to be ±0.1 molecules HD created per molecule CH₃D destroyed.

4.1 Introduction

In the stratosphere, H₂O is created primarily by oxidation of CH₄ and H₂ by OH, Cl and O(¹D). Likewise, oxidation of CH₃D and HD creates HDO, but these deuterated species show a sufficiently different reactivity with surrounding chemical species so that their measurement can provide additional insight into the stratospheric hydrogen budget. In understanding the deuterium budget of the stratosphere, and relating it to the hydrogen budget, two questions arise. First, what is the lifetime of CH₃D compared to that of CH₄, or put another way, how does the D/H ratio in methane vary with the CH₄ mixing ratio? Second, is the rate of stratospheric HDO production balanced by that of CH₃D oxidation? If not, could there be a net production or destruction of HD, and possibly a change in the D/H ratio of molecular hydrogen, due to a difference in the rates between CH₃D oxidation to HD and HD oxidation to HDO? (Discussion of the D/H ratio in stratospheric water is in an accompanying paper by *Moyer et al.* [1995].)

We address these questions using spectroscopic measurements of stratospheric CH₃D and HDO mixing ratios using data from the ATMOS instrument. The ATMOS instrument, described in detail by *Farmer et al.* [1987], is a Fourier-transform infrared (FTIR) interferometer that gathered spectral absorption measurements at a resolution of 0.01 cm⁻¹ from solar occultations on board four Space Shuttle missions (see *Gunson et al.* [1995]). A previous report of HDO and CH₃D results from ATMOS was made by *Rinsland et al.* [1991] with data from the Spacelab 3 mission; however, their report did not relate changes in the CH₃D mixing ratio to those in HDO. With the combined Spacelab 3 and ATLAS-1, -2, and -3 missions, a much broader latitudinal coverage and many more vertical mixing ratio profiles were obtained, allowing characterization of the CH₃D and HDO budget on a more global basis. An important addition to previous reports is measurement of CH₃D inside the Arctic and Antarctic vortices. Above 28 km, the

molecular density of CH₃D is normally too low to gain an adequate spectral absorption signal with the ATMOS instrument. However, the descent of upper stratospheric air to altitudes below 28 km inside the vortices [Abrams et al., 1995a,b] allows measurement of CH₃D in "old" air that would be otherwise impossible to measure. As stratospheric dehydration is not the focus of this letter, we do not report measurements of HDO inside the polar vortices. Analyses are continuing on these vortex HDO measurements, and they will be the focus of an upcoming paper.

In this paper, the delta notation is used to describe isotopic fractionation as the difference in parts per thousand of an isotopically labelled species with respect to a standard, e. g.,

$$\delta D_{\text{sample}} (\%o) = 1000 \times \frac{(D/H)_{\text{sample}} - (D/H)_{\text{standard}}}{(D/H)_{\text{standard}}}$$
(4.1)

Standard Mean Ocean Water (SMOW) is used for the standard D/H ratio (see IUPAC [1983]) and we use the recommended value of (155.76±0.05)x10⁻⁶ [Hagemann et al., 1970].

4.2 Observations and data analysis

Information about ATMOS on the shuttle missions, and discussion of the mixing ratio retrieval process, can be found in *Gunson et al.* [1995]. For the results described here, spectral lines of the v_2 band of HDO and the v_1 band of CH₃D were analyzed, and spectral intervals used for these analyses are described in Table 4.1. HDO is best observed in filters 2 (1100-2000 cm⁻¹) and 9 (600-2450 cm⁻¹), while CH₃D could only be observed in filter 3 (1580 - 3400 cm⁻¹). However, CH₄ can be analyzed in all of these filters and, with a correlation of CH₃D to CH₄ in filter 3, we are able infer the mixing ratio of CH₃D in filters 2 and 9 from CH₄ mixing ratios. At polar latitudes, determination as to whether a

CH₃D measurement was inside or outside the vortex was based on visual inspection of colocated mixing ratio profiles of CH₄ and N₂O, as these gases show markedly smaller mixing ratios inside the vortex than outside at similar altitudes (see *Abrams et al.* [1995a,b]). "Vortex edge" observations were not used. CH₃D results are presented from tangent pressures of 100mb to 17mb, and HDO results are from 100mb to 10mb. Before final analyses, data were filtered by rejecting any observations with an estimated random error greater than 30% for CH₃D and HDO, or 10% for CH₄. This step eliminated about 16% of the filter 3 CH₃D data and about 19% of the filter 2 and 9 HDO data. Systematic biases, mostly from errors in line intensities, are estimated to be 7% for CH₃D, 6% for HDO, and 5% for CH₄ (see *Gunson et al.* [1995]).

4.3 CH₃D measurement and the D/H ratio in methane

Rate constants for reactions of OH or Cl with CH₃D are lower than those for CH₄ [DeMore et al., 1994; Wallington and Hurley, 1992], but any isotope effect between the rate constants of the excited $O(^{1}D) + CH_{3}D$ and that of $O(^{1}D) + CH_{4}$ is expected to be relatively minor [Kaye, 1987]. To estimate these combined effects on the D/H ratio of methane, we begin by assuming that in the mid to lower stratosphere CH₄ and CH₃D are only destroyed by OH, Cl, and $O(^{1}D)$ oxidation, and photolysis is negligible. The oxidation of CH₃D and CH₄ can then be described by:

$$\frac{1}{[CH_3D]} \frac{d[CH_3D]}{dt} = -k_{OH}^*[OH] - k_{CI}^*[CI] - k_{O(^1D)}^*[O(^1D)]$$
 (4.2)

$$\frac{1}{[CH_4]} \frac{d[CH_4]}{dt} = -k_{OH}[OH] - k_{CI}[CI] - k_{O(^1D)}[O(^1D)]$$
(4.3)

Let $\gamma_i(T)$ equal the ratios of the rate constants. That is, $\gamma_{OH}(T) = k_{OH}^*/k_{OH}$, $\gamma_{CI}(T) = k_{CI}^*/k_{CI}$, and $\gamma_{O(^1D)}(T) = k_{O(^1D)}^*/k_{O(^1D)}$, where T is temperature. Subtraction of Equation (4.3) from (4.2) gives:

$$\frac{1}{[CH_3D]} \frac{d[CH_3D]}{dt} - \frac{1}{[CH_4]} \frac{d[CH_4]}{dt} = -(\gamma_{OH} - 1)k_{OH}[OH] - (\gamma_{CI} - 1)k_{CI}[CI] - (\gamma_{O(^1D)} - 1)k_{O(^1D)}[O(^1D)]$$
(4.4)

But we note that

$$k_{OH}[OH] = \frac{-1}{[CH_4]} \frac{d[CH_4]}{dt} f_{OH}(z)$$
 (4.5)

where $f_{OH}(z)$ is the fraction of all CH_4 destroyed at altitude z that is destroyed by OH. Similar definitions for $f_{Cl}(z)$ and $f_{O(^1D)}(z)$, substitution in (4.4) for k_{OH} , k_{Cl} , and $k_{O(^1D)}$ (noting that they sum to unity), and rearrangement yield:

$$\frac{1}{[CH_3D]} \frac{d[CH_3D]}{dt} = \left(\gamma_{OH} f_{OH} + \gamma_{CI} f_{CI} + \gamma_{O(^1D)} f_{O(^1D)} \right) \frac{1}{[CH_4]} \frac{d[CH_4]}{dt}$$
(4.6)

or:

$$\frac{1}{[CH_3D]} \frac{d[CH_3D]}{dt} = \kappa(T,z) \frac{1}{[CH_4]} \frac{d[CH_4]}{dt}$$
 (4.7)

where $\kappa(T,z)$ is the term inside the parentheses of (4.6). Note that the left-hand side of Equation (4.7) is the time constant for destruction of CH_3D , or equivalently, its inverse lifetime. Thus, for a $\kappa(T,z) < 1$, the lifetime of CH_3D is longer than that of CH_4 . We estimate an "average" κ in the mid to lower stratosphere by assuming average fractions for destruction (i.e., f_{OH} , f_{Cl} , and $f_{O(^1D)}$), and assuming that the ratios of the rate constants, γ ,

are constant with the temperature range in the mid to lower stratosphere and that mixing effects are minor. Integrating (4.6), assigning boundary conditions, and dividing by total molecular concentration to get mixing ratios (χ) yield:

$$\ln \chi(CH_3D) = \ln \chi_0(CH_3D) + \kappa_{av} \ln \left[\frac{\chi(CH_4)}{\chi_0(CH_4)} \right]$$
 (4.8)

or equivalently,

$$\frac{\chi(\text{CH}_{3}\text{D})}{\chi_{0}(\text{CH}_{3}\text{D})} = \left(\frac{\chi(\text{CH}_{4})}{\chi_{0}(\text{CH}_{4})}\right)^{\kappa_{av}}$$
(4.9)

where $\chi_0(\text{CH}_3\text{D})$ and $\chi_0(\text{CH}_4)$ are the initial mixing ratios of these gases as they enter the stratosphere. A rough estimate of κ_{av} can be made if we set γ_{OH} =0.67 (averaged from 190K to 250K using the rate constants reported by *DeMore et al.* [1994]), γ_{Cl} = 0.735 (from the evaluation by *Wallington and Hurley* [1992] at 295K), and $\gamma_{\text{O(}^1\text{D})}$ ≈1 (see *Kaye* [1987]). Model calculations by one of us (RJS) indicate that between about 16 km and 30 km, about 50% of the destruction of CH₄ is by OH, 29% by Cl and 21% by O(^1D). Substituting these laboratory data and model results, κ_{av} ≈0.78, suggesting that the lifetime of CH₃D is significantly longer than that of CH₄, and enrichment of deuterium in methane can occur as the CH₄ mixing ratio decreases. We emphasize that this is only a rough estimation as none of the experiments measuring the kinetic rate constants of OH or Cl with CH₃D were made at the cold temperatures typical of the lower stratosphere, and we are unaware of any laboratory measurements of the O(^1D)+ CH₃D rate constant.

ATMOS measurements of the mixing ratios of CH_3D are plotted against co-located measurements of CH_4 in the upper panel of Figure 4.1. We assume that $\chi_0(CH_4) =$

(1.71 \pm 0.03)x10⁻⁶ (from global 1992 tropospheric measurements reported in *WMO* [1995], and where we have assumed a 2% error). From a best fit line of $\ln \chi(CH_3D)$ vs $\ln \chi(CH_4)$, we observe $\kappa_{av} = (0.84 \pm 0.02)$ (1 σ precision), higher than our initial estimate of 0.78. Note that the error in κ_{av} is not affected by systematic errors in the measured stratospheric mixing ratios of CH_3D and CH_4 (which instead affect the error in the intercept). On average, the stratospheric lifetime of CH_3D should be κ_{av}^{-1} (=1.19 \pm 0.02) times that of CH_4 , or about 190 years based on a stratospheric CH_4 lifetime of 160 years [*Prather and Spivakovsky*, 1990]. From the fitted line at a tropospheric CH_4 mixing ratio of (1.71 ± 0.03) x10⁻⁶, we find the mixing ratio of CH_3D entering the stratosphere to be (9.9 ± 0.8) x10⁻¹⁰ (1 σ combined precision and systematic error).

The lower panel of Figure 4.1 shows the enrichment of CH_3D with CH_4 mixing ratio, as well as the average enrichment calculated using the fitted line from the upper panel. Although there is considerable scatter in the data, it is seen that methane becomes progressively enriched in deuterium as the mixing ratio of CH_3D decreases. With a tropospheric CH_4 mixing ratio of $(1.71\pm0.03)\times10^{-6}$, the average δD in methane entering the stratosphere is $-(71\pm74)\%$ (1σ combined precision and systematic error). This is within error of the ATMOS Spacelab 3 measurements of *Rinsland et al.* [1991], who found δD in stratospheric methane near latitudes of 30°N and 49°S to be $-(49\pm44)\%$ and $+(24\pm125)\%$ respectively. This is also within error of free tropospheric measurements by *Ehhalt* [1973], who reported values of -86% and -94‰, and *Wahlen et al.* [1987], who reported $-(80\pm8)\%$.

4.4 Measurement of HDO vs CH₃D and implications for HD

In the mid to lower stratosphere, atomic H is distributed primarily among CH₄, H₂O and H₂, with negligible amounts among other species. It appears that oxidation of H₂ is roughly balanced by its production via oxidation of CH4 through a short-lived CH2O intermediate [Dessler et al, 1994; Abbas et al., 1995]. Similarly, deuterium is most likely distributed mainly among CH₃D, HDO, and HD (with HD produced by oxidation of CH₃D and destroyed by oxidation to HDO). However, in the mid to lower stratosphere, it may not necessarily be the case that the mixing ratio of HD is as weakly varying as that of H₂. As suggested by Ehhalt et al. [1989], the lower reaction rate constant of OH and HD compared to that of OH and H₂ may serve to enrich tropospheric hydrogen gas in deuterium. However, stratospheric destruction of HD is regulated not only by rates of OH attack, but also to a large extent by those of O(1D), and the rate constant of the O(1D) reaction is not expected to be significantly affected by deuterium substitution [Kaye, 1987]. (Reaction with Cl is a much more minor sink for hydrogen than for methane using the rate constants given by DeMore et al. [1994].) For production of HD, account must be taken of not only the oxidation rates of CH₃D, but also the partitioning of D in the shortlived species in the reaction pathway from CH₃D to HD (e.g. the yield of CH₂D vs CH₃, or that of CHDO vs CH₂O). In the absence of direct measurement of HD, a test for changes in the HD mixing ratio is examination of those for HDO and CH₃D. Assuming the sum of the mixing ratios of HDO, CH₃D, and HD constant, then:

$$\frac{d\chi(\text{HDO})}{d\gamma(\text{CH,D})} + \frac{d\chi(\text{HD})}{d\gamma(\text{CH,D})} = -1 \tag{4.10}$$

and any deviation of $d\chi(HDO)/d\chi(CH_3D)$ from -1 through the mid and lower stratosphere implies a changing mixing ratio of HD.

In order to compare HDO and CH₂D mixing ratios, we use the relationship in Equation 4.8 (with $\kappa = 0.84 \pm 0.02$, $\chi_0(\text{CH}_4) = (1.71 \pm 0.03) \times 10^{-6}$, and $\chi_0(\text{CH}_3\text{D}) =$ (9.9±0.8)x10⁻¹⁰) to estimate the CH₃D mixing ratio in filters 2 and 9 from measurements of CH₄. Figure 4.2 is a scatter plot of the HDO mixing ratio versus this derived CH₃D mixing ratio. We calculate $d\chi(HDO)/d\chi(CH_3D)$ by a least-squares straight line fit, but fitted only where CH_4 mixing ratios were less than 1.4×10^{-6} to avoid seasonal effects in HDO in the lower stratosphere. The calculated slope, $-(1.0\pm0.1)$ (1 σ combined precision and systematic error), indicates that HDO production is in near balance with CH₃D destruction. Large effects are still possible for HD within the error, however. As about 10% of stratospheric deuterium is in HD, small deviations in $d\chi(HDO)/d\chi(CH_3D)$ from -1 can have significant effects in the HD mixing ratio, similar to the case for H₂ (e.g., Dessler et al. [1994]). Large effects may also occur in the D/H ratio of stratospheric hydrogen, particularly if the H₂ mixing ratio does not change in the mid to lower stratosphere. While we cannot find statistically significant evidence for changes in HD, the results in this report can provide some constraints below about 10 mb. To illustrate this, we assume a constant H_2 mixing ratio of 0.5×10^{-6} and a δD in hydrogen entering the stratosphere of +(70±30)‰ [Friedman and Scholz, 1974], giving an initial HD mixing ratio of (1.67 ± 0.05) x 10^{-10} . Results here indicate that if CH₃D enters the stratosphere at a mixing ratio of 9.9×10^{-10} , then when half of the CH_3D is destroyed, the change in the HD mixing ratio is constrained to be $\pm 5 \times 10^{-11}$ and the change in δD in hydrogen can be constrained to ±320%.

4.5 Conclusions

We have analyzed ATMOS spectra for mid to lower stratospheric HDO and CH_3D mixing ratios. The average D/H ratio in methane entering the stratosphere was found to be -(71 \pm 74)‰, but as the lifetime for CH_3D is greater than that of CH_4 by a factor of (1.19 \pm 0.02), this D/H ratio increases as methane becomes oxidized. Stratospheric production of HDO is (1.0 \pm 0.1) times that of CH_3D loss, and assuming deuterated species other than HDO, CH_3D and HD are in negligible abundance, changes in HD abundance are thus constrained to be \pm 0.1 molecules HD per molecule CH_3D destroyed (1 σ combined precision and systematic error). To better understand these phenomena, research is warranted into the partitioning of deuterium in CH_3D destruction products (including a comparison of the photolysis rates of CHDO vs that of CH_2O), and direct stratospheric measurements of the HD mixing ratio.

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Table 4.1: Spectral intervals and lines used for HDO and CH₃D analyses

Spectral Interval Center (cm ⁻¹)	Interval Width (cm ⁻¹)	Altitude Range (km)	Line Center (cm ⁻¹)	Line Intensity at 296 K (cm molecule ⁻¹)	Air- broadened half-width (cm ⁻¹ /atm at 296k	Ground- state energy () (cm ⁻¹)	Temperature Sensitivity of Line Intensity at 230K (% K ⁻¹)
CH ₃ D - 2950.84 3061.56 3078.34 3098.91 HDO -	0.24 0.20 0.17 0.15	10-28 15-36 15-28 10-24	2950.8514 3061.4148 3078.3125 3078.3551 3098.8832	2.81x10 ⁻²³ 5.21x10 ⁻²³ 1.61x10 ⁻²³ 2.09x10 ⁻²³ 1.58x10 ⁻²³	0.072 0.077 0.070 0.072 0.068	266.3 89.9 217.1 184.7 346.0	0.07 -0.4 -0.07 -0.2 0.3
1408.35 1421.62 1439.93 1451.40 1469.43 1474.09 1475.62	0.25 0.40 0.32 0.34 0.23 0.31	10-34 13-42 15-40 10-29 17-40 19-40 20-33	1408.3914 1421.6073 1439.8887 1451.4597 1469.3658 1474.1110 1475.5917	9.62x10 ⁻²⁴ 1.26x10 ⁻²³ 1.54x10 ⁻²³ 1.21x10 ⁻²³ 2.53x10 ⁻²³ 1.30x10 ⁻²³ 1.37x10 ⁻²³	0.102 0.077 0.095 0.093 0.095 0.094 0.096	29.8 233.1 150.1 265.2 156.4 156.4 150.2	-0.6 0.02 -0.2 0.07 -0.2 -0.2
1479.96 1484.11 1488.16 1494.86 1497.85	0.42 0.25 0.48 0.29 0.34	10-26 15-40 20-40 15-29 15-40	1480.0941 1484.1065 1488.0252 1488.1937 1494.8598 1497.8807	5.81x10 ⁻²⁴ 2.44x10 ⁻²³ 1.09x10 ⁻²³ 1.33x10 ⁻²³ 8.34x10 ⁻²⁴ 2.08x10 ⁻²³	0.093 0.092 0.093 0.100 0.095 0.085	225.9 225.9 221.9 32.5 221.8 308.6	-0.2 -0.04 -0.04 -0.6 -0.05 0.02

Note: The temperature dependence for air-broadened half-widths is T-0.75 for CH₃D and T-0.64 for HDO. Line parameters for CH₃D are from *Rinsland et al.* [1991]. HDO line parameters are discussed in *Brown et al.* [1995].

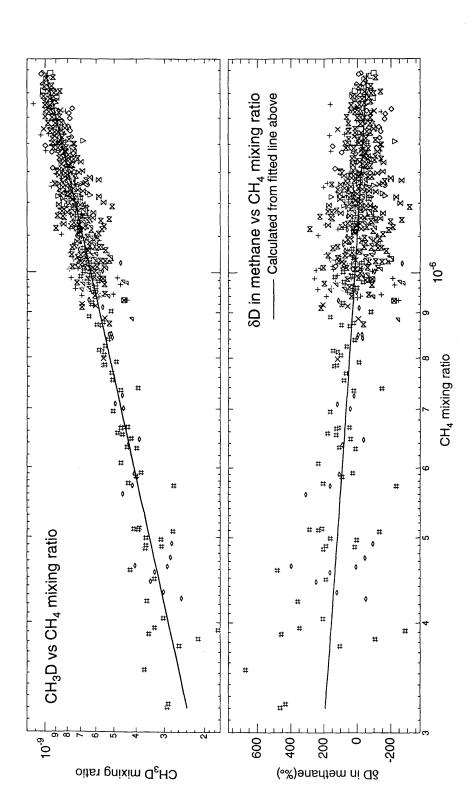


Figure 4.1: CH₃D mixing ratio vs CH₄ mixing ratio (upper panel) and δD in methane vs CH₄ mixing ratio (lower panel). Numbers in parentheses x 75°S-65°S outside vortex (86), # 75°S-65°S inside o 50°S-□ 0°-15°N (22), ◇ 15°N-30°N (58), ⋈ 30°N-50°N (216). Note: Some polar profiles had regions inside and outside the vortex. indicate the number of observations used: Spacelab 3: 8 48°S (9), × 26°N-31°N (30); ATLAS-1: + 50°S-20°S (94); ATLAS-2: 65°N-70°N inside vortex(63); ATLAS-3: △ 65°N-70°N outside vortex, vortex (86), 25°S (28),

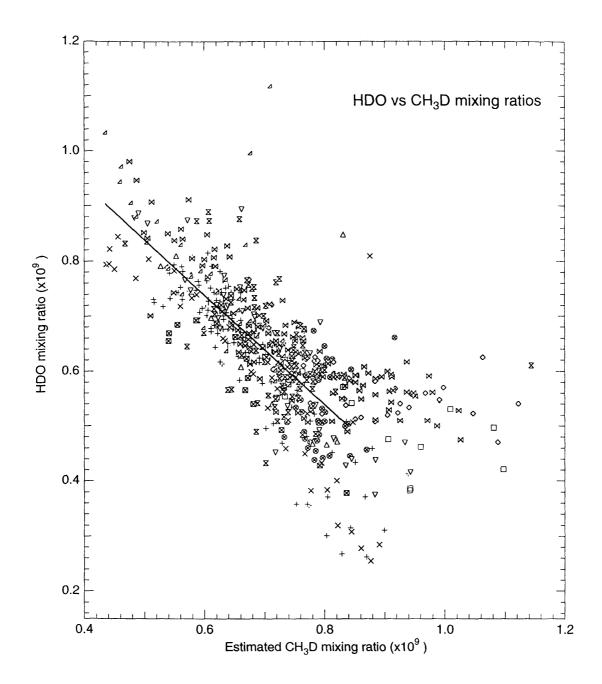


Figure 4.2: HDO mixing ratio vs CH₃D mixing ratio. The line is fitted only where the CH₄ is less than 1.4×10^{-6} . Number in parentheses indicate the number of observations used: Spacelab 3: $\approx 49^{\circ}$ S (13), $\times 26^{\circ}$ N-31°N (33); ATLAS-1: $+ 55^{\circ}$ S-30°S (118), $\approx 10^{\circ}$ S-15°N (36), $\triangle 15^{\circ}$ N-30°N (10); ATLAS-2: $\nabla 50^{\circ}$ S-30°S (40), $\triangle 65^{\circ}$ N-70°N, outside vortex (35); ATLAS-3: $\times 75^{\circ}$ S-65°S, outside vortex (82), $\square 12^{\circ}$ N (4), $\triangle 15^{\circ}$ N-30°N (25), $\bowtie 30^{\circ}$ N-50°N (122).

Appendix A

A brief note on error sources and their treatment

This appendix briefly reviews some error sources and their treatment in analyses of ATMOS spectra. Discussion of the instrumental errors of Fourier-transform interferometers is given by *Chamberlain* [1979] and *Beer* [1992]. A full description of the equations used in analyzing ATMOS spectra can be found in *Norton and Rinsland* [1991]. *Shaffer et al.* [1984] discuss systematic sources of error in solar occultation-type measurements in general, while discussions about ATMOS error determinations in particular can be found in *Abrams et al.* [1995] and *Gunson et al.* [1995].

A1.1 Spectral intensities and ground state energies

A full spectra and spectral lineshapes can be found elsewhere (e.g., Webster et al. [1988]), but a quick look at spectral line intensities may be helpful in understanding retrieval errors. The spectral intensity of a rotational-vibrational transition at frequency v is given by

$$S = S_0 \frac{Q_r(T_0)}{Q_r(T)} \frac{Q_v(T_0)}{Q_v(T)} \exp\left(\frac{hcE''}{k} \left[\frac{1}{T_0} - \frac{1}{T}\right]\right) \frac{1 - \exp(-hcv/kT)}{1 - \exp(-hcv/kT_0)}$$
(A1.1)

where S, S_0 = line intensity at temperature T and T_0 , respectively (T_0 is usually 296 K)

 Q_v , Q_R = vibrational and rotational partition functions, respectively

E'' = ground state energy (cm⁻¹)

v = frequency of the line

c =speed of light

h = Planck's constant

k = Boltzmann's constant

Absorption coefficients across a spectrum for an individual line are calculated by multiplying the line intensity by an appropriate lineshape function (e.g., a Voigt or Lorentz profile), followed by convolution with the instrumental function. See *Norton and*

Rinsland [1991]). Spectral databases, such as the ATMOS linelist [Brown et al., 1995], contain information on S_0 , E" and v for over 250,000 lines of some 50 atmospheric gases. Most of the systematic error in a mixing ratio retrieval will be caused by errors in S_0 . For example, a line intensity that is 10% too high will result in a retrieved mixing ratio that is 10% too low. Spectral intensity uncertainties for gases studies by ATMOS can range from 2-4% for minor gases such as CO_2 , H_2O or N_2O to as high as 15% for HNO₃ and 20% for CCl_4 . Brown et al. [1995] lists estimates of line intensity uncertainties for gases retrieved by ATMOS.

Errors in the temperature profile affect the calculated intensity through the partition functions and the ground-state energy, E", but at stratospheric temperatures, it is only the E" term that can lead to significant error in the line intensity. As mentioned in Section 1.1.5, ATMOS trace gas profile retrieval methods utilize a 150 km/150 layer pressure temperature model developed individually for each occultation. The temperature error in such a model profile can range to ± 4 K. Therefore, wherever possible, lines are chosen with ground-state energies between 100 cm⁻¹ and 400 cm⁻¹ to keep the temperature sensitivity of the intensities on the order of 1% K⁻¹.

A1.2 Pressure-temperature profiles

The precision and accuracy of ATMOS retrievals from space is greatly aided by CO_2 absorption features within spectra that can be analyzed for determining precise tangent pressures and temperatures, the CO_2 mixing ratio being nearly constant through most of the atmosphere. This makes ATMOS a "self-contained" instrument that does not need external instrumentation to gather temperature and pressure information. ATMOS determination of these profiles is described by *Rinsland et al.* [1992], *Stiller et al.* [1995] and *Abrams et al.* [1995b]. A near-constant CO_2 mixing ratio profile is assumed (e.g., for 1992, 360 ppmv in the troposphere decreasing to a constant 347 ppmv from 21 km to the turbopause

[Gunson et al., 1995]). The software used to determine tangent pressures for individual spectra is the same used to determine mixing ratios, except here the mixing ratio of the target gas, CO_2 , is fixed but the pressure is allowed to vary. Temperature sensitive CO_2 are used lines to determine temperatures, while maintaining hydrostatic equilibrium. (An iterative pressure-temperature determination using temperature-sensitive CO_2 lines is used for filters 1 and 3 because they lack adequate temperature-insensitive CO_2 lines. See Stiller et al. [1995].)

However, like any other gas that ATMOS measures, errors from noise and spectral parameter inaccuracies affect retrievals of CO₂. As they affect the molecular density within a pressure-temperature model, random errors from pressure soundings become systematic errors in the determination of gas mixing ratios for the occultation from which they were derived. For ATMOS profiles, the random error in a pressure sounding retrieval is determined as

$$\varepsilon_{\text{Pressure}}(z) = \frac{P(z)}{\vartheta_{\text{co}_2}(z)} \left[\Sigma_i \frac{\left(\frac{\partial T_{i,c}(v_i, z)}{\partial \vartheta_{\text{co}_2}} \right)^2}{\left(\frac{1}{\text{SNR}} \right)^2 + R_i^2} \right]^{-1/2}$$
(A1.2)

where: v_i = frequency at spectral grid point i (cm⁻¹)

z = altitude (km)

 $T_{i,c}(v_i,z)$ = calculated transmission at spectral grid point i

P(z) = pressure at altitude z

 ϑ_{CO_2} = assumed mixing ratio for CO_2

SNR = signal-to-noise ratio

Note that the addition of the signal-to-noise error assures that (A1.2) can never be zero, even in the case of a perfect fit. Systematic errors in the CO_2 line intensities add a 4%

error, while an estimated 2% error is from the assumed CO_2 profile. Systematic and precision errors are combined in quadrature to produce a total error varying from about 10% for all filters decreasing to about 5% near 25 km. Tangent pressure errors tend to increase substantially below 100 mb due to CO_2 line saturation; at 10 km, the tangent pressure error is about 30% in filter 9, 10% in filter 3, and about 5-6% in filters 1, 2, 9 and 12.

A1.3 Signal-to-noise and residual errors

The signal-to-noise ratio (SNR) varies for the spectral filters used by ATMOS by the amount of incident radiation on the detector; the broader the spectral bandpass, the lower the SNR. The trade-off is that limiting the spectral bandpass limits the number of co-located gases measurable. While a precise signal-to-noise ratio (SNR) can (and is) calculated by examining out-of-bandpass regions of the spectrum, Filter 1 (600-1200) generally has been shown to have a high SNR of ≈300:1, while with its broad bandpass, Filter 9 has an SNR of ≈100:1. From Table Mountain, however, the ability to take measurements within a narrow range of solar zenith angles allows averaging, and an increased SNR:

$$SNR_{average} = SNR_{individual} \sqrt{number of individual spectra}$$
 (A1.3)

Such averaging must be done judiciously, however, because as the solar zenith angle changes, so does the number of molecules in the instrument's line of sight.

The signal-to-noise error will be manifested in fitted spectra's residual of the fitted and observed transmission. The residual error in a mixing ratio retrieval is calculated by:

$$\varepsilon = \sqrt{\frac{\sum_{i} \left[T_{o}(v_{i},z) - T_{s}(v_{i},z) \right]^{2}}{N \sum_{i} \left[\partial T_{s}(v_{i},z) / \partial \vartheta \right]^{2}}}$$
(A1.4)

where: T_o , T_s = observed and synthetic transmission, respectively, at frequency v_i of spectral grid point i and altitude z.

 ϑ = the target gas mixing ratio

N = the number of inflection points in $(\partial T_s/\partial \vartheta)$, usually about the number of target lines in a microwindow.

The residual error is combined into a total root-sum-of-squares error with estimated errors from pressure sounding, zero transmission uncertainty, and propagated error from uncertainties in mixing ratio retrievals from higher altitudes. These latter two errors tend to be small. Note that SNR ratio is implicitly counted twice: once from its contribution to the tangent pressure error and once from its contribution to the residual. This tends to produce a conservative error, even with an excellent spectral fit.

A1.4 Averaging and combined errors

As mentioned in Section 1.1.5, when several microwindows are available for a mixing ratio retrieval, an averaged result is determined weighted by the inverse error of precision:

$$\vartheta_{mn} = \frac{\sum_{n} \left(\vartheta_{n} / \varepsilon_{n}^{2} \right)}{\sum_{n} \left(1 / \varepsilon_{n}^{2} \right)}$$
 (A1.5)

The precision of this weighted average is calculated two ways. The first is the reduced standard error of the mean:

$$\varepsilon_{\rm rsa} = \frac{1}{\sum_{\rm n} 1/\varepsilon_{\rm n}^2} \tag{A1.6}$$

The second method is the error on the estimation of the mean,

$$\varepsilon_{\text{std}}^{2} = \frac{\Sigma_{\text{n}}(\vartheta_{\text{n}}^{2}/\varepsilon_{\text{n}}^{2}) - \frac{\Sigma_{\text{n}}(\vartheta_{\text{n}}/\varepsilon_{\text{n}}^{2})^{2}}{\Sigma_{\text{n}}(1/\varepsilon_{\text{n}}^{2})}}{(M-1)\Sigma_{\text{n}}(1/\varepsilon_{\text{n}}^{2})}$$
(A1.7)

where M is the number of "effective" microwindows:

$$M = \frac{\left[\sum_{m}(1/\epsilon_{n})\right]^{2}}{\sum_{m}(1/\epsilon_{n}^{2})}$$
 (A1.8)

M is less than the actual number of microwindows except where all the precision errors are the same. The final precision quoted is the greater of (A1.6) or (A1.7).

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Appendix B

Additional research

This section briefly describes contributions by the author to works in addition to those in Chapters 2, 3, and 4.

B.1 Stratospheric trace gas column burdens from Table Mountain Observatory.

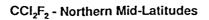
Perturbations to stratospheric chemistry can be long-term, such as the effects of atmospheric loading of the photolysis products of chlorofluorocarbons, and shorter-term such as increased aerosol loading by events such as volcanic eruptions. Such effects may be detected in the mixing ratios of several "active" species, such as O₂, and "sink" species, such as HCl, HF, and HNO₃, all of which primarily reside in the stratosphere and can be measured from ground-based infrared solar absorption techniques. During the delay of shuttle missions due the Challenger accident, the ATMOS instrument was at Table Mountain Facility, Wrightwood, CA (TMF; 34.4°N, 117.7°W), where it took a series of solar absorption spectra spanning from 1985 to 1990. The use of the TMF site has been described in Chapters II and III of this thesis. Measurements of O₃, HCl, HF, and HNO₃ column burdens from TMF were made in order to evaluate the precision attainable by solar absorption Fourier-transform spectrometry for this purpose and evaluate the suitability of TMF as a site for detection of stratospheric change. This research is described in Gunson and Irion [1991]. While the time interval in which measurements were taken did not allow a good evaluation of long-term trends in the stratosphere for these gases, the TMF site was shown to be adequate for this purpose, and the spectra and measurements provide a baseline from which future measurements can be compared and trends determined.

B.2 Collation and analyses of previously measured chloroflurocarbon, hydrochlorofluorocarbon and halon mixing ratio profiles in the stratosphere

The loading and location of active chlorine and bromine in the stratosphere depends not only on the stratospheric loading of their source gases, such as the chlorofluorocarbons and halons, but also the stratospheric lifetimes of these source gases; longer-lived gases can reach and be destroyed at higher altitudes. Stratospheric lifetimes of halogenated source gases, with lifetimes larger than transport time scales, are usually not measured directly. They are instead measured relative to that of N_2O by determining the slope of the stratospheric mixing ratio of the source gas plotted against co-located N_2O mixing ratios. In regions where the slope is linear, their lifetimes are related by:

$$\frac{\tau_{\text{source gas}}}{\tau_{\text{N}_2\text{O}}} = \frac{d\vartheta_{\text{N}_2\text{O}}}{d\vartheta_{\text{source gas}}} \frac{\vartheta_{\text{source gas}}^*}{\vartheta_{\text{N}_2\text{O}}^*}$$
(B1)

where τ is the lifetime, ϑ is mixing ratio, and ϑ^* is a mixing ratio representative of that in the troposphere. (See Plumb, R. A. and M. K. W. Ko, Interrelationships between mixing ratios of long-lived stratospheric constituents, *J. Geophys. Res.*, 97, 10145-10156, 1992.) In an effort to quantify stratospheric mixing ratios and lifetimes for several source halocarbons, and intercompare results by different techniques, previously measured results from in-situ and remote sensing techniques for some 20 gases were compiled and compared and, where possible, their mixing ratios plotted against those of N₂O and the slope determined. From these, stratospheric lifetimes with respect to N₂O could be calculated. An example of these plots, for CCl_2F_2 , are shown in Figures B.1 and B.2. Full results are given in *Fraser et al.* [1994].



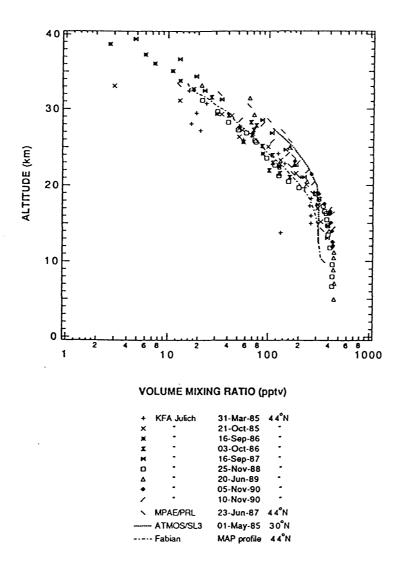


Figure B.1: CCl₂F₂ measurements from northern mid-latitudes (from Fraser et al. [1994]).

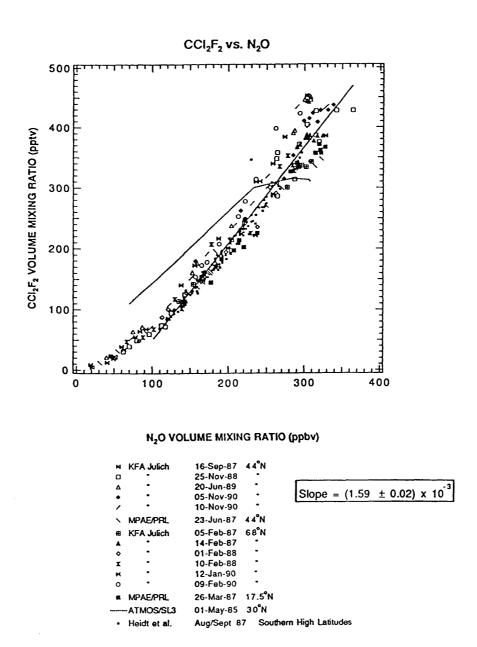


Figure B.2: Simultaneous measurements of CCl₂F₂ and N₂O. The slope is calculated from the data over the range indicated, but excludes measurements by ATMOS. The error is given as 1σ precision. From *Fraser et al.* [1994].

B.3 The use of laboratory spectral cross-sections to infer chlorofluorocarbon and hydrochlorofluorocarbon trace gas mixing ratios

Usually, the modelling of solar absorption spectra requires the calculation of spectral cross-sections based on knowledge of precise spectral line frequencies, line intensities and ground-state energies for the molecules of interest. These spectral line parameters can be difficult to obtain for chlorofluorocarbons, hydrochlorofluorocarbons and N_2O_5 as these molecules show broad, unresolved features in their infrared absorptions even at the best available resolutions. Calculation of their line intensities and frequencies from theory are difficult. Determination of their stratospheric mixing ratios using ATMOS softwate previously relied on empirical pseudo-line parameters, but these were prone to systematic errors in the band intensities and measurement uncertainties from poorly modelled temperature and pressure-broadening effects.

Rather than calculating cross-sections for these gases from pseudo-line parameters, a more efficient, if less flexible approach was to use laboratory-derived cross-sections directly in modelling atmospheric absorptions. This required the integration of new code into existing ATMOS software so that it can simultaneously use both linelists and tables of cross-section data to model absorptions of several gases simultaneously within a spectral window. For a spectral band of a particular molecule, a series of laboratory-derived empirical cross-section data, taken at temperatures and pressures exemplary of the vertical structure of the stratosphere, were first evaluated for suitability and consistency and then converted to a standard binary format. When these cross-sections were used in ATMOS retrievals for a particular model temperature and pressure, the methodology was to interpolate the cross-sections by temperature (or log pressure) to match model parameters, and then to interpolate across wavenumber to match the spectral grid. Absorptions by other molecules (calculated from spectral linelists) were then added into the window to model the total atmospheric absorption. The benefit of this technique was that it was computationally

very fast and avoided the need for psuedo-line parameters. The drawback was that uncertainties increased if an occultation's termperature/pressure profile was highly different from the conditions under which the cross-sections were originally measured, and could give poor results if the cross-section data were of poorer resolution than that of the ATMOS instrument. Overall, however, the approach has been successful for retrievals of various species, and its use is continually extended and modified as new laboratory data become available. (See, for example, the absorption of HCFC-142b in Figure 1.2.1.) The ATMOS linelist and cross-section data currently used are described in *Brown et al.* [1995], and representative ATMOS retrievals using the cross-section algorithm can be found in *Rinsland et al.* [1995a,b], and *Zander et al.* [1995].

B.4 Analyses of stratospheric H₂O and CH₄

There were several motivations to investigate the stratospheric abundances of H₂O and CH₄ from ATMOS measurements. First, as discussed in Section 1.5, the study of the deuterated isotopomers HDO and CH₃D in the stratosphere cannot be done outside the context of the more abundant H₂O and CH₄; it is the variations of deuterium in total water or methane, and the reasons for such variations, that are of interest,. Second, the partitioning of hydrogen among H₂O, CH₄ and H₂ in the upper stratosphere and lower mesosphere was not well measured. Assuming that the total amount of hydrogen residing in species other than H₂O, CH₄ and H₂ is inconsequential, then co-located measurements of H₂O and CH₄ can be used to deduce variations in the vertical H₂ mixing ratio profile. Third, and again as discussed in Section 1.5, the exact mechanisms for transport of water from the troposphere to the stratosphere are not well understood. Much as any proposed mechanism must account for the abundance of HDO in the stratosphere, so must any mechanism account for any seasonal variation of H₂O in the lower stratosphere. As ATMOS can simultaneously determine CH₄ and H₂O, increases in the H₂O mixing ratio

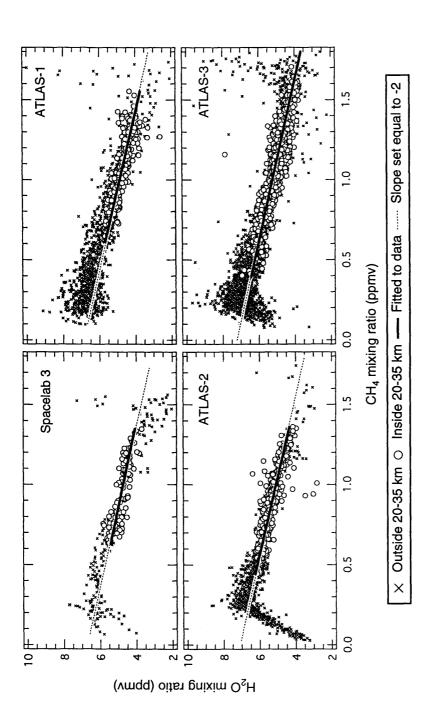
from methane oxidation can be subtracted, and the seasonal variation of water entering the stratosphere through the tropics deduced.

As a contribution in the study of these issues, spectral features of H₂O, HDO, and CH₃D absorptions in ATMOS spectra were selected and tested for their precision in mixing ratio determination and to evaluate their optimal vertical range. (Spectral intervals for CH₄ analyses were not selected by the author.) For the work described by *Moyer et al.* [1995], briefly summarized in Section 1.5, results of ATMOS mixing ratio retrievals were used to determine deuterium depletions in water, both corrected and uncorrected for methane oxidation. When modelled depletions of deuterium in water were compared to ATMOS results, the amounts of deuterium in stratospheric water suggested transport through the troposphere in fast convective updrafts, and not solely by gradual uplift through the troposphere and across the tropopause.

In connection with studies of the stratospheric hydrogen budget, co-located retrievals of H_2O and CH_4 for each mission were plotted against each other over the range of ATMOS measurements. Examples of these are shown in Figure B.3. In order to determine the amount of H_2O produced by the stratospheric oxidation of CH_4 , lines were fitted within the 20-35 km range. Again, assuming that the total amount of hydrogen residing in species other than H_2O , CH_4 and H_2 is minor, deviations of the slopes of H_2O vs. CH_4 correlations from -2 indicate net production or destruction of H_2 . However, the proximity of the slopes of the solid fitted lines in Figure B.3 to -2 indicate that H_2 oxidation does not consitute a significant net source of H_2O within the 20-35 km range, and that H_2O production is effectively balanced by CH_4 oxidation. In regions where the CH_4 mixing ratio is ≈ 250 pptv (about 50 km), the mixing ratios of water are generally higher than that expected by CH_4 oxidation (shown by the dashed curves in Figure B.3), indicating additional net production of H_2O from H_2 oxidation. At even lower CH_4 mixing ratios, the

amount of H₂O decreases because of H₂O photolysis. This study was described in detail by *Abbas et al.* [1995a].

The seasonal variation of water injected into the stratosphere was investigated by analyses of ATMOS retrievals of CH_4 and H_2O in the lower tropical stratosphere. Averaged retrieval results are shown in Figure B.4. As airmasses enter the stratosphere across the tropical tropopause throughout the year and slowly move upwards, a significant seasonal variation is imprinted on the mixing ratio of H_2O , but not on CH_4 . Assuming that the minimum at 40 mb (about 22.5 km) in the water vapor mixing ratio corresponds to entry across the tropopause at 100 mb during January and February, and the maximum at 55 mb (about 19 km) corresponds to entry during June and July, an average upward velocity of \approx 7-8 km year-1 was inferred for the region near 22 km. Full details of the study are given in *Abbas et al.* [1995b].



in the above plots are fitted to data between 20 and 35 km, while the dashed lines are fitted to the same region, except that the slope is forced to equal -2 Figure B.3: Stratospheric H₂O vs. CH₄ mixing ratios from ATMOS. Only measurements outside the polar vortices are shown. The solid lines and the line is extrapolated to lower and higher altitudes. Slopes of the solid lines are -(1.89±0.12) for Spacelab 3, -(2.08±0.08) for ATLAS-1, -(2.08±0.09) for ATLAS-2, and -(1.98±0.07) for ATLAS-3 (1σ precision).

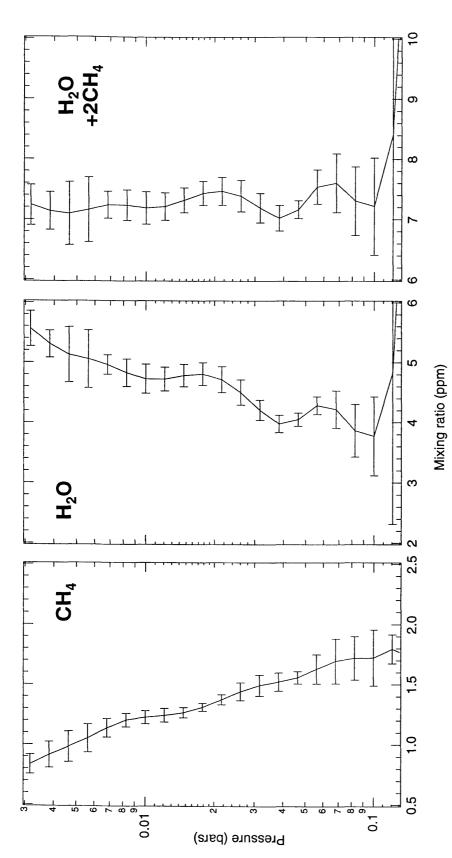


Figure B.4: ATMOS/ATLAS-3 measurements of tropical (8°N-20°N) CH₄, H₂O and H₂O+2CH₄. The structure in the H₂O and H₂O+2CH₄ profiles is indicative of the seasonal variation in water injection into the lower stratosphere and its propagation upwards with time. The vertical velocity of tropical upwelling in the lower stratosphere (≈7 km yr⁻¹ near 22 km) can be estimated from the wavelength of this structure.

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Appendix C

Column and profile data from ATMOS on the Space Shuttle and at Table Mountain

Note: ATMOS space-based retrieval data for CH₄, H₂O, HDO, CH₃D and most other gases are publicly available through the internet:

remus.jpl.nasa.gov; login anonymous

ftp: WWW: ftp://remus.jpl.nasa.gov/pub/results/vers2/

Table C.1: Average column burdens over Table Mountain Facility

Columns have not been scaled to sea level. Read col. error=random error in column; $^{668}O_3$ enrich= $^{16}O^{16}O^{18}O$ enrichment; $^{668}O_3$ enr. err=random error in $^{16}O^{16}O^{18}O$ enrichment.

Year	Day of	$^{48}O_{3}$	$^{48}O_{3}$	$^{668}O_{3}$	$^{668}O_{3}$	$668O_{3}$	668O3	CHF ₂ Cl	CHF ₂ Cl
	Year	column	col. error	column	col. error	enrich	enr. err	column	col. error
	(x	10 ¹⁸ mole	ec. cm ⁻²)	(x10 ¹⁵ mo	olec. cm ⁻²)	(%)	(%)	(x10 ¹⁵ mo	lec. cm ⁻²)
1985	274	7.96	0.43	3.58	0.16	12.2	7.9	1.05	0.08
1986	345							1.21	0.10
	346	7.13	0.43	3.42	0.17	20.1	9.3		
1987	58	8.96	0.40	4.28	0.26	19.5	9.0	1.16	0.09
	84	8.96	0.34	4.29	0.21	19.8	7.5		
	85	8.65	0.45	4.42	0.28	27.7	10.4	1.16	0.09
	139	9.31	0.28	4.56	0.39	22.5	11.1		
	140	9.60	0.38	4.52	0.39	17.7	11.1	1.29	0.10
	141	9.47	0.32	4.58	0.28	20.9	8.4	1.20	0.10
	154	8.86	0.43	4.26	0.28	20.2	9.8	1.22	0.10
	168	8.34	0.27	4.08	0.25	22.4	8.5	1.21	0.10
	238	8.19	0.35	3.88	0.30	18.5	10.4	1.30	0.10
	272	7.98	0.31	3.75	0.19	17.4	7.4	1.21	0.10
	280	7.39	0.34	3.54	0.30	19.9	11.7		
	288	7.53	0.34	3.52	0.17	16.8	7.8	1.25	0.10
	314	7.78	0.34	3.62	0.22	16.3	8.7	1.24	0.10
1988	77	7.71	0.36	3.65	0.18	18.2	8.1	1.22	0.10
	84	8.20	0.40	3.90	0.19	19.0	8.3	1.25	0.10
	90	9.47	0.65	4.40	0.27	16.9	10.7	1.24	0.10
	124	8.76	0.38	4.15	0.21	18.8	7.8	1.29	0.10
	130	8.31	0.34	3.77	0.32	13.4	10.8		
	154	8.25	0.38	4.08	0.35	23.7	12.1		
	176	8.05	0.49	3.78	0.19	17.4	9.2	1.22	0.10
	182	8.12	0.30	3.72	0.22	14.5	7.9	1.28	0.10
	190	7.81	0.51	3.70	0.18	18.6	9.7	1.27	0.10
	196	7.98	0.30	3.77	0.21	17.9	7.8	1.39	0.11
	209	7.92	0.39	3.74	0.23	18.1	9.2	1.30	0.10
	216	7.77	0.42	3.50	0.31	12.6	11.6	1.26	0.10
	223	7.94	0.35	3.79	0.19	19.1	8.0	1.32	0.11
	231	7.82	0.28	3.69	0.21	18.1	7.8	1.39	0.11
	252	8.02	0.25	3.63	0.18	13.2	6.6	1.38	0.11
	259	7.43	0.45	3.52	0.17	18.4	9.3	1.36	0.11
	293	7.62	0.32	3.47	0.17	13.9	7.5	1.32	0.11
	301	7.11	0.25	3.30	0.20	16.0	8.2	1.36	0.11
	321	7.38	0.30	3.37	0.17	14.0	7.3	1.43	0.11

Table C.1 continued

Year	Day of Year (x	48O ₃ column 10 ¹⁸ mole		668O ₃ column (x10 ¹⁵ mo	col. error	668O ₃ enrich (%)	668O ₃ enrich. (%)	CHF ₂ Cl column (x10 ¹⁵ mo	col. error
1989	9	8.14	0.45	3.55	0.31	9.1	11.1		
	31	7.89	0.25	3.66	0.31	15.8	10.6		
	206	7.89	0.45	3.56	0.31	12.7	11.6		
	207	8.15	0.58	3.73	0.19	14.3	9.9	1.41	0.11
	208							1.33	0.11
	209	7.89	0.46	3.67	0.18	16.2	8.9	1.36	0.11
	243	7.81	0.25	3.68	0.18	17.0	6.9	1.42	0.11
	250	7.90	0.51	3.80	0.19	20.2	9.8	1.39	0.11
	257	7.72	0.33	3.61	0.18	17.0	7.7	1.45	0.12
1990	122	8.45	0.41	3.97	0.20	17.3	8.2		
	123	8.42	0.48	3.90	0.13	15.8	7.6	1.45	0.12
	124	8.15	0.37	3.81	0.26	16.9	9.6	1.52	0.12
	201	7.76	0.41	3.63	0.18	16.9	8.5	1.45	0.12
	204	7.76	0.34	3.51	0.30	13.1	10.9		
	205	7.95	0.30	3.89	0.33	22.4	11.5		
	206							1.50	0.12

Table C.2: ATMOS ozone volume mixing ratios and heavy ozone enrichments.

This table contains data only from 26 to 2.6 mb retrievals only where $^{16}O^{16}O^{18}O$ and/or $^{16}O^{18}O^{16}O$ were simultaneously measurable with $^{16}O^{16}O^{16}O$.

Mixing ratios and random errors are in units of 10⁻⁶ for ¹⁶O¹⁶O. Latitude and longitude are given for the 30 km tangent height. Random errors for the enrichments were calculated as:

Enrichment error (%) =
$$100 \times \left(\frac{\delta_s (\%)}{100} + 1\right) \sqrt{\left(\frac{RE_s}{MR_s}\right)^2 + \left(\frac{RE_u}{MR_u}\right)^2}$$

where: RE_s and RE_u are the mixing ratio random errors of the substituted ozone molecule ($^{16}O^{18}O$ or $^{16}O^{18}O^{16}O$) and $^{48}O_3$, respectively,

 $\mathrm{MR_s}$ and $\mathrm{MR_u}$ are the mixing ratios of the substituted ozone molecule and $^{48}\mathrm{O_3}$, respectively,

 δ_{s} is the enrichment of the substituted ozone molecule in %.

Read: sl3=Spacelab 3, at1=ATLAS-1, at2=ATLAS-2, at3=ATLAS-3

Alt=altitude, Press=pressure, vmr=volume mixing ratio, err=random error

sr=sunrise occultation, ss=sunset occultation

 $^{48}O_3 \text{ vmr} = ^{16}O^{16}O^{16}O \text{ volume mixing ratio x } 10^6$

 $^{48}O_3$ err = $^{16}O^{16}O^{16}O$ volume mixing ratio random error x 10^6

 $^{668}O_3 \text{ enr} = ^{16}O^{16}O^{18}O \text{ enrichment } (\%)$

 $^{668}O_3$ enr err = $^{16}O^{16}O^{18}O$ enrichment random error (%)

 $^{686}O_3$ enr= $^{16}O^{18}O^{16}O$ enrichment (%)

 $^{686}\text{O}_3$ enr err = $^{16}\text{O}^{18}\text{O}^{16}\text{O}$ enrichment random error (%)

Table B.2: ATMOS ozone volume mixing ratios and heav Alt Temp Press $48O_3$ $48O_3$ $668O_3$ 668 O_3 686 O_3 $686O_3$ $686O_3$ $686O_3$ $686O_3$ $686O_3$ $686O_3$	MOS ozone volume mixing ratios and heavy ozone enrichments. (See notes on page 135.)	Alt Temp Press 48O ₃ 48O ₃ 668O ₃ 668O ₃ 686O ₃	(km) (K) (mh) ymr om on one
	\mathbf{AT}	Alt Temp Press 48O ₃ 48O ₃ 668O ₃ 668O ₃ 686O ₃ 686O ₃	(mb)

sl3 sr06 Latitude:-46.9 Longitude:309.9 Filter: I	sl3 ss05 Latitude: 31.4 Longitude: 0.4 Filter: I
2.6 6.80 0.98 16.8 26.3 21.5	26.3
3.2 6.68 0.97 15.9 26.5 24.0	3.2 5.61 0.82 20.0 27.8 17.2
37.4 230.3 3.8 6.56 0.94 16.6 26.2 23.1 22.7	38.2 248.3 3.8 6.18 0.89 21.4 27.6 12.7 23.7
4.6 6.44 0.91 17.5 25.8 20.3	4.6 6.56 0.93 21.5 27.4 14.1
5.6 6.31 0.90 17.1 26.0 16.5	5.6 6.81 0.97 20.8 26.6 18.1
6.8 6.08 0.85 16.1 25.3 12.4	6.8 7.24 1.01 18.9 25.6 19.5
8.3 5.75 0.76 16.5 24.2 9.7	8.3 7.96 1.06 16.6 24.1 16.9
10.0 5.48 0.68 17.3 23.1 9.0	10.0 8.60 1.07 14.7 22.5 13.4
12.1 5.36 0.61 16.7 21.6 9.5	12.1 8.62 0.99 13.9 20.9 12.1
14.7 5.35 0.56 14.5 20.3 10.3	14.7 8.01 0.84 14.3 19.6 13.1
17.8 5.26 0.50 13.6 18.8 11.8	17.8 7.11 0.67 14.2 18.3 12.7
21.5 5.00 0.43 14.8 18.0 14.1	21.5 6.23 0.53 9.6 16.9 5.9
26.1 4.57 0.35 15.3 17.3 14.6	26.1 5.33 0.41 1.7 15.4 -4.6
Latitude: 32.8 Longitude: 115.5	Latitude: 29.9 Longitude: 245.4
2.6 4.57 0.65 17.7 29.5 22.9	2.6 4.61 0.66 24.1 30.4 27.2
3.2 5.71 0.83 18.7 27.2 21.6	3.2 4.89 0.72 22.8 28.3 26.7
3.8 6.75 0.97 19.0 26.7 19.4	3.8 5.35 0.77 22.0 27.8 25.1
4.6 7.44 1.06 20.0 26.6 17.0	4.6 6.05 0.86 21.8 27.4 21.3
5.6 7.91 1.13 21.1 26.3 14.8	5.6 6.92 0.99 21.0 26.3 16.7
6.8 8.43 1.17 19.8 25.5 12.3	6.8 7.77 1.08 19.2 25.3 13.2
8.3 9.04 1.20 15.5 23.6 9.3	8.3 8.56 1.14 17.3 23.9 11.6
10.0 9.47 1.18 12.0 21.8 7.7	10.0 9.07 1.13 16.0 22.5 11.2
12.1 9.40 1.08 13.0 20.5 9.7	12.1 9.04 1.04 15.4 20.9 11.1
14.7 8.83 0.92 16.6 20.3 13.0	14.7 8.47 0.89 15.6 19.7 11.4
17.8 7.88 0.75 15.0 18.7 10.7	17.8 7.63 0.72 14.8 18.2 10.6
21.5 6.77 0.58 5.1 16.2 0.2	21.5 6.72 0.57 8.9 16.7 5.4
26.1 5.61 0.43 -4.8 14.3 -9.8	-1.3 15.0 -3.7

enr enrerr enrenrerr

vmr err

(mb)

(km) (K)

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

 Table B.2 continued. (See notes on page 135.)

 Alt Temp Press 48O3 48O3 668O3 668O3 686O3 (660)

 (km) (K) (mb) vmr err enr enr enr enr enr enr enr enr

at	4	36	Ä	3	Š	Ϋ́	33,	33	a	4	č	ਲ ਲ	ĕ	33	ř	32	8	a	4	33	ñ	Ř	33	ж.
lter: I														Filter: 1	17.1	19.1	19.8	21.5	19.3	22.9	28.1	24.2	20.0	25.5
Longitude:300.7 Filter: 1															11.5		9.3					4.2		42.0
tude:3	23.9	23.4	24.2	22.7	22.5	22.2	22.9	19.3				15.8		Longitude: 184.8			23.2	22.0	20.4	21.3	22.7	19.7	17.5	20.6
	0.15 19.6	0.15 16.1	0.23 15.4		0.21 15.5	0.20 15.2	0.18 14.0	0.20 12.8	0.21 12.4	0.20 12.0		0.13 3.8	0.12 -0.9	' Longi	0.14 15.2	0.19 18.4	0.22 18.0	0.26 14.8	0.26 13.2		0.23 11.8	0.21 7.5	0.21 11.5	0.20 37.1
Latitude:-28.2					7.85 (9.38				5.02	Latitude:-16.7	6.12 (09.9	7.40 (8.32 (8.99	9.31	_		10.2	8.32 (
Latitu	5.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7
103	41.0 250.1	248.7	246.3	243.5	240.1	236.0	234.0			228.7		224.3	220.8	604	41.2 252.7	250.4		243.2	238.3		233.6	231.9	229.8	227.5
atl sr03	41.0	39.5	38.1	36.8	35.4	34.0	32.7	31.4	30.1	28.8	27.5	26.2	24.9	at1 sr09	41.2	39.8	38.4	37.0	35.6	34.3	32.9	31.6	30.3	29.0

0									0									0								
Filter:	24.2	25.9	19.9	19.5	20.5	22.6	23.0	26.1	Filter:	17.3	15.6	13.9	13.4	16.6	24.5	17.8	18.3	Filter:	22.9	23.6	20.3	20.0	24.0	18.6	20.9	21.3
4.2	15.0	19.1	11.3	8.1	11.4	13.9	13.2	11.6	1.5	32.8	21.8	10.3	7.8	13.7	19.8	20.6	15.5	2.9	1.2	9.6	17.1	16.4	12.5		9.5	
ude:11	21.9	23.4	7	15.4	14.8	15.0	15.1	16.4	, de: 9	13.1	12.0	10.6	8.6	11.2	15.3		11.1	4de:25	25.9	23.2	17.3	16.9	17.0		13.3	13.4
Longitude: I	11.9	10.3		22.5	22.9	20.5	17.5	13.0	ongiti	45.1	37.3	23.1	14.8	13.8	18.3	23.5	22.7	ongiti	16.1	19.8	28.7	7	17.8	15.9	17.0	15.7
7 8:	0.21	0.22	Ġ	0.22	0.23	0.23	0.30	0.32	4.		0.21	0.21	0.21	0.21	0.24	0.22	0.21	7 9	0.21	0.19	0.34	0.40	0.37	0.32	0.25	0.26
<u>.</u> ک	6.61	7.16	7.73	8.86	9.72	96.6	10.2	10.5	de: -5	6.70	7.52	8.20	8.87	9.44	9.63	9.53	9.51	de: -1	6.48	6.4	6.73	7.96	9.46	10.2	10.3	10.3
Latitude	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0
20	2	2	249.8	3	237.8	235.4	234.4	233.7	21	256.8	252.2	247.5	242.9	239.7	9	238.2	236.1	22	256.0	2	246.9	242.8	238.9	236.4	235.0	234.0
at I sr20	41.0	39.5	38.1	36.7	35.3	34.0	32.6	31.3	atl sr2	41.0	39.5	38.1	36.7	35.4	34.0	32.6	31.3	at I sr	40.9	39.5	38.0	36.7	35.3	33.9	32.6	31.3

enr enr err enr err

vmr err

(qm)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (Km) (Km) (km) vmr err err err err err err err

Filter: 9	17.2	17.0	16.4	17.3	22.8	26.5	20.8	Filter: 1	22.0	22.3	19.6	19.6	19.1	23.7	21.8	23.0	22.5	19.8	Filter: 9	17.1	18.6	18.3	19.0	17.3	14.7	14.4	16.0
V	14.9	15.4	12.9	11.7	11.5	8.8	5.6	02.2 I	15.6	14.1	17.4	17.5	14.3	13.1	12.8	11.1	9.6	11.3	79.5	-3.5	5.3	22.3	28.6	19.0	9.2	7.9	13.2
ide:27		11.9	11.4	12.4	15.3	16.8	12.4		25.2	24.3	21.5	20.9	20.9	21.8	19.6	19.1	17.9	16.2	7	14.8	14.6	14.8	14.7	13.2	9.3	9.1	9.6
0 Longitude:270.2	20 23	0.20 11.6	0.20 8.7	0.20 15.4	0.24 21.0	0.24 17.8	0.31 10.9	5 Longitude:2	0.17 19.5	0.21 14.4	0.23 15.8	0.22 18.7	0.24 18.5	0.24 16.3	0.26 14.1	0.25 12.0	0.20 10.6	0.18 12.4	Longitude:	0.21 23.6	0.20 22.3	0.19 24.8	0.18 24.1	0.19 19.0	0.21 14.8	0.22 13.4	0.21 12.9
Latitude: -1.0	7.2	(4.6 8.32	5.6 8.92	09.6 8.9	8.3 10.3	10.0 10.7	Latitude: 0.6	2.6 6.36	3.2 7.65	3.8 7.83	4.6 8.02	5.6 8.69	6.8 9.50	8.3 10.3	10.0 10.7	12.1 10.2	14.7 9.28	Latitude: 1.1	2.6 7.28	3.2 7.30	3.8 7.02	4.6 7.31	5.6 8.44	6.8 9.72	8.3 10.4	10.0 10.3
at1 sr23	253	38.1 249.9	36.7 246.1	35.3 242.5	34.0 240.2	32.6 236.0	31.3 232.5	atl sr26	40.9 255.1	39.4 252.1	38.0 247.7	36.6 243.4	35.3 240.5	33.9 238.6	32.5 237.0	31.2 234.5	29.9 232.1	28.6 229.4	at1 sr27	41.0 258.6	39.6 253.0	38.1 248.4	36.7 245.2	35.4 242.1	34.0 239.4	32.6 236.8	31.3 234.4

Filter	19.2	17.5	18.5	20.3	19.7	20.5	20.8	23.3	20.0	21.0	Filter.	23.2	22.8	17.9	9	20.7	19.1	18.4	25.3	Filter.	23.2	16.7	16.9	16.4	16.0	21.1	21.5	22.6
<i>o</i> ;	18.8	14.8	10.9	12.2	15.4	13.7	10.0	8.8	9.2	10.5	0.3	3.7	18.0	15.2	9.4	6.5	6.3	9.2	12.1	7.6	7.4-	11.7	24.0	23.4	14.8	4.6	-1.7	-0.6
tde: 88		α i	Ξ.	21.7	20.6	20.2	19.5	19.4	17.1	17.0	tde:29	21.6	20.0	12.7	13.3	14.8	13.0	11.9	15.8	ide:26	24.5	13.8	12.1	11.6	11.9	14.4	14.5	14.5
Longitude	19.6	18.2	16.1	14.9	14.7	13.8	12.8	12.1	11.7	12.3	ongiti	1.1	3.4	14.0	22.0	22.0	18.4	15.5	16.9	ongiti	22.9	18.1	16.1	14.5	16.4	19.3	19.0	15.6
2 T	5	0.22	0.25	0.25	0.24	0.24	0.24	0.26	0.24	0.19	7 Z	0.19	0.21		0.19	0.21	0.20	0.19	0.22	7 Z			0.20				0.26	0.25
le: 3.	∹:	9.	8.33	ϵ	8.49	9.37	10.5	6.01	9.01	89.6	te: 6.	6.22	7.09	7.39	8.00	8.78	8.99	9.07	9.61	te: 7.	6.40	7.09		•	9	9.56	0.01	10.1
Latitue	2.6	3.2		4.6	5.6	8.9	8.3	10.0	12.1	14.7	Latitue			3.8	4.6	5.6	8.9	8.3	10.0	Latituc	2.6	3.2		4.6	9.9	8.9	8.3	10.0
31	251.5	248.8	45.	241.8		237.6	236.6	234.6	233.0	230.2	32	258.5	250.5	245.3	242.2	239.9	238.1	235.8	233.4	33	257.3	253.5	250.2	246.8	243.6	241.7	239.0	236.0
atl sr3	40.8	39.4	38.0	36.6	35.2	3.9	32.5	31.2	29.8	28.5	atl sr.	40.9	Ś	0	6.7	-	33.9		31.3	atl sr.	41.1	9.6	8.2	36.8	35.4	34.0	32.7	31.3

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

enr enr err enr enr err

vmr err

(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃
(km) (K) (mb) vmr err enr enr enr enr err

Filter: 1	19.4	19.2	19.3	19.7	20.4	20.6	20.4	22.5	18.5	19.8	Filter: 1	19.5	18.5	19.0	19.5	18.8	18.8	18.8	21.9	24.2	19.0	19.0
	14.7	14.2	12.2	10.6	11.1	12.9	12.5	8.9	6.3	8.9	77.0	11.7	9.4	14.1	17.3	15.4	12.0	10.7	11.3	10.0	5.9	7.4
Longitude: 222.3	23.3	23.5	22.7	21.1	20.2	19.9	19.6	19.3	16.6	16.3	ide:17	23.6	22.7	22.0	21.5	20.5	19.7	18.9	19.0	18.9	16.1	15.7
ongitu	14.8	19.7	17.8	11.3	7.5	10.7	15.0	14.2	10.8	8.6	Longitude: 1	18.3	18.2	18.8	18.7	17.3	15.7	14.9	14.2	12.2	8.5	10.9
	0.16 14.8	0.21	0.23	0.25	0.23	0.25	0.26	0.26	0.23	0.19		0.12 18	0.17	0.21	0.25	0.24	0.23	0.23	0.25	0.22	0.18	0.17
Latitude: 8.2	5.87	99.9	7.60	8.48	9.13	89.6	10.3	8.01	10.7	9.95	de: 9.2	4.71	5.74	7.13	8.32	86.8	9.45	10.0	10.5	10.3	9.40	8.46
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0 10.8	12.1	14.7	Latitude:	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8
35	257.6	252.4	247.4	242.6	239.1	237.1	236.6	234.7	232.8	230.8	37	259.9	253.3	246.7	240.6	237.3	236.0	235.4	233.8	231.9	229.2	226.0
atl sr35	40.9	39.5	38.0	36.6	35.3	33.9	32.6	31.2	29.9	28.6	atl sr37	40.8	39.4	37.9	36.5	35.2	33.8	32.5	31.2	29.9	28.5	27.3

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Filter:	26.9	20.0	21.8	24.0	20.9	22.0	23.3	21.8	Filter:	22.5	24.3	20.4	19.4	21.2	20.8	20.9	23.9	Filter:	18.6	17.1	18.0		23.9	24.2	19.9	19.5
1.7	4.8	11.0	19.5	17.1	11.5	13.0	14.1	8.4	5.3	6.4	18.2	14.0	7.9	5.8	8.5	12.8	15.7	5.1	4.0	9.9	10.3	14.3	16.4	15.1	12.1	9.7
4de:13	23.7	18.2	18.7	17.4	14.6	14.1	13.8	12.8	1de: 80	22.8	21.4	15.3	15.2	15.5	14.1	13.1	14.7	4de:26	17.4	13.8	13.2	14.5	16.8	16.0	12.1	12.2
Longitude: I	8.7	0		32.6	22.9	17.1	13.9	11.8	ongiti	37.8	27.4	20.2	22.2	24	21	17.3	18.0	Longiti	15.8	-0.1	-3.4	10.7	22.5		21.2	22.6
.2 1	0.18	0.20	0.21	0.22	0.21	0.20	0.20	0.28	<i>I I</i> :	0.20	0.22	0.20	0.21	0.23	0.26	0.29	0.27	7 8:	0.17	0.17	0.17	0.18	0.21	0.24	0.25	0.24
de: 10	5.94	6.90		8.48	8.66	8.96	9.57	10.2	de: 11	5.79	6.52	7.63	8.36	8.79	9.44	10.2	10.5	de: 14	5.99	6	6.35		8.76	9.76	10.3	10.4
Latitude:		3.2	3.8	4.6	5.6	8.9	8.3	10.0	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0
39	258.7	253.4	248.9	249.3	246.9	241.2	238.5	236.1	7	258.6	259.4	254.3	248.7	244.1	240.3	237.1	232.8	42	256.1	253.1	250.0	246.9	241.5	237.5	235.6	234.4
atl sr.	41.2	39.7	33	36.9	35.5	34.1	32.7	31.4	atl sr4	41.0	39.6	38.1		35.3			31.2	atl sr	41.0	39.5	38.1	36.7	35.3	33.9	32.6	31.2

 $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$

Alt Temp Press

enr enr err enr err

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vmr

(mb)

 Table B.2 continued. (See notes on page 135.)

 Alt Temp Press 48O3 48O3 668O3 668O3 686O3 686O3 (Km) (Km) (mb) vmr err enr enr enr enr enr enr

9	_	~	₹+	~	2	0	•	_	: 1	7	~	7	_	~	-	~1	~	_	~	6:	2		7	~	10	-	•	
Filter.	21.1	20.3	18.	18.3	15.0	16.0	15.9	23.	Filter	20.7	20.3	20.7	19.0	18.8	18.4	19.	19.2	20.0	19.8	Filter:	26.6	24.0	22.7	20.8	21.	17.	16.9	15.0
_	3	11.8	2.9	1.4	6.1	10.2	11.3	11.0	8.6	16.0	14.4	13.0	13.4	14.3	14.0	12.4	10.1	7.8	8.9	06.5	29.3	20.7	11.3	9.0	13.2	19.2	19.6	10.5
Longitude: 242.4	22.1	21.5	16.5	15.4	10.3	10.2	10.1	14.0	ide:21	24.3	24.4	23.4	21.9	21.2	19.6	18.7	17.8	17.2	16.6		20.8	19.1	18.5	15.7	13.1	10.2	9.7	9.0
ongiti	45.6	41.8	26.6	17.6	16.1	16.8	17.1	17.1	Longitude: 2	15.6	17.0	20.9	22.8	20.3	15.9	13.2	12.2	11.0	10.2	ongitude: 1	26.8	17.8	14.2	27.5	31.5	28.4	22.4	13.4
	0.18	0.19	0.20	0.21	0.20		0.20	0.21	7 L	0.14	0.17	0.21	0.24	0.24	0.23	0.27	0.26	0.21	0.20	7 8:	0.24	0.19	0.20	0.22	0.20	0.18	0.17	0.17
atitude: 15.3	5.92	6.78	7.64	8.37	8.96	9.52	10.0	0.2	le: 15.	4.61	5.99	7.57	8.45	8.79	9.40	0.4	11.1	10.9	9.78	17	6.21	6.64	7.33	8.09	8.57	8.73	8.84	9.24
Latituc	2.6	3.2	3.8	4.6	5.6	8.9	8.3 1	10.010	Latitude:	5.6	3.2	3.8	4.6	5.6	8.9	8.3 1	10.01	12.1	14.7	Latitude:	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0
43	260.3	254.6	249.3	244.3	241.6	239.6	237.0	233.8	44	256.8	253.5	250.2	245.7	239.5	235.3	233.6	232.9	231.4	\sim	49	260.0	259.2	257.7	253.1	246.4	241.6	237.7	234.1
atl sr43	41.1	39.6	38.2	36.8	35.4	34.0	32.7	31.3	atl sr44	40.9	39.5	38.0	36.6	35.2	33.9	32.6		29.9	28.6	atl sr49	41.8	40.3	38.8	37.4	36.0	34.6	33.2	31.9

10.011.3 0.27 2.3 8.2 -2.5 12.9 Latitude: 27.8 Longitude: 237.7 Filter: 9 2.6 5.95 0.18 16.0 21.7 0.1 31.5 19.3 17.7 18.2 19.1 20.3 20.4 20.9 14.4 26.6 25.4 Latitude: 18.2 Longitude: 83.9 14.6 2.6 5.40 0.18-19.9 15.4 14.8 15.1 14.1 0.21 28.0 0.19 22.1 0.22 23.8 0.28 14.1 0.16 25.1 0.15 30.1 0.17 23.0 0.23 14.3 0.21 15.1 3.2 5.59 3.8 6.51 4.6 7.83 6.8 9.90 2.6 5.95 3.2 5.70 3.8 5.71 5.6 8.98 8.3 10.8 38.4 251.7
37.0 245.7
35.6 240.0
34.2 237.1
32.9 234.1
31.6 231.0
at1 sr56
40.9 262.4
39.5 258.2
38.0 252.2
36.6 246.2
35.2 243.2
33.8 242.6
32.5 237.5 39.8 256.6

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E

vmr

(qm)

(km) (K)

Latitude:-42.0 Longitude: 95.2

0.18 18.8

0.18 21.2 0.18 17.8 0.21 12.8

10.0

25.5 22.1

18.5

0.12 15.2

5.45

21.5

0.14 15.2

0.17 12.2

14.7

26.1 4.81 0.13 9.6 16.3 5.3 22.1 Latitude:-44.5 Longitude:251.2 Filter: 2.6 5.57 0.16 18.1 24.8 18.5 21.0

16.4

24.3

0.18 19.2

2.6 5.57 3.2 6.13

0.21 18.7 0.21 17.1

6.63 86.9 7.09 6.94 6.92 6.80

16.5

0.25 16.0 0.19 16.4 0.15 17.3

20.3 19.3

11.8 10.6

0.15 16.7 0.16 14.3

10.0

19.0

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135	668O ₃ 686O	enr err
age	3,668	
s on I	66803	enr
d. (See notes on page 135.	⁴⁸ O ₃	еп
d. (Se	⁴⁸ O ₃	vmr
ontinue	Press	(mb)
Fable B.2 continued	Temp	(K
Table	Alt T	(km)

atl ss18 L 39.5 249.9 38.1 246.0	36.8 242.1 35.4 239.6 34.0 237.8 32.7 235.2	32.7 233.2 31.4 232.3 30.0 230.1 28.7 228.2	26.2 224.3 26.2 224.3 24.9 222.4 2 at 1 ss25 L	40.8 250.2 39.4 248.5 38.0 244.6 36.6 239.9	35.3 237.3 33.9 234.8 32.6 232.0 31.3 229.5	30.0 227.6 28.7 225.3 27.4 222.9 26.2 221.0 24.9 219.9
Filter: 9				Filter: 9		
8.7 Longitu 0.20 46.0 0.21 30.2	3.8 8.17 0.22 19.5 10.9 4.6 8.52 0.21 15.0 10.3 5.6 8.45 0.20 16.6 10.5 6.8 8.30 0.10 18.7 15.0	6.8 8.39 0.19 18.7 15.9 1 8.3 8.75 0.25 15.9 10.9 2 10.0 9.20 0.27 12.0 9.1 3 12.1 9.13 0.27 11.8 9.0	14.7 8.51 0.52 15.8 7.1 17.8 7.48 0.19 13.7 14.4 21.5 6.18 0.15 7.6 14.0 26.1 4.91 0.11 2.7 14.8	Latitude:-31.8 Longitude: 10.2 2.6 5.76 0.19 22.8 24.0 3.2 6.20 0.21 12.1 22.0 3.8 7.33 0.21 8.7 15.9	0.21 10.6 0.22 16.7 0.22 22.3 0.23 18.9	10.0 8.82 0.24 10.7 10.6 12.1 8.29 0.22 7.5 10.3 14.7 7.30 0.17 11.4 13.5 17.8 6.57 0.14 14.9 15.2 21.5 6.07 0.20 8.2 18.1 26.1 5.22 0.14 -2.3 11.0
•	38.3 251.4 36.9 246.8 35.5 242.2 34.1 230.6	34.1 239.0 32.8 237.1 31.4 234.2 30.1 231.3	26.8 226.3 27.5 224.9 26.3 221.6 25.0 219.4	atl ss08 41.1 259.3 39.6 252.4 38.2 246.1	36.8 242.1 35.5 240.3 34.1 238.6 32.8 235.4	31.4 232.5 30.1 229.8 28.8 227.4 27.5 225.6 26.3 224.2 25.0 221.8

vmr err enr enr enr enr err

(qm)

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (Km) (K) (mb) vmr err err err err err err err err

Filter: I	19.1	19.2	19.3	20.7	19.0	18.7	19.6	19.8	18.8	19.0	19.4	22.2	24.9	Filter: 9	19.6	19.6	19.0	18.9	19.7	21.4	20.0	19.2	18.9	18.1	17.8	22.6	20.2
	17.1	17.8	15.5	15.1	17.0	16.4	12.2	7.9	6.7	7.8	9.3	9.2	7.0	Ŋ	6.6	8.9	8.9	9.9	10.3	15.6	17.6	15.5	11.8	9.1	8.4	9.1	8.7
ide: 97	23.9	23.8	23.1	22.2	20.3	20.2	20.0	19.1	17.5	16.7	16.3	17.8	19.0	tde: 24	21.8	19.7	17.9	15.9	15.1	14.9	13.2	12.7	14.2	13.4	12.0	15.5	14.8
.2 Longitude: 92.5	0.16 17.4	0.20 20.6	0.20 17.8	0.21 14.1	0.18 13.8	0.19 15.5	0.17 15.9	0.16 14.0	0.16 11.2	0.16 10.3	0.15 11.8	0.12 13.0	0.11 10.6	8 Longitude:	0.19 30.9	0.21 36.0	0.21 27.0	0.19 14.9	0.16 7.3	∞	0.16 13.8	0.15 16.6	0.17 13.9	0.19 10.3	0.18 10.4	0.19 13.7	0.14 15.3
le:-46	5.77	5.94	6.14	6.41	6.75	7.14	7.54	7.86	7.82	7.33	6.55	5.74	4.96	le:-46.	6.61	7.42	7.71	7.45	6.87	6.53	6.67	7.05	7.23	7.22	7.02	6.34	5.17
Latitude:-46.2	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitud	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
atl ss26	41.0 246.5	39.7 243.7	38.3 241.9	36.9 240.9	35.5 239.0	34.2 235.7	32.9 234.0	31.5 232.9	30.2 231.6	28.9 229.8	27.6 227.5	26.3 225.5	25.1 224.0	atl ss27	40.9 248.8	39.5 246.3	38.1 244.0	36.8 240.8	35.4 237.8	34.1 235.5	32.7 233.3	31.4 231.4	30.1 229.6	28.8 228.1	27.5 226.4	26.2 224.9	25.0 223.6

atl ss29	Latitu	atitude:-47	.3 L	ngiti	Longitude:33	39.1	Filter: 9
40.9 254.2	2.6	6.29	0.19	6.0	19.1	15.9	20.5
39.5 249.4	3.2	6.94	0.20	13.4	20.3	15.7	20.8
38.0 247.6	3.8	7.45	_	21.3	18.9	17.9	24.5
36.6 244.7	4.6	7.70	0.25	24.9	15.0	16.5	20.7
35.3 240.5	5.6	7.46	0.24	<u></u>	14.6	14.6	19.9
33.9 236.9	8.9	7.06	0.18	29.5	14.7	14.8	20.5
32.6 234.4	8.3	7.18	0.15	24.2	14.1	13.7	20.7
31.3 232.7	10.0	7.98	0.24	12.9	11.2	9.6	16.9
29.9 231.5	12.1	8.59	0.25	8.9	10.5	6.7	16.3
28.6 230.2	14.7	8.47	0.25	7.3	12.7	9.9	19.7
27.3 229.0	17.8	7.57	0.21	11.0	13.7	8.7	21.0
26.0 228.4	21.5	6.27	0.15	13.4	15.5	9.6	23.0
24.7 225.9	26.1	4.99	0.11	12.4	19.5	6.7	27.7
atl ss31	Latitue	de:-47	7 L	Longitu	ude:29	3.8	Filter: I
40.5 248.5	2.6	4.76	0.13	19.1	23.5	25.2	20.0
39.1 244.4	3.2	5.37	0.16	17.0			19.6
37.8 240.6		6.05	0.18	17.4	21.7	16.7	18.7
36.4 237.8	4.6	6.61	0.19	17.8	21.6	11.9	18.0
35.1 236.4	5.6	98.9	0.20	17.4	21.8	11.1	19.3
33.7 234.0	8.9	6.84	0.18	16.2	20.2	12.8	17.8
2.4	8.3		0.16	14.5	19.1	13.5	17.4
31.1 227.5	10.0		0.14	12.9	18.6	11.0	18.5
9.8 2	12.1	9.	0.12	11.9	18.5	7.9	20.1
28.5 223.6	14.7	6.22	0.12	11.7	18.2	6.3	20.4
27.3 221.2	17.8	5.55	0.12	11.7	18.0	6.3	21.0
26.0 218.7	21.5	4.88	0.10	11.0	18.1	6.1	21.4
24.8 217.1	26.1	4.25	0.10	9.5	16.2	4.9	18.9

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

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(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃686O₃ (86O₃ (68O₃ (68O₃

Filter: 9	26.5	22.7	17.3	16.3	17.3	18.4	19.6	19.6	21.7	22.5	22.3	24.3	21.6	Filter: 9	20.0	19.7	19.2	19.4	18.9	18.2	19.2	24.4	22.5	20.0	19.0	19.7	20.4
	31.9	22.8	8.6	0.2	9.0	7.1	14.6	14.7	9.0	9.9	6.7	10.6	7.6	59.2	30.6	27.8	14.8	4.3	3.1	8.8	15.2	18.5	15.7	9.2	5.5	7.0	7.7
tde:27	25.1	19.9	13.8	13.8	13.0	12.9	13.1	13.5	15.4	16.0	15.5	16.9	16.2	ide:35	17.1	16.8	16.0	15.0	14.9	13.7	13.6	15.6	14.1	13.1	13.4	13.9	14.4
.9 Longitude: 271.2	0.20 16.2	0.21 10.1	0.20 8.7	0.19 11.0	0.20 13.0	0.19 12.0	0.16 10.7	0.15 14.6	0.17 20.4	0.19 20.9	0.14 16.1	0.13 11.8	0.14 8.4	3 Longitude: 3	0.20 18.3	0.19 14.9	0.18 14.2	0.18 15.7	0.18 18.5	0.20 20.2	0.18 18.8	0.15 15.4	0.14 14.4	0.14 17.0	0.17 18.8	0.16 16.5	0.15 9.9
Latitude:-47.9	6.15	92.9	7.19	7.26	7.05	6.70	6.31	6.05	9.00	6.05	5.93	5.52	4.88	te:-50.	6.73	6.62	6.91	7.34	7.74	7.81	7.41	6.80	6.53	6.71	6.97	87.9	5.94
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude:-50	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
atl ss32	40.5 252.3	39.1 249.5	37.7 243.6	36.3 236.9	35.0 233.4	33.7 232.5	32.4 230.1	31.1 226.7	29.8 224.7	28.5 223.5	27.3 222.4	26.0 220.8	24.8 219.1	atl ss36	40.9 249.9	39.5 246.5	38.1 243.5	36.7 242.4	35.4 241.8	34.0 239.2	32.7 236.5	31.3 235.3	30.0 233.9	28.7 230.6	27.4 228.7	26.1 228.5	24.8 227.9

Filter:	18.3	18.0	18.1	19.0	19.1	19.5	19.5	17.2	17.3	21.0	22.1	24.1	26.8	Filter:	18.7	19.7	19.5	19.8		20.9	19.5	18.3	18.0	23.9	23.8	4	26.9
6.5	•	13.5		18.4	18.3	14.5	9.6	6.5	6.7	8.4			3.2	3.8		7.7	\sim i	7.	23.0	22.7	14.6	5.8		•	•	0.8	•
Longitude:33	$\tilde{\omega}$	<u>.</u> ;	21.8		21.0	20.5	19.7	17.8	16.9	17.8	17.7	18.8	20.6	14e:31		18.4		15.0	15.4	•	14.3	12.9		16.5	17.4	17.4	0
ongit	•	14.3	16.5	17.4	16.8	15.7	14.3	12.9	11.9	11.5	11.5	10.9	8.7	ongiti		20.6	14.6	\sim	20.0	26.0	20.7	11.2	8.6	16.5	21.1	17.8	12.1
7 9.	0.13	0.19	•	0.20	0.20	0.19	0.17	0.13	0.11	0.12	0.11	0.10	0.11	.7 1	0.20	0.22	0.20	0.17	0.15	0.15	0.14	0.15	0.14	0.10	0.0	0.0	0.08
de:-50	5.51	6.40	6.99	7.21	7.24	7.26	7.29	7.18	69.9	5.94	5.22	9.	4.22	de:-50	6.43	7.23	7.23	6.74	6.23	5.95	5.90	5.83	5.51	4.94	4.34		3.59
Latitu			3.8		5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitu	2.6	3.2	3.8	4.6	5.6		•	10.0	12.1	14.7	17.8	21.5	26.1
\sim	50.	4	244.7	4	4	239.5	238.1	236.1	233.4	230.1	226.4	224.0	221.8	38	S	251.5	247.9	245.4	242.8	239.2	235.1	230.7	226.3	2	220.7	219.4	218.3
atl ss.	40.9	39.5	∞i	36.7	35.4	34.0	32.7	31.3	30.0	28.7	27.4	26.1	24.8	atl ss	o.	39.3	37.9	36.5	35.1	33.8	32.4	31.1	29.8		27.3	26.0	24.8

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vmr

(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (km) (K) (mb) vmr err err err err err err err err

							-																				
Filter: 9	22.5	24.6	21.6	20.1	13.6	13.8	13.8	13.5	17.6	18.8	18.4	19.3	20.3	Filter: I	15.9	16.2	16.1	16.6	16.4	17.0	17.0	15.0	14.7	17.2	17.9	19.0	21.5
	-15.6	-7.0	5.6	15.5	19.1	19.3	18.4	16.0	11.5	6.2	4.4	6.1	9.9	45.9 I	15.4	16.9	16.4	14.3	12.2	11.0	9.4	7.3	0.9	6.2	7.2	8.3	8.5
1de:26	19.2	16.4	14.4	13.9	9.6	10.1	10.1	6.6	12.5	13.5	14.0	14.6	15.6	tde:24	22.5	21.7	21.7	20.9	20.7	20.2	19.5	17.4	16.3	16.6	16.3	16.6	18.0
Longitude: 268.6	18.7	6.9	1.4	5.2	14.0	20.3	21.0	19.0	16.3	13.5	11.7	11.3	11.4	Longitude:2	15.4	17.9	15	17.9	16.3	15.4	14.5	12.7	11.2	10.5	10.6	11.5	12.0
	0.17	0.19	0.19	0.18	0.15	0.13	0.12	0.11	0.14	0.15	0.15	0.13	0.11	7 0:	0.17	0.21	0.20	0.22	0.22	0.17	0.14	0.13	0.13	0.14	0.14	0.11	0.11
de:-50	5.87	89.9	7.29	7.14	6.59	6.23	6.16	6.13	6.04	5.95	5.82	5.46	4.82	te:-51	6.91	7.21	7.26	7.18	7.04	6.90	98.9	88.9	6.74	6.36	5.81	5.20	4.55
Latitude:-50.8	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitud	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
40	244.1	240.7	238.7	237.3	234.7	231.7	228.7	225.7	223.8	222.6	221.6	220.7	219.8	11	239.4	236.5	233.7	231.4	229.6	228.3	227.4	226.3	225.0	223.6	222.1	221.0	220.1
atl ss40	40.4	39.1	۲.	36.3	35.0	33.7	32.4	31.1	29.8	28.6	27.3	26.0	24.8	atl ss41	40.4	39.1	37.7	36.4	35.1	33.8	3	31.2	29.9	28.7	27.4	26.1	24.9

15.4 15.3 15.7 15.8 18.0 18.9 20.5 Latitude:-53.8 Longitude:107.6 Filter: Latitude:-53.6 Longitude:130.3 Filter: 0.19 24.5 22.4 22.1 19.5 11.8 0.14 6.9 15.8 0.13 11.7 15.9 1 0.12 21.6 16.6 1 17.9 16.6 22.7 15.5 14.5 13.2 0.14 12.6 0.19 26.8 0.19 23.0 0.16 16.0 0.15 17.0 0.17 14.4 0.13 9.9 0.18 14.5 0.15 15.2 0.18 19.8 0.18 19.9 0.16 13.6 0.11 15.0 0.19 19.8 0.18 17.2 0.16 14.3 0.10 16.7 17.8 6.08 21.5 5.58 26.1 5.23 2.6 5.68 5.89 6.29 14.7 6.82 12.1 10.0 10.0 14.7 28.5 228.6 27.3 226.0 26.0 223.3 24.7 222.0 atl ss44 L 33.8 233.2 32.5 232.9 31.2 232.4 29.9 230.8 40.6 248.2 39.2 245.8 37.8 243.8 36.4 242.0 35.1 240.2 33.7 238.3 32.4 236.4 31.0 234.2 29.7 231.7 35.1 234.6 28.4 229.8 27.1 229.0

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E

vmr

(mb)

(km) (K)

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

16.9

7.4

24.6 225.2 26.1 5.31 0.12 10.1 11.5

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃ 668O₃ 668O₃ 686O₃ (km) (K) (mb) vmr err enr err err err err err err

Filter: 9	19.0	17.3	17.1	17.7	19.6	19.3	16.3	15.4	17.1	18.5	13.8	13.2	13.8	Filter: I	12.9	15.2	15.4	15.0	14.7	11.8	11.0	11.1	11.4	11.7	18.2	18.6	18.7
	10.4	8.5	8.0	12.6	19.5	16.6	6.1	2.2	9.1	13.5	9.4	9.9	6.4	1.0 F	8.7	7.7	10.2	12.6	11.2	8.9	3.4	3.8	6.1	7.9	8.9	9.0	5.6
Longitude: 39.6	21.7	15.4	13.8	14.2	13.0	13.5	12.0	11.3	12.0	12.8	10.0	8.6	10.3	tde:24	20.4	21.6	21.7	20.3	19.6	17.2	16.0	15.4	14.9	14.3	16.8	16.5	16.0
ongitu	37.6	21.2	8.2	6.7	21.0	26.9	23.9	20.2	19.0	15.2	9.3	8.6	0.16 11.6	Longitude:2	0.14 14.5	19.2	21.1	18.6	13.7	9.0	6.9	8.1	11.0	13.3	13.9	12.5	8.0
	0.19	0.19	0.18	0.16	0.14	0.14	0.17	0.19	0.16	0.13	0.19	0.18	0.16	.5	0.14	0.18	0.19	0.17	0.16	0.15	0.13	0.12	0.11	0.11	0.12	0.10	0.00
de:-54	6.28	6.80	7.02	6.74	6.23	6.14	6.48	82.9	6.75	6.70	87.9	6.58	5.87	te:-55	90.9	6.80	7.29	7.46	7.43	7.29	7.10	6.90	99.9	6.33	5.88	5.35	4.75
Latitude:-54.2	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude:-55	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
47	249.7	246.2	243.0	239.7	236.9	234.8	233.1	231.9	230.8	229.0	227.2	226.0	224.8	48	238.8	235.9	233.1	230.5	228.2	226.3	225.3	224.3	223.4	222.4	221.4	220.6	220.0
atl ss47	40.9	39.5	38.1	36.7	35.4	34.1	32.7	31.4	30.1	28.8	27.5	26.2	24.9	atl ss48	40.2	38.9	37.6	36.2	34.9	33.6	32.4	31.1	29.8	28.6	27.3	26.0	24.8

at2 sr02 40.1 238.3 38.8 234.4		63 48 48	.5 Lo 0.17 1 0.19 1	ongitu 17.2 16.9	Longitude:21 7 17.2 20.4 9 16.9 20.6	. છ	Filter: 1 13.3 13.3
230		6.72	.19 1	7.0	20.3	11.4	
228.		6.94			19.3	12.7	13.9
.8 227.		7.10	Ξ,	•	•		
.6 226		7.12	:	•	18.5	8.5	13.2
2.3 225.0	8.3	6.92	0.13 1	2.2	17.7	0.9	13.9
.0 223.4	10.0	95.9	0.12 1	3.8	17.2	5.9	14.0
9.7 221.9	12.1	6.12	0.11	6.5	16.7	7.4	14.2
28.5 221.3	14.7	5.68	0.11 1	7.7	16.9	8.1	15.9
7.2 221.0	17.8	5.33	0.11 1	5.1	15.7	6.4	15.7
26.0 221.0	21.5	4.99	0.13 1	1.5	14.9	4.9	15.4
24.7 221.3	26.1	4.64	0.13 1	10.6	14.5	6.3	15.6
at2 sr05	Latitu	$\boldsymbol{\mathcal{A}}$	8 Lo	Longiti	tde: 14	5.7	Filter:12
40.4 243.6		5.25	0.13 1	16.2	13.8	15.4	15.2
39.0 240.4	3.2	5.84	0.13 1	4.6	13.5	16.5	15.1
7.6 237.5	3.8	6.29	0.18 1	5.5	11.4	18.4	15.4
36.3 234.6	4.6	92.9	0.20	0.9	11.2	18.6	15.4
5.0 232.1	5.6	ε.	0.21	5.6	11.0	16.5	15.0
.62	8.9	7.83	0.17 1	5.5	10.2	13.7	15.1
2.3 229.8	8.3	8.04	0.16 1	5.5	10.0	11.4	14.9
.0 229.0	10.0	7.88	0.16 1	4.9	9.6	9.2	14.4
9.7 228.1	12.1	7.50	0.16 1	4.2	9.3	7.8	14.0
8.5 227.3	14.7	7.02	0.15 1	4.6	9.4		14.2
.2 226.4	17.8	6.48	0.14 1	5.7	10.3	6.6	15.3
5.9 225.8	21.5	5.90	0.13	4.7	11.9	10.3	17.2

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err

vmr

(mb)

(km) (K)

Alt Temp Press 48O₃ 48O₃ 668O₃ 668O₃ 686O₃ 686O₃ enr enr err enr err Table B.2 continued. (See notes on page 135.) err vmr (mb) (km) (K)

vmr

(mb)

Table B.2 continued. (See notes on page 135.) Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ enr enr err enr enr err еп vmr (mb) (km) (K)

Filter: 1	14.2	13.7	13.8	13.7	13.4	14.9	14.9	14.1	13.5	13.5	14.6	16.7	16.8	Filter:12													
	2.6	6.9	10.3	8.9	9.9	6.4	7.0	9.9	5.9	5.7	6.3	6.9	7.0	<i>ن</i> .													
ide:25	21.5	20.5	20.2	19.7	19.3	19.0	18.3	17.1	15.9	15.0	14.8	15.5	15.1	ide:161	15.4	12.8	12.6	12.2	11.1	11.2	11.2	11.3	11.4	11.1	10.9	11.6	12.2
7 Longitude: 252.3	0.22 16.1	0.20 17.2	0.17 18.6	0.17 16.2	0.17 13.1	0.16 12.1	0.15 12.1	0.13 11.6	0.12 11.1	0.11 11.1	0.12 11.5	0.15 12.4	0.16 13.3	0 Longitude: 1	0.16 20.2	0.15 15.5	0.15 13.8	0.14 15.5	0.15 17.6	0.14 18.5	0.13 18.7	0.10 19.1	0.09 19.6	0.09 18.9	0.09 16.4	0.10 13.7	0.09 11.9
le: 65.	8.11	7.38	95.9	6.43	89.9	98.9	6.77	6.53	6.25	00.9	5.73	5.40	4.97	le: 66.	6.03	6.10	6.22	6.43	6.61	9.60	6.31	5.87	5.50	5.30	5.22	5.10	4.87
Latitude: 65.7	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 66.0	5.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at2 sr24	40.1 244.4	38.7 240.3	37.3 235.8	36.0 231.0	34.7 226.3	33.4 224.9	32.2 224.1	30.9 223.2	29.6 221.6	28.4 219.8	27.1 218.8	25.9 219.5	24.7 220.0		40.3 244.1	38.9 240.3	37.6 236.3	36.3 233.0	34.9 231.0	33.6 229.3	32.3 227.7	31.0 226.5	29.8 225.3	28.5 224.4	27.2 223.9	26.0 223.4	24.7 222.9

1. 258.5	•	7														\sim													
sr30 Latitude: 66.1 Longitude: 93.3 1 258.5 2.6 5.52 0.14 24.9 22.7 6.6 7 255.1 3.2 6.26 0.15 19.9 21.9 4.4 2 251.4 3.8 6.55 0.15 20.3 21.5 6.9 2 251.4 3.8 6.58 0.16 20.2 21.5 6.9 2 246.4 4.6 6.68 0.16 20.6 20.8 10.7 4 241.3 5.6 6.89 0.16 20.6 20.8 10.7 4 241.3 5.6 6.89 0.16 20.6 20.8 10.7 4 241.3 5.6 6.89 0.16 20.6 20.8 10.7 4 231.1 8.3 6.99 0.13 13.4 18.9 4.9 4 235.7 10.0 6.53 0.12 11.9 17.8 4.3 5 223.9 17.8 4.60 0.12 11.9 17.8 8.6 2 221.8 21.5 4.06 0.12 15.4 16.3 8.6 2 221.8 21.5 4.06 0.12 15.4 16.5 8.2 2 247.1 3.2 5.13 0.12 15.8 15.3 16.4 2 247.1 3.2	;	tilter:		16.0	•	16.9	16.9	17.0	17.5	17.2	16.4	16.8	7	20.0	19.6	Filter:		17.6	17.9		16.8			16.2	17.0	17.0	17.1	16.7	17.2
sr30 Latitude: 66.1 Longitude: 1 258.5 2.6 5.52 0.14 24.9 22.7 2 251.4 3.8 6.55 0.15 19.9 21.2 2 251.4 3.8 6.55 0.15 20.3 21.2 2 251.4 4.6 6.68 0.16 20.6 20.2 4 241.3 5.6 6.89 0.16 20.6 20.2 4 241.3 5.6 6.89 0.16 20.6 20.2 1 238.1 6.8 7.08 0.15 16.7 19.3 2 237.1 8.3 6.99 0.13 13.4 18.3 2 237.1 8.3 6.99 0.13 13.4 18.3 2 237.1 8.3 6.99 0.13 13.4 18.3 2 221.8 1.2.1 5.90 0.14 12.0 16.7 2 222.9 17.8 4.60 0.12 11.9 17.5 2 221.8 21.5 4.06 0.12 15.4 16.5 2 247.1 3.2 5.13 0.12 15.8 15.5 2 247.1 3.2 5.13 0.12 17.6 15.8 2 247.1 3.2 5.13 0.12 21.6 14.8 4 241.8 4.6 5.61 0.12 21.6 14.8 4 233.5 8.3 5.	,	نہ		•	•				4.9	4.3	5.4		8.6	8.4		1 1.9	16.4	16.6	18.3			9.7	8.1	8.7	9.7	10.1	8.9	5.6	2.2
sr30 Latitude: 66.1 1 258.5 2.6 5.52 0.14 2 251.4 3.8 6.55 0.15 2 251.4 3.8 6.55 0.15 2 251.4 4.6 6.68 0.16 4 241.3 5.6 6.89 0.16 1 238.1 6.8 7.08 0.15 1 238.1 6.8 7.08 0.15 2 257.1 8.3 6.99 0.15 2 257.2 10.0 6.53 0.12 2 257.3 17.8 4.60 0.12 2 257.5 14.7 5.25 0.13 3 251.3 2.15 4.06 0.12 2 27.8 21.5 4.06 0.12 2 247.1 3.2 5.13 0.12 3 251.3 2.6 4.79 0.12 4 241.8 4.6 5.61 0.12 4 241.8 4.6 5.61 0.12 4 233.5 8.3 5.79 0.10 5 255.0 14.7 4.70 0.06 5 225.0 14.7 4.70 0.06 5 221.9 17.8 4.36 0.06 2 221.9 17.8 4.36 0.06		٠.							18.9	17.8	16.7		15.8		16.3	ide: 25				13.7	12.6	11.4	11.2	11.6	12.0	•	•		13.2
sr30 Latitude: 66.1 1 258.5 2.6 5.52 0.14 2 251.4 3.8 6.55 0.15 2 251.4 3.8 6.55 0.15 2 251.4 4.6 6.68 0.16 4 241.3 5.6 6.89 0.16 1 238.1 6.8 7.08 0.15 1 238.1 6.8 7.08 0.15 2 257.1 8.3 6.99 0.15 2 257.2 10.0 6.53 0.12 2 257.3 17.8 4.60 0.12 2 257.5 14.7 5.25 0.13 3 251.3 2.15 4.06 0.12 2 27.8 21.5 4.06 0.12 2 247.1 3.2 5.13 0.12 3 251.3 2.6 4.79 0.12 4 241.8 4.6 5.61 0.12 4 241.8 4.6 5.61 0.12 4 233.5 8.3 5.79 0.10 5 255.0 14.7 4.70 0.06 5 225.0 14.7 4.70 0.06 5 221.9 17.8 4.36 0.06 2 221.9 17.8 4.36 0.06	•	ongitu	4	6	0	22.1	Ö	6.			12.0	13.4		16.3		ongitu				23.5		7	15.9	15.9			15.7	12.2	8.1
sr30 Latitude: 7.258.5 2.6 5.5 7.258.1 3.2 6.2 5.5 7.258.1 3.2 6.5 5.5 7.258.1 3.8 6.5 6.8 8.246.4 4.6 6.6 6.8 7.0 6.5 7.237.1 8.3 6.9 7.237.1 8.3 6.9 7.237.2 14.7 5.2 5.22.8 21.5 4.0 6.251.3 2.6 4.7 5.2 5.1 3.7 5.1 5.3 5.3 5.1 5.3 5.1 5.3 5.1 5.3 5.1 5.3 5.1 5.3 5.3 5.1 5.3 5.1 5.3 5.3 5.1 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.1 5.3 5.3 5.3 5.3 5.3 5.3 5.3 5.3 5.3 5.3	,	_	Ξ.		Ξ.	1.	Ξ.	Τ.	-:	Ξ:	Τ.	Ξ.			0.13	.3 L	Ξ.	Ξ.			0.11		$\overline{}$	0.10			0.08	0.08	0.0
sr30 2 251.4 2 251.4 2 251.4 4 241.3 1 238.1 1 238.1 1 238.1 1 238.1 1 238.1 1 238.1 2 221.8 0 219.7 2 221.8 1 239.2 1 239.2 1 239.2 1 239.2 2 221.8 8 244.2 8 244.2 8 247.1 8 247.1 8 227.8 1 239.6 1 239.6 1 239.6 2 221.8			ن.	4	ĸ.	89.9	68.9	7.08		3.			4.60	4.06	\mathcal{C}	te:	4.7	5.1	2	5.61	5.81	5.91	5.79	5.48		4.70	ι	0.	3.74
sr30 258.5 7 258.5 7 258.5 8 246.4 4 241.3 1 238.1 7 237.1 7 237.1 7 237.1 8 244.2 8 244.2 9 221.8 9 247.1 1 239.2 1 239.2 1 239.2 1 239.2 2 221.8 8 227.8 8 244.2 9 247.1 8 244.2 9 251.8 1 239.2 1 239.2 1 239.2 1 239.2 1 239.2 1 239.2 2 221.8 8 227.8		Latitu	2.6	3.2	3.8					10.0	12.1		17.8	21.5	26.1	Latitu	2.6	3.2	3.8					10.0	12.1	14.7		21.5	26.1
$\frac{1}{2}$ 1 L 2 8 4 1 L 4 0 L 2 2 0 0 $\frac{1}{2}$ 8 4 1 L 4 1 8 2 2 0	,	30	58.	5	51	46	41	38	3	35	31	27	23	21	19.7		51.3	-	2	241.8	39	236.4	33.	30.	27.	7	21.	18.	214.8
	,	at2 sr	40.1	∞	۲.	S	4	33.1	31.7	0	29.0	27.7	9	YO.	24.0	at2 sr	39.6	7	∞.	5.4	34.1	۲.		30.1	∞	۲.	26.2		23.7

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

enr enr err enr err

vmr err

ξ

(km)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃ 668O₃ 686O₃
(km) (K) (mb) vmr err enr enr enr err enr err

Filter: 1	18.8	17.8	17.6	18.3	18.6	18.0	17.6	17.8	18.3	18.5	18.4	17.7	17.3	Filter: 12													
	10.8	9.0	9.0	10.0	9.5	0.9	3.2	4.0	6.9	7.8	5.0	0.0	-3.5	.3 F													
ıde: 89	22.6	22.4	22.4	21.4	20.9	20.2	19.3	18.4	17.6	16.7	16.6	15.7	15.1	de: 21	17.3	15.9	14.1	14.0	12.4	12.1	10.5	10.5	10.7	12.4	12.6	13.8	15.1
I Longitude: 89.4	0.12 10.4	0.16 20.1	0.17 22.1	0.18 19.2	0.19 16.4	0.18 15.7	0.15 15.6	0.13 14.8	0.11 13.3	0.10 13.0	0.10 13.5	0.12 10.8	0.13 5.8	2 Longitude:	0.12 19.5	0.12 17.2	0.13 16.8	0.13 19.0	0.11 20.2	0.11 18.4	0.10 15.0	0.09 13.9	0.08 16.2	0.08 19.9	0.08 21.7	0.08 19.8	0.08 15.4
Latitude: 67.1	4.96	5.43	5.90	6.37	08.9	7.02	6.87	6.27	5.52	4.96	4.65	4.41	4.10	le: 67.2	4.62	4.81	5.03	5.39	5.70	5.71	5.40	5.05		4.56	4.30	4.04	3.80
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude	5.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0		14.7	17.8	21.5	26.1
at2 sr40	40.2 259.3	38.7 254.0	37.3 249.0	35.9 245.8	34.5 243.3	33.1 241.1	31.7 239.4	\sim	29.1 233.9	27.7 230.1	4	25.2 224.6	23.9 221.3	at2 sr43	39.5 248.5	38.1 245.2	∞.	35.4 239.9	_	32.7 235.7	31.4 231.7	30.1 227.0	28.8 222.8	27.5 220.0	26.3 217.7	25.1 215.7	23.8 214.0

Filter: 1	6.3	5.8	5.0	5.1	5.0	4.6	4.4	4.3	4.4	4.9	5.2	6.5	8.4	ter:12													
	3.7	-0.2	2.1 1	5.9	7.1 1	6.4	5.4	4.8	4.7	4.9	5.8 1	6.4	4.2	1 Filte													
ide:312.9	21.7	21.7	20.6	20.3	20.0	19.0	18.2	17.1	16.3	15.7	15.1	15.2	15.9	de:245	15.7	14.1	13.6	12.5	12.2	11.3	11.2	10.7	10.3	10.3	11.0	11.1	13.4
Longitude:3	12.3	10.8	12.8	14.8	14.4	13.1	12.1	12.1	12.3	12.5	13.2	13.6	11.2	Longitu	25.4	S	23.1	19.7	17.3	16.6	16.1	14.9	13.7	13.4	14.1	14.8	14.3
رم	0.17	0.17	0.18	0.19	0.19	0.18	0.17	0.15	0.14	0.14	0.14	0.15	0.16	4.	0.15	0.16	0.16	0.14	0.13	0.11	0.10	0.10	0.00	0.09	0.0	0.00	0.09
6	5.92		6.50		7.20	7.45	7.39	7.05	6.70	6.41	6.12	5.73	5.26	de: 67.	5.67	6.44	6.81	92.9	6.48	6.10	5.81	5.63	5.49	5.31	5.08	4.80	4.50
Latitude:		3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitue	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8		26.1
sr46	3	36	34	232.3	230.2	228.5		226.0	225.3	224.9	224.8	224.6	224.2	46	239.9	236.5	233.3	231.1	228.9	226.8	224.6	222.6	220.9	219.2	218.6	218.8	219.0
at2 sr	40.5	9	37.8	36.4	35.1	33.8	32.5	1.2	0.0	28.7	7.4	26.1	24.9	at2 sr4	40.0	7	7.3	36.0	34.7	3.4	32.2	6.0	9.	8.4	27.1		24.7

(mb) vmr err

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err enr enr enr err err

Filter: 1	16.4	15.7	15.8	16.2	17.3	16.9	16.4	16.0	16.6	16.8	15.4	15.3	18.6	Filter: 1	21.2	20.0	19.6	18.7	18.5	20.4	19.0	18.5	19.2	20.3	20.4	23.6	23.1
	9.6	12.5	13.4	13.4	12.6	10.4	7.5	5.7	5.2	5.0	4.5	3.8	3.0	3	15.0	10.7	8.3	7.5	6.9	9.9	6.5	6.5	7.0	7.9	8.1	5.0	3.2
tde:17	21.5	20.5	20.5	20.9	21.3	21.0	19.3	18.0	17.3	16.5	15.3	14.6	16.3	ide: 82	24.7	22.9	23.1	22.0	21.7	21.5	19.8	18.6	18.1	17.9	17.2	18.6	18.7
Longitude: 176.9	10.8	10.6	12.1	15.7	, 19.0	19.0	16.1	. 13.3	12.1	12.0	12.3	12.4	11.5	Longitude:	17.0	18.9	. 21.9	22.9	20.8	17.9	16.2	15.6	15.1	14.8	14.7	13.3	14.0
	0.15	0.19	0.19	0.19	0.17	0.17	0.14	0.14	0.13	0.12	0.13	0.15	0.15	4	0.12	0.14	0.14	0.14	0.15	0.16	0.13	0.11	0.10	0.11	0.11	0.11	0.13
de: 67	5.83	5.87	5.94	90.9	6.22	6.41	6.56	6.53	6.30	5.95	5.58	5.26	4.95	te: 68.	4.81	5.10	5.54	6.14	69.9	6.85	6.56	5.97	5.36	4.86	4.52	4.33	4.02
Latitude: 67.6	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude:	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at2 sr52	40.3 242.9	38.9 238.7	37.6 234.6	36.2 231.6	34.9 229.8	33.6 228.2	32.3 227.2	31.0 226.2	29.8 225.4	28.5 224.6	27.2 224.0	25.9 223.7	24.7 223.4	at2 sr64	40.3 261.6	38.8 257.2	37.4 252.6	35.9 248.1	34.5 244.3	33.2 243.6	31.8 241.9	30.4 237.6	29.1 233.1	27.8 228.8	26.5 225.2	25.2 222.9	24.0 220.7

r: I	7.	4.	6.	7	ι.	4.	∞.	7	z.	9.	0.	7	7	7:	ϵi	0.	7	-:	∞.	6.	9.	9.	7	ε:	3	7.	4.
Filte	17	17	16	17	16	16	16	18	17	17	18	18	18	ilte	19	6	19	20		17	16	16	17	17	14	11	12
0	4.5	5.5	1.0	8.7	2.8	16.5	2.5	7.0	2.9	0.7	3.3	2.9	9.01	~	8.7	7.1	8.9	0.7	9.0	5.8	3.3	2.5	4.4	4.2	3.3	1.7	8.7
38.	T	1	_		Τ		_				_	_	_	11.2				_							'	-1	'
rde:2	16.7	15.0	14.6	13.5	12.2	11.9	12.4	13.6	12.8	12.8	13.1	13.6	13.8	tde: 1	24.3	23.2	23.2	22.7	21.9	20.9	19.2	18.1	17.6	17.2	14.5	12.0	12.0
Longitu	16.9		12.8	15.1	19.7	22.7	20.7	16.6	14.3	15.4	17.3	17.0	15.5	ongitı	16.6	14.2	15.4	17.5	18.0	16.6	13.7	12.9	15.7	15.6	6.4	-4.1	-1.7
	0.15	0.15	0.15	0.15	0.14	0.13	0.12	0.12	0.11	0.10	0.10	0.10	0.10	ľ	0.11	0.13	0.13	0.16	0.16	0.16	0.16	0.15	0.14	0.13	0.13	0.15	0.15
68.8	~	_	7	_	Ī.	2		_	2	_		_	_	68.9				-					9		~	7	7
_		5.99	4	96.9	7.12	6.9	69.9	6.37	5.9	3	5.10	4.70	4.30	de: 6	4.7	5.06	5.46	5.98	6.54	7.04	7.48	7.80	7.	7.29	6.77	6.12	5.3%
Latitude:	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
2	42.	238.2			229.5	224.8	222.9	222.7	221.9	220.9	220.2	219.9	219.6	81	249.5	245.2	240.8	236.8		232.0	229.5	227.6	226.3	224.7	223.3	222.5	221.8
2 sr	=	8.7	7.4	-:	4.8	3.5	2.2	0.9	9.7	4.	7.2	6	7	2 sr	3	9	7.5		∞.	3.5	2.2	0.9	9.6	8.3	7.0.7	5.8	4.5
at2	40	38	'n	36	35	33	37	3	53	28	2	25	24	at2	40	38	3	36	%	č	37	3(53	7	6	5	7

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$

enr enr err enr err

E

(mb) vmr

(km) (K)

	3 686O ₃	enr err
<u>2</u>	098	enr
age 135.	3668O ₃ 686O ₃	enr err
s on p	66803	enr
d. (See notes on page 1	4803	еп
d. (Se	4803	vmr
continued	Press	(mb)
Table B.2 co	Temp	(<u>K</u>
Table	Alt 7	(km)

Filter: 12													lter: 1	15.2	15.0	15.2	15.3	15.8	15.6	14.5	14.1	14.0	15.6	16.1	16.9	18.8	
													.9 FI	7.8	6.6	11.5	12.3	11.8	6.7	7.0	5.2	4.7	5.1	9.6	0.9	5.2	
Longitude: 295.8	15.4	11.7	11.8	11.5	5 6.94 0.13 19.0 10.8	10.7	9.3	9.3	9.5	12.7	13.0	15.0	de:316	21.8	20.8	50.6	50.6	20.4	19.7	17.9	16.7	15.7	14.8	14.0	14.3	16.3	
ongitu	11.2	15.8	18.7	19.4	19.0	17.4	15.5	15.7	19.0	22.7	22.4	20.0	ongitu	17.7	17.6	17.8	18.9	19.4	17.5	14.0	11.4	10.3	6.6	10.5	12.4	13.0	
	0.15	0.16 15.8	0.16 18.7	0.15	0.13	0.13	0.13	0.12	0.11	0.29	0.31	0.52	.3 L	0.15	0.18	0.18	0.18	0.17	0.17	0.15	0.15	0.15	0.14	0.14	0.15	0.15	
Latitude:-49.5	5.49	6.32	6.82	6.94	6.94	6.94	6.94	6.82	6.55	6.26	6.24	6.39	de:-47	6.65	98.9	6.99	7.07	7.14	7.29	7.49	7.62	7.60	7.36	6.81	5.94	5.03	
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	∞.	10.	12.	14.7	17.8	21.5	Latit	5.	3.7	3.8	4.	5.0	9.9	∞	10.0	12.1	14.7	17.8	21.5	26.1	
505	240.6	237.0	233.6	230.7	9 228.2	225.9	224.3	223.4	222.6	221.8	220.9	220.0	H	247.1	243.7	240.0	36.5 236.3	232.5	228.7	226.1	225.8	225.4	224.2	222.7	221.3	220.4	
at2 ss05	40.3	38.9	37.6	36.2	34.9	33.7	32.4	31.1	29.8	28.6	27.3	26.1	at2 ss	40.6	39.3	37.9	36.5	35.2	33.9	32.6	31.3	30.0	28.8	27.5	26.2	25.0	
ilter: I	20.9	20.0	19.0	19.6	21.8	21.7	21.6	22.1	22.0	22.3	23.1	24.5	24.9	Iter: I	15.3	15.4	14.9	14.3	16.0	15.8	14.8	14.4	14.2	14.7	14.9	15.2	18.0
.8 Filter: I	9.		9.5 19.0		11.0 21.8	10.4 21.7	8.4 21.6		5.1 22.0		3.5 23.1	3.1	3.4	2	9.1						9.7 14.8					5.7 15.2	
	9.	7.0	9.5	11.0	11.0				5.1	4.5	3.5	3.1	3.4	2	9.1	20.8	17.7	13.4	10.1		6.7						
	22.4 25.9 7.6	21.1 24.7 7.0	17.6 22.4 9.5	16.3 22.0 11.0	22.6 11.0	21.8 10.4	21.1 8.4	20.3 6.2	19.3 5.1	18.6 4.5	18.7 3.5	3.1	3.4	2	9.1	21.1 20.8	20.3 17.7	20.0 13.4	20.1 10.1	19.8 9.0	18.4 9.7	17.0 10.4	15.8 9.2	15.1 6.8	11.4 14.5 5.2	12.1 13.9 5.7	
I Longitude: 98.8	0.12 22.4 25.9 7.6	0.14 21.1 24.7 7.0	0.17 17.6 22.4 9.5	0.17 16.3 22.0 11.0	0.18 17.4 22.6 11.0	0.16 18.7 21.8 10.4	0.15 18.4 21.1 8.4	0.13 16.5 20.3 6.2	0.13 14.4 19.3 5.1	0.12 13.8 18.6 4.5	0.11 14.6 18.7 3.5	0.12 15.5 19.6 3.1	0.12 15.5 19.6 3.4	0 Longitude: 26.2	0.14 22.3 22.0 19.1	0.16 19.4 21.1 20.8	0.16 16.1 20.3 17.7	0.17 15.6 20.0 13.4	0.16 17.4 20.1 10.1	0.16 18.2 19.8 9.0	0.15 17.1 18.4 9.7	0.14 15.3 17.0 10.4	0.14 13.3 15.8 9.2	0.13 11.9 15.1 6.8	0.13 11.4 14.5 5.2	0.15 12.1 13.9 5.7	0.15 12.8 15.2 7.2
I Longitude: 98.8	0.12 22.4 25.9 7.6	0.14 21.1 24.7 7.0	0.17 17.6 22.4 9.5	0.17 16.3 22.0 11.0	0.18 17.4 22.6 11.0	0.16 18.7 21.8 10.4	0.15 18.4 21.1 8.4	0.13 16.5 20.3 6.2	0.13 14.4 19.3 5.1	0.12 13.8 18.6 4.5	0.11 14.6 18.7 3.5	0.12 15.5 19.6 3.1	0.12 15.5 19.6 3.4	0 Longitude: 26.2	0.14 22.3 22.0 19.1	0.16 19.4 21.1 20.8	0.16 16.1 20.3 17.7	0.17 15.6 20.0 13.4	0.16 17.4 20.1 10.1	0.16 18.2 19.8 9.0	0.15 17.1 18.4 9.7	0.14 15.3 17.0 10.4	0.14 13.3 15.8 9.2	0.13 11.9 15.1 6.8	0.13 11.4 14.5 5.2	0.15 12.1 13.9 5.7	0.15 12.8 15.2 7.2
Latitude: 69.1 Longitude: 98.8	2.6 5.21 0.12 22.4 25.9 7.6	3.2 5.62 0.14 21.1 24.7 7.0	3.8 5.91 0.17 17.6 22.4 9.5	4.6 6.22 0.17 16.3 22.0 11.0	5.6 6.60 0.18 17.4 22.6 11.0	6.8 6.97 0.16 18.7 21.8 10.4	8.3 7.08 0.15 18.4 21.1 8.4	10.0 6.80 0.13 16.5 20.3 6.2	12.1 6.27 0.13 14.4 19.3 5.1	14.7 5.71 0.12 13.8 18.6 4.5	17.8 5.19 0.11 14.6 18.7 3.5	21.5 4.59 0.12 15.5 19.6 3.1	26.1 4.01 0.12 15.5 19.6 3.4	Latitude:-50.0 Longitude: 26.2	2.6 5.66 0.14 22.3 22.0 19.1	3.2 5.71 0.16 19.4 21.1 20.8	3.8 5.98 0.16 16.1 20.3 17.7	4.6 6.37 0.17 15.6 20.0 13.4	5.6 6.77 0.16 17.4 20.1 10.1	6.8 7.08 0.16 18.2 19.8 9.0	8.3 7.26 0.15 17.1 18.4 9.7	10.0 7.26 0.14 15.3 17.0 10.4	12.1 7.14 0.14 13.3 15.8 9.2	14.7 6.94 0.13 11.9 15.1 6.8	17.8 6.61 0.13 11.4 14.5 5.2	21.5 6.04 0.15 12.1 13.9 5.7	26.1 5.26 0.15 12.8 15.2 7.2
Latitude: 69.1 Longitude: 98.8	2.6 5.21 0.12 22.4 25.9 7.6	3.2 5.62 0.14 21.1 24.7 7.0	3.8 5.91 0.17 17.6 22.4 9.5	4.6 6.22 0.17 16.3 22.0 11.0	5.6 6.60 0.18 17.4 22.6 11.0	6.8 6.97 0.16 18.7 21.8 10.4	8.3 7.08 0.15 18.4 21.1 8.4	10.0 6.80 0.13 16.5 20.3 6.2	12.1 6.27 0.13 14.4 19.3 5.1	14.7 5.71 0.12 13.8 18.6 4.5	17.8 5.19 0.11 14.6 18.7 3.5	21.5 4.59 0.12 15.5 19.6 3.1	26.1 4.01 0.12 15.5 19.6 3.4	Latitude:-50.0 Longitude: 26.2	2.6 5.66 0.14 22.3 22.0 19.1	3.2 5.71 0.16 19.4 21.1 20.8	3.8 5.98 0.16 16.1 20.3 17.7	4.6 6.37 0.17 15.6 20.0 13.4	5.6 6.77 0.16 17.4 20.1 10.1	6.8 7.08 0.16 18.2 19.8 9.0	8.3 7.26 0.15 17.1 18.4 9.7	10.0 7.26 0.14 15.3 17.0 10.4	12.1 7.14 0.14 13.3 15.8 9.2	14.7 6.94 0.13 11.9 15.1 6.8	17.8 6.61 0.13 11.4 14.5 5.2	21.5 6.04 0.15 12.1 13.9 5.7	26.1 5.26 0.15 12.8 15.2 7.2
Latitude: 69.1 Longitude: 98.8	0.12 22.4 25.9 7.6	3.2 5.62 0.14 21.1 24.7 7.0	3.8 5.91 0.17 17.6 22.4 9.5	4.6 6.22 0.17 16.3 22.0 11.0	5.6 6.60 0.18 17.4 22.6 11.0	6.8 6.97 0.16 18.7 21.8 10.4	8.3 7.08 0.15 18.4 21.1 8.4	10.0 6.80 0.13 16.5 20.3 6.2	12.1 6.27 0.13 14.4 19.3 5.1	14.7 5.71 0.12 13.8 18.6 4.5	17.8 5.19 0.11 14.6 18.7 3.5	21.5 4.59 0.12 15.5 19.6 3.1	26.1 4.01 0.12 15.5 19.6 3.4	Latitude:-50.0 Longitude: 26.2	2.6 5.66 0.14 22.3 22.0 19.1	3.2 5.71 0.16 19.4 21.1 20.8	3.8 5.98 0.16 16.1 20.3 17.7	4.6 6.37 0.17 15.6 20.0 13.4	5.6 6.77 0.16 17.4 20.1 10.1	6.8 7.08 0.16 18.2 19.8 9.0	8.3 7.26 0.15 17.1 18.4 9.7	10.0 7.26 0.14 15.3 17.0 10.4	12.1 7.14 0.14 13.3 15.8 9.2	14.7 6.94 0.13 11.9 15.1 6.8	17.8 6.61 0.13 11.4 14.5 5.2	21.5 6.04 0.15 12.1 13.9 5.7	26.1 5.26 0.15 12.8 15.2 7.2

enr enr err enr err

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(mb)

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(km)

Alt Temp Press 48O₃ 48O₃ 668O₃ 668O₃ 686O₃686O₃

Table B.2 continued. (See notes on page 135.) Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ 686

Filter: 12	16.0	15.4	15.2	15.3	16.1	17.1	16.9	16.7	17.2	18.4	26.2	30.1	Filter: I	14.4	14.6	15.0	16.1	16.2	15.5	14.7	13.8	13.0	12.9
1.6	17.4	14.6	13.0	13.7	16.8	20.1	18.6	12.1	8.9	16.6	37.6	58.1	1.3	8.6	9.3	12.6	17.3	18.4	15.1	9.7	4.2	-0.3	-1.2
xde:24	14.7	13.1	16 15.6 11.7 13.0	11.6	11.4	11.3	11.4	11.7	12.3	12.2	21.3	23.4	18: 18	20.7	20.7	20.6	20.1	20.5	19.6	18.4	17.0	15.5	14.6
ongitı	0.16 25.3	19.3	15.6	15.1	17.4	20.7	21.5	19.1	16.2	15.2	18.9	30.4	ongitı	15.8	14.4	14.8	16.6	18.1	18.1	16.5	13.0	9.4	9.2
7 /:	0.16	0.17	0.16 15.6	0.16	0.14	0.12	0.11	0.10	0.00	0.00	0.40	0.35	. Z	0.20 15.8	0.20	0.19	0.15	0.14	0.14	0.12	0.12	0.13	0.13
de:-46.	6.23	6.73	66.9	6.94	6.61	6.21	5.92	5.73	5.43	4.93	4.31	3.81	de:-46.	6.62	87.9	6.65	6.36	6.29	6.50	8.78	68.9	6.81	6.55
Latitue	2.6	3.2	3.8 6.99 0.16	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	Latitue	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7
14	248.8	245.0		236.1		230.2	228.0	226.1	224.3	222.7	221.4	220.3	17	241.6	237.7	233.9	230.9	228.3	225.8	224.3	223.0	222.2	221.8
at2 ss	40.5 248.8	39.1	37.7	36.3	35.0	33.7	32.4	31.1	29.8	28.6	27.3 221.4	26.0	at2 ss	40.4	39.0	37.7	36.4	35.1	33.8	32.5	31.2	30.0	28.7

16.8 Latitude: -45.3 Longitude: 45.8 Filter: 16.3 15.2 14.515.5 0.15 18.5 22.3 18.1 20.8 18.0 20.6 15.9 Latitude:-45.8 Longitude:113.5 Latitude:-43.6 Longitude: 202.5 18.0 19.2 10.3 20.7 14.0 12.6 12.2 12.0 13.3 12.2 12.5 0.16 12.4 0.14 12.5 0.18 16.9 0.16 19.0 0.12 17.5 0.19 9.2 0.16 14.8 0.16 17.8 0.16 16.7 0.15 11.9 0.11 14.8 0.13 10.7 0.15 19.7 0.16 19.3 0.16 13.7 0.15 18.7 0.15 23.2 2.6 6.07 3.2 6.25 2.6 5.68 4.6 6.29 6.37 17.8 6.12 3.8 6.34 10.0 6.59 12.1 6.63 14.7 6.54 5.6 34.1 230.4 32.8 229.4 31.5 228.6 30.2 227.4 27.6 224.8 at2 ss23 34.4 236.9 33.0 235.4 at2 ss32 38.4 240.3 37.1 238.2 38.9 241.6 37.5 238.6 28.9 225.9 41.2 246.9 40.3 246.1 39.8 243.6 35.7 237.6 36.2 236.3 34.8 234.0 35.4 231.7 30.9 225.7

Table B.2 continued. (See notes on page 135.)

Temp
(km) (K) (mb) vmr err enr enr enr err
at3 sr02 Latitude:-64.6 Longitude: 50.9 Filter:12
41.5 253.0 2.6 6.19 0.19 27.0 16.6 12.2 16.9
40.0 250.2 3.2 6.43 0.17 24.2 14.9 12.3 16.8
6.55 0.15 20.8 13.2 11.6
0.14 18.6 12.0 12.5
6.8 6.69 0.14 17.6 11.5 12.4
8.3
0.14 13.5 10.4 8.3
10.3 8.9
239.3 14.7 6.57 0.12 16.8 10.5
237.8 17.8 6.30 0.12 19.1 12.0 18.1
236.6 21.5 6.18 0.11 14.4 11.6
235.5
at3 sr03 Latitude:-64.8 Longitude: 27.9 Filter: 9
6.38 0.27 77.6 38.0 50.9
18.5
254.3 3.8 7.41 0.27 36.3 21.2 -
7.56 0.25 20.6 19.0 -24.7
7.42 0.27 9.2 16.1 -15.1
6.8 7.34 0.28 0.6 14.2 -3.8
8.3 7.57 0.27 -1.8 13.9
10.0 7.91 0.19 0.7 10.9 -3.4 1
12.1 8.10 0.18 6.4 11.1 -3.3 1
14.7 8.14 0.18 15.2
17.8 8.08 0.19 22.7 12.8 5.9
242.5 21.5 7
1.2

enr enr err enr enr err

vmr err

(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (Km) (Km) (mb) vmr err enr enr enr enr err

Filter: 12	21.7	20.3	20.7	21.3	21.0	20.2	21.7	22.4	22.1	ilter:12	19.2	18.3	18.3	18.1	Filter: 9	23.3	22.0	19.0
	7.7	17.6 10.9	14.5	18.3	18.2	14.9	12.6	5.4	5.6	3.6	13.7	11.7	8.9	9.0	0.7	16.9	9.0	-6.8
1de:34	20.7	17.6	17.4	14.1	13.3	12.7	12.6	13.9	13.8	ide:27	13.3	12.8	12.6	12.6	ide:25	18.1	17.5	16.3
Latitude:-64.9 Longitude:342.3	0.13 16.9	0.12 33.4	0.12 33.9	0.12 27.0	0.12 20.3	0.12 16.8	0.12 16.7	0.06 8.9	0.9 90.0	Latitude:-65.2 Longitude:273.6	0.07 22.8	0.07 18.0	0.06 12.9 12.6 8.9	0.05 4.6	3 Longitude: 250.7	0.0	0.08 30.7	0.10 1.7
de:-64	4.35	4.60	3.8 5.00	5.46	5.85	6.12	60.9	3.12	2.84	de:-65.	3.74	3.27	2.83	2.42	atitude:-65.3	17.8 3.28	2.96	2.61
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	21.5	26.1	Latitue	14.7	17.8	21.5 2.83 (26.1	Latitue	17.8	21.5	26.1
at3 sr05	41.0 274.6	39.4 271.1	37.9 267.5	36.4 265.4	34.9 265.5	33.4 265.6	31.9 263.7	24.8 232.3	23.5 226.6	at3 sr08	27.2 236.5	25.9 230.2			at3 sr09	26.3 233.2	25.0 231.3	23.7 228.1

at3 sr11	Latitu	Latitude:-66.4	·	ongitı	Longitude: 290.5		Filter:12
	2.6	3.23	0.09 24	24.0	26.9	30.2	29.3
38.8 277.9	3.2	3.75	0.10	30.7	22.2	17.2	26.8
37.3 277.6	3.8	4.32	0.10	33.8	19.7	7.4	24.8
35.7 275.1	4.6	4.86	0.11	29.5	18.7	0.9	24.2
34.1 269.0	5.6	5.22	0.11	23.3	16.4	8.6	23.5
32.6 262.2	8.9	5.23		19.3	15.8	10.5	23.7
31.2 254.0	8.3	4.90	0.10	18.2	13.0	11.2	20.2
29.8 245.3	10.0	4.42		19.9	13.3	12.6	20.6
28.4 237.9	12.1	3.86		23.7	13.9	15.2	21.5
27.1 232.9	14.7	3.37	0.07	23.5	15.7	14.4	23.9
25.8 228.2	17.8	3.05	90.0	14.6	14.6	7.4	22.6
24.5 223.0	21.5	2.82	90.0	3.0	13.1	0.1	19.1
23.2 217.4	26.1	2.46	0.05	10.0	14.0	9.2	20.8
at3 sr12	Latitu	Latitude:-66.	.S. L	ngiti	Longitude: 26	67.6 I	Filter:12
26.0 229.7	17.8	3.00	0.06 23.9	23.9	13.7	15.9	20.5
24.7 225.9	21.5	2.95	90.0	5.8	12.0	1.0	17.3
23.4 221.2	26.1	2.69	0.05	9.0	11.6	-0.7	16.9

enr enr err enr enr err

vmr err

(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (km) (K) (mb) vmr err err err err err err err err

a	4	4	38	'n	36	κ	$\tilde{\omega}$	32	3(23	7	7	7	at	4	4	33	'n	3,	8	ξ	33	3	23	7	7	7
Filter: 12	20.4	18.3	18.6	18.8	19.0	17.8	16.1	15.9	16.6	17.1	18.0	19.0	18.1	Filter: 9	39.1	30.4	30.9	22.2	20.7	21.0	20.6	19.9	16.4	15.5	15.9	20.6	20.5
	5.1	4.3	8.9	13.4	17.7	14.3	9.4	7.4	8.3	11.5	16.4	18.0	12.8	5.9	6.5	33.4	38.0	25.1	14.5	15.8	20.7	16.2	4.9	-0.8	3.4	11.4	6.6
4de: 15	21.6	16.8	16.6	15.1	13.9	12.4	10.7	10.5	10.5	10.5	11.0	11.9	11.2	17:apr	15.5	25.8	29.0	20.7	21.1	21.5	16.0	15.3	11.9	11.3	11.2	14.0	13.5
.9 Longitude: 198.6	0.11 26.8	0.11 25.4	0.11 24.8	_	0.11 25.6	0.11 20.9	0.12 16.2	0.12 14.5 10.5 7.4 15.9	0.11 15.3	0.11 17.4	0.12 20.2	0.13 19.8	0.11 13.1	9 Longiti	0.19-37.1	0.20 10.4	0.19 28.4	0.21 31.3	0.25 32.0	0.26 35.8	0.24 38.4	0.25 32.4	0.19 21.3	0.19 14.9	0.18 15.8	0.16 18.7	0.14 13.9
Latitude:-66.9	4.02	4.39	4.65	4.80	5.08	5.66	6.25	6.51	6.48	6.55	98.9	6.90	6.26	Latitude:-66.	4.89	5.11	5.29	5.46	5.64	5.79	5.99	6.42	86.9	7.31	7.29	7.04	6.63
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1		17.8			Latitue		3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 sr15	41.3 260.4		38.3 254.6	36.9 251.3	35.5 247.7	34.1 244.6	32.7 243.9	31.3 244.0	29.9 243.2	28.6 241.3	27.2 239.5	25.9 238.2	24.5 237.3	at3 sr16	41.8 260.4	40.3 256.2	38.8 251.1	-	36.0 246.7	34.6 246.0	33.2 243.9	31.9 241.4	30.5 239.4	29.1 240.5	27.8 241.6		25.0 242.5

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Filter:	26.2	27.0	21.6	16.9	17.8	19.1	20.8	19.8	20.7	19.8	19.4	19.7	19.5	Filter	17.0	16.7	16.8	17.0	17.9	18.6	18.4	17.7	17.2	16.7	16.9	18.3	21.8
.7	4	9.3	1.9	-11.4	-6.2	13.0	27.1	21.1	3.8	-3.0	7.0	17.3	11.3	9.1	5.7	7.1	9.1	10.3	11.3	12.9	14.6	14.4	11.9	10.1	11.4	13.3	9.7
10 squ	5 10.9 31.0	28.4	24.1	20.2	18.2	16.6	14.9	15.2	13.6	14.0	13.6	12.8	12.9	1de: 6	16.7	14.9	13.4	13.1	12.7	12.6	12.0	11.0	10.7	10.2	10.3	11.3	14.0
ongiti	10.9	36.6	0	49.1	32.6	11.3	-2.4	-0.5	11.0	20.0	20.5	15.6	9.5	ongiti	28.5		17.8	15.3	14.8	16.1	18.3	18.9	16.7	14.8	16.0	17.7	13.8
7		0.23	0.21	0.20	0.19	0.24	0.25	0.24	0.16	0.15	0.14	0.14	0.13	7.2 T	0.17	0.16	0.15	0.14	0.13	0.12	0.12	0.12	0.12	0.12	0.12	0.11	0.0
de:-67	•	5.92	•	5.79	5.93	6.14	6.30	6.36	6.37	6.46	6.58	6.50	5.96	de:-67	5.81	6.18	6.42	6.46	6.33	6.23	6.29	6.48	69.9	92.9	9.90	6.27	5.93
Latitude		3.2	•	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	•	26.1	Latitu	•	3.2	•	4.6	5.6	6.8	8.3	10.0	12.1	14.7	17.8	21.5	26.1
-	253.6	249.7	246.1	244.2	242.9	241.8	240.5	239.3	238.6	238.3	238.0	237.6	237.3	61	256.4	252.0	248.3	244.9	243.2	242.8	242.1	240.5	238.8	237.7	237.1	236.6	236.4
at3 sr.	41.7	40.3	38.9	37.5	36.1	34.7	33.3	32.0	30.6	29.3	27.9	26.6	25.2	at3 sr	41.5	40.0	38.6	37.2	35.8	34.4	33.0	31.7	30.3	29.0	27.6	26.3	24.9

enr enr enr enr err

CII

vmr

(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (km) (K) (mb) vmr err err err err err err err

Filter:12	20.4	20.1	20.8	21.1	23.0	22.8	21.8	21.1	20.1	20.6	27.9	Filter:12	27.7	29.4	25.3	23.3	22.2	23.7	23.0	21.3	21.7	21.0	21.4	21.2	20.6
	0.7	8.6	14.4	17.0	17.5	16.8	14.9	12.1	10.1	11.0	10.3	_	13.9	29.3	28.4	19.9	13.7	11.2	9.2	8.1	9.3	12.7	14.8	14.2	11.5
ide:35	20.4	17.0	16.6	15.8	14.2	14.6	13.0	12.8	12.0	11.6	16.3	ide:28	22.7	23.3	17.6	17.7	16.6	15.4	15.2	14.0	14.3	14.3	13.9	14.2	14.1
.5 Longitude: 352.7	0.12 20.6	0.12 25.6	0.13 26.0	0.13 19.6	0.13 14.1	0.13 18.2	0.13 26.4	0.13 26.8	0.14 19.6	0.12 14.3	0.07 10.9	.8 Longitude: 284.0	0.09 -4.0	0.09 8.2	0.10 25.5	0.10 30.8	0.11 24.3	0.10 18.0	0.10 17.9	0.09 21.4	0.07 23.2	0.08 21.2	0.07 18.1	0.06 16.7	0.05 15.4
de:-67.	4.27	4.59	4.95	5.39	5.95	6.51	98.9	6.92	6.70	6.07	3.38	te:-67.	3.24	3.53	3.92	4.38	4.81	5.06	5.04	4.67	3.99	3.31	2.95	2.83	2.58
Latitude:-67	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	26.1	Latitude:-67	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 sr22	41.1 265.6	39.6 263.5	38.1 262.0	36.6 261.1	35.1 261.9	33.7 262.5	32.2 260.5	30.7 258.2	29.3 253.1	27.8 246.2	23.8 234.4	at3 sr24	40.7 281.9	39.1 277.9	37.5 274.6	36.0 271.8	34.5 268.7	32.9 265.1	31.5 260.5	30.0 251.1	28.6 241.9	27.3 235.9	25.9 231.2	24.6 226.0	23.4 220.1

10.0 5.49 0.10 21.1 12.2 15.7 19.3 12.1 5.83 0.10 18.9 12.4 12.6 20.8 Latitude:-68.2 Longitude:169.3 Filter: 9 27.3 25.0 19.5 20.6 20.5 20.5 20.5 30.0 20.3 -0.8 -9.5 Latitude:-68.0 Longitude:215.3 0.10 21.0 12.3 16.5 37.9 27.7 24.0 26.0 18.4 16.9 14.9 13.9 27.7 0.20 27.4 0.20 -1.7 0.10 17.4 0.10 14.0 0.18 21.8 0.18 24.5 0.10 13.7 0.18 10.2 0.14 14.6 0.19 -3.7 0.17 20.2 0.13 12.0 3.8 4.00 4.6 4.31 5.6 4.67 2.6 4.99 5.02 5.22 5.43 5.54 6.8 4.97 5.52 10.0 5.58 38.4 257.2 37.0 255.4 35.5 255.6 34.1 255.6 32.6 254.0 31.2 252.5 29.8 251.2 37.4 246.6 36.0 246.3 34.6 245.2 33.2 243.4 31.8 241.9 30.5 241.3 29.1 240.7 40.2 251.3 38.8 248.1 41.4 264.6 41.7 254.7 27.7 239.9 26.4 239.3 at3 sr28

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vmr

(mb)

 Table B.2 continued. (See notes on page 135.)

 Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃

 (km) (K) (mb) vmr err err err err err err err err

at	41	39	38	37	35	34	32	31	30	28	27	25	24	at	41	40	38	37	36	34	33	32	30	29	27	26	25
:12	, •	_	, C				_		_			_,	_	:12			- \			_	_					_	
ilter	16.6	17.(17.6	17.5	16.8	16.7	18.0	19.1	18.9	18.3	18.1	19.2	18.9	ilter	19.6	18.2	17.2	17.5	19.3	19.7	19.0	18.4	18.2	18.1	18.6	19.5	19.6
	0	9.6	3.5	4.1	3.1	3.4	5.8	7.2	5.1	1.7	1.0	2.3	9.0	_	0.1	7.1	5.3	7.8	2.9	15.3	13.3	8.6	8.3	9.1	11.8	13.3	10.5
Longitude: 123.6	17.6	16.1	14.8	14.6	14.3	14.3	13.6	0.11 20.3 12.7 1	12.3	11.5	11.4	12.2	12.0	1de: 54	17.5	15.1	13.9	13.7	13.0	12.8	11.9	11.6	11.3	11.0	11.2	12.2	11.9
ngit	32.2	25.2	16.8	14.1	15.0	17.2	19.5	20.3	17.7	14.3	14.2	16.1	14.0	ngiti	8.4	12.6	20.0	20.7	18.1	17.0	17.8	17.9	15.9	14.4	15.9	17.4	13.4
	0.16 32.2	0.14	0.13	0.13	0.13	0.12	0.12	0.11	0.11	0.11	0.11	0.11	0.11	9 P	0.15	0.15	0.16	0.16	0.15	0.14	0.13	0.12	0.12	0.13	0.12	0.10	0.09
Latitude:-68.3	5.49	5.53	5.55	_	5.69	5.83	5.03	5.12	.98	.79	5.75	.78	5.63	·	5.30	5.82	6.24	9.90	6.92	7.06	7.01	6.92	68.9	98.9	6.72	6.38	5.86
Latitue	2.6	3.2	3.8	4.6			8.3	10.0	12.1	1.7	8.	.5	2.1	tituc	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
30	252.7	248.7	246.7	245.1	244.0	242.9	242.0			235.8	233.0	232.4	232.3	33	259.3	256.1	253.8	251.7	249.2	246.5	243.7	240.8	238.6	237.8	237.5	236.8	235.9
at3 sr30	41.4	40.0	38.6	37.2	35.8	4.	33.0		α	28.9	27.6	26.3	25.0	at3 sr.	41.4 259.3	39.9	38.4		35.6	34.2	32.8	31.4		28.7	27.4	26.0	24.7

Filter: 9	29.1	31.0	28.1	23.4	24.6	23.3	22.7	20.2	19.3	19.7	20.6	21.9	21.7	ilter:12	18.7	19.0	18.6	19.0	18.7	18.7	19.1	19.1	19.4	19.3	20.0	20.4	19.5
I	11.2	22.3		0.5	4.7	17.4	16.1	4.9	-0.2	3.4	10.1	13.1	10.4	9.6 F	3.5	7.7	6.7	11.9	14.0	15.4	16.1	16.0	15.1	13.5	11.9	10.7	10.2
Longitude: 32.	31.2	30.8	27.6	20.2	20.5	18.1	16.5	11.0	11.0	11.5	12.8	14.5	14.1	tde:13	18.2	17.7	15.0	15.4	14.0	13.9	13.1	12.8	12.5	12.3	12.6	12.8	12.6
ongitu	37	35.5	$^{\circ}$	10.7	13.0	19.0	8.6	-8.2	-8.8	7.7	23.6	24.2	14.1	Longitu	11.0	14.5	16.5	20.7	24.8	25.4	23.0	20.2	18.1	16.5	14.8	13.7	13.6
ر	0.23	0.25	0.24	0.22	0.20	0.19	0.19	0.22	0.22	0.15	0.13	0.17	0.18	<i>.</i> 3	0.14	0.14	0.14	0.14	0.12	0.12	0.12	0.12	0.10	0.10	0.0	0.10	0.10
atitude:-68	οó	6.36	6.59	6.51	6.36	6.48	7.09	7.83	7.91	7.35	6.79	6.48	6.03	de:-69		5.31	5.38	5.41	5.55	5.75	5.90	5.87		5.63		5.87	5.78
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
sr34		262.2	259.6	255.2	250.2	247.2	246.1	245.6	246.2	246.7	246.7	246.1	243.6	36	251.7	∞	245.9	243.8	242.6	241.7	240.7	239.6	238.7	237.9	237.3	236.8	236.2
at3 sr	41.4		8.5		35.6	34.1	32.7	31.4	30.0	28.6	27.2	25.8	24.4	at3 sr.			9		36.1	34.7	33.3	32.0	30.6	29.3	27.9	56.6	25.2

enr enr err enr enr err

err

vmr

(mb)

 Table B.2 continued. (See notes on page 135.)

 Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃

 (km) (K) (mb) vmr err enr enr enr enr enr err

Filter:12	21.3	18.8	18.8	18.9	18.5	18.9	18.6	18.8	19.1	19.5	19.3	18.2	17.5	Filter: 9	37.7	33.4	29.8	28.2	24.7	25.2	25.1	23.9	21.9	22.0	24.3	23.6	22.4
	19.4	15.2	14.8	16.7	15.5	11.6	9.5	10.6	12.7	13.5	12.7	10.2	5.9	6	9.0	2.1	1.2	6.3	12.5	17.1	17.6	11.2	4.0	4.9	10.8	11.8	7.9
ıde: 71	19.6	15.7	16.0	13.3	13.0	12.4	12.3	11.9	11.8	12.0	12.1	11.7	11.3	tde: 47.	43.1	45.8	36.4	24.8	21.7	19.8	17.8	17.1	13.9	14.2	15.6	15.1	14.2
4 Longitude: 71.1	0.13 10.3	0.14 23.6	0.15 26.2	0.13 19.9	0.13 16.9	0.13 18.2	0.12 18.8	0.12 17.6	0.11 16.5	0.11 16.6	0.11 17.3	0.10 16.2	0.09 11.6	6 Longitude:	0.21 15.2	0.23 44.9	0.24 57.5	0.23 43.6	0.23 25.7	0.22 16.3	0.21 12.5	0.19 10.4	0.20 10.8	0.20 14.4	0.21 18.5	0.16 18.7	0.12 13.3
Latitude:-69.4	4.95	5.36	5.62	5.86	6.19	6.52	6.64	6.51	6:36	6.42	6.51	6.42	6.04	Latitude:-69.	5.30	5.87	6.34	6.79	7.13	7.18	7.02	6.83	6.80	7.02	7.23	7.05	6.48
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitue	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 sr39	41.4 259.1	39.9 254.1	38.5 249.0	37.1 246.8	35.7 246.4	34.3 245.5	32.9 244.1	31.5 242.2	30.1 240.0	28.8 237.8	27.4 235.6	26.1 234.4	24.8 234.1	at3 sr40	41.5 264.8	40.0 262.2	38.5 258.5	37.1 255.2	35.6 252.6	34.2 250.0	32.8 247.0	31.4 244.3	30.0 242.4	28.7 241.3	27.3 241.4	25.9 241.3	24.6 240.1

at3,		Latitu	de:-69	.9 7	niguc	tde:29	3.1	
25.5		17.8 3.32 0.07	3.32	0.07 17.1 15.5 10.6	17.1	15.5	10.6	24.5
24.6		21.5	2.85	90.0	16.1	15.6	10.0	
23.3		26.1	2.61	90.0	16.2	20.0	10.4	
at3	at3 sr46	Latitu	Latitude:-70.0	0.	ngiti	Longitude: 269.8	9.8	Filter: 9
40.5	40.9 280.3		3.91	0.17	0.4	63.7	-15.0	42.4
26.1	236.0		3.14	0.11	18.9	20.9	10.4	30.1
24.7	, 232.0		21.5 2.87	0.09 22.3 20.3 1.4	22.3	20.3	1.4	25.1
23.4	226.9	26.1	2.66	0.08	16.9	19.1	-1.3	23.8
at3	at3 sr48	Latitu	Latitude:-70.0	0.	ngiti	tde:22	4.4	Filter:12
41.0	41.0 266.6		2.6 3.78	0.11	31.5	0.11 31.5 25.2 14.2	14.2	24.1
39.5	39.5 263.6		3.92	0.11	29.3	0.11 29.3 24.4	7	25.0
at3	sr51		de:-70	.3 L	ngiti	3 Longitude: 109.4		Filter:12
41.8	3 256.1		4.98	0.14	30.4	20.6	ω,	20.1
40.3	40.3 252.4		5.27	0.13	13.6	17.9	8.0	19.4
38.9	249.2		3.8 5.55	0.14 11.1	11.1	15.8	10.1	19.6
37.5	246.4	4.6	5.86	0.15	17.0	15.3	10.1	19.4
36.1	245.2	5.6	6.26	0.14	21.0	15.6	7.7	19.9
34.7		8.9	9.90	0.14	19.8	15.4	9.9	21.0
33.3	3 244.3	8.3	6.74	0.13	16.1	12.8	8.7	19.2
31.5	244.1	10.0	6.62	0.12	17.1	12.9	11.6	19.7

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

enr enr err enr err

vmr err

(mp)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃
(km) (K) (mb) vmr err enr enr enr enr enr err

Filter: 12	24.5	23.6	22.6	23.2	23.7	21.0	20.8	23.9	21.5	20.6	ilter:12	26.3	25.5	25.8	27.7	25.8	24.2	26.3	27.0	24.2	24.8	31.6	31.1	29.3
	17.9	16.9	13.8	12.5	12.2	11.9	11.7	11.4	10.8	10.3	1.4 F	7.5	7.4	8.6	14.7	17.4	16.3	13.4	11.7	12.5	14.9	16.8	15.3	9.1
Longitude: 39.8	21.5	19.4	18.8	17.7	17.1	13.8	13.4	14.4	12.3	11.7	ide:33.	26.9	25.3	22.3	20.2	16.9	15.6	16.5	16.8	14.8	14.7	18.8	18.7	18.3
ongiti	2.8	9.5	18.0	21.4	23.8	25.3	21.8	15.6	12.6	13.4	Longitude:3	0.10 44.6	29.1	12.9	9.0	13.6	18.9	21.3	22.2	22.2	20.3	18.5	18.1	16.2
	0.12	0.12	0.13	0.14	0.14	0.13	0.13	0.14	0.13	0.13	7	0.10		0.11	0.12	0.12	0.11	0.11	0.11	0.10	0.0	0.0	0.08	0.07
Latitude:-70.6	4.40	4.91	5.55	6.19	6.53	6.62	98.9	7.18	7.25	7.21	de:-70	3.73	4.04	4.51	5.02	5.36	5.53	5.58	5.35	4.83	4.33	3.87	3.40	2.96
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	Latitude:-70.	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
	263.7	261.7	260.4	257.7	254.2	252.4	251.1	250.0	248.5	.4 246.5	22	269.2	267.2	265.9	266.0	264.6	261.2	258.2	255.2	248.9	242.7	238.6	234.2	229.2
at3 sr54	41.4	39.9	38.4	36.9	35.5	34.0	32.6	31.2	29.8	28.4	at3 sr.	40.9	39.4	37.9	36.3	34.8	3	31.9 2	30.4	29.0 2	27.6	26.2	24.9	23.6 2

Filter: 9	41.8	34.1	36.4	36.5	30.4	28.4	30.4	30.9	28.2	29.1	32.2	29.0	27.1	Filter:12	24.9	26.3	22.1	20.8	22.8	24.3	20.8	19.2	18.7	20.2	22.0	22.0	22.5
	-14.2	8.9	17.2	11.3	10.4	12.1	10.1	11.6	18.6	21.9	16.2	6.7	-0.8	9.9 F	14.0	15.9	12.9	9.6	11.4	15.1	15.0	11.9	10.4	11.5	12.1	10.8	9.4
ude:30	39.3	37.0	32.4	35.6	25.0	22.3	23.0	21.1	18.9	18.1	23.0	21.8	21.7	tude:16	24.7	22.1	18.2	17.1	17.7	18.0	14.6	13.0	12.1	12.9	13.9	14.1	14.8
3 Longitude:308.2	0.17 37.2	0.18 26.7	0.18 10.3	0.19 1.4	0.20 14.8	0.19 33.0	0.16 35.3	0.15 24.2	0.17 12.5	0.15 5.7	0.09 7.8	0.09 14.7	0.08 20.1	Longiti	0.13 6.0	13 1	0.13 25.2	0.13 26.7	0.12 22.2	0.11 20.1	0.10 20.7	0.10 18.5	0.10 15.1	0.10 14.1	0.11 14.2	.10 14.1	.09 14.1
atitude:-70.8		4.35	4.81	5.49	5.91					4.25 (3.61	2	2.74 (de:-71.1	4.61 (5.10	5.19	5.36			5.48 (_	5.83 0	5.62
Latitu	2.6	3.2		4.6	5.6	8.9	8.3		12.1	14.7	17.8	21.5	26.1	Latitud	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 sr58	80	4.	9 272	.3	34.8 268.0	.3 264	_	30.3 256.5	.9 250	27.5 244.7	26.1 239.5	8	23.4 233.3	Ó	_	.1 25	7.	37.3 246.9	35.9 244.4	34.5 243.0	-:	31.8 238.9	30.4 237.7	29.1 236.6	7.7 23	26.4 234.9	25.1 234.6

enr enr err enr err

E

vmr

(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (km) (K) (mb) vmr err err err err err err err

6														12		12						
Filter:	18.9	22.1	29.6	33.3	29.3	22.6	19.6	20.5	22.6	20.6	21.4	20.8	20.8	Filter: I	29.3	Filter: 1	31.4	28.0	26.7	24.8	25.0	24.3
	-57.0	-49.2	-2.2	37.1	32.5	13.8	6.0	-1.5	3.4	10.9	12.7	7.6	1.6	3	7.4		9.5	8.4	7.1	11.1	15.4	15.7
Longitude:147.1	47.9	39.1	23.4	25.7	23.6	18.4	15.3	15.6	16.9	15.4	15.2	14.7	15.4	Longitude: 299.	18.3	Longitude: 229.9	32.4	27.2	23.4	20.0	19.3	17.6
ongiti	23.6	10.0	4.1	16.3	24.0	9.1	-5.3	6.4	26.2	33.4	28.7	21.6	13.3	ongiti	0.06 16.5	ongiti	31.8	36.1	25.6	20.4	23.8	23.5
	0.23 23.6	0.23	0.20	0.19	0.22	0.21	0.20	0.18	0.16 26.2	0.20 33.4	0.16 28.7	0.13	0.13 13.3	7 9	90.0	7 · T	0.10 31.8	0.11	0.11	0.10	0.10	0.10
Latitude:-71.1	5.44	5.50	5.41	5.55	5.80	5.91	6.05	6.31	6.36	6.22	6.20	6.15	5.72	de:-71.	3.01	de:-71.	3.71	4.12	4.36	4.40	4.52	4.82
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude:-71	26.1 3.01	Latitude:-7	2.6	3.2	3.8	4.6	5.6	8.9
54	260.1	254.0	249.4	246.4	244.2	242.4	240.9	239.9	239.7	238.9	238.0	237.4	236.7	99	230.8	65	267.7	266.1	262.3	260.6	259.6	256.8
at3 sr64	41.8 260.	40.4	38.9	37.5	36.1	34.7	33.4	32.0	30.6	29.3	27.9	26.6	25.2	at3 sr66	23.6 230.8	at3 sr69	41.1	39.6	38.1	36.6	35.1	33.7 ;

21.5 6.52 0.19 16.2 16.7 9.9 25.3 26.1 6.05 0.16 9.9 14.7 3.3 22.2 Latitude: 71.8 Longitude: 161.3 Filter: 12 Latitude:-71.8 Longitude:206.9 Filter: 9 0.13 15.2 24.9 18.2 15.3 11.8 11.8 9.7 15.6 11.7 20.4 19.2 17.6 0.17 26.0 14.8 0.16 24.0 0.16 16.5 0.21 23.4 0.12 22.6 0.15 6.4 0.13 20.4 0.13 14.8 0.12 15.3 0.11 18.6 0.11 19.5 0.11 21.2 0.11 20.5 0.14 2.6 4.66 4.79 5.08 5.99 12.1 6.16 17.8 6.60 10.0 14.7 41.6 258.0 40.1 253.2 34.1 248.7 32.7 245.0 31.3 242.3 30.0 241.4 28.6 240.3 27.2 239.2 25.9 237.8 24.6 235.5 at3 sr72 37.3 249.0 35.9 244.8 34.5 241.5 33.1 241.0 31.7 240.5 30.4 238.8 29.0 236.7 235.3 37.0 252.4 35.5 251.3 38.7 249.8

Alt Temp Press $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

enr enr err enr err

vmr err

(mb)

₹

(km)

35.3

23.7 229.5 26.1 2.68 0.06 17.6 22.3 10.5

 Table B.2 continued. (See notes on page 135.)

 Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃

 (km) (K) (mb) vmr err enr enr enr enr enr err err

24.8	29.9	24.1	23.5	23.1	24.4	25.1	25.3	26.1	26.7	27.8	28.8	^c ilter:12	34.5	33.1	31.9	31.7	30.3	24.6	26.9	28.9	29.8	29.4
9.9	13.7	17.8	14.8	13.2	14.5	13.4	12.2	13.1	13.7	14.0	14.9	Ι:	17.4	8.3	6.5	11.0	14.7	14.3	14.1	15.3	14.4	11.7
23.2	26.6	18.1	17.5	17.0	16.3	15.7	15.5	15.4	15.8	16.4	16.8	7										
0.12 18.9	0.12 19.2	0.15 20.2	0.14 21.9	0.13 23.0	0.12 22.0	0.12 19.9	0.11 18.1	0.10 17.3	0.09 16.9	0.10 16.4	0.10 16.9	I Longitu	0.12	0.12	0.12	0.13	0.14	0.14	0.16	0.14	0.13	0.13
4.20	4.67	5.20	5.49	5.71	5.90	5.86	5.55	5.27	5.20	5.29	5.27	le:-72.	4.11	4.43	4.75	5.16	5.88	6.54	6.84	6.91	6.95	6.73
	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	Latituc	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7
41.4 267.3	39.9 264.4	38.4 262.5	36.9 260.3	35.4 258.8	34.0 257.5	32.5 255.1	31.1 251.5	29.7 247.3	28.3 244.7	26.9 243.1	25.5 242.4	at3 sr81	41.2 270.2	39.7 268.9	38.2 267.0	36.7 263.5	35.2 260.0	33.7 258.0	32.3 257.9	30.8 255.9	29.4 253.5	27.9 251.2
	267.3 2.6 4.20 0.12 18.9 23.2 6.6	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8	267.3 2.6 4.20 0.12 18.9 23.2 6 264.4 3.2 4.67 0.12 19.2 26.6 13 262.5 3.8 5.20 0.15 20.2 18.1 17 260.3 4.6 5.49 0.14 21.9 17.5 14 258.8 5.6 5.71 0.13 23.0 17.0 13	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 255.1 8.3 5.86 0.12 19.9 15.7 13.4	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 255.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 255.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.20 0.09 16.9 15.8 13.1	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 255.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 255.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 242.4 21.5 5.27 0.10 16.9 16.8 14.9	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 16.1 242.4 21.5 5.27 0.10 16.9 16.8 14.9 16.1 242.4 21.5 5.27 0.10 16.9 16.8 14.9	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 260.3 4.6 5.49 0.14 21.9 17.5 14.8 250.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 81 Latitude:-72.1 Longitude:313.1 F 270.2 2.6 4.11 0.12	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 270.2 2.6 4.11 0.12 17.4 8.3	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 8.1 21.5 5.27 0.10 16.9 16.8 14.9 268.9 3.2 4.43 0.12 8.3 267.0 3.8 4.75 0.12 8.3	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 14.2 25.2 24.3 0.12 2.0 16.9 16.8 14.9 25.2 24.3 0.12 2.0 16.9 16.8 14.9 25.2 24.3 0.12 2.0 2.6 4.11 0.12 17.4 268.9 3.2 4.43 0.12 8.3 267.0 3.8 4.75 0.12 1.5 11.0	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.29 0.10 16.4 16.4 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 270.2 2.6 4.11 0.12 2.6 4.13 0.12 2.6 4.13 0.12 2.6 4.13 0.12 2.6 4.13 0.12 2.6 4.14 0.12 2.6 4.14 0.12 2.6 4.14 0.12 2.6 5.8 0.14 11.0 2.9 2.6 5.8 0.14 11.0 2.6 5.8 0.14 11.0 2.6 5.8 0.14 11.0 2.9 2.9 2.9 2.9 2.9 2.9 2.9 2.9 2.9 2.9	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 13.7 243.1 17.8 5.27 0.10 16.9 16.8 14.9 763.1 Latitude:-72.1 Longitude:313.1 F 267.0 3.8 4.75 0.12 6.5 267.0 3.8 4.75 0.12 6.8 266.0 3.8 4.75 0.12	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.1 8.3 5.86 0.12 19.9 15.7 13.4 257.1 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.27 0.10 17.3 15.4 13.1 244.7 14.7 5.20 0.09 16.9 15.8 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 81 Latitude:-72.1 Longitude:313.1 F 250.2 3.2 4.43 0.12 8.3 265.0 3.8 4.75 0.12 6.5 260.0 3.8 4.75 0.12 <	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.1 8.3 5.86 0.12 19.9 15.7 13.4 257.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.20 0.09 16.9 15.8 13.7 247.4 14.7 5.20 0.09 16.9 15.8 14.0 247.4 14.7 5.20 0.09 16.9 16.8 14.9 81 Latitude:-72.1 Longitude:313.1 F 267.0 3.8 4.75 0.12 6.5 267.0 3.8 4.75 0.12 6.5 268.0 4.6 5.16 0.13 <	267.3 2.6 4.20 0.12 18.9 23.2 6.6 264.4 3.2 4.67 0.12 19.2 26.6 13.7 262.5 3.8 5.20 0.15 20.2 18.1 17.8 260.3 4.6 5.49 0.14 21.9 17.5 14.8 258.8 5.6 5.71 0.13 23.0 17.0 13.2 257.5 6.8 5.90 0.12 22.0 16.3 14.5 257.1 8.3 5.86 0.12 19.9 15.7 13.4 251.5 10.0 5.55 0.11 18.1 15.5 12.2 247.3 12.1 5.20 0.10 17.3 15.4 13.1 247.4 14.7 5.20 0.09 16.9 15.8 14.0 242.4 21.5 5.27 0.10 16.9 16.8 14.9 48.1 17.8 5.29 0.10 16.9 16.8 14.9 48.9 3.2 4.43 0.12 0.12 8.3 260.0 3.8 4.75 0.12 0.13 11.0 260.0

at3 sr82	Latitu	atitude:-72.2	.2 Longitude: 288.4	ude:28		Filter: 9
41.3 275.9	2.6	4.04	0.17-21.4	75.6	16.4	54.2
39.7 273.8	3.2		0.20-21.5	64.4	-35.3	30.3
38.2 270.2	3.8	5.03	0.20 8.3	40.9	-34.8	24.0
36.7 266.8	4.6		0.21 30.8	40.5	-9.2	34.3
35.2 265.6	5.6	5.91	0.20 23.7	24.5	4.1	28.1
33.6 267.8	8.9		0.22 10.6	21.1	9.4	56.6
32.1 267.2	8.3	6.83	0.22 13.8	21.2	13.8	31.4
30.6 266.8	10.0	6.44	0.20 22.5	22.2	13.3	32.3
29.1 260.5	12.1	5.70	0.16 24.6	23.8	9.5	37.1
27.7 255.5	14.7	4.93	0.12 22.7	18.3	6.9	30.3
26.2 249.1	17.8	4.08	0.12 21.0	24.8	9.9	38.0
24.8 239.6	21.5	3.39	0.10 17.6	28.0	6.9	44.0
23.5 231.7	26.1	2.95	0.07 16.3	24.3	7.3	36.5
at3 sr84	Latitu	xde:-72	.2 Longitude: 243.0	ude:24	_ `	Filter:12
26.3 240.5	17.8	3.50	0.07 18.5	19.9	14.8	32.6
25.0 235.1	21.5	3.04	0.07 17.4	23.4	12.0	37.8

 $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$ $^{686}O_3$

Alt Temp Press

enr enr err enr err

ᇊ

vmr

(qm)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃
(km) (K) (mb) vmr err enr enr err err err

r:12	5	6	3	8	6	5	5	5	0	7	7	4	7	7:9	7	9	9	4	Ε.	4	2	9.	4	2	6	2	8:
Filter: I.	25	28	25	26	29	23.	25.	26.	29.	22.7	28	24	30	Filter:	40.7	44.6	33.6	30.4	29.1	31.	32.2	30.6	27.4	26	24.3	31.2	27
		10.7	10.0	14.2	16.6	16.1	16.4	15.1	10.6	7.2	8.0	11.2		150.0	-16.8	-4.6	-0.5	8.6	15.4	9.5	10.4	18.3	12.4	0.3	3.0	12.7	11.5
Longitude: 173.7	26.0	25.8	22.4	22.5	23.1	16.3	17.0	17.8	18.7	14.3	18.2	15.1	19.3	tde:15	_	64.6	42.2	31.6	25.7	30.0	29.3	21.8	19.3	19.7	17.7	20.6	18.8
ongiti	7.2	15.4	27.2	28.2	21.0	15.9	16.9	19.3	18.0	15.3	13.3	12.4	13.0	Longitude: 1	22.2	33.8	23.0	11.8	17.1	20.1	23.6	35.0	35.5	17.9	5.8	7.5	15.8
	0.12	0.13	0.13	0.14	0.13	0.12	0.12	0.12	0.14	0.13	0.12	0.10	0.10	3	0.23	0.22	0.21	0.18	0.18	0.20	0.20	0.16	0.15	0.18	0.17	0.20	0.12
atitude:-72.3	4.16	4.79	5.15	5.18	5.38	5.70	5.93	6.16	6.40	6.49	6.35	5.93	5.36	le:-72.	5.37	5.47	5.32	5.08	5.25	5.71	5.94	5.97	6.16	6.47	6.52	6.19	5.66
Latituc	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitud	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
sr87	259.4	258.1	256.2	251.1	248.3	246.9	244.8	243.2	239.8	236.0	235.4	234.6	232.3	sr88	262.7	253.8	250.9	250.2	247.7	244.6	242.8	242.6	241.8	239.9	239.0	238.3	237.6
at3 sr	41.4	40.0	38.5	37.1	35.6	34.2	32.8	31.5	30.1	28.7		26.1	24.8	at3 sr	41.7	40.3	∞.	37.4		34.6	33.2	31.8	30.5	29.1	27.7	26.4	25.0

9														12													
Filter:	33.6	41.0	34.1	31.9	39.3	32.1	36.2	30.9	26.7	36.4	31.4	27.7	34.0	Filter:	13.3	13.3	13.4	13.6	13.5	13.3	13.1	13.0	13.4	14.1	14.3	14.9	18.3
7.0	-12.9	20.4	8.7	-13.1	0.4	14.3	4.6	-1.8	9.9	11.8	7.0	5.5	14.7	1.0 1	9.0	10.8	13.8	16.1	15.1	12.4	10.4	10.0	10.5	11.4	12.6	13.4	11.6
Longitude:34	65.6	\sim	33.1	29.6	33.7	4	29.9	23.0	16.9	22.2	18.1	17.1	19.5	ude:21	13.4	13.0	11.3	11.1	11.1	10.1	9.6	9.5	9.6	8.6	6.6	10.6	13.7
ongiti	29.2	11.2	4.2	16.1	24.5	31.5	28.0	16.6	13.0	14.1	14.4	14.7	13.6	ongiti	14.0	11.3	12.5	14.2	14.3	14.1	14.5	14.5	13.6	12.9	13.9	16.5	17.0
.4 I	\sim	0.23	0.22	0.21	0.26	0.25	0.25	0.27	0.19	0.20	0.18	0.22	0.21	7 0:	0.20	0.19	0.17	0.16	0.15	0.15	0.15	0.14	0.12	0.11	0.10	0.0	0.08
de:-72	4.80	5.14	5.47	5.90	6.28	6.53	6.77	7.02	7.32	7.59	7.74	7.79	7.17	de: 49	7.28	7.37	7.05	6.77	6.88	7.19	7.40	7.30	6.97	6.52	6.04	5.49	4.83
Latitude:	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
76	269.3	266.0	260.8		255.0	3	251.6	251.4	252.5	253.1	253.2	253.3	253.4	02	229.7	226.6	224.2	222.7	221.3	220.7	221.0	221.3	221.2	220.8	220.4	219.5	218.0
at3 sr94		40.2		37.2	35.8	34.4	32.9		30.1	28.6	27.2		24.3	at3 ss		6.8		36.4	35.1	33.9	2.6	1.3	30.1	28.8	27.6	26.3	

 $^{48}O_3$ $^{48}O_3$ $^{668}O_3$ $^{668}O_3$ $^{686}O_3$

Alt Temp Press

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(km)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err err err err err err

at3 ss03 Latitude: 48.9 Longitude: 188.4 Filter: 9	at3 ss10 Latitude: 48.2 Longitude: 30.1 Filter:12
2.6 7.12 0.29 41.3 22.4 4.1	2.6 6.25 0.17 12.4 14.7
3.2 7.00 0.27 37.2 21.4 2.5	3.2 6.75 0.17 8.7 14.0 20.8
37.8 228.4 3.8 6.77 0.24 26.1 20.5 0.1 18.0	37.0 231.7 3.8 6.74 0.16 12.8 12.6 19.1 15.3
4.6 6.48 0.21 17.3 19.5 -0.2	4.6 6.50 0.15 18.3 11.4 14.9
5.6 6.30 0.20 13.1 19.1 3.5	5.6 6.22 0.13 19.3 11.5 11.2
6.8 6.18 0.20 11.1 15.1 10.2	6.8 5.89 0.12 15.8 10.9 10.6
8.3 6.08 0.20 11.2 12.4 17.2	8.3 5.52 0.10 12.5 10.1 12.3
10.0 6.08 0.19 15.2 12.6 19.8	10.0 5.24 0.09 13.7 10.1 14.8
12.1 6.19 0.16 21.2 12.7 16.9	12.1 5.12 0.09 19.4 10.5 16.7
14.7 6.26 0.13 23.4 12.2 11.8	14.7 5.12 0.09 22.5 10.8 14.8
17.8 6.13 0.12 19.4 11.5 8.2	17.8 5.09 0.09 17.4 10.3 8.1
21.5 5.68 0.15 12.6 12.3 8.4	21.5 4.92 0.08 7.1 9.8 -0.1
26.1 4.99 0.13 7.7 11.9 9.2	26.1 4.54 0.07 0.0 9.4 -3.6
Latitude: 48.6 Longitude: 120.6	Latitude: 48.2 Longitude: 7.5
2.6 6.17 0.17 17.9 13.4 16.0	2.6 7.36 0.32 11.3 23.0 29.0
3.2 6.42 0.16 15.7 13.0 13.7	3.2 7.14 0.29 20.8 23.0 22.8
3.8 6.50 0.16 15.4 12.2 14.4	3.8 6.59 0.25 27.6 16.2 16.2
4.6 6.53 0.16 16.9 12.3 16.3	4.6 6.23 0.23 28.9 16.3 9.3
5.6 6.50 0.15 19.2 12.5 18.1	5.6 6.06 0.22 24.9 16.0 2.6
6.8 6.43 0.13 21.6 11.4 19.7	6.8 5.74 0.19 19.6 15.1 -4.1
8.3 6.62 0.12 20.9 11.0 18.3	8.3 5.27 0.14 16.6 13.9 -8.9
10.0 7.17 0.14 15.7 10.0 12.7	10.0 4.94 0.12 17.0 13.9 -7.0
12.1 7.64 0.16 11.0 9.1 7.9	12.1 4.90 0.12 19.7 14.1 3.4
14.7 7.56 0.16 11.1 9.0 8.5	14.7 5.06 0.17 19.5 12.9 14.1
17.8 6.79 0.13 16.9 10.0 15.5	17.8 5.15 0.17 13.7 12.1 17.1
21.5 5.85 0.10 20.7 12.3 20.6	21.5 5.01 0.11 5.2 11.1 13.8
26.1 5.04 0.09 14.8 11.7 15.1	26.1 4.58 0.08 0.9 10.8 12.0

Alt Temp Press

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(km)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err enr enr enr enr enr err

23 840	:13	Latitu	de: 46	ionol 6	tude: 14	911	Filter: 12	at3.ss17	Latit
40.3	40.3 234.1	2.6	5.71	1 2.6 5.71 0.18 15.5 13.5 20.6	13.5	20.6	17.1	40.3 244.3	2.6
39.0	231.7	3.2	5.93	0.17 15.3	13.4	16.7	16.5	38.9 242.3	
37.7	230.4	3.8	9.00	0.17 14.5	13.2	16.8	16.4	37.5 240.2	
36.4	231.2	4.6	9.00	0.16 13.4	11.8	19.1	16.8	36.2 234.8	
35.0	232.1	5.6	5.98	0.15 12.8	11.7	20.0	16.9	34.8 229.1	
33.7	231.4	8.9	5.91	0.15 13.7	11.9	19.0	16.9	33.5 223.6	
32.4	229.2	8.3	5.83	0.13 16.1	10.8	17.5	16.6	32.3 223.2	
31.1	226.8	10.0	6.01	0.13 16.9	10.8	15.8	16.3	31.0 223.0	
29.9	224.6	12.1	6.43	0.13 15.7	8.6	14.4	14.9	29.8 221.7	
28.6	222.5	14.7	6.73	0.13 14.7	9.3	14.1	14.2	28.5 218.4	
27.3	220.5	17.8	6.50	0.12 16.1	9.4	15.5	14.3	27.3 215.1	
26.1	219.2	21.5	5.82	0.10 18.6	10.9	17.2	16.0	26.1 213.4	
24.8	218.7	26.1	5.01	0.08 18.1	12.9	15.6	18.3	24.9 213.3	
at3 ss	91.	Latitu	de: 46	.5 Longi	tude: 7.	3.8	Filter: 12	at3 ss19	
40.0	245.0	2.6	6.01	0.16 15.3	15.0	14.4	15.8	39.8 242.0	
38.6	240.8	3.2	6.24	0.16 13.1	14.6	14.3	15.7	38.4 237.1	
37.2	236.7	3.8	6.28	0.16 16.2	12.1	17.2	16.1	37.1 232.1	
35.9	232.7	4.6	6.35	0.15 18.2	12.2	18.1	16.1	35.8 228.5	
34.6	229.6	5.6	6.47	0.15 17.9	12.1	15.7	15.8	34.5 226.8	
33.3	228.7	8.9	6.47	0.13 16.9	11.9	12.7	16.8	33.2 225.0	
32.0	228.3	8.3	6.20	0.12 15.9	11.8	11.2	16.6	32.0 222.0	
30.7	226.3	10.0	5.75	0.11 14.5	10.6	11.3	14.9	30.7 219.4	
29.4	222.4	12.1	5.32	0.09 13.7	10.3	12.3	14.6	29.5 217.4	
28.2	218.4	14.7	5.07	0.09 14.9	10.4	13.8	14.8	28.3 215.5	
27.0	215.9	17.8	4.90	0.08 16.7	11.4	14.8	15.8	27.1 213.5	
25.7	214.8	21.5	4.67	0.07 15.8	12.0	12.8	16.2	25.9 211.6	
24.5	213.6	26.1	4.29	0.06 9.5	11.4	6.3	6.3 15.4	24.7 210.8	

itude: 46.1 Longitude: 6.1 Filter:12 .6 5.50 0.15 24.2 18.6 16.2 16.8 .2 5.73 0.18 23.5 14.3 23.5 16.5 15.8 20.2 26.2 10.4 tude: 46.5 Longitude: 51.2 16.0 14.2 13.8 0.08 16.0 14.7 12.7 12.5 11.4 0.15 18.6 11.8 0.11 19.0 13.1 0.14 5.8 0.16 8.6 0.16 14.8 0.19 20.6 0.22 25.6 0.18 23.3 0.14 16.6 0.13 14.4 0.11 15.2 0.20 30.8 0.18 23.8 0.09 10.6 0.16 11.7 0.09 5.07 1 4.34 5.99 6.36 6.50 89.9 5.83

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Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃ 668O₃686O₃ 686O₃ (km) (K) (mb) vmr err enr enr enr enr enr err

Filter: 12	15.2	13.6	13.7	13.3	13.5	13.3	13.2	13.2	13.2	13.5	13.1	13.0	13.1	Filter: 9	28.2	26.2	23.4	20.9	20.5	15.9	17.0	17.7	14.2	14.1	14.6	15.3	14.7
	13.6	8.4	8.4	13.5	17.8	16.1	11.7	9.1	10.5	12.6	10.5	4.4	-0.1	5.7	20.0	9.7	-4.2	-15.4	-18.1	-12.0	-4.8	-0.1	2.9	6.7	11.9	14.6	9.1
.tde:29	13.2	11.6	11.8	10.5	10.2	10.0	8.6	9.6	9.6	10.5	10.2	10.4	10.7	\sim	24.6	24.4	20.8	17.0		12.0	12.0	12.1	12.4	12.6	12.9	13.5	13.0
6 Longitude: 298.4	0.18 13.2	0.18 11.5	0.17 14.6	0.15 19.5	0.14 22.3	0.13 20.0	0.13 15.9	0.12 13.9	0.12 15.6	0.12 17.6	0.11 14.6	6.9 60.0	0.07 1.0	5 Longitude: 2	0.30 25.4	0.28 24.5	0.26 19.4	0.24 13.9	0.22 10.1	0.17 8.8	0.15 9.6	0.13 11.8	0.13 14.8	0.12 17.9	0.12 20.8	0.14 21.1	0.12 15.5
Latitude: 45.6	92.9	7.21	7.26	7.06	6.90	6.93	7.00	6.99	6.87	6.59	6.13	5.47	4.67	te: 45.	7.27	7.02	6.75	6.41	6.02	5.70	5.56	5.58	5.65	5.56	5.23	4.71	4.09
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 45	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss22	39.6 231.0	38.3 228.7	37.0 227.2	35.7 224.9	34.4 221.6	33.2 218.3	32.0 216.2	30.7 214.8	29.5 213.7	28.3 212.8	27.1 212.1	25.9 211.8	24.7 212.3	at3 ss23	39.7 236.2	38.4 231.3	37.1 226.9	35.8 223.6	34.5 220.4	33.3 217.7	32.0 215.8	30.8 213.8	29.6 213.0	28.4 213.1	27.2 213.1	26.0 212.9	24.8 212.7

21.5 5.31 0.08 19.1 12.0 15.9 15.6 26.1 4.59 0.07 15.8 11.7 13.0 15.3 Latitude: 44.7 Longitude: 162.9 Filter:12 Latitude: 45.2 Longitude: 230.6 Filter: 12 19.8 19.9 16.8 14.5 13.6 14.0 16.2 0.17 17.2 14.5 14.5 12.4 0.13 11.0 10.1 0.14 17.2 0.13 11.9 0.11 15.3 0.16 17.1 0.15 21.9 0.13 19.2 0.13 13.2 0.13 11.6 0.16 20.3 0.15 20.7 0.12 18.2 0.12 16.8 0.09 17.5 0.13 11.5 0.12 15.7 0.12 16.5 2.6 5.90 3.2 6.08 17.8 5.94 21.5 5.31 10.0 6.64 12.1 6.54 6.54 14.7 6.35 3.8 6.22 4.6 6.37 6.45 37.4 224.5 36.2 222.3 34.9 220.2 33.6 218.8 32.4 218.8 31.2 218.9 29.9 218.3 28.7 217.2 27.5 216.0 26.2 215.5 25.0 215.4 40.3 232.9 38.9 231.7 37.6 230.5 36.3 229.4 35.0 228.5 33.7 228.0 32.4 227.4 31.1 226.0 29.9 223.7 28.6 221.5 27.4 219.8

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Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err err err err err err err

Filter: 9	26.5	17.8	18.0	21.8	22.3	24.0	19.4	16.0	15.4	15.8	16.2	16.5	18.2	Filter:12	17.2	16.7	17.0	16.1	16.8	16.6	16.8	16.5	16.1	15.9	16.4	17.6	16.8
	-19.9	-25.5	-22.2	-2.2	26.9	37.3	24.7	6.9	1.3	4.0	7.5	8.2	5.7	7	_	4.7	0.9	14.1	19.2	18.7	16.2	14.0	12.0	11.3	12.1	11.4	5.1
ude: 14	28.6	18.2	18.4	19.4	18.8	18.5	13.8	11.6	11.6	12.3	12.6	12.7	14.6	tde: 95	17.6	14.5	14.8	12.1	11.9	11.6	10.8	10.7	10.4	10.4	11.1	12.3	11.7
6 Longitude: 140.3	0.30 13.2	0.30 13.7	0.26 18.3	0.24 26.4	0.20 34.6	0.19 32.5	0.22 21.0	0.22 10.5	0.22 10.8	0.16 16.9	0.12 20.6	0.11 16.8	0.11 8.5	2 Longitude: 95.	0.15 11.8	0.16 17.5	0.16 21.7	0.16 22.5	0.16 19.9	0.16 16.2	0.14 14.3	0.14 14.5	0.13 15.8	0.12 17.0	0.11 17.5	0.09 15.1	0.08 7.5
Latitude: 44.6	7.38	7.17	6.79	6.57	6.49	6.51	6.57	6.61	6.61	6.54	6.32	5.81	5.02	le: 44.	5.61	6.43	99.9	6.40	6.32	6.58	7.00	7.32	7.43	7.21	6.59	5.78	4.92
Latitue	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 44	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss29	40.9 240.8	39.5 240.4	38.2 240.1	36.8 239.8	35.5 238.9	34.1 237.4	32.8 235.4	31.4 232.1	30.1 228.6	28.8 225.9	27.6 224.5	26.3 223.1	25.0 221.5	at3 ss31	40.4 245.5	39.0 244.3	37.6 243.2	36.2 239.2	34.9 234.7	33.6 231.0	32.3 228.8	31.0 226.7	29.7 225.2	28.4 223.9	27.1 222.6	25.9 221.3	24.6 220.0

Latitude: 43.4 Longitude: 342.4 Filter: 14.2 21.1 20.0 21.0 21.0 21.0 0.30 -5.9 18.1 -6.9 0.28 -5.0 18.6 0.7 0.24 4.4 14.6 12.0 14.3 -2.6 14.4 -13.7 -9.5 10.3 11.2 Latitude: 43.6 Longitude: 4.9 15.9 14.3 0.30 4.6 19.9 10.7 0.07 2.1 11.4 11.1 0.09 14.5 0.19 31.4 0.17 22.5 0.14 12.5 0.12 14.0 0.11 14.6 0.14 13.2 0.10 15.4 0.08 9.8 0.22 16.3 0.21 27.6 0.15 5.9 2.6 6.46 2.6 7.59 26.1 4.46 7.65 17.8 5.64 21.5 5.09 4.6 7.39 14.7 6.06 12.1 6.27 8.9 30.9 220.4 29.6 218.1 28.4 215.9 27.2 214.3 26.0 213.0 24.8 212.0 37.3 235.8 35.9 229.8 34.7 224.9 33.4 223.3 32.1 222.2 38.8 237.0 37.5 235.4 36.2 232.1 34.9 228.0 33.6 224.5 32.3 223.3 31.0 222.2 29.8 220.9 28.5 219.3 26.1 216.7 24.9 215.5 40.2 238.6 27.3 217.9 at3 ss35

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Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (km) (K) (mb) vmr err err err err err err err

Filter: 9	27.8	27.5	26.1	25.1	23.1	21.4	20.9	17.1	16.6	16.1	15.9	17.2	18.6	Filter:12	20.9	18.3	15.2	14.4	14.7	15.1	15.1	15.1	15.1	16.0	16.3	17.4	17.6
	23.6	24.1	16.2	10.9	10.9	13.2	14.5	13.1	9.2	4.3	3.6	8.9	13.1	1.7	42.9	26.3	11.1	5.3	6.5	6.6	10.9	10.1	9.4	10.0	12.4	13.3	7.7
<i>ide: 2</i> (24.3	25.4	20.6	19.7	16.7	13.7	14.1	12.3	12.7	13.2	12.4	12.6	13.5	tde:34	17.8	15.6	11.2	10.7	10.6	10.8	10.7	10.2	10.1	11.0	11.4	12.8	13.5
Longitude: 26.7	.25 -5.2	1.76 1.1	0.26 13.6	0.25 18.1	0.24 9.4	0.23 -0.2	0.21 7.3	0.18 22.6	0.18 29.4	0.19 25.7	.17 17.7	.14 13.1	.10 11.9	Longitude:34	0.16 41.5	.18 25.6	.18 9.8	0.17 5.1	.16 8.9	.15 14.7	0.13 15.4	0.12 13.0	0.10 11.6	0.09 13.0	0.08 16.6	0.09 18.1	.09 12.6
Latitude: 40.5	6.20	0 99.9	7.32 0	7.41	6.72	6.03	5.98	6.25	6.24 0	5.87 0	5.43 0	5.08 0	4.66	de: 40.0	5.22 0	6.47 0.	7.54 0.	7.93	7.80 0.	7.44 0.	7.06 0	6.65 0	6.15 0	5.61 0	5.15 0	4.79 0	4.43 0
Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitud	2.6	3.2		4.6	5.6	6.8	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss42	40.5 249.2	39.1 245.6	242	36.3 239.4	35.0 235.9	33.7 231.3	32.4 226.7	31.1 223.3	29.8 221.0	28.6 219.5	27.3 219.6	26.1 219.2	24.9 216.7	at3 ss44	40.2 245.6	38.8 240.9	.5 236	36.1 232.8	34.8 229.5	33.5 227.3	32.3 225.5	31.0 223.4	29.7 221.3	28.5 219.7	27.2 218.4	26.0 216.8	24.8 214.8

Latitude: 39.4 Longitude:274.0 Filter:12 26.1 4.43 0.07 13.4 13.4 10.4 16.2 Latitude: 39.1 Longitude: 251.5 Filter: 9 15.8 34.4 32.3 27.4 20.9 22.2 19.4 19.7 117.9 36.4 13.6 30.7 1.9 29.9 -6.9 22.7 -10.0 22.1 -4.5 19.9 6.6 14.4 12.5 11.7 18.3 0.09 14.3 0.09 14.4 0.12 12.0 0.10 13.4 0.10 13.9 0.26 3.0 0.11 12.2 0.27 28.9 0.15 3.6 0.14 10.8 0.15 15.5 0.26 18.5 0.25 26.0 0.21 12.1 0.14 13.1 0.19 0.27 17.8 5.36 21.5 5.03 26.1 4.43 2.6 6.31 3.2 6.63 6.59 14.7 5.54 4.6 6.35 25.9 209.5 24.7 210.0 at3 ss48 L 39.9 242.4 38.5 237.7 37.2 233.6 35.9 230.0 28.2 209.5 27.1 208.7 34.6 226.4 33.3 222.7 32.1 219.1 30.8 216.7 29.6 214.9 28.4 214.2 33.1 222.1 31.9 217.9 30.6 213.9 29.4 211.5 25.9 214.9 24.7 213.9 35.7 231.8 34.4 227.0 27.2 214.8

Alt Temp Press

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(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err enr enr err enr err err

at3	40.8	39.4	38.0	36.6	35.3	33.9	32.6	31.2	30.0	28.7	27.4	26.1	24.9	at3	40.5	39.2	37.8	36.4	35.1	33.7	32.4	31.1	29.8	28.5	27.2	26.0	24.8
12	4	~	2	∞.	•	_		9.9	•	10	10	~		::12	_	∞	,,	3.	6.	∞	10	•	,,		-	~	~1
Filter: 12	17.4	15.8	15.	15.	15.9	16.	15.	15.0	15.9	14.5	14.5	16.8	17.6	Filter: 1	14.0		17.6	17.	17.3	16.8	16.5	16.9	17.0	17.	15.4	15.	19.2
	ς.	13.0	13.1	18.2	18.9	13.2	9.0	9.3	11.2	12.1	13.5	14.2	8.6	8.9	7.8	13.0	19.4	19.1	13.9	10.5	11.2	13.8	15.0	12.4	8.5	8.1	10.3
Longitude: 206.4	17.1	14.3	13.1	12.2	12.4	11.6	11.4	11.3	11.2	10.2	10.3	12.5	13.5	$\tilde{\omega}$	13.9	14.3	14.5	14.3	13.0	12.4	11.4	11.3	11.5	11.6	10.8	11.0	14.5
onoiti	31.4	15.2	10.3	13.7	16.4	14.8	13.1	13.5	14.1	14.2	16.0	17.5	12.6	Longitude: l	16.0	18.5	22.1	23.7	21.2	16.0	11.4	10.5	12.1	13.0	13.1	15.3	17.8
	.1.	0.19	0.18	0.15	0.14	0.13	0.13	0.13	0.12	0.12	0.11	0.0	0.07	7 6	0.17 16.0	0.16 18.	0.16	0.15	0.13	0.13	0.14	0.13	0.12	0.11	0.0	0.08	0.08
de: 38	5.53	6.47	6.87	6.77	6.65	6.71	6.74	6.64	6.50	6.33	5.98	5.39	4.62	te: 37.	6.28	5.91	5.82	6.01	6.36	6.64	6.71	6.62	6.47	6.25	5.90	5.36	4.63
Latitude: 38.7	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 37	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
0	237.8	233.7	230.0	226.7	224.2	223.5	223.1	221.8	219.9	218.5	217.7	216.1	213.4	ري	236.1	234.0	234.1	235.1	235.6	235.9	32.9	228.6	24.8	221.4	219.4	219.1	217.0
013.8850	39.9		37.3	36.0	34.7	33.4	32.1	30.9			27.2		24.7	at3 ss53	40.3			ω.	35.0 2	33.7	32.3	0.	29.7	28.5	27.2	26.0 2	24.8

Filter: 9		22.1			20.7	19.5	19.2	19.8	20.8	17.9	17.2		22.1	ilter:12	18.5	17.9	17.6	17.7	17.9	18.0	18.3	18.1	18.0	20.8	19.8	18.2	18.8
6.4	-20.4	-1.1	11.8	10.4	5.6	7.1	11.5	13.2	10.5		3.9	8.9	8.5	1.3 F	12.7	9.3		8.9	11.8	14.7	16.7	17.6	17.0	14.8	11.8	6.6	10.7
Longitude:11	33.3	24.7	24.6	17.6	17.3	17.2	17.0	16.1	14.6	12.9	12.9	14.7	17.0	ude: 7.	14.5	16.2	15.4	14.5	13.4	12.4	12.2	11.5	11.5	13.0	12.9	13.0	13.8
ongit	15	∞	7	11.9	_	22.6	23.3	19.3	13.5	10.1	12.7	18.6	17.8	Longita	7.1	24.2		24	18.9	15.4	14.6	15.5	15.9	15.8	15.6	15.1	13.6
5	0.30		0	Ö	0	o.	0.23	0.21	0.20	0.26	0.25	0.20	0.14	0.	0.16	0.16	0.16	0.16	0.15	0.15	0.15	0.16	0.16	0.13	0.11	0.0	0.07
de: 37	7.05	6.74	6.46	6.48	6.82	7.26	7.54	7.52	7.35	7.09	6.58	5.74	4.80	de: 37	5.9	6.2	6.29	6.34	6.52	6.75	6.94	7.08	7.20	7.13	6.55	5.64	4.69
Latitue	2.6	3.2		4.6	5.6	8.9	8.3	0	12.1	14.7	17.8	21.5	26.1	Latitu	2.6	3.2		4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
54	245.7	244.7	244.1	243.0	240.9	238.0	233.4	229.2	227.0	224.9	223.3	221.1	217.7	26	245.4	3.1	241.4	240.1	238.0	234.9	231.7	228.6	226.1	224.9	223.4	218.9	214.8
at3 ss54	40.8	39.4	38.0	36.6	35.3	33.9	32.6	31.2	30.0	28.7	27.4	26.1	24.9	at3 ss.	40.5	39.2	37.8	36.4	Τ:	33.7	32.4	31.1	29.8	28.5	27.2	26.0	24.8

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(qm)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err enr enr enr err err

Filter: 12	20.1	18.5	19.1	20.9	19.1	17.5	17.7	19.9	18.3	17.6	16.4	16.2	20.7	Filter: 9	37.1	28.5	25.5	22.2	24.3	24.4	23.0	19.7	19.6	24.0	23.8	23.0	22.3
	9.4	17.5	22.3	22.1	19.1	15.2	12.7	12.7	14.1	14.5	12.6	10.7	9.7	w.	42.8	10.6	-2.7	4.2	16.1	19.4	14.8	6.6	9.6	11.5	7.9	-1.0	-7.7
ıde: 3.	18.7	14.6	15.5	15.3	13.7	12.1	11.9	13.0	11.9	11.6	11.4	11.4	15.1	de:34	47.5	28.3	24.1	17.1	18.4	18.0	17.8	14.1	12.9	12.1	12.7	16.5	18.1
I Longitude: 3.8	0.15 7.9	0.15 14.0	0.15 22.3	0.16 26.5	0.15 25.7	0.14 21.7	0.13 17.1	0.13 14.5	0.11 14.3	0.10 15.7	0.09 16.7	0.08 16.0	0.08 13.4	8 Longitude: 341	0.24 37.9	0.28 23.3	0.28 3.5	0.30 -7.5	0.29 0.4	0.25 17.4	0.23 25.5	0.33 19.9	0.31 11.7	0.19 7.8	0.16 8.5	0.18 11.2	0.12 13.7
le: 36.	5.62	5.82	6.31	6.93	7.13	7.03	96.9	87.9	6.34	5.85	5.54	5.22	4.70	le: 35.8	5.72	6.99	7.49	7.28	7.29	7.78	8.24	8.20	7.81	7.27	6.63	5.81	4.85
Latitude: 36.I	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude:		3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss59	40.5 250.9	Ξ.	37.7 242.3	36.3 240.4	35.0 237.7	33.6 231.9	32.3 226.9	31.0 225.2	29.8 222.8	28.5 218.7	27.3 215.9	26.1 214.7	24.9 214.2	at3 ss60	40.6 254.9	39.1 248.7	37.7 241.8	36.4 236.7	35.1 233.6	33.7 230.7	32.4 227.9	31.1 225.9	29.9 224.3	28.6 222.1	27.4 220.0	26.1 219.6	24.9 219.6

Latitude: 34.2 Longitude: 228.7 Filter: 12 26.1 4.43 0.08 14.9 16.5 8.8 21.8 Latitude: 34.0 Longitude: 206.2 Filter: 9 30.4 28.2 26.0 33.8 22.5 28.5 21.0 15.9 0.11 15.2 12.1 12.8 0.09 16.1 15.1 11.1 0.08 14.9 16.5 18.3 16.0 16.7 15.6 15.3 14.2 0.12 13.2 0.29 31.7 0.29 9.8 0.14 11.9 0.16 19.7 0.15 17.4 -6.7 0.25 6.3 0.26 23.6 0.24 24.5 0.22 16.0 0.14 13.5 0.19 13.4 0.17 18.2 0.27 26.1 4.43 2.6 6.91 3.2 7.11 21.5 5.15 7.04 7.04 7.15 6.97 10.0 7.43 14.7 6.62 17.8 10.0 5.6 36.0 229.2
34.7 225.5
33.5 222.3
32.2 220.9
31.0 220.6
29.7 219.7
28.5 218.4
27.2 217.2
26.0 216.1
24.8 214.8
at3 ss66
L
40.1 246.2
38.8 241.7
37.4 236.9
36.1 232.3
34.8 228.2
33.5 225.7
32.2 224.8
30.9 223.7
29.6 222.1
28.4 220.2
27.2 218.2
25.9 216.4
24.7 214.5

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(mb) vmr

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃
(km) (K) (mb) vmr err enr enr enr enr err err

Filter: 12	15.7	15.8	16.5	17.5	17.8	16.2	15.8	17.3	18.1	16.7	16.8	19.8	21.9	Filter: 12	19.7	20.4	18.4	18.4	19.8	20.7	19.2	18.3	17.4	17.6	22.1	22.3	22.9
	7.1	9.5	15.7	19.4	18.6	15.2	13.0	12.7	12.5	11.7	12.0	13.9	15.1	.3	22.5	10.5	7.0	12.2	16.5	15.5	12.8	12.1	13.1	13.7	12.2	10.6	10.6
16:16	14.8	13.4	14.1	14.1	13.4	11.1	10.8	11.7	12.2	11.5	11.6	14.0	15.6	de: 71	21.8	19.5	16.1	14.9	14.7	14.2	12.4	11.7	10.8	11.0	14.4	15.1	16.0
3 Longitude:161.2	0.19 3.1	0.20 10.5	0.19 20.9	0.18 24.0	0.17 21.0	0.15 16.8	0.14 14.1	0.13 13.0	0.12 13.0	0.13 13.8	0.12 14.8	0.10 15.5	0.08 15.4	.8 Longitude: 7	0.14 45.1	0.16 26.9	0.18 16.9	0.19 15.8	0.17 16.6	0.16 15.7	0.16 14.4	0.16 13.6	0.15 14.1	0.13 15.6	0.11 16.4	0.10 15.6	0.09 14.1
te: 33.	6.84	6.97	96.9	7.02	7.19	7.39	7.35	7.01	6.53	6.09	99.5	5.14	4.45	le: 31.	5.15	00.9	6:39	6:39	6.58	7.28	8.05	8.09	7.53	6.93	6.42	5.76	4.88
Latitude: 33.3	5.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 31	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss68	40.0 241.2	38.7 237.8	37.3 233.7	36.0 230.1	34.7 227.4	33.4 225.5	32.1 224.7	30.9 223.7	29.6 222.6	28.4 220.5	27.1 217.8	25.9 215.7	24.7 213.7	at3 ss71	40.6 245.3	39.2 243.8	37.9 242.3	36.5 240.4	35.1 238.7	33.8 237.0	32.4 235.2	31.1 232.6	29.8 227.2	28.5 222.3	27.3 220.5	26.0 218.6	24.8 216.4

Latitude: 31.3 Longitude: 3.6 Filter:12 20.2 20.3 19.5 0.14 11.6 19.8 13.8 0.15 22.4 18.6 17.1 6.2 24.8 14.5 23.0 0.20 12.3 14.9 11.8 0.15 9.4 15.0 14.2 29.1 Latitude: 31.5 Longitude: 48.8 0.11 10.5 16.2 14.9 12.6 0.26 10.7 12.1 0.24 11.6 12.2 0.17 25.0 15.2 13.7 0.23 23.2 0.16 22.6 0.23 28.1 0.22 15.1 0.17 17.4 0.17 13.4 0.16 12.6 0.17 14.3 0.13 15.4 0.16 15.7 0.12 14.4 0.23 13.7 2.6 5.01 3.2 5.67 10.0 8.54 12.1 8.34 26.1 4.85 21.5 5.86 17.8 6.92 3.8 6.00 6.16 6.85 5.6 6.8 8.3 14.7 8.9 10.0 4.6 38.0 239.2 36.7 238.2 35.3 236.9 34.0 235.2 32.7 232.8 31.4 230.4 30.1 229.0 28.8 227.4 27.5 225.6 26.2 222.0 25.0 217.9 40.6 255.6 39.2 251.1 37.8 243.9 36.4 238.5 35.0 236.0 33.7 234.1 32.4 231.9 31.1 226.7 29.8 222.2 28.6 220.1 27.3 218.7 26.1 217.3 24.9 215.8 39.4 244.2

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(qm)

 Table B.2 continued. (See notes on page 135.)

 Alt Temp Press 48Q3 48Q3 668Q3 668Q3 686Q3 (86Q3 (80Q3 (8

Filter: 9	34.3	33.9	32.1	28.0	27.5	20.2	19.7	19.1	20.8	20.9	22.7	20.5	25.7	Filter: 12	21.3	21.5	21.1	20.7	19.9	19.3	19.8	19.4	19.2	21.0	20.0	17.1	16.8
	7.	31.8	32.5	13.0	-5.1	-17.6	-12.6	4.4	14.2	14.1	12.7	12.9	8.6	7:	13.4	17.0	18.0	15.6	13.2	12.1	11.0	9.6	9.2	10.2	11.5	11.5	9.0
.de:13	48.1	34.0	29.0	28.8	24.0	20.6	17.1	14.1	15.5	16.4	17.7	15.2	18.0	ide: 94	20.5	18.7	16.8	14.8	14.3	13.3	12.7	12.4	12.5	13.7	13.2	12.1	11.9
4 Longitude: 139.0	0.30 42.8	0.33 19.9	d)	0.26 2.6	0.32 22.8	0.25 21.4	0.24 7.4	0.22 9.6	0.26 20.9	0.24 27.9	0.17 24.4	0.18 15.4	0.10 6.5	5 Longitude: 94	0.16 21.9	6 20	0.17 17.1	0.15 16.2	0.15 17.4	0.14 16.3	0.16 13.4		0.14 15.5	0.13 16.8	0.12 15.7	0.10 14.2	0.08 12.3
Latitude: 26.4	7.23	7.75	7.54	6.97	6.52	6.45	7.00	7.76	8.00	7.47	6.47	5.43	4.44		5.60	9.00	98.9	6.57	92.9	7.08	7.43	7.53	7.35	7.04	6.59	5.90	4.92
Latitue	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 25	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss78	40.4 245.9	39.0 243.1	37.7 240.2	36.3 235.7	35.0 233.2	33.7 231.6		31.0 229.9	29.7 228.5	28.4 226.1	27.2 221.6	25.9 218.1	24.7 215.9	at3 ss80	40.7 252.3	3	37.8 248.4	36.4 245.7	35.0 242.5	33.7 238.1	32.3 233.6	31.0 229.4	29.7 225.6	28.5 222.7	27.2 220.7	26.0 218.9	24.7 217.1

26.1 4.76 0.09 9.6 16.7 6.1 24.0 Latitude: 24.0 Longitude: 4.1 Filter: 9 0.18 13.3 12.7 13.1 0.15 14.1 12.6 13.2 0.16 14.4 14.2 13.0 14.8 10.9 16.9 15.8 14.4 0.7 14.9 0.14 13.0 15.4 11.2 Latitude: 24.2 Longitude: 26.7 0.12 11.2 18.2 0.09 9.6 16.7 0.24 52.7 53.3 0.26 49.9 46.6 35.1 25.8 19.3 15.2 14.9 0.16 16.9 0.20 13.9 0.26 32.9 0.24 12.5 0.24 -8.9 0.23 -9.1 0.23 10.7 2.6 5.84 (3.2 6.24 (3.8 6.64 (17.8 7.12 21.5 6.03 6.81 6.97 10.0 12.1 14.7 10.0 8.9 37.9 246.7
36.5 245.0
35.1 241.4
33.7 237.4
32.4 233.4
31.1 229.5
28.5 223.7
27.3 222.2
26.0 220.0
24.8 215.7
at3 ss84 1
40.8 257.1
39.4 253.2
37.9 247.6
36.5 242.7
35.2 239.2
33.8 236.7
32.5 234.4
31.2 230.4
29.9 226.0
28.6 223.5
27.3 222.2

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(mb)

Table B.2 continued. (See notes on page 135.)
Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃ (km) (K) (mb) vmr err err err err err err err err

Filter: 12	27.5	25.5	21.6	18.1	17.5	21.9	23.0	22.4	18.5	20.1	21.3	23.1	21.2	Filter: 12	24.6	22.7	23.6	24.4	22.9	19.9	21.3	23.7	20.3	19.9	24.2	31.2	30.3
_	21.4	27.3	25.3	18.6	15.8	15.1	11.9	8.9	8.7	8.6	10.5	11.0	11.0	5.0	4.2	7.0	14.0	18.3	16.5	13.4	10.9	9.0	8.6	10.7	13.3	13.1	10.2
ide:31	28.6	23.4	16.2	12.4	12.2	14.0	14.5	13.9	11.4	12.7	13.7	15.8	14.9	3	26.1	20.8	19.4	18.5	15.5	13.1	13.6	15.0	12.8	12.4	15.3	20.2	20.9
8 Longitude:319.3	0.14 42.7	0.14 36.6	0.15 23.3	0.16 21.5	0.17 21.0	0.18 17.8	0.20 14.8	0.17 13.2	0.15 13.6	0.14 15.3	0.13 15.6	0.11 13.8	0.08 11.6	9 Longitude:2	0.15 15.3	0.14 15.3	0.18 19.3	0.21 19.1	0.18 16.5	0.18 15.9	0.17 16.4	0.18 15.4	0.17 14.2	0.15 14.3	0.13 14.0	0.12 12.8	0.09 12.5
Latitude: 22.8	4.94	5.29	5.62	5.99	6.47	7.20	8.16	8.81	89.8	7.93	96.9	5.92	4.77		5.20	5.47	9.00	6.94	7.87	8.39	8.65	8.82	8.62	7.87	6.85	5.77	4.63
Latitue	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1	Latitude: 20.	5.6	3.2	3.8	4.6	9.9	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
at3 ss86	40.8 257.8	4.	37.9 249.2	36.5 244.8	35.1 241.7	33.8 238.9	32.4 236.5	31.1 233.6	29.8 229.2		27.2 222.1	26.0 218.7	24.7 217.0	at3 ss89	40.6 254.3	39.2 251.7	37.8 247.4	36.4 243.7	35.0 241.2	33.6 237.1	32.3 234.3	31.0 232.5	29.7 229.7	28.4 225.6	27.1 221.0	25.9 217.9	24.7 214.5

21.5 5.84 0.16 22.2 24.3 2.3 23.2 26.1 4.38 0.12 16.8 21.8 8.3 25.7 Latitude: 19.5 Longitude:184.6 Filter:12 22.4 21.5 16.9 14.8 13.9 14.2 Latitude: 20.0 Longitude: 229.7 16.9 0.25 13.1 14.4 16.5 19.5 15.7
 17.8
 7.25
 0.16
 20.9
 18.8

 21.5
 5.84
 0.16
 22.2
 24.3

 26.1
 4.38
 0.12
 16.8
 21.8
 0.17 20.3 22.3 14.1 14.5 0.18 21.1 0.18 15.2 0.17 13.3 0.16 16.6 0.16 16.6 0.15 14.6 0.28 17.3 0.18 24.3 0.28 15.6 0.17 14.5 2.6 6.01 6.50 6.8 8.06 8.3 8.28 7.00 14.7 8.18 5.6 7.90 10.0 37.9 250.4 36.5 243.5 35.1 238.6 33.8 235.6 32.5 233.2 31.2 231.4 29.8 229.2 28.6 226.5 27.3 223.5 26.0 219.8 24.8 215.1 at3 ss92 I 40.5 252.9 39.0 248.3 37.6 243.4 36.3 238.5 34.9 235.4 33.6 234.1 32.3 233.9 30.9 232.7 29.6 229.0 28.3 224.1 27.1 219.2

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vmr

(mp)

(km) (K)

Alt Temp Press 48O₃ 48O₃ 668O₃ 668O₃ 686O₃ 686O₃ enr enr err enr err Table B.2 continued. (See notes on page 135.) vmr (qm) (km) (K)

at3 ss95 Latitude: 17.0 Longitude: 117.5 Filter: 12	at3 ss98 Latitude: 15.0 Longitude: 50.2 Filter:12
2.6 6.07 0.18 11.3 26.1 9.1	2.6 6.11 0.18 25.1 31.9 6
3.2 6.71 0.17 23.3 17.9 15.3	3.2 6.54 0.18 24.2 20.9 7
3.8 6.91 0.18 24.3 18.5 19.9	3.8 6.55 0.17 24.2 20.6 11
4.6 6.95 0.17 19.1 17.9 18.1	4.6 6.55 0.16 24.0 20.5 15
34.9 236.6 5.6 7.08 0.16 17.2 18.4 13.5 24.8	34.9 241.1 5.6 7.08 0.16 20.6 17.0 17.6 23.9
6.8 7.48 0.17 16.6 17.9 12.2	6.8 7.93 0.17 15.8 14.2 14
8.3 8.11 0.21 15.5 13.9 12.2	8.3 8.40 0.18 14.3 16.3 11
10.0 8.38 0.20 16.0 15.5 12.1	10.0 8.35 0.16 15.7 13.3 10
12.1 8.11 0.18 17.0 14.0 13.8	12.1 8.12 0.17 17.0 17.0 12
14.7 7.71 0.17 15.7 13.9 15.1	14.7 7.87 0.15 16.6 14.3 15
17.8 7.24 0.15 12.9 18.5 12.9	17.8 7.38 0.13 15.0 13.1 13
21.5 6.30 0.12 11.6 17.1 9.9	21.5 6.41 0.13 13.1 18.8 10
26.1 4.91 0.09 12.2 16.7 8.6	26.1 5.00 0.10 11.9 19.9 9
Latitude: 16.2 Longitude: 95.1	Latitude: 11.6 Longitude: 298.1
2.6 5.86 0.26 4.9 49.8 8.6	2.6 6,32 0.28 16.8 38.4 -6
3.2 6.45 0.28 38.1 52.5 31.5	3.2 7.18 0.31 30.0 52.1 -18
3.8 7.23 0.28 36.6 28.5 8.4	3.8 7.88 0.30 2.4 28.5 -18
4.6 7.76 0.27 36.6 23.5 -4.2	4.6 8.22 0.28 -9.2 18.8 -5
5.6 7.94 0.30 42.9 24.5 -3.6	5.6 8.47 0.30 9.8 24.0 5
6.8 8.04 0.35 32.5 26.4 -1.6	6.8 8.63 0.34 22.6 19.6 10
8.3 8.10 0.23 15.6 16.3 5.6	8.3 8.58 0.32 22.5 16.7 14
10.0 8.21 0.29 10.2 17.6 12.6	10.0 8.49 0.26 18.7 18.7 9
12.1 8.19 0.28 12.8 17.0 13.5	12.1 8.38 0.23 14.3 14.7 6
14.7 7.77 0.21 16.7 15.6 10.8	14.7 8.10 0.23 11.1 18.3 9
17.8 6.98 0.32 17.1 22.3 6.0	17.8 7.38 0.24 11.3 15.8 13
21.5 5.99 0.15 12.4 18.9 1.7	21.5 6.23 0.19 11.8 21.9 9
26.1 4.79 0.10 5.7 16.9 0.9	26.1 4.96 0.13 11.3 24.2 3

Table B.2 continued. (See notes on page 135.)

Alt Temp Press 48O₃ 48O₃ 668O₃668O₃686O₃ 686O₃

(km) (K) (mb) vmr err err err err err err

Filter:12	25.4	27.0	27.8	21.9	23.4	21.9	22.0	26.7	21.3	21.1	27.8	25.2	41.7
3.3	16.5	16.4	14.9	13.9	14.8	15.1	13.7	11.7	10.1	10.0	12.3	14.9	15.2
ude:25	24.4	25.1	22.6	15.9	16.4	14.4	14.0	16.3	12.8	12.6	17.0	15.7	26.6
0 Longit	0.16 14.5	0.18 17.9	0.22 18.7	0.22 19.2	0.21 19.9	0.20 19.1	0.21 16.8	0.21 15.0	0.17 14.1	0.15 13.3	0.17 12.4	0.16 12.4	2 26.1 5.08 0.13 13.1 26.6 15.2
de: 10.	5.29	6.04	7.05	7.71	8.08	89.8	9.37	9.53	9.16	8.51	7.58	6:36	5.08
Latitue	5.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
	249.	247.	246.	243.4	238.5	235.6	234.4	233.3	229.8	227.2	225.1	220.1	216.2
at3 ss104	40.6	39.2	37.8	36.4	35.1	33.7	32.4	31.1	29.7	28.4	27.2	25.9	24.7

		i	\sim													
668O3 686O3686O3	enr enr err		Filter: 1	28.8	22.2	23.2	24.5	29.5	23.6	35.4	25.4	22.1	33.0	26.8	24.0	23.7
36860			•	13.0	17.2	17.5	16.3	16.4	15.6	13.8	13.9	14.3	12.8	11.5	11.3	10.4
	enr err		Longitude: 187.4	29.1	19.5	18.5	18.8	20.2	15.7	22.5	15.7	13.2	19.7	16.7	15.6	15.8
66803	enr		ongiti	9		13.6	14.5			16.9					14.0	14.8
48 03	сп			0.18	0.17	0.17	0.17	0.22	0.20	0.21	0.18	0.16	0.18	0.15	0.12	0.09
8 03	vmr		Latitude: 3.4	6.20	6.54	6.67	7.12	7.71	8.26	8.76	8.88	8.68	8.43	7.75	6.56	5.23
Press	(qm)		Latitu	2.6	3.2	3.8	4.6	5.6	8.9	8.3	10.0	12.1	14.7	17.8	21.5	26.1
Temp	(X		107	253.0	248.8	244.0	240.7	238.1	234.6	233.5	231.7	227.9	227.1	224.9	218.6	215.7
Alt	(km)		at3 ss107	40.6	39.2	37.8	36.4	35.0	33.7	32.4	31.1	29.7	28.5	4	25.9	24.7
	i															